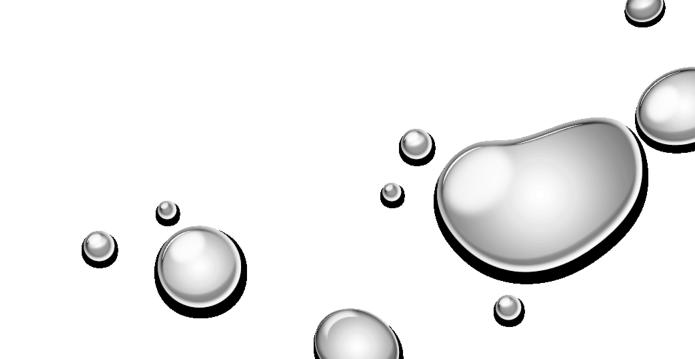


# **ENVIRONMENTAL ASPECTS**



# **GREEN CHEMISTRY, SUSTAINABLE CHEMISTRY**

The design of chemical products and processes that reduce or eliminate the use or generation of hazardous substancese;

Systematically pursued from the 90s of the 20th century;

1952 – Upjohn – the synthesis of cortisone – just 10 step synthesis of cortisone using fermentation for a key hydroxylation reaction at position 11 of progesterone



Characterizes the "greeness" of a synthetic process by calculating the number of atoms from all of the reactants that make it into the final product;

Does not address the hazard, reaction yield, stoichiometry, the amount of solvent, ...

% Atom Economy = Molecular weight of the product / Molecular weight of all products x 100

Trost, B. *Science 254*, 1471 (**1991**)







Environmental Factor, *E*-Factor

The ratio of waste over product

It is usual to calculate *E*-factor without process water

E-Factor = the amount of waste (kg) / the amount of the product (kg)

Industry Segment	Volume (t/y)	<i>E</i> -Factor
Oil Refining	10 <sup>6</sup> - 10 <sup>8</sup>	< 0.1
Bulk Chemicals	10 <sup>4</sup> - 10 <sup>6</sup>	< 1 - 5
Fine Chemicals	10 <sup>2</sup> - 10 <sup>4</sup>	5 - 50
Pharmaceuticals	10 - 10 <sup>2</sup>	25 - 100



Reaction Mass Efficiency (RME)

The idea is to keep the simplicity of the atom economy concept, but avoid the high impact of solvents which are found in the *E*-factor; RME takes into account reaction yield, stoichiometry and the use of catalysts or other reagents;

% Reaction Mass Efficiency = Mass of desired product / Mass of all reactants x 100

Process Mass Intensity (PMI)

PMI measures the mass of materials used to make 1 kg of the API

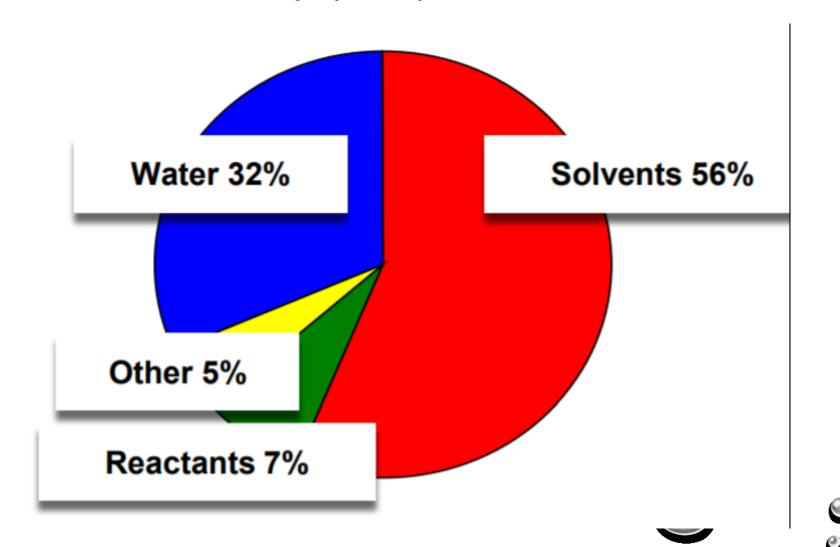
% Process Mass Intensity = Mass of all material used to make the product (kg) / Mass of product (kg) x100

Allows companies to benchmark and quantify improvements to the efficiency and sustainability of their production

Reasonable target of a single synthetic step is any value between 10 and 40

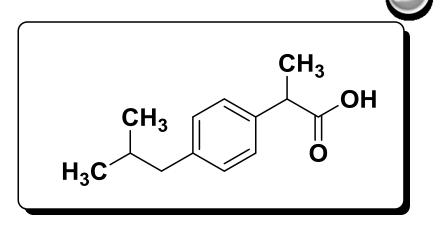


Process Mass Intensity (PMI)





### IBUPROFEN EXAMPLE



Originator – Boots Group (1960s)
Original name – Brufen (Aspro, Panadol, Nurofen)
Generic names – Motrin, Advil, Nuprin, Ibalgin

Sold as the racemate despite the fact that (S)-(+)-Ibuprofen is the active form;

Fast epimerization in vivo







# **IBUPROFEN EXAMPLE**

Original synthesis – overall yield 40% Annual production in Great Britain 3000 tones

### IBUPROFEN EXAMPLE

New synthesis – BHC company (1990)

Overall yield 77%

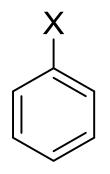
Presidential Green Chemistry Challenge Greener Synthetic Pathways Award v roce 1997

- Myths of biocatalysis:
  - Expensive
  - Unstable
  - Not readily available
  - Sensitive to reaction conditions
  - Not good enough or wrong selectivity
  - Give poor volumetric productivity
  - Difficult work-up

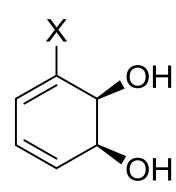
Nowadays, all of them are mostly wrong







Pseodomonas putida



X = CI, Br

Microbial oxidation using bacterial dioxygenase

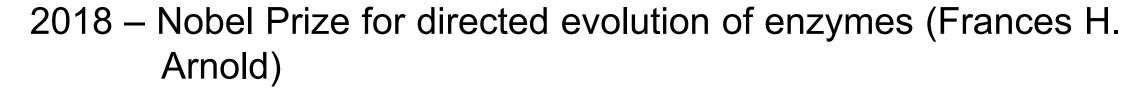
Hudlicky, T. et al J.Am.Chem.Soc. 116, 5108 (1994)







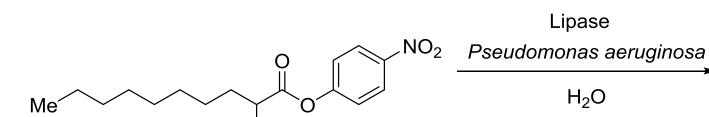




Iterative change of amino acids in the enzyme until the desired property (acitivity, stability, selectivity) is achieved;

Directed evolution mimics the processes of Darwinian evolution in a test tube





Ме

Fast hydrolysis

No hydrolysis







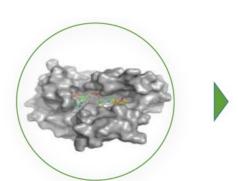




Computer modeling & Al-driven structure-function mapping deeply mines vast data generated; increasingly predictive of real world protein performance



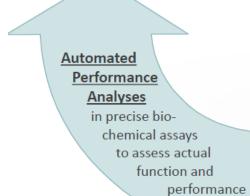




**Starting Protein** 

from nature, Codexis libraries,

or in silico inspired







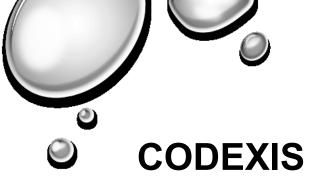
CodeEvolver® **Engineered Protein** tailored for specific end use application

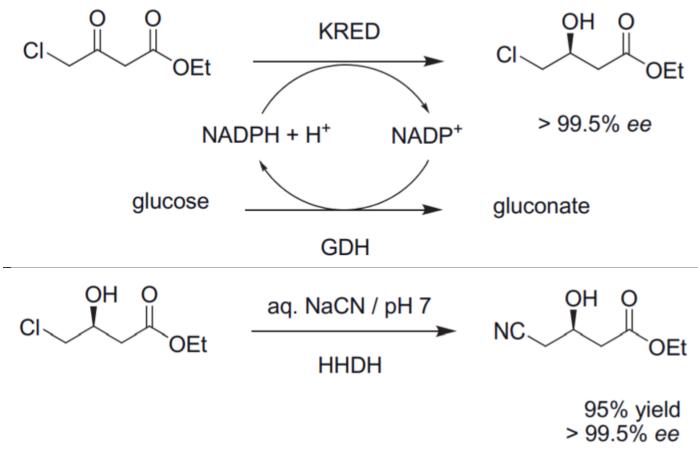
https://www.youtube.com/watch?v=up5QUdTLsBU&feature=youtu.be







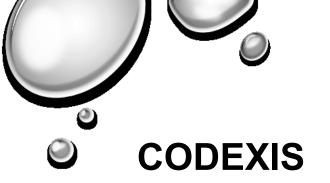




KRED = ketoreductase

GDH = glucose dehydrogenase HHDH = halohydrin dehalogenase





Atorvastatin (Lipitor®)

2006 - Presidential Green Chemistry Challenge Award

Substantial waste reduction

Overall yield > 90%

Purity > 98%

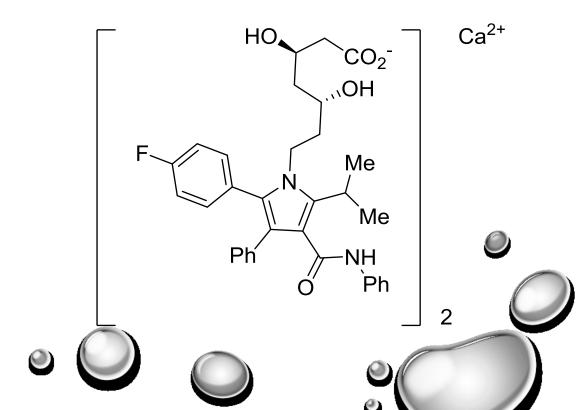
ee > 99.9%

Low loading of enzymes

Solvent recycling

E-Factor is 5.8 (without used water)

E-Factor is 18 (with used water)





# GREEN CHEMISTRY, SUSTAINABLE CHEMISTRY



"Green chemistry is not just a mere catch phrase; it is the key to the survival of mankind"

