Pharmacokinetics

- General principles of the fate of the drug in the body
- Overview of pharmacokinetic processess: Drug absorption,

distribution, metabolism and elimination

Pharmacokinetics

Occupation theory: The intensity of pharmacological response (E) is proportional to the conentration of reversible drug-receptor complex

 Action of a drug requires presence of a certain concentration in the fluid bathing the target tissue.

Pharmacokinetics deals with the processes of

absorption, A distribution, D invasion metabolism M excretion of the drug E "ADME"

And their relationship with their biological (pharmacological) effect

"WHAT DOES ORGANISM DO WITH THE DRUG"

What does influence the movements of the drug in the body?

physico-chemical properties

lipophilic/hydrophilic properties, molecule

structure, pKa, charge...

AH \leftrightarrows A- + H+ B + H+ \leftrightarrows BH+ permeation across the membranes

lipophilic – difusion (passive) hydrophilic – through the pores active transport

bonds of the drugs to:

plasma proteins

blood cells in the circulation

tissue

receptors

perfusion of the tissues

a) brain, heart, liver, kidneyb) fat tissue





A bound drug has no effect!

Amount bound depends on:

- 1) free drug concentration
- 2) the protein (binding sites) concentration
- 3) affinity for binding sites

% bound: <u>[bound drug]</u> x 100 [bound drug] + [free drug]

ABSORPTION

F

Absorption – permeation of the soluted drug into the body fluids from the site of administration – necessary for the general (systemic) effect

Local effect – on the skin, mucous membranes...

mouth, rectum, vagina

- absorption is fault, can cause difficulties, adverse

effects)

(local aenesthetics, corticosteroids)

Rate and extent of absorption are described by the parameters :

C max - max. concentration of the drug in the plasma after single administration

T max - time after administration, when is Cmax

- bioavailability (extent of absorption)

Bioavailability

• The fraction of the dose of a drug (F) that enters the general circulatory system,

F= <u>amt. of drug that reach systemic circul</u>. Dose administered

F = AUCp.o./AUCi.v.

Area under curve (AUC)

Is a measure of bioavailability



 $F = AUC_{p.o.} / AUC_{iv}$

Bioavailability

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Extravascular route - 0-100% (resp. 0-1).
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Intravenous - 100\% = 1
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If F is 0-20% = 0-0,2 – not suitable route of administration

(in spite of that fact, some drugs are administered, even if the F < 2-5 %, such as SET, bisphosphonates).

F = AUCpo/AUCiv

(the same drug, same dose, same patient)

 $M \vdash I$

Bioavailability

- A concept for oral (extravascular) administration
- Useful to compare two different drugs or different dosage forms of same drug
- depends, in part, on rate of dissolution (which in turn is dependent on chemical structure, pH, partition coefficient, surface area of absorbing region, etc.) Also first-pass metabolism is a determining factor





First pass effect, presysthemic elimination



Other factors influencing the absorption

gender, body weight, plasma volume, gastric amptying rate,

age - pH, bile, enzyme levels and activity

patophysiological state – liver disseases, inflammation

simultaneously eaten meal -

acceleration/decelaration

chemical incompatibilities

function of the GIT

Distribution

= permeation from the blood to the tissues and site of the action is dynamic process

rate - depends on:

bonds (with the plasmatic proteins...) permeation across the membrabes blood perfusion through the organ

state - distribution equilibrium; the the proportion of the free (unbounded) fractions of the drug in the blood and in the tissues are the same

Barriers – the distribution is limited

blood-brain barrier ("leaky areas" – area postrema),

penicilines X aminoglycosides

placental barrier...

Volume of Distribution

Volume of distribution – apparent, hypotethical the proportion of the quantity of the drug and reached plasmatic concentration

 $V_d = D/C$

- $-V_d$ is the apparent volume of distribution
- C= Conc of drug in plasma at some time
- D = Total quantity (dose) of drug in system

V_d gives one as estimate of how well the drug is distributed.
 Value < 0.071 L/kg indicate the drug is mainly in the circulatory system. Values > 0.071 L/kg indicate the drug has gotten into specific tissues.

Volumes of the water in human body celková tělesná voda 421 voda extracelulární voda intracelulární 141 281 plazma voda intersticiální $10\,1$ 41

ELIMINATION

Biotransformation – metabolism

Sites of biotransformation

anywhere, where the enzymes are present: plasma, kidney, lungh GIT, brain, but especially **liver**

Enzymatic

- biodegradation
- bioactivation (prodrug)

enalapril-enalaprilate

codein-morphine

bromhexin - ambroxol

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Cytochrom P450, dehydrogenases

2. Phase : conjugation – metabolites are more soluble in the water

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Metabolite - effective ("more / less / in other way")

- ineffective

- toxic





Genetic polymorphism

Genetic polymorphism = the existence of several (At least two) alleles for the gene from which At least part has a population frequency of at least 1 %

Pharmacogenetics

focuses on the study of genetically conditioned variability in the response to a drug

•Pharmacogenomics examines the relationship of drug effect on the level of the whole genome, respectively transcriptome

Genetic polymorphism of biotransformation enzymes

Polymorphism in the gene of *N* - *acetyltransferase*

- Inactivation of drugs in the liver : slow x fast acetylators
- Isoniazide, procainamide, hydralazine
- Peripheral neuropathy (prevention pyridoxine)

Polymorphism of *thiopurine S - methyltransferase*

- the metabolism of azathioprine
- commercially available genetic test for determining the polymorphisms, prevention of serious adverse reactions



INDUCERS of CYP 450

- dexamethason
- phenobarbital
- rifampicine
- phenytoin
- St. John's Wort (*Hypericum perforatum*)
- Ginkgo biloba



INHIBITORS of CYP 450

- antidepressants (fluoxetine, fluvoxamine, paroxetine)
- quinine, quinidine
- chloramphenicol, erythromycin
- ketoconazol, itraconazol
- grapefruit juice



Phase I of biotransformation

hydroxylation oxidation O-dealkylation N-dealkylation N-oxidation oxidative deamination $\begin{array}{rcl} -\mathrm{CH}_{2}\mathrm{CH}_{3} & \rightarrow -\mathrm{CH}_{2}\mathrm{CH}_{2}\mathrm{OH} \\ -\mathrm{CH}_{2}\mathrm{OH} & \rightarrow -\mathrm{CHO} \rightarrow -\mathrm{COOH} \\ -\mathrm{CH}_{2}\mathrm{OHCH}_{2} \rightarrow -\mathrm{CH}_{2}\mathrm{OH} + -\mathrm{CHO} \\ -\mathrm{N}(\mathrm{CH}_{3})_{2} & \rightarrow -\mathrm{NHCH}_{3} + \mathrm{CH}_{3}\mathrm{OH} \\ -\mathrm{NH}_{2} & \rightarrow -\mathrm{NHOH} \\ -\mathrm{CH}_{2}\mathrm{CHCH}_{3} \rightarrow -\mathrm{CHCOCH}_{3} + \mathrm{NH}_{3} \\ \mathrm{NH}_{2} \end{array}$

Other non-microsomal biotransformations

- hydrolysis of esters in plasma (suxamethonium by cholinesterase)
- dehydrogenation of alcoholic and aldehydic group in cytosol in the liver (ethanol)
- MAO in mitochondria (tyramine, noradrenaline, dopamine, amines)
- xanthinoxidase (6-merkaptopurine, uric acid)
- enzymes with distinct function (tyrosine-hydroxylase, dopadecarboxylase, etc.)



Phase II of biotransformation

CONJUGATION

Glucuronides -OH, -SH, -COOH, -CONH wih glucuronyl acid (UDP- GlcUAc) Sulphates: with -OH functional group

Acetylates: acetyl CoA with NH_2 , -CONH₂, s aminoacid- group

with gluthathion with -halogen- or -nitrate functional groups, epoxides sulphates





saliva, skin, hair, breast milk...

Kidney

- MW < 60.000 D (MW albumin = 68.000 D)
- tubular secretion
 - organic acids
 - furosemid
 - thiazide diuretics
 - penicilins
 - glucuronides
 - organic bases
 - Morphine
 - Atropine
 - Histamine...
- tubular reabsorption

acidification

acetazolamid (inhibitor of CA) ammonium chloride

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alcalization

sodium bicarbonate

Liver

Billiar excretion, clearance.

enterohepatic circulation



ELIMINATION = biotransformation + excretion

Kinetics of the first order = rate of elimination is descending with the descending concentration in the blood (linear kinetics)

Kinetics of the zero order
= rate of elimination is constant (nonlinear kinetics)

First Order Kinetics







Elimination (first order)

Elimination constant $k_e = lnc_1 - lnc_2 / t_2 - t_1$

Half-life of the elimination – the drug is completely eliminated after 4-5 t_{0,5}

 $t_{0,5} = \ln 2 / k^e = 0,7 / k_e$

clearance Volume of the blood in a defined region of the body that is cleared of a drug in a unit time

 $CI_{TOT} = D/AUC = k_e Vd$

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Clearance Cl

- Volume of blood in a defined region of the body that is cleared of a drug in a unit time.
- more useful concept in reality than k_{el} since it takes into account blood flow rate
- Clearance varies with body weight
- Also varies with degree of protein binding

PHARMACOKINETIC PARAMETERS

