RNDr. Danuše Lefnerová, Ph.D. Doc.Ing. Martin Krsek, CSc.



Act No. 258/2000 Coll. on public health protection:



"Drinking water is all water in its original state or after treatment intended for drinking, cooking, preparation of food and drinks, water used in the food industry, water intended for body care, cleaning of objects , which by their intended purpose come into contact with food or human body and for other purposes of human consumption, irrespective of its origin, state and mode of delivery. "

Function of water in body:

- transport (transmission of nutrients, waste products, heat, electrolytes, hormones)
- assistance in thermoregulation
- acts as a solvent and a suitable environment for chemical reactions taking place in the body
- protects the surroundings of joints, spinal cord and brain
- surrounds the fetus like amniotic fluid
- involved in maintaining homeostasis, ensuring a physically and chemically stable internal environment of the body





Water requirement:

Very individual, can not be standardized.

It depends on many factors - age, sex, weight, ambient temperature and air humidity, health, diet, physical activity

After birth, water is 75% of body weight, in adults 60% and in old age 50% of body weight

The need for water is partly covered by its natural content in foods ranging from 20-30% (very fatty products) and very often between 80-90% (fruits, vegetables, soups, sauces).

By the oxidative metabolism of organic macronutrients, 300 ml of water per day is produced in the human body.

Necessary water intake is average 2.5 l per day for an average adult (1.5 liters in the form of beverages and 1 l from regular food).

Literature - 22 ml - 50 ml per kg of body weight, regardless of climatic conditions

Children are dehydrated faster, so they should drink more - schoolchildren by half more than adult of the same weight.

Water output and intake should always be in equilibrium

Lack of water in the body (dehydration) - headaches, fatigue, malaise, decrease in physical and mental performance, including decreased concentration

- children - decreased ability to concentrate

Long-term fluid deficiency - renal impairment, kidney and urinary stones, risk of urinary tract infection,...

Biological (biogenic) value of drinking water

Drinking water must be health-conscious with a satisfactory biological value.

Mineral substances contained in drinking water are usually in ion form, perfectly dissolved, and are therefore easily resorbable and more usable for the organism.

Water is an important source of all required minerals (fluorine in the form of fluoride anions, iodine, sodium, potassium, the calcium / magnesium ratio, selenium, zinc and other macro and micro elements).



Drinking water sources - their protection

Czech Republic - surface sources: underground source = 1: 1 2016 - 2.305 water treatment plants - 600 million m³ (Prague 100 million) Brno dam - 7.6 - 10.8 million m³

When determining the range of hygiene protection, account shall be taken of:

- geological composition of soil, its permeability (sand x clay)
- slope of the land around the source
- yield of water source
- industrial activity
- agricultural activity
- transportation around the source



Campaign For Water Protection

Drinking water sources - a multi-barrier approach to their protection

To ensure that water is microbiologically safe, it is necessary to apply:

- 1. barrier consistent protection of water source (functional protection zone)
- 2. barrier the use of such water treatment technology, which corresponds to the quality of raw water
- 3. barrier water protection against secondary contamination during distribution/transport to the consumer
- 4. barrier internal water supply (domestic water supply) design of hygienically safe materials

Drinking water sources - water treatment

Category according to Decree of the Ministry of Health No. 428/2001 Coll., As amended by Decree No. 146/2004 Coll. Sources % Sources % surface underground A 1 - **Simple physical treatment** and disinfection, for example fast filtration and disinfection, or simple sand filtration, chemical deacidification or mechanical deacidification or removal gaseous components by aeration. 7.1 76.5 A 2 - common physical treatment, chemical treatment and disinfection, coagulation filtration, infiltration, slow biological filtration, flocculation, sedimentation, filtration, disinfection (final behavior), single- or two-stage jamming and demagnetization. 54 9 A 3 - **intensive physical and chemical treatment**, extended treatment and disinfection, for example chlorination to break point, coagulation, flocculation, sedimentation, filtration, adsorption (activated carbon), disinfection (ozone, final chlorination). Combination physically chemical and microbiological and biological treatment 38,9 14,5

Drinking water sources and their protection

Suspicion or finding that the water in the source is contaminated:

The source of the pollution is removed, building modifications are carried out, the hygienic protection zone is restored

Wells - the inner walls of the well are mechanically cleaned, the water is drained, the bottom is cleaned from sludge

One-off disinfection is carried out



Disinfection of drinking water

Chloramine - 2-3 g / m³ of water, Savo, Chlorine, Sagen (AgCl – oligodynamic effects of Ag), Ozone, UV, filtration

Boiling : bubbles on the whole surface of water (100 ° C), leave for 10 minutes stand and naturally cool down - do not put ice in it. **Water is not sterile !!! (**sterility only ensures sterilization in the **autoclave**).







Non-potable/running water

It is hygienically safe water, which is not used as drinking water and for cooking but only for washing, bathing and for production purposes.

Domestic hot water is produced from drinking water under the Public Health Act but is not considered drinking water (for ex. *Legionella pneumophila*).

Industrial water

Technological water

Water for irrigation



Condition: excretion of agent by excrement (humans and animals) and possibility of new food infections

Viruses:

Rotaviruses - diarrhoeal diseases Polioviruses - agents of poliomyelitis RNA viruses – hepatitis A, E, (F)





What is hepatitis? What vaccines are available Estimated cases world-How does the The A, B, C, D and E wide (per year) of hepatitis for which types of hepatitis? virus spread? Hepatitis B and C: There are five main hepatitis viruses Hepatitis A 400 million cases Hepatitis B Hepatitis C Gradual death: Hepatitis D Hepatitis A and E: An estimated Lack of food hygiene. Hepatitis E 1.4 million people die contaminated water and worldwide from hepatitis sub-standard sanitary facilities every year Treatment: Inflammation 00% of hepatitis C of the liver patients can be Hepatitis B. C and D: Blood, sperm and other healed within three to six months bodily fluids Source: WHO © DM



Viruses – cont.:

Norovirus - originally called the Norwalk virus (1972 Norwalk USA)



- RNA virus
- it causes an epidemic acute gastroenteritis
- diseases from water (Prague-Dejvice 2015), food, but also through direct contact

 - symptoms of the disease - nausea, vomiting, diarrhea and abdominal cramps, mild fever, fever, pain of muscles and headache, fatigue
 - it is also transmitted via a faecal oral route





Gram negative facultative anaerobic rods:

Escherichia coli, Klebsiella spp., Citrobacter spp., Salmonella typhi, Salmonella paratyphi - frequent water transfer

Shigella sonei, Shigella flexneri - bacillary dysentery

Yersinia entrocolitica - diarrheal diseases in children



Serratia marcescens - infection of the urogenital and respiratory tract *Proteus mirabilis, Proteus vulgaris* – urogenital tract, gastroenteritis in infants





Gram negative aerobic rods and cocci:

Klebsiella pneumoniae, Pseudomonas aeruginosa

- via aerosols
- inflammation of the upper respiratory tract, lung, urogenital tract

Neisseria gonorhoea, Treponema pallidum

- they do not survive in the waters, transmission is unlikely

Campylobacter spp. - diarrheal diseases











Legionella pneumophila (Legionella) – G-aerobic rod

Occurrence:

hot and cold water, surface water, water courses as part of biofilms, filters, refrigeration circuits of air conditioners.

Survives in humid environments It **lives** and multiplies at 25-50 °C.

Spread with air - inhalation of aerosol contaminated water **Respiratory** tract infections, pneumonia (Legionnaires' disease)







(182 - 29)

Leptospirosis – G- spirochete

Febrile bacterial disease, a **zoonotic** disease caused by genera *Leptospira*

The way of infecting people is contacting damaged skin, eyes or mucous membranes with water contaminated by the urine of the infected animal

This is an **antropozoonosis** – transfer from animal (rats, mice, cattle) to humans

Symptoms - high fever, chills, tremor, severe headache and muscle aches, abdominal pain, nausea, vomiting





Fungi and yeasts – water mains, air washers, air conditioning



Protozoa

Cryprosporidium - protozoa - occurrence in surface waters.

- it can also penetrate into drinking water chemical disinfection against oocysts is ineffective.
- causes diarrheal disease cryptosporidiosis.
- 1993 Milwaukee Cryptosporidiosis outbreak 400,000 patients
 104 death

Giardia intestinalis - human lamblia - it causes diarrheal disease – giardiasis



Infective germ can enter the body and cause the disease not only by the digestive tract, but also by the airways, skin abrasions and injuries.



DID YOU KNOW?

Cryptosporidium,

a diarrheal disease caused by parasites, is easily spread in water and is resistant to chlorine.

Here's how to avoid contacting and spreading crypto at swimming pools:

Don't swim if you have had diarrhea in the previous two weeks.

Avoid swallowing pool water.

Use swim diapers on babies, and check them often.

Make kids take bathroom breaks every 60 minutes.

Health aspect

The route of inhalation and dermal - may be more risky than the oral route

- substances after entry into the body can affect the target organs before liver biotransformation.
- **doses** received by inhalation and dermal route are, in their sum, at least equal to the dose obtained by ingestion of 2 liters of water.

Extra-intestinal and intestinal disease - extra-intestinal and intestinal infections

Water hardness and cardiovascular diseases

Water hardness - mainly composed of calcium and magnesium carbonate

History: The 50s of the last century - Japanese chemist Kobayashi pointed out that mortality on the cerebral vascular disease of the population is higher in the vicinity of the Japanese rivers, where the water is more acidic (softer) than the rivers with water harder (more basic) - from where the water was used for drinking purposes.

The relationship between drinking water hardness and mortality in cardiovascular

disease - demonstrated in many further studies.





Hardness minerals precipitate out of the water and form energy robbing scale.







Water hardness and cardiovascular disease

Calcium: necessary for the proper functioning of the heart's transfer system, for blood clotting and neuromuscular irritation

Magnesium: an important role as a cofactor and activator of over 300 enzymatic reactions including glycolysis, ATP metabolism, transport of elements such as Na, K, Ca through membranes.

It is also important for protein and nucleic acid synthesis.

Magnesium deficiency increases the risk of vascular spasms and promotes cardiac arrhythmias.

Protective effect of calcium and magnesium against the development of tooth decay

Water hardness and cardiovascular disease

Calcium and magnesium - beneficial antitoxic function

Ca and Mg also prevent the absorption of some toxic metals such as lead and cadmium.

Utilization of magnesium from the water is also higher than that from food.

Cooking in soft water causes considerable loss of elements from food, while cooking in hard water minimizes losses.

Water disinfection with chlorination – chlorine can react with higher level of organic substances in water – production of low-molecular substances (chloroform, chlorobenzene, heptachlor and other chlorinated compounds) - genotoxic activity.

Studies show a **connection** between drinking chlorinated drinking water and the occurrence of bladder, rectum, kidney **tumors**. Negative effects on **reproduction** have also been demonstrated.

Arsenic in drinking water - exposure is associated with the occurrence of various skin lesions (pigmentation, keratoses, skin tumors and malignant tumors). Arsenites show high embryotoxicity.

Nitrates - no clear epidemiological evidence that people who consume drinking water with increased nitrate content are at increased risk of cancer (reaction with amino acids to produce nitrosamines).

Only one of three UK studies has revealed a relationship between stomach cancer mortality and nitrate content in water. So far, the results are inconsistent. **Methemoglobinemia** may be more significant – especially in infants

Radionuclides – mainly natural radionuclides are important for drinking water. It is generally assumed that the ingestion of radon in drinking water is not associated with any significant risk of cancer.

Fluorides - fluorization of drinking water to prevent tooth decay has been reversed in the Czech Republic. Environmental studies have not confirmed the effect of fluorides on cancer.

Industrially produced **organic substances** (chlorophenols, trichlorethylene, volatile organic compounds) - besides impairment of the immune system, there was an increased incidence of bladder cancer.

Cyanobacterial toxins (cyanotoxins)

Cyanobacteria - in warmer waters with higher nutrient content (especially phosphorus - agriculture, detergents)



They can cause digestive tract disorders, allergic reactions, liver disease, immune system weakness, respiratory and contact dermatitis...

They have embryotoxic and genotoxic effects



Drinking water can theoretically contribute to the annual increase in the probability of developing cancers in the Czech Republic by only 1-2 cases to the total number of over 60,000 new cases of newly reported cancer.

Bottled water

Natural mineral water Spring water Infant water Drinking water



Natural mineral water: extracted from an underground source that must be approved and regularly inspected by the Ministry of Health

Mattoni, Magnesia, Podebrady







Bottled water

Spring water (table): from a protected underground source that does not need to be approved by the Department of Health.

It must not be modified in any way that would alter the characteristic composition

Toma natura, Bonaqua

Infant water: from a protected underground source, and more stringent requirements apply to it Mountain spring (Horský pramen)

Drinking water: It does not have to come from an underground source, it can be tapped from the public water supply, the quality is comparable to drinking water quality from the water supply

Infant water x normal drinking water

Parametr (mg/l)	Infant water	drinking water
Nitrates	10	50
Nitrites	0,02	0,5
Na	20	200
Fluorides	0,7	1,5
Chlorides	100	100
Antimon (Sb)	0,003	0,005
Cyanides	0,005	0,05
Arsenic	0,005	0,01
Chrome	0,025	0,05
Cadmium	0,002	0,005
Copper	0,2	1
Lead	0,005	0,01
Mercury	0,0005	0,001

Mineral bottled water

If you would drink only **mineral water**, you would get too much salt into your body, especially sodium.

The mineral waters - benefit mainly in heat, during heavy work and intense sports

Pregnant and lactating women should choose those with higher potassium and calcium levels

The sodium content should be mainly watched by the cardiac

Dyspnoeic **people**, people who suffer from swelling, who tend to develop kidney stones – no mineral water at all

Artificially supplied carbon dioxide does not bring anything to the body to benefit. It contributes to bloating and burying. Carbon dioxide is the waste product that our body must actually constantly get rid of.

Health risks - demineralized water

Demineralized water - it does not have the character of drinking water and its regular consumption or one-off consumption in large quantities must be considered as a health risk

Devices based on distillation or deionization must be rejected as the final stage of drinking water treatment:

- virtually zero intake of calcium and magnesium with water
- reduced intake of some essential elements and microelements
- high losses of calcium, magnesium and other essential elements from foods cooked in demineralized water



HOME / POTRAVINY / NÁPOJE

Oxy Water – Kyslíková voda – 250 ml

35Kč (vč. DPH)

Oxy Water Kyslíková voda je revoluční nápoj s vysokou dávkou kyslíku. Prémiový energetický nápoj – bez chemických přísad, bez cukru, vyroben z české vody.

Množství

PŘIDAT DO KOŠÍKU
Water enriched with oxygen

Water artificially enriched with pure oxygen - a food supplement

Manufacturers claim: increased oxygen concentration in blood, increased vitality and performance, blood pressure regulation, improved metabolism, increased immunity, increased resistance to stress, improved concentration and memory

Our body needs oxygen in blood and tissues but definitely not in digestive system, which is supposed to be anaerobic.

A safe concentration of oxygen in drinking water can be considered to be up to 25 mg per liter.

Increased oxygen saturation - the risk of oxidative stress and its cytotoxic effects

An increase in chromosome aberrations was observed in rats

Water enriched with oxygen

Oxygen-enriched water manufacturers have not yet submitted any scientific study that would confirm the safety of higher oxygen values in water (when consumed)

Further work is needed to confirm or eliminate the potential physiological and toxic effect of oxygen enriched water

No international recommendations (WHO, FAO, etc.) on the safe maximum concentration of oxygen in drinking water

Natural pure water has a maximum of about 10 mg of oxygen per liter

Water and health - Microbiological analysis of water

DECREE No. 83 of 30 April 2014 amending Decree No. 252/2004 Coll., laying down the sanitary requirements for drinking and warm water and the frequency and scope of drinking water control, as amended

Limit value (MH): Exceeding does not usually present an acute health risk. Unless otherwise indicated, this is the upper limit of the allowable range.

Highest limit value (NMH): The value of a health-related indicator of the quality of drinking water as a result of which the **use of water as a drinking water is excluded**, unless the public health authority decides otherwise under the law.

Microbiological analysis of water

Indicator (groups of) bacteria

Psychrophilic bacteria (22°C) - indicators of general contamination (200) **Mesophilic bacteria** (36°C) - indicators of general contamination (40)

Faecal contamination

Coliform bacteria - indicators of faecal contamination (*Escherichia* spp., *Salmonella* spp., *Shigella* spp., *Yersinia* spp.) **Enterococci** - indicators of faecal contamination

Clostridium perfringens Escherichia coli

Sampling procedure for microbiological analysis

If the well has not been used for a long period of time (eg in the winter season), it is necessary to drain the well prior to sampling (at least rinse well pipes from the well to the sampling point). If the well is drained, the water is allowed to rise and then sample of water is taken.

• A sample of water is taken from the point where water is normally used (from the distribution line, from the tap, from the pump, etc.). The sample can not be taken over the hoses used for watering and sprinkling.

• Prior to sampling, the water is drained for **1 to 5 minutes evenly** (according to the length of the pipeline) and then sampling bottle is filled with the sample as described below. The sample bottle is held in such a way that any impurity from the hands does not get into the sampler.

• Samples should be stored in a refrigerator and taken to the laboratory within 48 hours after sampling, sample for bacteriological analysis within 24 hours!

Sampling for chemical analysis:

The sample is taken up in polyethylene sample bottles (bottles). When sampling, the sample bottle, including the closure, is rinsed three times with tap water.

The sample bottle is filled to the edge.

Sample collection for bacteriological analysis:

☑ The sample is to be taken only into glass bottles with aluminum foil.

The sample bottle is opened just before sampling. The sample bottle is not flushed and it is filled so that between the surface and the plug there is about 2 cm of air.

☑ After closure of the bottle, the socket plug is again covered with aluminum foil.

Microbiological analysis of water Determination of indicators of general contamination



Microbiological analysis of water Determination of indicators of faecal contamination



Cultivation medium

Endo 37°C coliform bacteria

mFC 44°C termotolerant coliform bacteria

SB 37°C enterococci

Coliform bacteria



Escherichia coli



E. coli on EMB agar. Has a nice green metallic sheen. How...

HiCrome Chromogenic Coliform Agar (CCA)

HiChromogenic Coliform Agar is a selective medium recommended for the simultaneous detection of *Escherichia coli* and total coliforms in water samples (1).

The medium contains three chromogenic substrates.

The enzyme β -D-galactosidase produced by coliforms cleaves 6-chloro-3-indoxyl- β -D-galactopyranoside to form pink to red coloured colonies (3).

The enzyme β-D-glucuronidase produced by *E.coli*, cleaves 5-bromo-4chloro-3-indoxylβ-D-glucuronic acid (2).

Colonies of *E.coli* give dark blue to violet coloured colonies due to cleavage of both the chromogens.

The presence of the third chromogen IPTG enhances the colour reaction. Addition of L-Tryptophan improves the indole reaction thereby increasing the detection reliability.

Termotolerant coliform bacteria

mFC agar



Colonies that are light to dark blue, in whole or part, are counted as fecal coliforms

Enterococci

Slanetz & Bartley Medium

Originally intended as a medium for the enumeration of enterococci in water using Membrane Filtration, this medium has become more popular in many other areas such as food bacteriology.

The medium contains Tetrazolium Chloride, which is reduced by enterococci to the insoluble red dye Formazan resulting in **dark red colonies** of enterococci on the agar. It should be noted that this reaction is not exclusive to enterococci and colonies should be confirmed by additional testing e.g. aesculin hydrolysis.





Physico-chemical properties

рН	Oxidability
Alkalite	Ammonia
Total hardness	Calcium
Nitrates	Magnesium
Nitrites	Iron
Chlorides	Cadmium
Sulfates	Trihalomethanes
Phosphates	PAH - polyaromatic hydrocarbons

Organoleptic evaluation - temperature, color, turbidity, taste, smell

Accredited water analysis in Brno Water management services of the Czech Republic

Basic analysis of water with mikrobiology - 1290, - CZK

pН overall hardness COD-Mn (Chemical Oxygen Demand) nitrates nitrites iron manganese sulphates chlorides ammonium ions coliform bacteria - cfu Escherichia coli - cfu

Vyhláška č. 83/2014 Sb.

Decree amending Decree No. 252/2004 Coll. Laying down hygiene requirements for drinking and hot water and frequency and scope of drinking water control, as amended Unit (cfu/ml) limit

	č.	ukazatel	jednotka	limit	typ limitu
	1	Clostridium perfringens	KTJ/100ml	0	MH
Intestinal enterococci	2	Intestinální	KTJ/100ml	0	NMH
intestinal enterococci	2	enterokoky	KTJ/250 ml	0	NMH
	2	3 Escharichia cali		0	NMH
	5	Eschenchia coli	KTJ/250 ml	0	NMH
Coliform bacteria	4	kalifarmuf haktaria	KTJ/100ml	0	MH
	4	Kollormni baktene	KTJ/250 ml	0	MH
Abioseston – org.+anorg. particles	5	mikroskopický obraz - abioseston	%	10	MH
Number of organisms	6	mikroskopický obraz - počet organismů	jedinci/ml	50	MH
Number of live organisms	7	mikroskopický obraz - živé organismy	jedinci/ml	0	MH
	_	počty kolonií při 22	KTJ/ml	Bez abnormálních změn	MH
Number of bacterial colonies (cfu)	8	°C	KTJ/ml	200	DH
at 22 ° C		-	KTJ/ml	100	NMH
	počtv kolonií nři 36		KTJ/ml	Bez abnormálních změn	MH
Number of bacterial colonies (cfu)	9	°C	KTJ/ml	40	DH
		-	KTJ/ml	20	NMH
	10	Pseudomonas aeruginosa	KTJ/250 ml	0	NMH

Vyhláška č. 83/2014 Sb.

Decree amending Decree No. 252/2004 Coll. Laying down hygiene requirements for drinking and hot water and frequency and scope of drinking water control, as amended

č.	ukazatel	zkratka	jednotka	limit	typ limitu
11	1,2-dichlorethan		μg/l	3,0	NMH
12	akrylamid		µg/l	0,1	NMH
13	amonné ionty	NH4+	mg/l	0,50	MH
14	antimon	Sb	µg/l	5,0	NMH
15	arsen	As	µg/l	10	NMH
16	barva		mg/l Pt	20	MH
17	benzen		μg/l	1,0	NMH
18	benzo[a]pyren	BaP	µg/l	0,01	NMH
19	beryllium	Be	µg/l	2,0	NMH
20	bor	В	mg/l	1,0	NMH
21	bromičnany	BrO3-	µg/l	10	NMH
22	celkový organický uhlík	TOC	mg/l	5,0	MH
23	dusičnany	NO3-	mg/l	50	NMH
24	dusitany	NO2-	mg/l	0,50	NMH
25	epichlorhydrin		µg/l	0,10	NMH
26	fluoridy	F-	mg/l	1,5	NMH
27	hliník	AI	mg/l	0,20	MH
29	hořčík	Ma	ma/l	10	MH
20	HOICIK	wig	ng/i	20-30	DH
29	chemická spotřeba kyslíku (manganistanem)	CHSK -Mn	mg/l	3,0	MH
30	chlor volný	Cl ₂	mg/l	0,3	MH
31	chlorečnany	CIO3-	µg/l	200	NMH
32	chlorethen (vinylchlorid)		µg/l	0,50	NMH
33	chloridy	CI-	mg/l	100	MH
34	chloritany	CIO2-	µg/l	200	NMH
35	chrom	Cr	µg/l	50	NMH
36	chuť			přijatelná pro odběratele	MH

37	kadmium	Cd	µg/l	5,0	NMH
38	konduktivita	k	mS/m	125	MH
39	kyanidy celkové	CN-	mg/l	0,050	NMH
40	mangan	Mn	mg/l	0,050	MH
41	měď	Cu	µg/l	1000	NMH
42	microcystin-LR		µg/l	1	NMH
43	nikl	Ni	µg/l	20	NMH
44	olovo	Pb	µg/l	10	NMH
45	ozon	O ₃	µg/l	50	NMH
46	pach			přijatelný pro odběratele	MH
47	pesticidní látky	PL	µg/l	0,10	NMH
48	pesticidní látky celkem	PLC	µg/l	0,50	NMH
49	pН	pН		6,5-9,5	MH
50	polycyklické aromatické uhlovodíky	PAU	µg/l	0,10	NMH
51	rtuť	Hg	μg/l	1,0	NMH
52	selen	Se	Hg/I	10	NMH
53	sírany	SO42-	mg/l	250	MH
54	sodík	Na	mg/l	200	MH
55	stříbro	Ag	µg/l	25	NMH
56	teplota		°C	8-12	DH
57	tetrachlorethen	PCE	µg/l	10	NMH
58	trihalomethany	THM	μg/l	100	NMH
59	trichlorethen	TCE	µg/l	10	NMH
60	trichlormethan (chloroform)		µg/l	30	NMH
61	uran	U	µg/l	15	NMH
62	vápník	Ca	ma/l	30	MH
02	vapiik	Ca	mgn	40-80	DH
63	vápník a hořčík	Ca + Mg	mmol/l	2-3,5	DH
64	zákal		ZF (n)	5	MH
6 5	železo	Fe	mg/l	0,20	MH

Example of a real protocol of water analysis



Zdravotní ústav se sídlem v Ostravě Centrum hygienických laboratoří

Zkušební laboratoř č. 1393 akreditovaná ČIA podle ČSN EN ISO/IEC 17025:2005 Partyzánské náměstí 7, 702 00 Ostrava

PROTOKOL č. 32767/2017

Zákazník : Krajská hygienická stanice Jihomoravského kraje se sídlem v Brně Jeřábkova 1847/4 602 00 Brno

Číslo zakázky : Příjem vzorku : 31.5.2017 12:54 Vyšetření vzorku : 31.5.2017 - 15.6.2017 Číslo jednací : ZU/15317/2017 Číslo spisu : S-ZU/15317/2017 Spisový znak : 4.0.3

Číslo objednávky :2BM0173

Vzorek číslo :		n ander hen han einen her her einen her eine sehen her einen her
Datum odběru :	31.5.2017 Čas odb	ěru : 12:10
Název vzorku :	voda pitná - studna	
Množství vzorku :	cca 5 1	
Místo odběru :		
Matrice :	voda pitná	
Vzorkoval :	Mrázová Zina, Bc.	
Metoda vzork. :	SOP VZ OV 001 (ČSN EN ISO 5667-1, Č	SN EN ISO 5667-3; ČSN ISO 5667-5,
	ČSN ISO 5667-14; ČSN EN ISO 19458, Č	SN ISO 11731)
Způsob odběru :	prostý vzorek	
Účel odběru :	státní zdravotní dozor	

Místní měření					
Ukazatel	Hodnota	Jednotka	ТҮР	Použitá metoda	Nejistota
teplota vzorku	12,3	°C	Α	SOP OV 042	±1°C
chlor volný	<0,03	mg/l	A	SOP OV 008.01	-
pH	7,4		A	SOP OV 033	±0,3 j.pH

Vysi	edky zkoušení - che	mické vyšetře	ni		
Ukazatel	Hodnota	Jednotka	TYP	Použitá metoda	Nejistota
1,2-dichlorethan	<0,5	μg/l	A	SOP OV 344 5	
amonné ionty	<0,040	mg/l	A	SOP OV 064 7	
Sb (antimon)	<0,50	μg/l	A	SOP OV 201 5	-
As (arzen)	<1,0	µg/l	A	SOP OV 201 5	-
barva	<5	mg/l Pt	A	SOP OV 064.02 7	
benzen	<0,5	μg/l	A	SOP OV 344 5	
benzo(a)pyren	<0,004	μg/1	A	SOP OV 331	-
B (bor)	<0,050	mg/l	A	SOP OV 201 5	-
bromičnany	<3	μg/l	A	SOP OV 003 2	-
TOC	1,3	mg/l	A	SOP OV 307 7	±20%
dusičnany	6,94	mg/l	A	SOP OV 003 7	±15%
dusitany	<0,020	mg/l	A	SOP OV 064.04 7	-
fluoridy	0,246	mg/l	A	SOP OV 003 7	±15%
Al (hliník)	<0,005	mg/l	A	SOP OV 201 5	-
Mg (hořčík)	51,6	mg/l	Α	SOP OV 201 5	±20%
CHSK-Mn	. 1,5	mg/l	A	SOP OV 016 7	±25%
chloridy	288	mg/l	A	SOP OV 003 7	±15%
chloritany	<10,0	μg/l	A	SOP OV 003 2	-
Cr (chrom)	<1,0	μg/1	A	SOP OV 201 5	-
chuť	přijatelná	-	A	SOP OV 062 7	-
Cd (kadmium)	<0,20	μg/l	A	SOP OV 201 5	-
elektrická konduktivita (25°C)	170	mS/m	A	SOP OV 064.13 7	+10%

Ukazatel	Hodnota	Jednotka	TYP	Použitá metoda	Nejistota
kyanidy celkové	<0,0050	mg/l	A	SOP OV 022.01 6	-
Mn	<0,030	mg/l	A	SOP OV 050 7	
Cu (měď)	6,2	μg/l	A	SOP OV 201 5	±20%
Ni (nikl)	<2,0	μg/l	A	SOP OV 201 5	-
Pb (olovo)	<1,00	μg/l	A	SOP OV 201 5	-
pach	přijatelný	-	A	SOP OV 062 7	-
suma PAU	0	μg/l	A	SOP OV 331 5	-
Hg (rtuť)	<0,2	μg/l	A	SOP OV 200.03 5	
Se (selen)	<1,00	µg/l	A	SOP OV 201 5	-
sírany	175	mg/l	A	SOP OV 003 7	±15%
Na (sodík)	23,9	mg/l	A	SOP OV 201 5	±20%
tetrachlorethen	<0,5	μg/l	A	SOP OV 344 5	-
trihalomethany	<0,5	μg/l	A	SOP OV 344 5	-
trichlorethen	<0,5	μg/l	A	SOP OV 344 5	-
trichlormethan (chloroform)	<0,5	μg/l	A	SOP OV 344 5	
Ca (vápník)	240	mg/l	A	SOP OV 201 5	±20%
zákal	0,58	ZF(n)	A	SOP OV 044.01 7	±20%
Fe (železo)	<0,030	mg/l	A	SOP OV 051 7	
dichlorbrommethan	<0,5	μg/l	A	SOP OV 344 5	
dibromchlormethan	<0,5	μg/l	A	SOP OV 344 5	
bromoform	<0,5	μg/1	A	SOP OV 344 5	
chlorečnany	<10,0	μg/l	A	SOP OV 003 2	-
suma vápník a hořčík (tvrdost)	7.4	mmol/l	A	SOP OV 039 7	+8%

Výsledky zkoušení - pesticidní látky

Ukazatel	Hodnota	Jednotka	TYP	Použitá metoda	Nejistota
2,4-D	<0,025	μg/l	A	SOP OV 341.02 5	
acetochlor	<0,025	μg/l	A	SOP OV 341.02 5	
acetochlor ESA	<0,025	µg/l	A	SOP OV 341.02 5	
acetochlor OA	<0,025	µg/l	A	SOP OV 341.02 5	
alachlor	<0,025	µg/l	A	SOP OV 341.02 5	-
alachlor ESA	0,064	μg/l	A	SOP OV 341.02 5	±30%
alachlor OA	<0,025	µg/l	A	SOP OV 341.02 5	-
atrazin	<0,025	μg/l	A	SOP OV 341.02 5	
desethylatrazin	0,077	μg/l	A	SOP OV 341.02 5	±30%
atrazine-desisopropyl	<0,025	μg/l	A	SOP OV 341.02 5	-
hydroxyatrazin	<0,025	μg/l	A	SOP OV 341.02 5	-
azoxystrobin	<0,025	µg/l	A	SOP OV 341.02 5	
bentazone	<0,025	μg/1	A	SOP OV 341.02 5	
carbendazim	<0,025	µg/l	A	SOP OV 341.02 5	
carboxin	<0,025	ug/l	A	SOP OV 341.02 5	
clomazone	<0,025	μg/l	A	SOP OV 341.02 5	-
clopyralid	< 0.025	ug/l	A	SOP OV 341.02 5	
cyanazin	<0,025	μg/l	A	SOP OV 341.02 5	
cyproconazole	<0,025	μg/l	A	SOP OV 341.02 5	
cyprodinil	<0,025	μg/l	A	SOP OV 341.02 5	
desmedipham	<0,025	μg/l	A	SOP OV 341.02 5	-
dicamba	<0,025	μg/1	A	SOP OV 341.02 5	
dichlormid	<0,025	μg/l	Α	SOP OV 341.02 5	
dichlorprop	<0,025	μg/l	A	SOP OV 341.02 5	
dimethachlor	<0,025	μg/l	A	SOP OV 341.02 5	
dimethenamid	<0,025	μg/l	A	SOP OV 341.02 5	-
epoxiconazole	<0,025	μ <u>α</u> /1	A	SOP OV 341.02 5	-
ethofumesate	<0,025	μg/l	A	SOP OV 341.02 5	-
fenpropidin	<0,025	μg/l	A	SOP OV 341.02 5	-
fenpropimorph	<0,025	μg/l	A	SOP OV 341.02 5	
fluazifop-p-butyl	<0,025	μg/l	A	SOP OV 341.02 5	-

Strana 1 / 4

```
AZL 1393
```

Strana 2/4

Example of a real protocol of water analysis

Ukazatel	Hodnota	Jednotka	TYP	Použitá metoda	Nejistota
fluroxypyr	<0,025	μg/l	A	SOP OV 341.02 5	-
flusilazole	<0,025	μg/l	A	SOP OV 341.02 5	
haloxyfop-metyl	<0,025	μg/l	A	SOP OV 341.02 5	-
hexazinon	<0,025	μg/l	A	SOP OV 341.02 5	-
chloridazon (pyrazon)	<0,025	μg/l	A	SOP OV 341.02 5	-
chlorotoluron	<0,025	μg/l	Α	SOP OV 341.02 5	
chlorpyrifos	<0,025	μg/l	A	SOP OV 341.02 5	-
isoproturon	<0,025	μg/l	A	SOP OV 341.02 5	
kresoxim-methyl	<0,025	μg/1	A	SOP OV 341.02 5	
lenacil	<0,025	μg/1	A	SOP OV 341.02 5	
linuron	<0,025	μg/1	A	SOP OV 341.02 5	-
MCPA	<0,025	μg/l	A	SOP OV 341.02 5	
МСРВ	<0,025	μg/l	A	SOP OV 341.02 5	-
MCPP (mecoprop)	<0,025	μg/1	A	SOP OV 341.02 5	
mefenpyr-dietyl	<0,025	μg/1	A	SOP OV 341.02 5	-
metamitron	<0,025	μg/1	A	SOP OV 341.02 5	-
metazachlor	<0,025	μg/1	A	SOP OV 341.02 5	-
metazachlor ESA	<0,025	μg/l	A	SOP OV 341.02 5	
metazachlor OA	<0.025	ug/l	A	SOP OV 341.02 5	
metconazole	<0,025	ug/1	A	SOP OV 341.02 5	
metobromuron	< 0.025	ug/1	A	SOP OV 341.02 5	
metolachlor	< 0.025	ug/1	A	SOP OV 341.02 5	
metolachlor ESA	0,089	ug/1	A	SOP OV 341.02 5	±30%
metolachlor OA	<0,025	μg/l	A	SOP OV 341.02 5	-
metoxuron	<0,025	µg/1	A	SOP OV 341.02 5	-
pendimetalin	<0,025	μg/1	A	SOP OV 341.02 5	
pethoxamid	<0,025	ug/1	A	SOP OV 341.02 5	
phenmedipham	<0,025	µg/1	A	SOP OV 341.02 5	
prochloraz	<0,025	μg/1	A	SOP OV 341.02 5	-
propiconazole	<0,025	ug/l	A	SOP OV 341.02 5	
quinmerac	< 0.025	ug/l	A	SOP OV 341.02 5	
quinoxyfen	< 0.025	ug/1	A	SOP OV 341.02 5	
sebutylazin	< 0.025	ug/1	A	SOP OV 341.02 5	
simazin	< 0.025	ug/1	A	SOP OV 341.02 5	
spiroxamin	<0,025	µg/l	A	SOP OV 341.02 5	
tebuconazole	<0.025	ug/l	A	SOP OV 341.02 5	
terbuthylazin	<0.025	ug/1	A	SOP OV 341.02 5	1
thiophanate-methyl	<0.025	ug/l	A	SOP OV 341.02 5	+ :
nesticidní látky celkem	0.077			SOD OV 241 02 5	. 200/

Výsledky zkoušení - mikrobiologické vyšetření					
Ukazatel	Hodnota	Jednotka	TYP	Použitá metoda	Nejistota
Clostridium perfringens	0	KTJ/100ml	N	SOP OV 914.01 7	-
intestinální enterokoky	0	KTJ/100ml	A	SOP OV 906 7	
Escherichia coli	3	KTJ/100ml	A	SOP OV 900 7	1-9
koliformní bakterie	50	KTJ/100ml	A	SOP OV 900 7	36-64
počty kolonií při 22°C	1,44x10 ²	KTJ/ml	Α	SOP OV 908 7	$1,2x10^{2}$ - $1,68x10^{2}$
počty kolonií při 36°C	29	KTJ/ml	A	SOP OV 908 7	18-40
abioseston	3	%	A	SOP OV 916 7	30%
počet organismů	0	jedinci/ml	Α	SOP OV 916 7	-
živé organismy	0	jedinci/ml	A	SOP OV 916 7	-
mrtvé organismy	0	jedinci/ml	A	SOP OV 916 7	-

Poznámka k odběru : Odběr je předmětem akreditace, nedílnou součástí protokolu je Protokol o odběru č. 7404/2017

Poznámky k analýze :

Do sumy pesticidů jsou zahrnuty pouze relevantní metabolity. Ostatní nerelevantní metabolity jsou uvedeny pro informaci zákazníkovi. Postup pro hodnocení viz. metodika SZÚ. Suma PAU obsahuje: bezo(b)fluoranthen, benzo(k)fluoranthen, benzo(ghi)perylen, indeno(1,2,3-cd)pyren.

Upřesnění SOP :

e presenten o or r	
SOP OV 003	(ČSN EN ISO 15061, ČSN EN ISO 10304-1, ČSN EN ISO 10304-4)
SOP OV 008.01	(návod firmy HACH)
SOP OV 016	(ČSN EN ISO 8467)
SOP OV 022.01	(ČSN ISO 6703-2, ČSN 75 7415)
SOP OV 033	(ČSN ISO 10523)
SOP OV 039	(ČSN ISO 6059)
SOP OV 042	(ČSN 75 7342)
SOP OV 044.01	(ČSN EN ISO 7027-1)
SOP OV 050	(ČSN ISO 6333)
SOP OV 051	(ČSN ISO 6332)
SOP OV 062	(TNV 75 7340)
SOP OV 064.02	(návody firmy Thermo Scientific)
SOP OV 064.04	(návody firmy Thermo Scientific)
SOP OV 064.13	(návody firmy Thermo Scientific)
SOP OV 064	(návody firmy Thermo Scientific)
SOP OV 200.03	(ČSN 75 7440)
SOP OV 201	(ČSN EN ISO 17294-1,ČSN EN ISO 17294-2)
SOP OV 307	(ČSN EN 1484)
SOP OV 331	(ČSN EN ISO 17993)
SOP OV 341.02	(EPA 535, EPA 536)
SOP OV 344	(ČSN EN ISO 15680, ČSN EN ISO 10301)
SOP OV 900	(ČSN EN ISO 9308-1:2015)
SOP OV 906	(ČSN EN ISO 7899-2)
SOP OV 908	(ČSN EN ISO 6222)
SOP OV 914.01	(Vyhláška č. 252/2004 Sb., příloha č.6)
SOP OV 916	(ČSN 75 7712, ČSN 75 7713, ČSN 75 7717)

Místo provedení zkoušky (pracoviště) :

(2) - analýzy provedeny pracovištěm Ostrava (Partyzánské nám. 7, 702 00 Ostrava)

- ⁽⁵⁾ analýzy provedeny pracovištěm Olomouc (Wolkerova 6, 779 11 Olomouc)
- ⁶⁾ analýzy provedeny pracovištěm Jihlava (Vrchlického 57, 587 25 Jihlava)
- ⁽⁷⁾ analýzy provedeny pracovištěm Brno (Gorkého 6, 602 00 Brno)

Metody v sloupci TYP:"A" akreditovaná zkouška."N" neakreditovaná zkouška

< - výsledek pod mez detekce, > - výsledek je vyšší než uvedená hodnota

Výsledky se týkají pouze zkoušených vzorků.

Bez písemného souhlasu laboratoře se nesmí protokol reprodukovat jinak než celý.

Uvedené rozšířené nejistoty měření jsou součinem standardní nejistoty měření a koeficientu rozšíření k=2, což odpovídá hladině

spolehlivosti přibližně 95 %, nezohledňují vlivy odběrů vzorků.

Pro mikrobiologické ukazatele je nejistota měření vyjádřena jako 95% konfidenční meze vyjadřující variabilitu Poissonova rozdělení, nezohledňují vlivy odběrů vzorků.

 Vedoucí CHL:
 Doškářová Šárka, RNDr.

 Kontroloval:
 Komínková Jana, Bc.

 Protokol vyhotovil:
 Komínková Jana, Bc.

 Počet stran:
 4

 Dne:
 20.6.2017

AZL 1393

Ing. Dagmar Pecáková zástupce vedoucího Oddělení anorganických analýz

Strana 3 / 4

Waste water

- Ancient Greece, Rome the first sewer system, sewage drawn into rivers, or soak up.
- The Middle Ages a great decline.
- 18th century construction of sewerage systems (drainage of army objects, later church and public buildings).
- The end of the 19th century sewer systems in most European cities.
- 1865 England The creation of the "Royal Commission on River Pollution".
- 1860 First sewage WWTP sewage farms.
- 1880 First sedimentary treatment plant.
- 1900 First intermittent biofilters.
- 1910 USA experimental aeration of sewage
- 1912 England Invention of the activation system



Waste water

- The protection of watercourses is primarily aimed at ensuring the **required quality of effluent** on runoff from WWTP.
- Typical symptoms of **waste water content in streams** (odor, sludge, oxygen deficiency) caused by organic substances in European proportions are virtually eliminated.
- <u>New cleaning issues</u>: nitrogen, phosphorus, microfibers, pharmaceuticals.
- Act No. 254/2001 Coll. on water defines the concept of wastewater.
- Act No. 274/2001 Coll. about public water supply systems and sewers as amended by Act No. 76/2006 Coll.
- Government Decree No. 61/2003 Coll. in the act 229/2007 Coll. A 23/2011 Coll. about indicators and values of permissible surface water pollution, discharging requirements of sewage into surface water and sewage waters and about sensitive areas.
- Act No. 185/2001 Coll. on waste determines the treatment of waste from WWTPs (sludge, shrubs, sand, soil from root fields, etc.)

Waste water

Basic methods of wastewater treatment

- Mechanical
- Physical-chemical
- Biological
- In practice, combination of the three procedures are used

Sewage sorting

- Wastewater sewage
- Waste water industrial
- Wastewater rainfall (rainfall)

Waste water – WWTP scheme



The waste water first goes to the **gravel trap**. The next level is **ramp**

- removal of rough floating dirt (protection of other parts of the treatment plant)
- landfill, composting, combustion.

A sand trap is next

The last device for mechanical cleaning

 - the settling tank is a gravitational separation of suspended substances

Primary sludge is formed

- processed in the sludge management.



BIOLOGICAL CLEANING (SECONDARY)

- activation line
- secondary settling tank

Activation tank - microorganisms in aerobic conditions decompose organics

- biological oxidation of organic substrate, ammoniacal nitrogen, nitrification, denitrification, biological decomposition of phosphorus, or its chemical precipitation.

Secondary settling tank

separation of activated sludge from purified water

Sludge processing technology

- 1. Concentration
- 2. Stabilization
- 3. Drainage
- 4. Hygiene
- Stabilization of sludge reduction of degradable organic matter (microbes in aerobic or anaerobic conditions) and the destruction of pathogenic microorganisms
- Total decomposition and other sludge stabilization methods:
- drying at low temperatures,
- drying at high temperatures (105 ° C),
- total decomposition with oxygen at 160 ° C,
- combustion of sludge with other fuels in power stations or cement plants

Sludge hygiene

- heat treatment of sludge at high temperatures
- pasteurization of sludge
- chemical treatment of sludge liming
- anaerobic thermophilic processing methods
- composting
- special methods: ionizing radiation, ozone, decomposition

Category I - sludges can generally be applied to land used in agriculture **Category II -** sludge can be applied to agricultural land intended for the cultivation of technical crops, and land that will not grow for at least 3 years after the use of sewage sludge field vegetables and intensely fruit-bearing fruit planting

Sludge category	Acceptible microbial count per gram of dry matter of sludge		
Kategorie kalů	Přípustné množství mikroorganismů (KTJ*) v 1 gramu sušiny aplikovaných kalů		
	termotolerantní koliformní bakterie	enterokoky	Salmonella sp.
l.	< 10 ³	< 10 ³	negativní nález
П.	10 ³ - 10 ⁶	10 ³ - 10 ⁶	nestanovuje se

Limit values of concentrations of selected hazardous substances and elements in slurries for their use on agricultural land

Riziková látka	Mezní (maximální) hodnoty koncentrací v kalech (mg.kg ⁻¹ sušiny <u>)</u>	
As – arzén	30	
Cd - kadmium	5	
Cr - chrom	200	
Cu - měď	500	
Hg - rtuť	4	
Ni - nikl	100	
Pb - olovo	200	
Zn - zinek	2500	
AOX	500	
PCB (suma 6 kongenerů - 28+52+101+138+153+180)	0,6	

In dry matter

Waste is every movable thing the person discards, or has the intention or duty to discard it, and it belongs into one of the categories of waste listed in the catalog waste (Act No. 185/2001 Coll.)

Types by production:

Communal-mixed - occurring in the activity of natural persons, dangerous drugs, cans from paints and thinners, etc. - it is the largest amount of all wastes
 Agricultural - wastes from plant and livestock production (urine, manure), agrochemicals (fertilizers)
 Industrial - from industrial activities

According to the composition:

Inert - not subject to biodegradation - does not decompose/rot Construction waste, glass, has no dangerous properties
Biological - is capable of aerobic and anaerobic digestion. Food, paper, green mater
Toxic, radioactive - nuclear waste, old batteries, drugs, oils
Hospital - parts of human bodies, infectious waste, needles, plasma, cytostatics

Categories of waste from medical facilities

Specific waste

Sharp objects Pathological and biologically contaminated waste Discarded chemicals, drugs

Non-specific waste

Other wastes not requiring protection against the spread of infections

Exposure:

Dust

Contact

Mediated (contamination of water, soil, food chains, insects, rodents

Effects:

Irritation, toxic smoke

Parasitic infections - helminthoses, amoebae, cercariae (maggot stage, the evolutionary stage of the motolic, leaving the guts and looking for the host - the human)

Zoonoses

Hepatitis, AIDS

Poisoning

Action:

Minimizing quantity (recycling, use of one – off packaging)

Selection of suitable sites and landfill security

Separation of toxic, infectious, radioactive waste

Education of the population

Waste disposal:

Landfills Composting Combustion Chemical destruction Storage Feeding to animals Recycling of sorted waste (glass, paper, plastics, metals)

Preparation of landfills – must be waterproof


Preparation of landfills – must be waterproof

But also drained – any water from waste and rainfall – otherwise there would be a lake soon



When the landfills is full



Once the landfills is covered we must remove gases – methane etc – burn it

