

Kidneys in regulation of homeostasis

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Homeostasis

= maintenance of stable conditions in the internal body environment

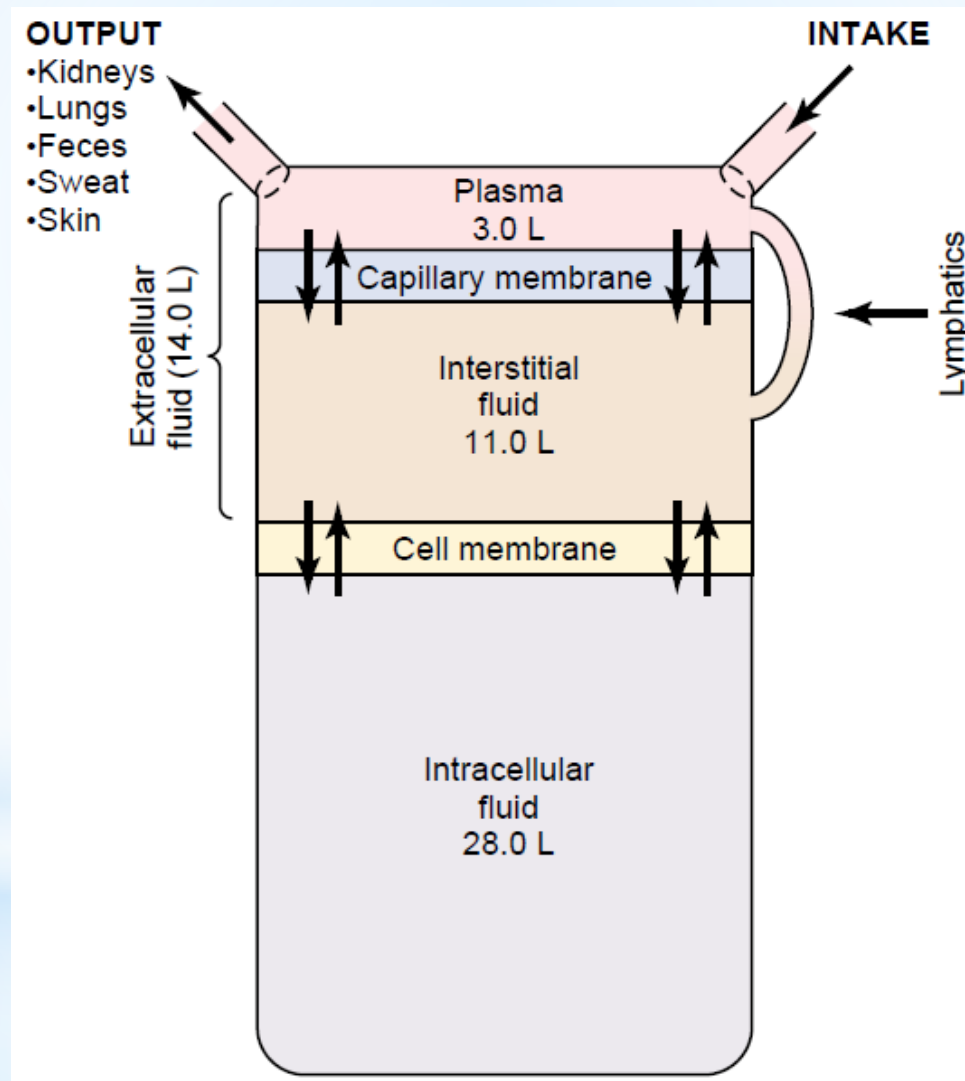
Maintenance of Constant Volume and Composition of Body Fluids

Maintenance of Acid-Base Balance

Constant Volume and Composition of Body Fluids - Regulation by Kidneys -

Body Fluids – Types and Volumes

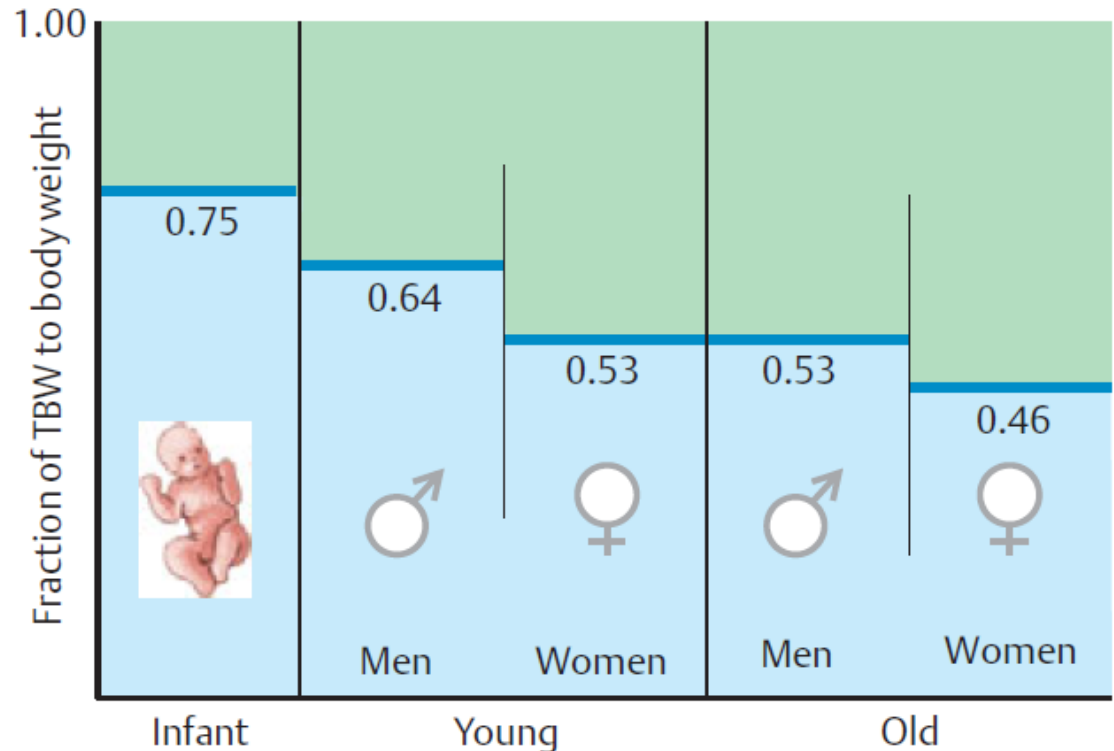
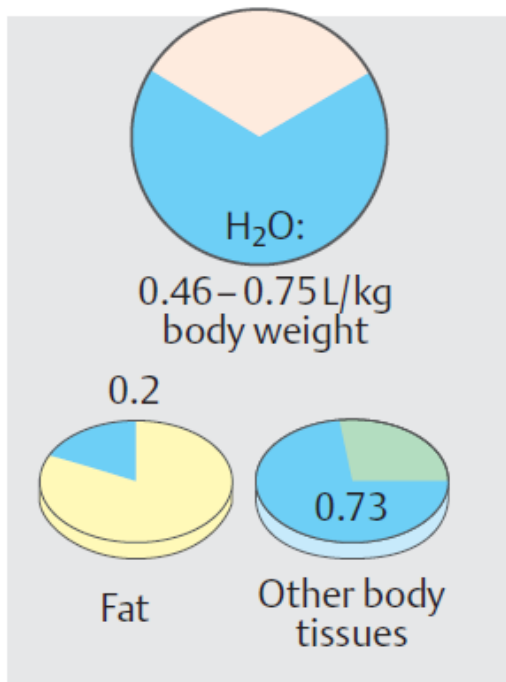
Body fluids occupy ~60% of the body weight.



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Body Fluids – Types and Volumes

B. Total body water (TBW) content

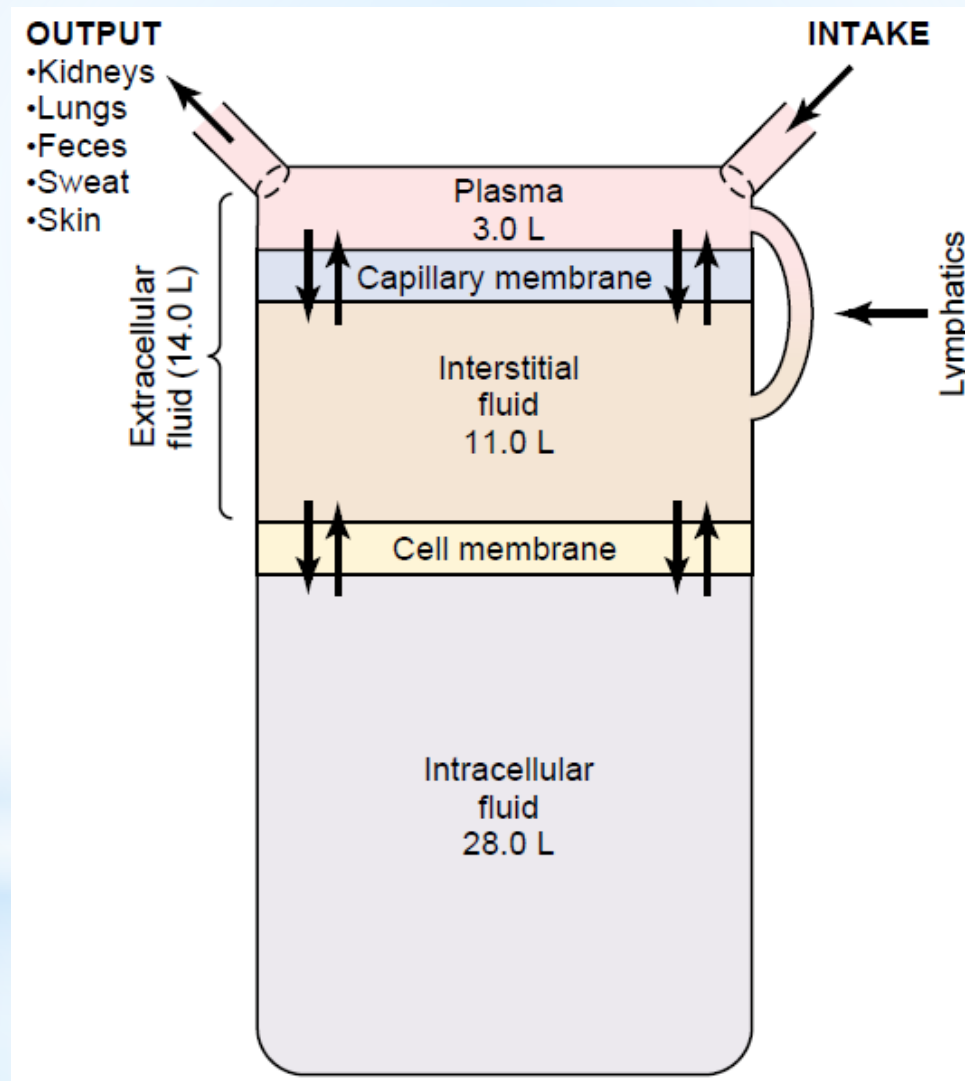


Despopoulos, Color Atlas of Physiology © 2003

Body Fluids – Types and Volumes

Body fluids occupy ~60% of the body weight.

Transcellular fluid (1-2 l) - special type of ECF. (peritoneal, pericardial, synovial, cerebrospinal and intraocular fluid)



5% of the body weight

15% of the body weight

40% of the body weight

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Body Fluids – Types and Volumes

Balance between Input and Output of Fluid

Daily Intake and Output of Water (ml/day)

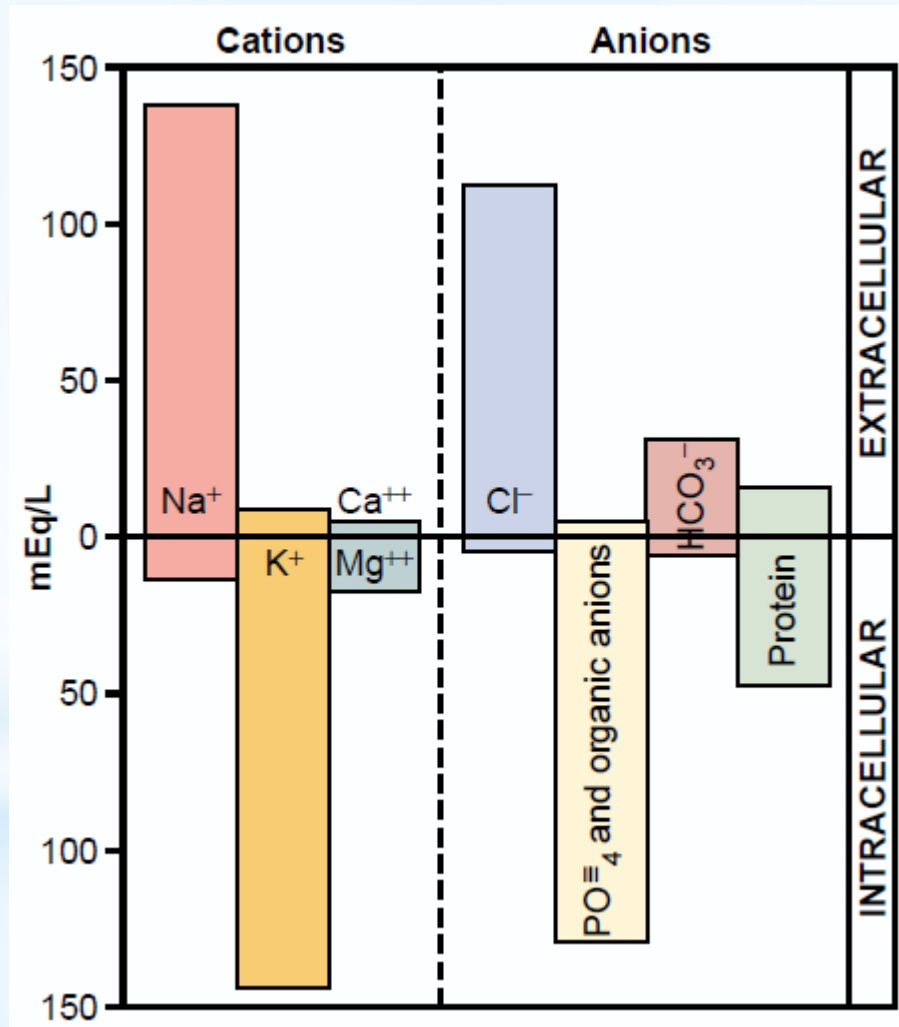
	Normal	Prolonged, Heavy Exercise
Intake		
Fluids ingested	2100	?
From metabolism	<u>200</u>	<u>200</u>
Total intake	2300	?
Output		
Insensible—skin	350	350
Insensible—lungs	350	650
Sweat	<u>100</u>	<u>5000</u>
Feces	100	100
Urine	<u>1400</u>	<u>500</u>
Total output	2300	6600

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Body Fluids – Composition

ECF vs. ICF



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Body Fluids – Composition

plasma vs. ISF

	Plasma (mOsm/L H ₂ O)	Interstitial (mOsm/L H ₂ O)
Na ⁺	142	139
K ⁺	4.2	4.0
Ca ⁺⁺	1.3	1.2
Mg ⁺	0.8	0.7
Cl ⁻	108	108
HCO ₃ ⁻	24	28.3
HPO ₄ ⁻ , H ₂ PO ₄ ⁻	2	2
SO ₄ ⁻	0.5	0.5
Phosphocreatine		
Carnosine		
Amino acids	2	2
Creatine	0.2	0.2
Lactate	1.2	1.2
Adenosine triphosphate		
Hexose monophosphate		
Glucose	5.6	5.6
Protein	1.2	0.2
Urea	4	4
Others	4.8	3.9

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Body Fluids – Composition

osmolality

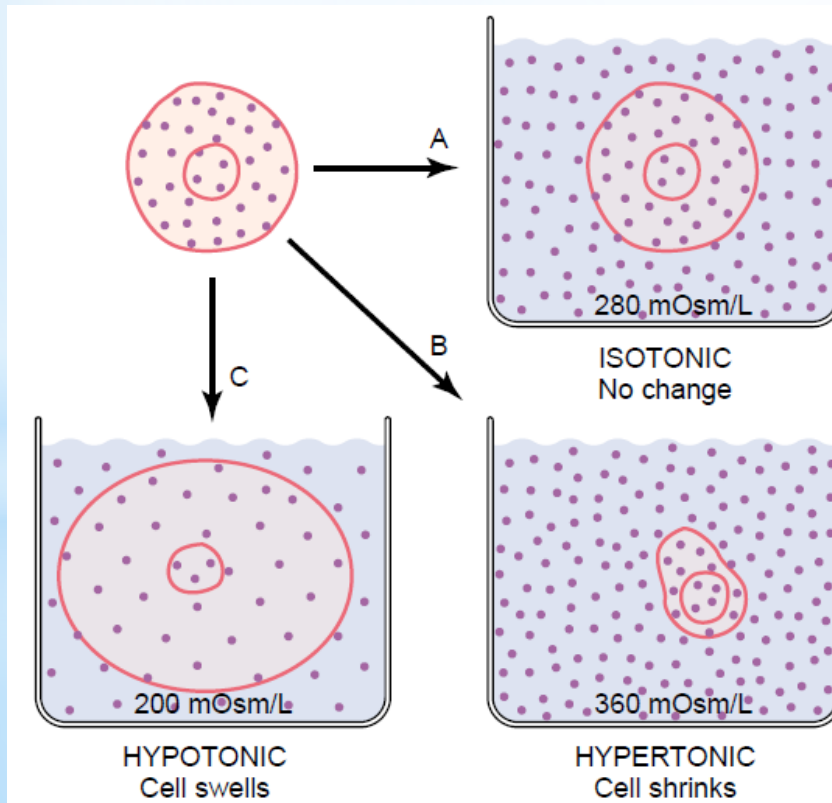
285 mosm/kg H₂O



↑ NaCl intake, loss of water → water leaves cells
(shrinking of cells)



↓ NaCl intake, ↑ water input → water sucked into cells
by osmosis (cell edema)



*Guyton & Hall. Textbook
of Medical Physiology.*



Body Fluids – Composition

osmolality

285 mosm/kg H₂O



↑ NaCl intake, loss of water → water leaves cells
(shrinking of cells)



↓ NaCl intake, ↑ water input → water sucked into cells
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Precise regulation of osmolality of ESF is necessary!

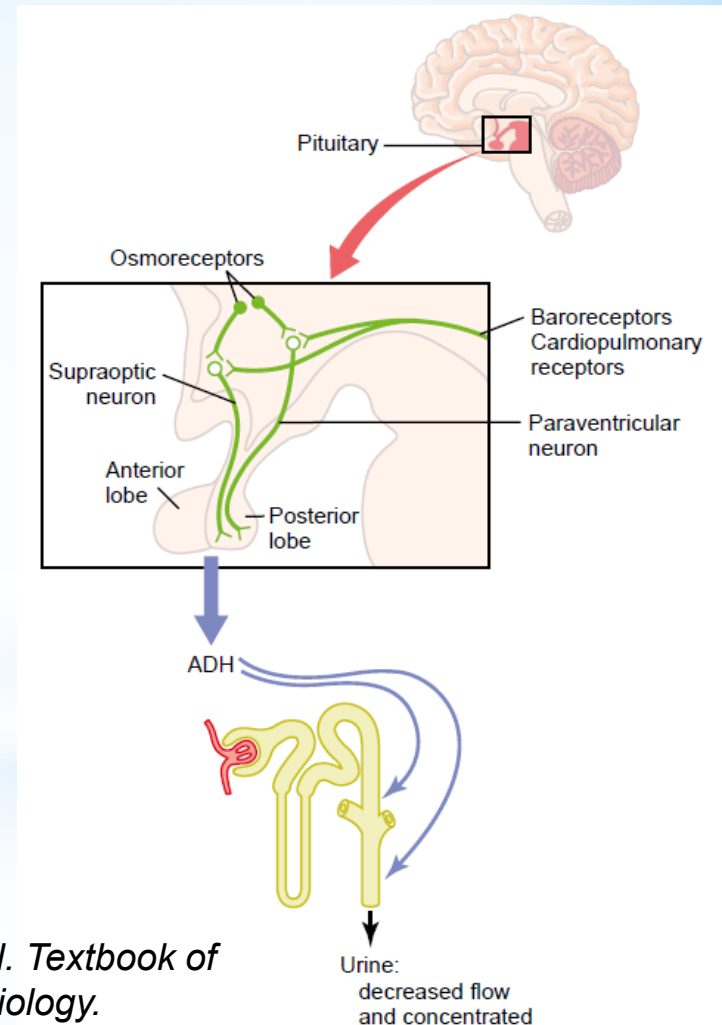
- **osmoreceptors**
 - **kidneys** (target organ for the action of hormones below)
 - **antidiuretic hormone**
-
- **aldosteron**
 - **natriuretic peptides**

Humoral Regulation of Body Fluids

Antidiuretic Hormone (*vasopressin*)

- effects:

- water reabsorption in kidneys (collecting duct, aquaporin 2)
- control of blood pressure (water reabsorption, vasoconstriction)
- ↑ glycogenolysis, mediator in the brain, ↑ secretion of ACTH in adenohypophysis



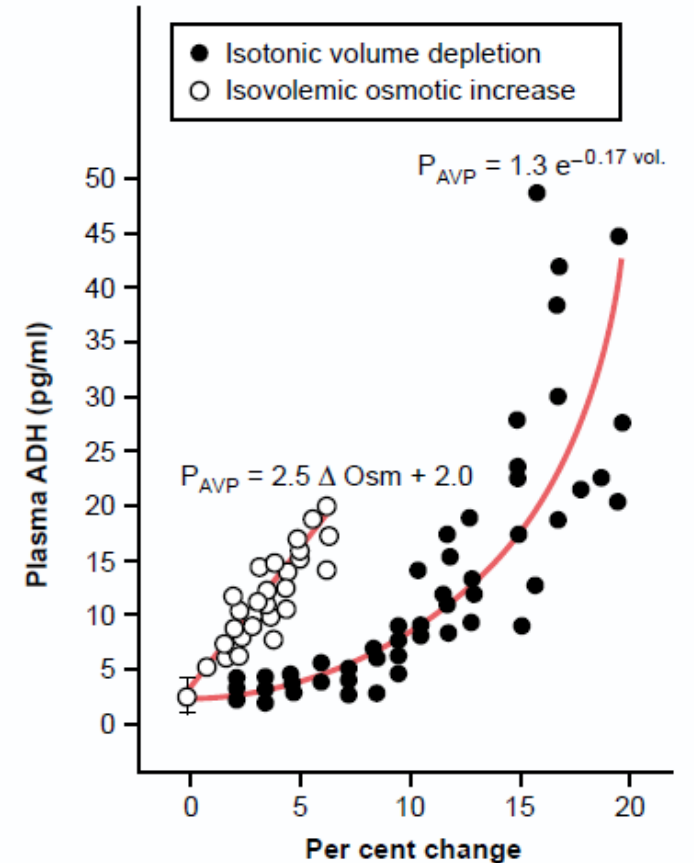
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Humoral Regulation of Body Fluids

Antidiuretic Hormone (*vasopressin*)

- regulation of secretion:

- ↑ - ↑ osmolality
- ↓ volume of ECF
- pain, emotions, stress (surgical), physical exertion; standing
- nausea, vomiting
- angiotensin II
- morphine, nicotine, barbiturates, ...
- ↓ - ↓ osmolality, ↑ volume of ECF
- alcohol; antagonists of opioids



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Humoral Regulation of Body Fluids

Antidiuretic Hormone (*vasopressin*)

- pathology:

↑ SIADH

↓ *diabetes insipidus*

Humoral Regulation of Body Fluids

Aldosteron

- the most important steroid with the mineralocorticoid effect

- **mechanism of action:**

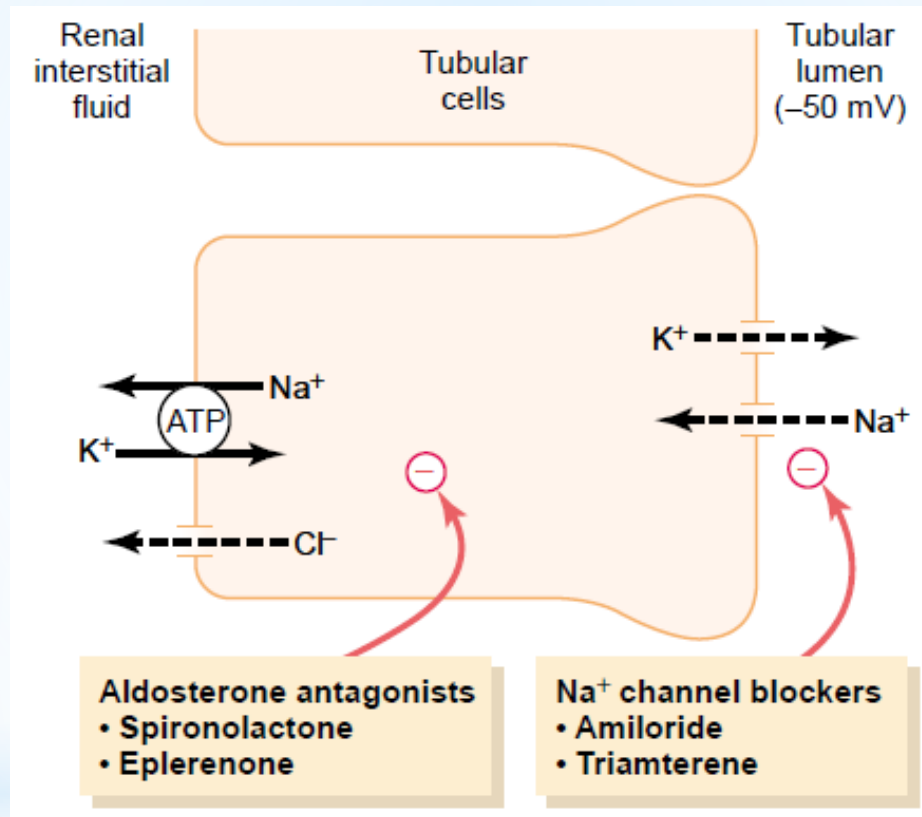
binding to the mineralocorticoid receptor → binding of the hormone-receptor complex to DNA → mRNA → synthesis of proteins:

- namely **Na⁺/K⁺-ATPase**

- ↑ number of amiloride-inhibited **Na⁺-channels** in the membrane of target cells

Start of the effect even 10 – 30 min after release of the hormone!





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Humoral Regulation of Body Fluids

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- namely **Na⁺/K⁺-ATPase**
- ↑ number of amiloride-inhibited **Na⁺-channels** in the membrane of target cells
- ↑ activity of **H⁺-pump** in collecting ducts of the renal cortex
- ↑ activity of **Na⁺/H⁺-antiport** in both distal and proximal parts of nephrons

Start of the effect even 10 – 30 min after release of the hormone!

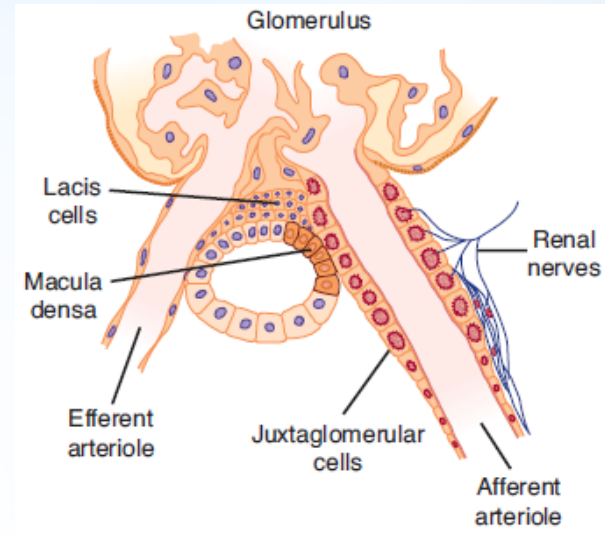
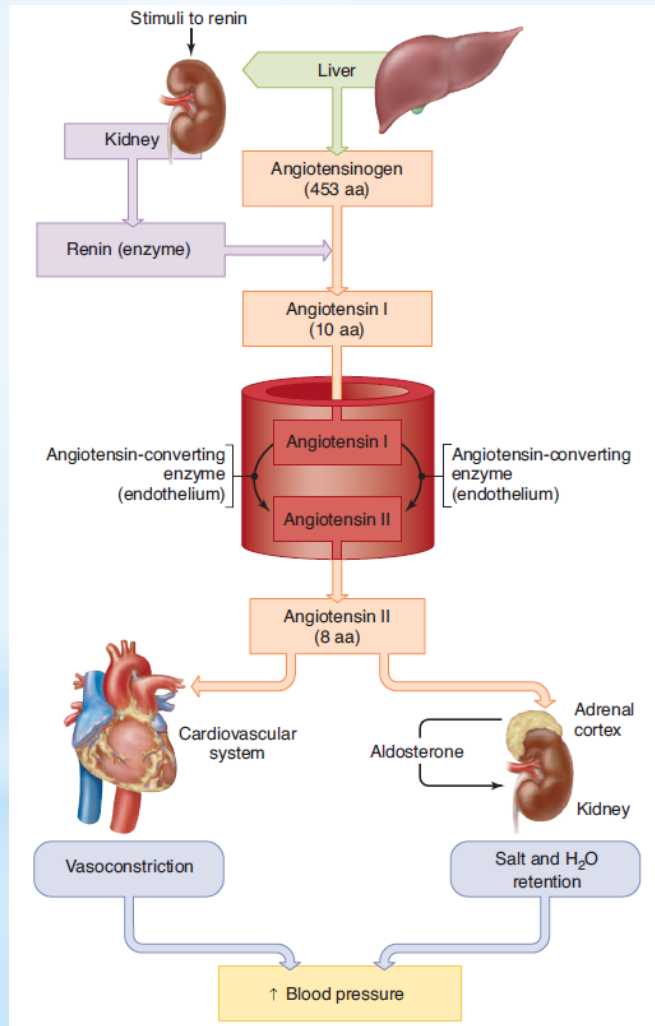
Humoral Regulation of Body Fluids

Aldosteron

- the most important steroid with the mineralocorticoid effect
- **effects:**
 - ↑ Na⁺ reabsorption from urine, sweat, saliva, gastric juice
 - ↑ K⁺ urine excretion, ↑ acidity of urine (exchange for Na⁺)
 - ↑ K⁺ content and ↓ Na⁺ content in muscle and brain cells
- **regulation of its secretion:**
 - **ACTH** from the adenohypophysis (transient effect)
 - direct stimulatory effect of ↑ **plasmatic concentration of K⁺** (even a small change – even after a meal rich for K⁺ - fruit, vegetable) **and ↓ Na⁺** (only a big change)
 - **renin-angiotensine-aldosteron system**

Humoral Regulation of Body Fluids

Renin-Angiotensin-Aldosterone System



Humoral Regulation of Body Fluids

Aldosteron

- the most important steroid with the mineralocorticoid effect
- regulation of its secretion:
 - ACTH from the adenohypophysis (a transient effect)
 - direct stimulatory effect of \uparrow plasmatic concentration of K^+ (even a small change – even after a meal rich for K^+ - fruit, vegetable) and \downarrow Na^+ (only a big change)
 - renin-angiotensine-aldosteron system
 - atrial natriuretic peptide (inhibition of renin secretion, \downarrow reactivity of *zona glomerulosa* to angiotensine II)
 - other hormones of adenohypophysis (besides ACTH; maintenance of reactivity of *zona glomerulosa*)

Humoral Regulation of Body Fluids

Aldosteron - Pathology

Primary hyperaldosteronism (Conn's syndrome)

- tumors of adrenal cortex which secretes aldosteron
 - heavy K^+ depletion
 - hypertension
 - **ECF expansion** (without edemas, without marked hypernatremia – redundant salts released by the so called escape phenomena)
- at a prolonged K^+ depletion:
 - renal damage → polyuria (the hypocalcemic nephropathy)
 - muscle weakness
 - metabolic alkalosis → ↓ plasmatic concentration of Ca^{2+} → latent or fully developed tetany
 - glucose intolerance

Humoral Regulation of Body Fluids

Aldosteron - Pathology

Primary hyperaldosteronism (Conn's syndrome)

- tumors of adrenal cortex which secretes aldosteron

Secondary hyperaldosteronism

- patients with the congestive heart failure, nephrosis, liver cirrhosis, renal artery constriction, hypertension, with the salt-losing form of adrenogenital syndrome

Hyporeninemic hypoaldosteronism

Pseudohypoaldosteronism

Humoral Regulation of Body Fluids

Atrial Natriuretic Peptide

- one of natriuretic peptides (BNP – cardiac ventricles, CNP – brain)
- receptors (ANPR-A – the highest affinity to ANP, ANPR-B – CNP, ANPR-C – all NP)
- short half-life
- secreted by atrial cardiomyocytes, found also in the brain

Humoral Regulation of Body Fluids

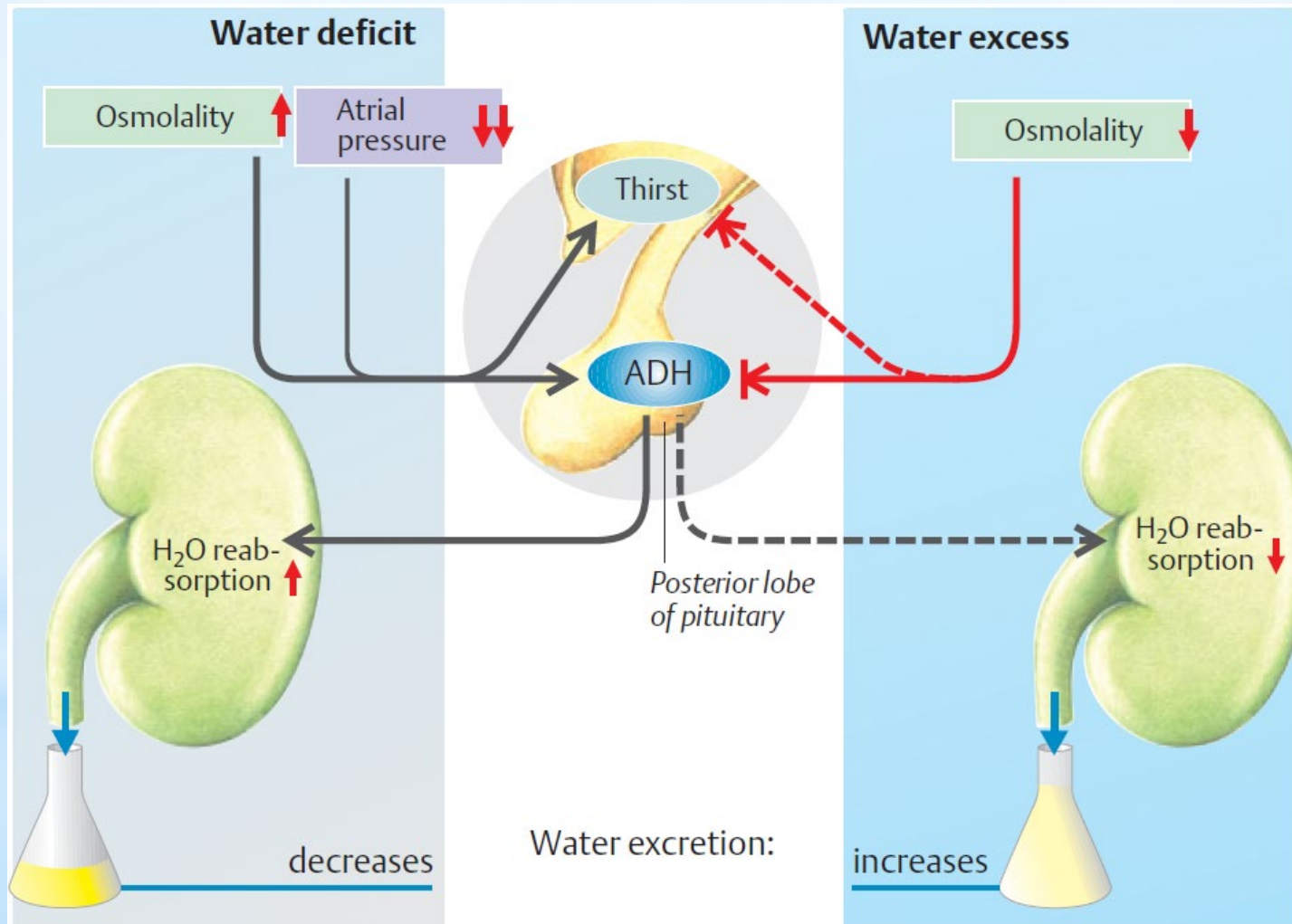
Atrial Natriuretic Peptide

- one of natriuretic peptides (BNP – cardiac ventricles, CNP – brain)
- **effects (through \uparrow cGMP): $\rightarrow \downarrow$ BP** (also through the brain stem)
 - \rightarrow natriuresis (1. \uparrow GFR – increased area for the filtration through relaxation of mesangial cells, 2. \uparrow Na^+ excretion – decrease tubular Na^+ reabsorption)
 - $\rightarrow \downarrow$ reactivity of vascular smooth muscles for vasoconstrictive substances
 - \rightarrow inhibition of renin secretion, \downarrow reactivity of *zona glomerulosa* for stimuli \uparrow aldosteron secretion
 - \rightarrow inhibition of ADH secretion $\rightarrow \uparrow$ water excretion
- **regulation of its secretion:**
 - \uparrow - \uparrow ECF volume (atrial cells' stretch at higher atrial filling)
 - \downarrow - \downarrow CVP at orthostasis

Humoral Regulation of Body Fluids

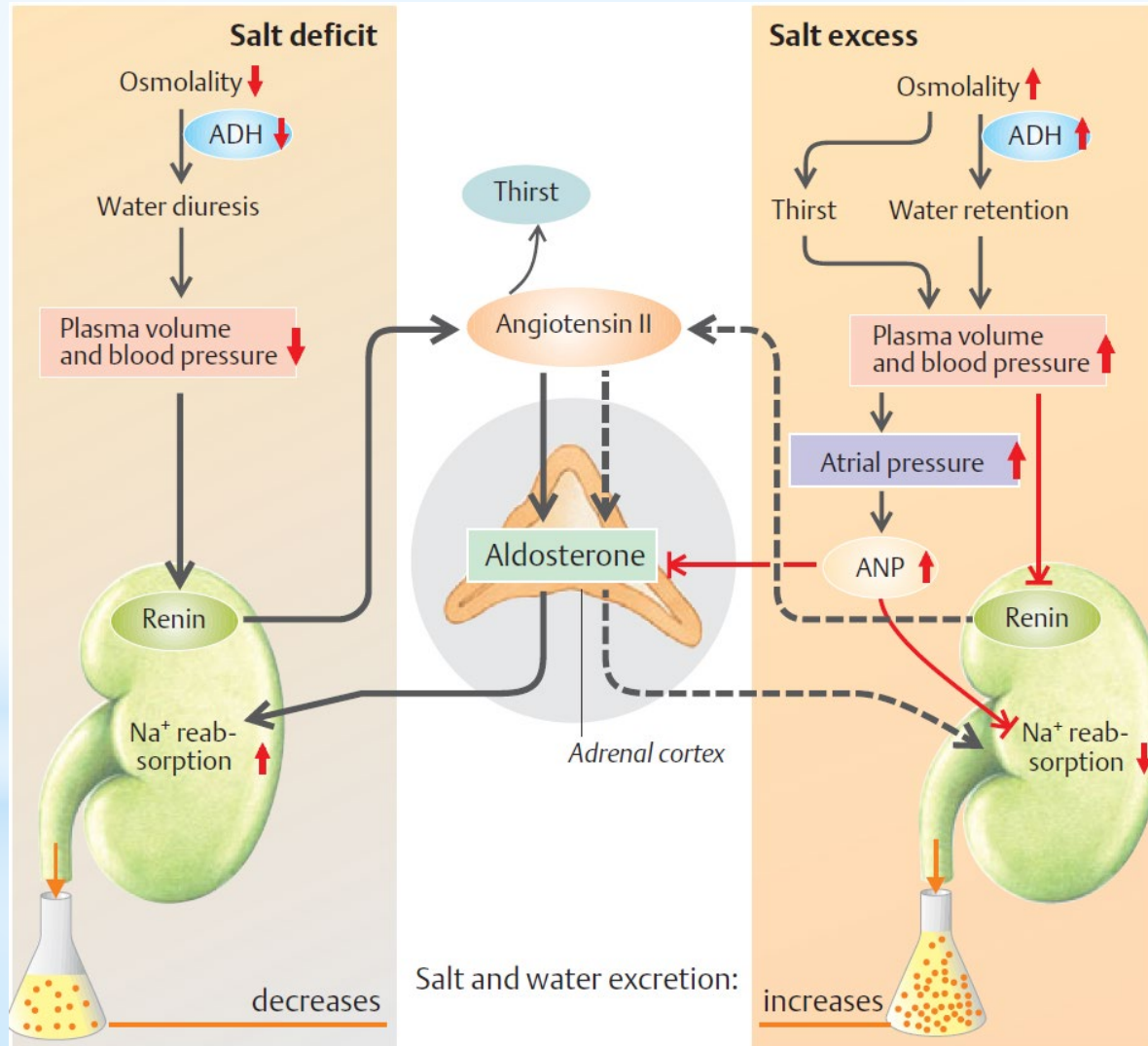
Water Homeostasis

water intoxication



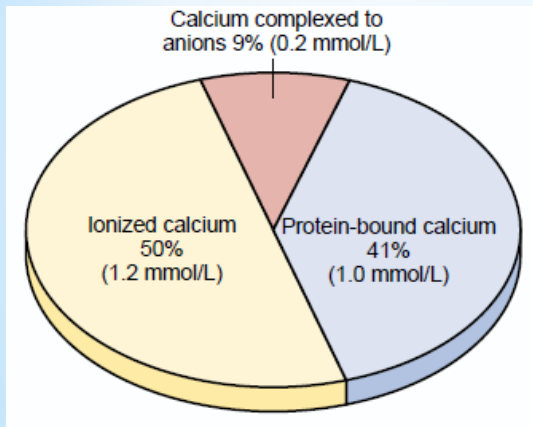
Humoral Regulation of Body Fluids

Salt Homeostasis

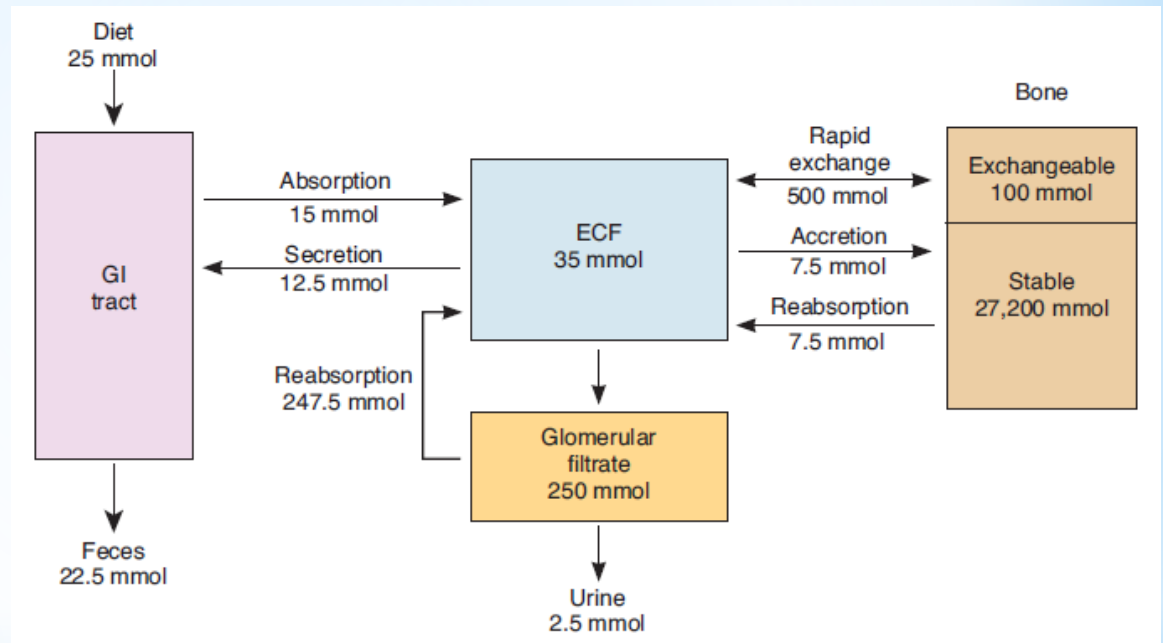


Humoral Regulation of Body Fluids

Calcium in the Body



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Ganong's Review of Medical Physiology

hypocalcemia

hypercalcemia

Humoral Regulation of Body Fluids

Hormonal Regulation of Calcemia

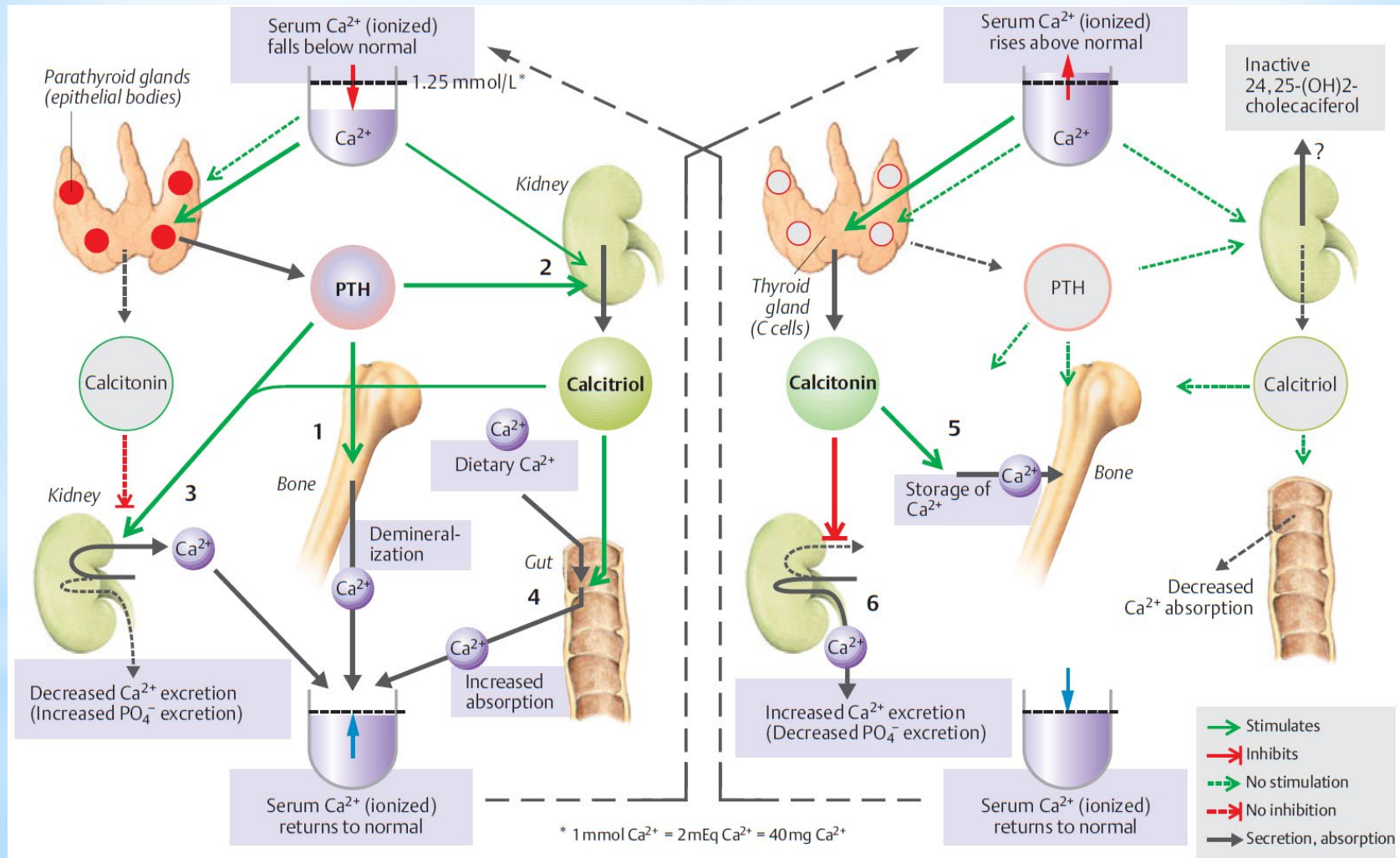
Parathormone

Vitamin D

Calcitonin

Humoral Regulation of Body Fluids

Hormonal Regulation of Calcemia



Acid-Base Balance

- Regulation by Kidneys -

Acid-Base Balance and its Regulation

- acid** substance releasing H^+ (e.g. $H_2CO_3 \rightarrow H^+ + HCO_3^-$)
- base** substance binding H^+ (e.g. $HCO_3^- + H^+ \rightarrow H_2CO_3$; proteins)
- $[H^+]$**
- influences activity of almost all enzymatic systems
 - very low compared to the concentration of other ions;
 $[H^+] = 40 \text{ nEq/l}$ but for example $[Na^+] = 142 \text{ mEq/l}$
 - thus, its changes has to be much smaller (3-5 nEq/l)
 \Rightarrow **precise regulation of $[H^+]$ is necessary!**

Since $[H^+]$ is a very small number, its negative logarithm is used:

$$pH = -\log [H^+] = -\log 0.000\ 000\ 040 = 7.4$$

The value of pH is thus inversely proportional to $[H^+]$.
Change of pH by 1 \sim change of $[H^+]$ 10-times bigger!

Acid-Base Balance and its Regulation

Physiological value of pH:

- arterial blood pH = 7.4
- venous blood pH = 7.35 (CO₂ from tissues)
- ICF pH = 6.0 - 7.4 (according to the cell type)
- urine pH = 4.5 - 8.0

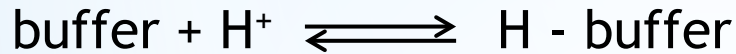
The value of pH 6.8 - 8.0 can be survived for several hours!

Acid-Base Balance and its Regulation

Acid-base balance is regulated by:

1) Buffers

- fast regulation (seconds)
- pH changes attenuated by binding and release of H⁺:



↑[H⁺] direction to the right favoured till free buffer is available

↓[H⁺] direction to the left favoured, H⁺ released

2) Lungs

- fast regulation (minutes even hours)
- elimination of CO₂ from the body ($\text{H}_2\text{CO}_3 \rightarrow \text{H}_2\text{O} + \text{CO}_2$)

3) Kidneys

- slower regulation (hours even days) but the most powerful
- elimination of acids and bases from the body

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Buffers

1) Bicarbonate buffer

- the most important buffer system
- weak acid H_2CO_3 and its salt NaHCO_3
$$\text{CO}_2 + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{CO}_3 \rightleftharpoons \text{H}^+ + \text{HCO}_3^- + \text{Na}^+$$
- the most powerful (despite not expected to be so powerful, $\text{pK} = 6.1$)

2) Phosphate buffer

- an important buffer system of the renal tubular fluid and of the intracellular fluid (high concentration + pH nearer to $\text{pK} = 6.8$)
- H_2PO_4^- , HPO_4^{2-}

3) Protein buffer

- an important buffer of an important buffer system of (conc. + pK)

60 - 70% of the buffer capacity of body fluids sites in the cells and is dependent on proteins!

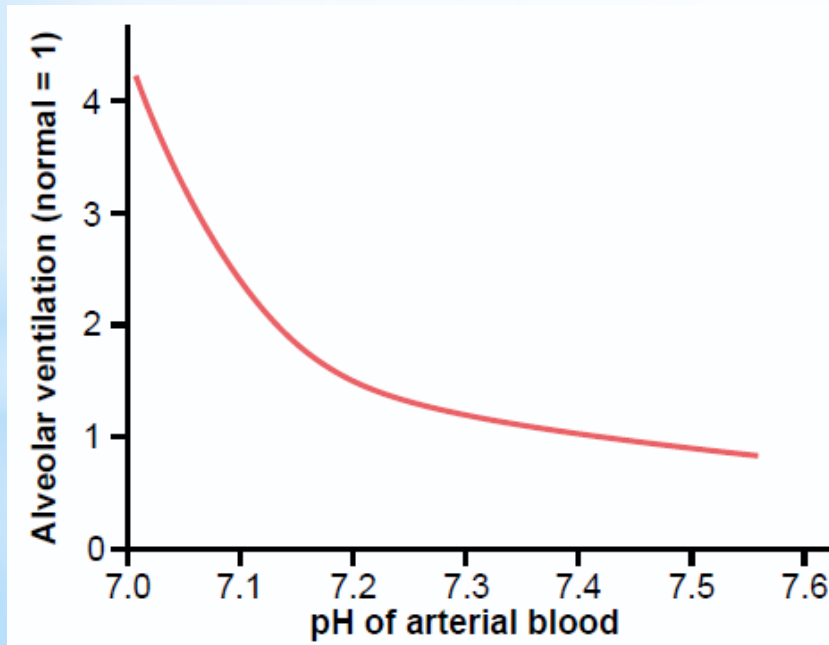


Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Lungs

- by the hyper- or hypoventilation

$$\text{pH} = 6.1 + \log \frac{\text{HCO}_3^-}{0.03 \times \text{P}_{\text{CO}_2}}$$



$\uparrow[\text{H}^+] \rightarrow \uparrow \text{Alveolar ventilation}$
 $\ominus \uparrow \downarrow \text{P}_{\text{CO}_2}$

Acid-Base Balance and its Regulation

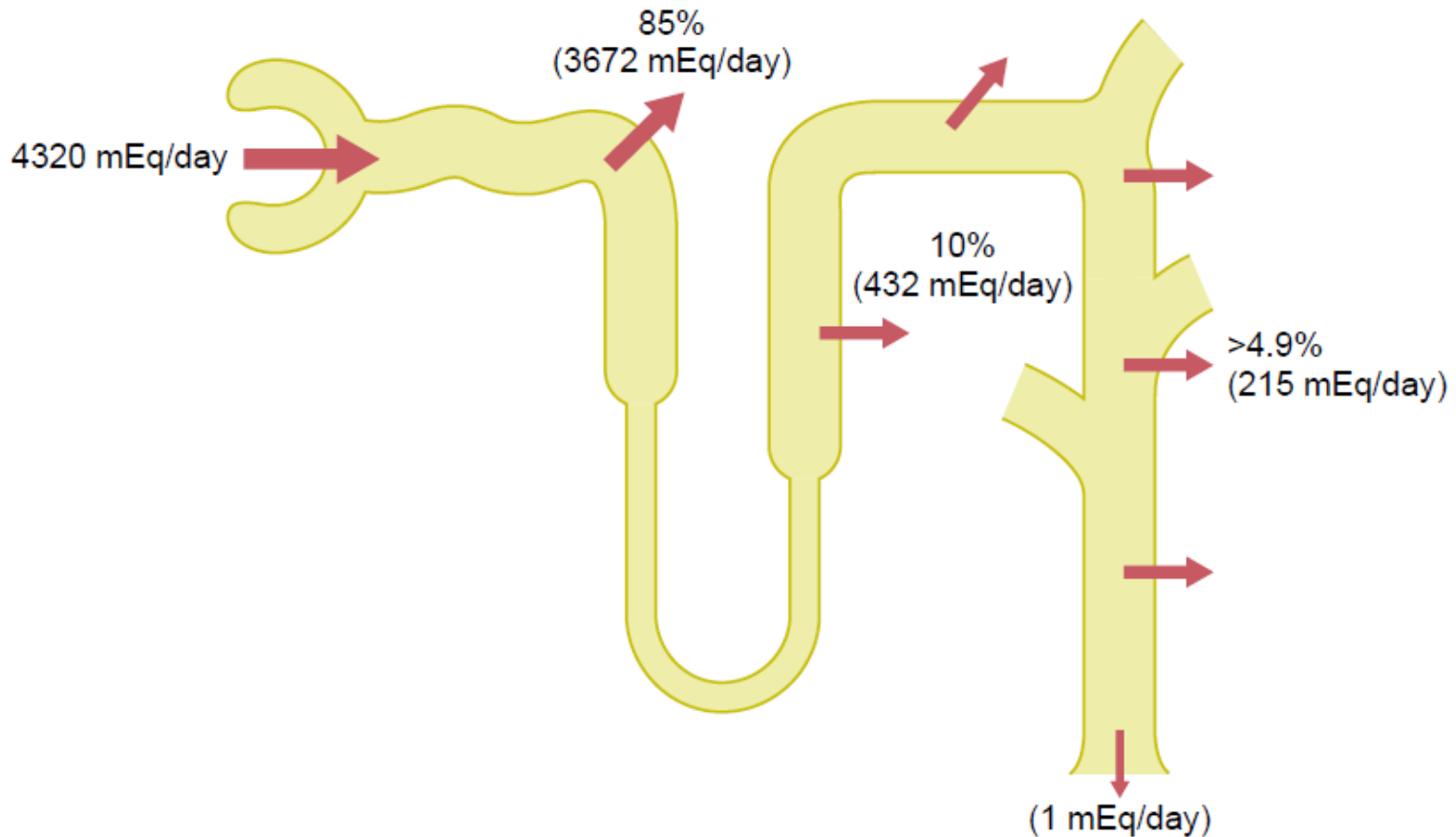
Regulation of Acid-Base Balance by Kidneys

- by excretion of acid or alkalic urine
- a high amount of HCO_3^- still filtered in the glomerulus
GFR 180 l/day, $[\text{HCO}_3^-]_{\text{plasma}} 24 \text{ mEq/l} \rightarrow 4320 \text{ mEq HCO}_3^-$
filtered per day - almost all ordinarily reabsorbed
- a high amount of H^+ still secreted in renal tubules
about 80 mEq of non-volatile acids are formed in the course of
metabolic processes per day - have to be excreted by kidneys
- filtered HCO_3^- / secreted H^+

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

- 1) Secretion of H^+
- 2) Reabsorption of HCO_3^-

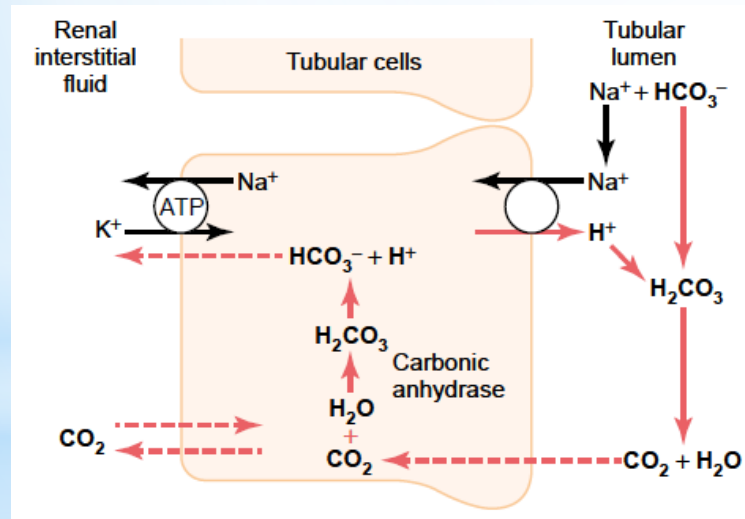


Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

- 1) Secretion of H^+
- 2) Reabsorption of HCO_3^-

❖ in the proximal tubule, thick loop of Henle and at the beginning of the distal tubule



Na^+ / H^+ -antiport

>90% HCO_3^- reabsorbed - only a slight acidification of the urine!

Reabsorption of HCO_3^- across the basolateral membrane facilitated by:

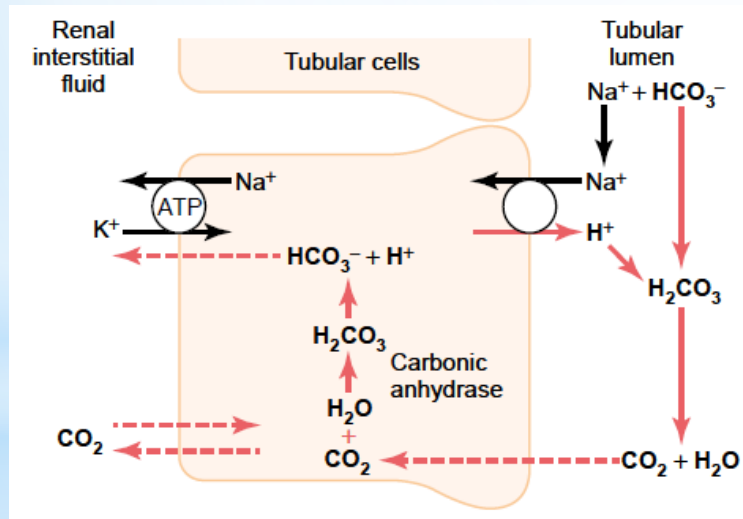
- Na^+ - HCO_3^- co-transport (the proximal tubule)
- Cl^- - HCO_3^- exchanger (the end of proximal tubule and the following parts of tubulus except for the thin loop of Henle)

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

- 1) Secretion of H^+
- 2) Reabsorption of HCO_3^-

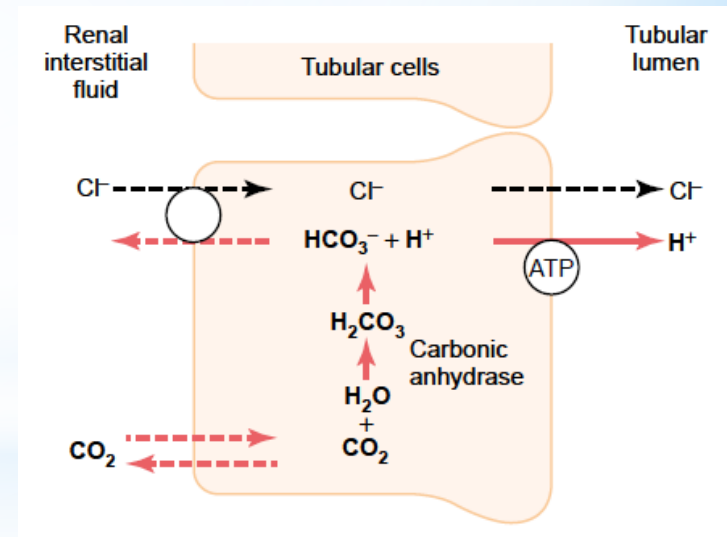
❖ in the proximal tubule, thick loop of Henle and at the beginning of the distal tubule



Na^+/H^+ -antiport

>90% HCO_3^- reabsorbed - only a slight acidification of the urine!

❖ in the final part of distal tubule and in the collecting duct



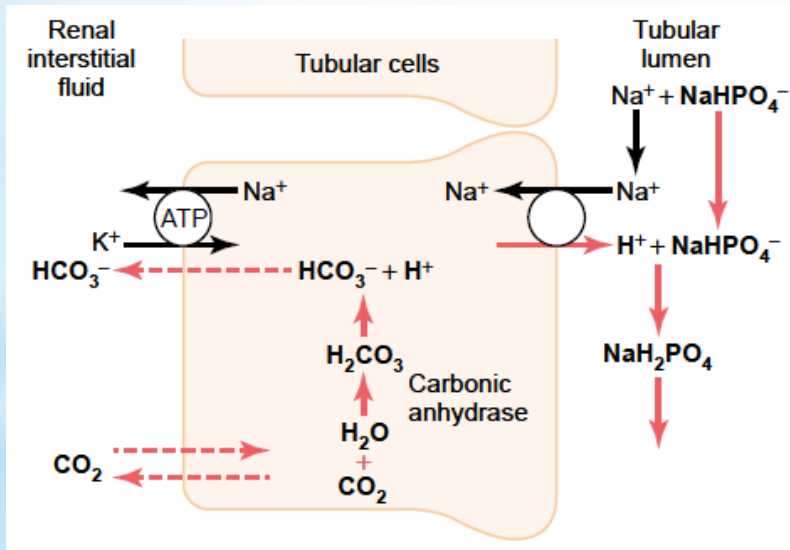
primary active transport of H^+
(intercalated cells)
acidification of urine

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

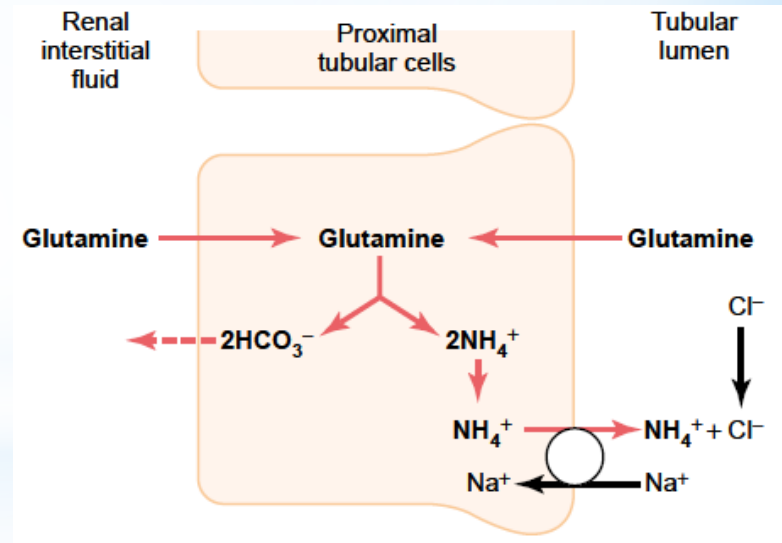
- 1) Secretion of H^+
- 2) Reabsorption of HCO_3^-
- 3) Production of HCO_3^- *de novo*

❖ Phosphate buffer (HPO_4^{2-} , $H_2PO_4^-$)



HPO_4^{2-} and $H_2PO_4^-$ are reabsorbed less than water \Rightarrow their concentration in the tubular fluid gradually rises

❖ Ammonium buffer (NH_3 , NH_4^+)



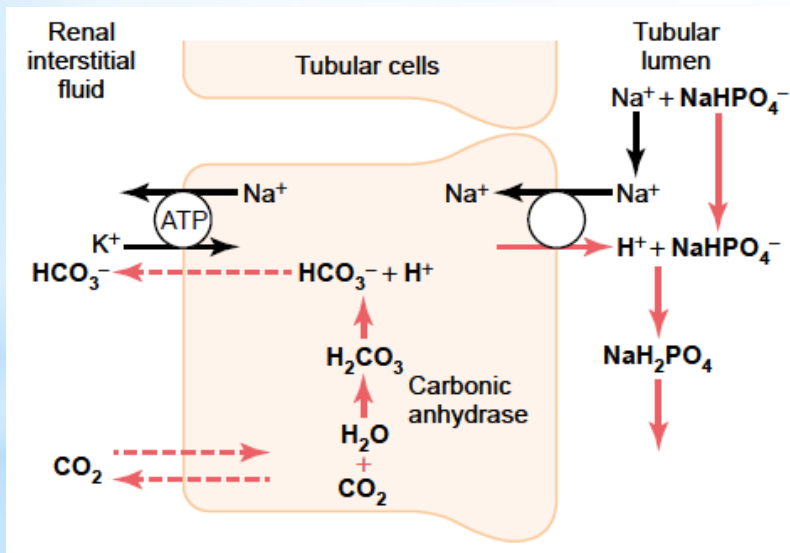
NH_4^+ originates from glutamine - the proximal tubule, thick ascending loop of Henle and distal tubule

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

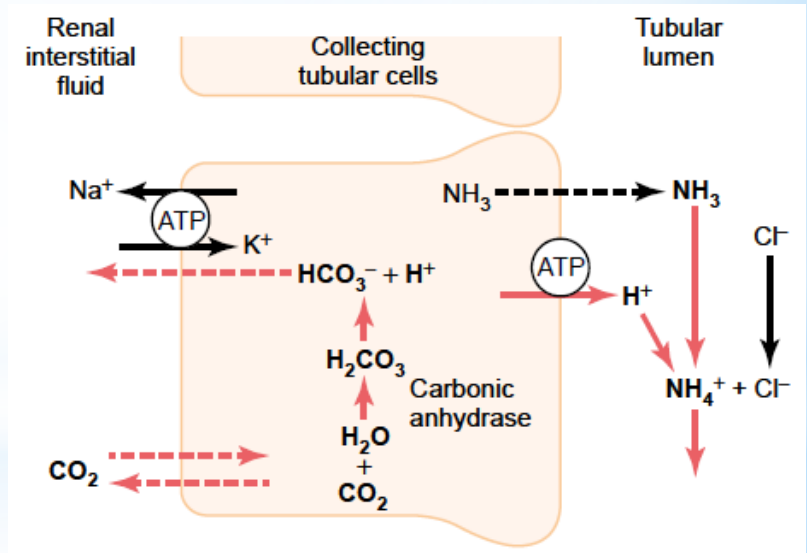
- 1) Secretion of H^+
- 2) Reabsorption of HCO_3^-
- 3) Produkce nového HCO_3^-

❖ Phosphate buffer (HPO_4^{2-} , $H_2PO_4^-$)



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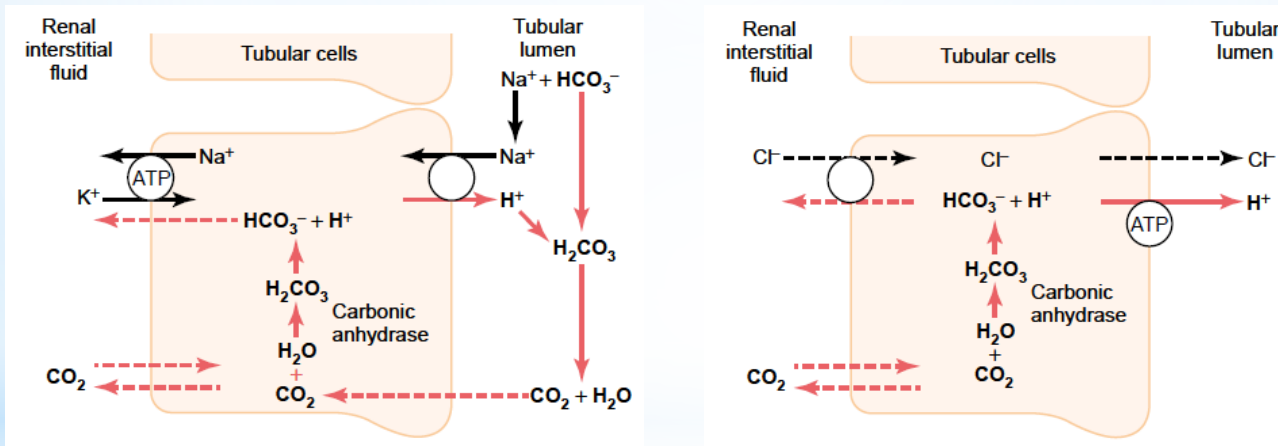
the collecting duct (permeable for NH_3 but far less for NH_4^+ - excreted by urine)
50% of H^+ secretion and HCO_3^- formed *de novo*!

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

Regulation of H⁺ secretion

- ↑ - ↑ pCO₂ in ECF (respiratory acidosis; direct stimulation due to ↑ formation of H⁺ in tubular cells)



- ↓ pH in ECF (respiratory or metabolic acidosis)
- ↑ secretion of aldosterone (stimulates active H⁺ secretion in intercalated cells of collecting ducts, also through Na⁺/H⁺ antiport; Conn's syndrome - alkalosis)

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

Regulation of H⁺ secretion

Factors That Increase or Decrease H⁺ Secretion and HCO₃⁻ Reabsorption by the Renal Tubules

Increase H⁺ Secretion and HCO₃⁻ Reabsorption

↑ PCO₂

↑ H⁺, ↓ HCO₃⁻

↓ Extracellular fluid volume

↑ Angiotensin II

↑ Aldosterone

Hypokalemia

Decrease H⁺ Secretion and HCO₃⁻ Reabsorption

↓ PCO₂

↓ H⁺, ↑ HCO₃⁻

↑ Extracellular fluid volume

↓ Angiotensin II

↓ Aldosterone

Hyperkalemia

RAS

↑ activity of Na⁺/H⁺ antiport

↑ activity of H⁺ ATPase



tendency to alkalosis

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

Acidosis - correction by kidneys

$$\downarrow \text{pH} = 6.1 + \log \frac{\text{HCO}_3^-}{0.03 \times P_{\text{CO}_2}} \downarrow$$

- **metabolic acidosis:** due to $\downarrow \text{HCO}_3^-$
renal correction: $\downarrow \text{HCO}_3^-$ in ECF \rightarrow \downarrow filtered HCO_3^- \rightarrow complete reabsorption of HCO_3^- + its formation *de novo* (HCO_3^- not excreted) + $\uparrow \text{H}^+$ excretion \rightarrow pH normalization
- **respiratory acidosis:** due to $\uparrow P_{\text{CO}_2}$ (hypoventilation)
renal correction: $\uparrow P_{\text{CO}_2}$ in ECF \rightarrow $\uparrow P_{\text{CO}_2}$ in tubular cells \rightarrow \uparrow formation of H^+ and HCO_3^- in tubular cells \rightarrow $\uparrow \text{H}^+$ secretion + $\uparrow \text{HCO}_3^-$ reabsorption \rightarrow pH normalization

Acid-Base Balance and its Regulation

Regulation of Acid-Base Balance by Kidneys

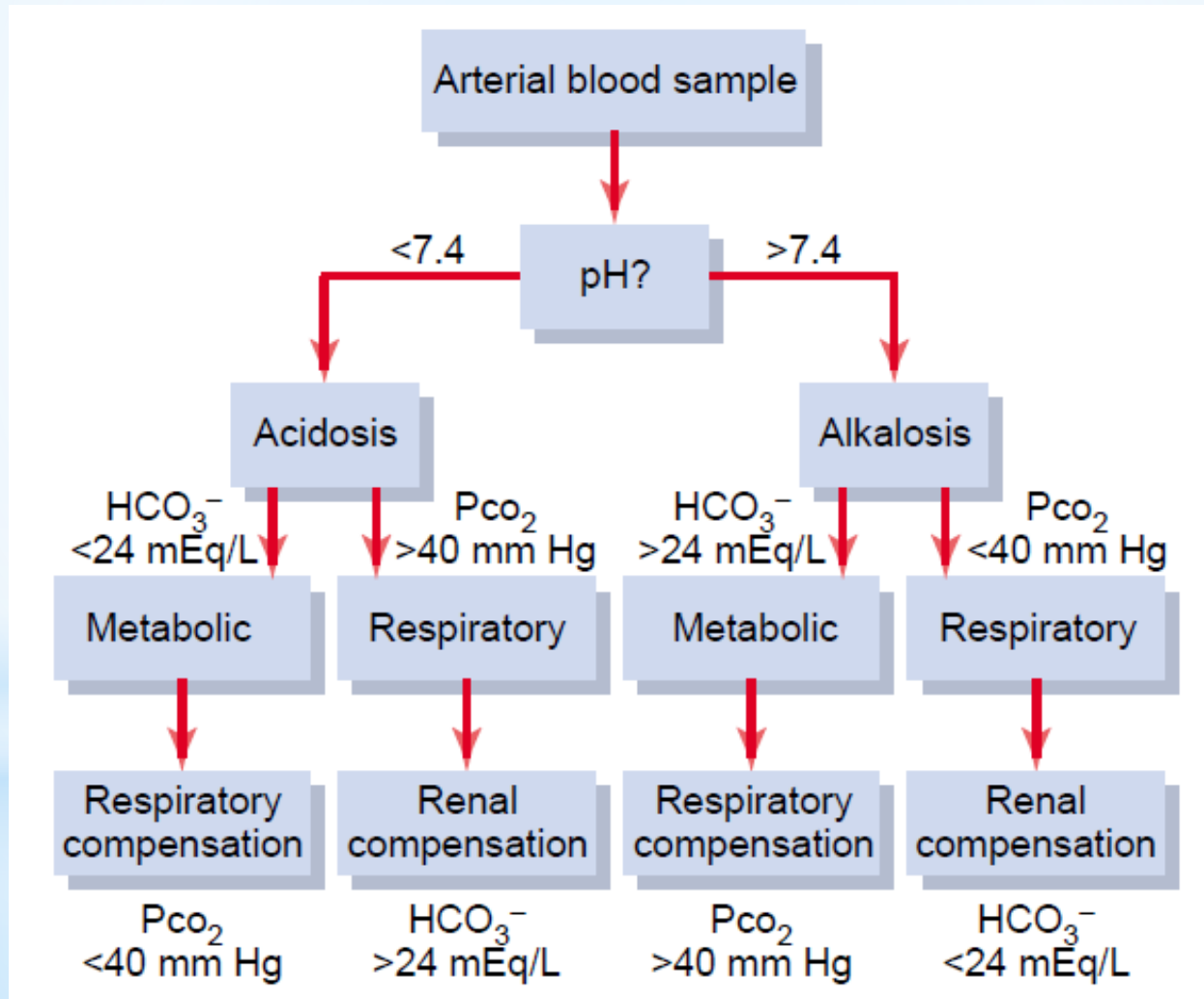
Alkalosis - correction by kidneys

$$\uparrow \text{pH} = 6.1 + \log \frac{\text{HCO}_3^-}{0.03 \times P_{\text{CO}_2}} \uparrow$$

- **metabolic alkalosis:** due to $\uparrow \text{HCO}_3^-$
renal correction: $\uparrow \text{HCO}_3^-$ in ECF \rightarrow \uparrow filtered HCO_3^- \rightarrow incomplete HCO_3^- reabsorption (lack of H^+) \rightarrow $\uparrow \text{HCO}_3^-$ excretion by urine \rightarrow pH normalization
- **respiratory alkalosis :** due to $\downarrow P_{\text{CO}_2}$ (hyperventilation)
renal correction: $\downarrow P_{\text{CO}_2}$ in ECF \rightarrow $\downarrow P_{\text{CO}_2}$ in tubular cells \rightarrow \downarrow formation of H^+ and HCO_3^- in tubular cells \rightarrow $\downarrow \text{H}^+$ secretion + $\downarrow \text{HCO}_3^-$ reabsorption \rightarrow pH normalization

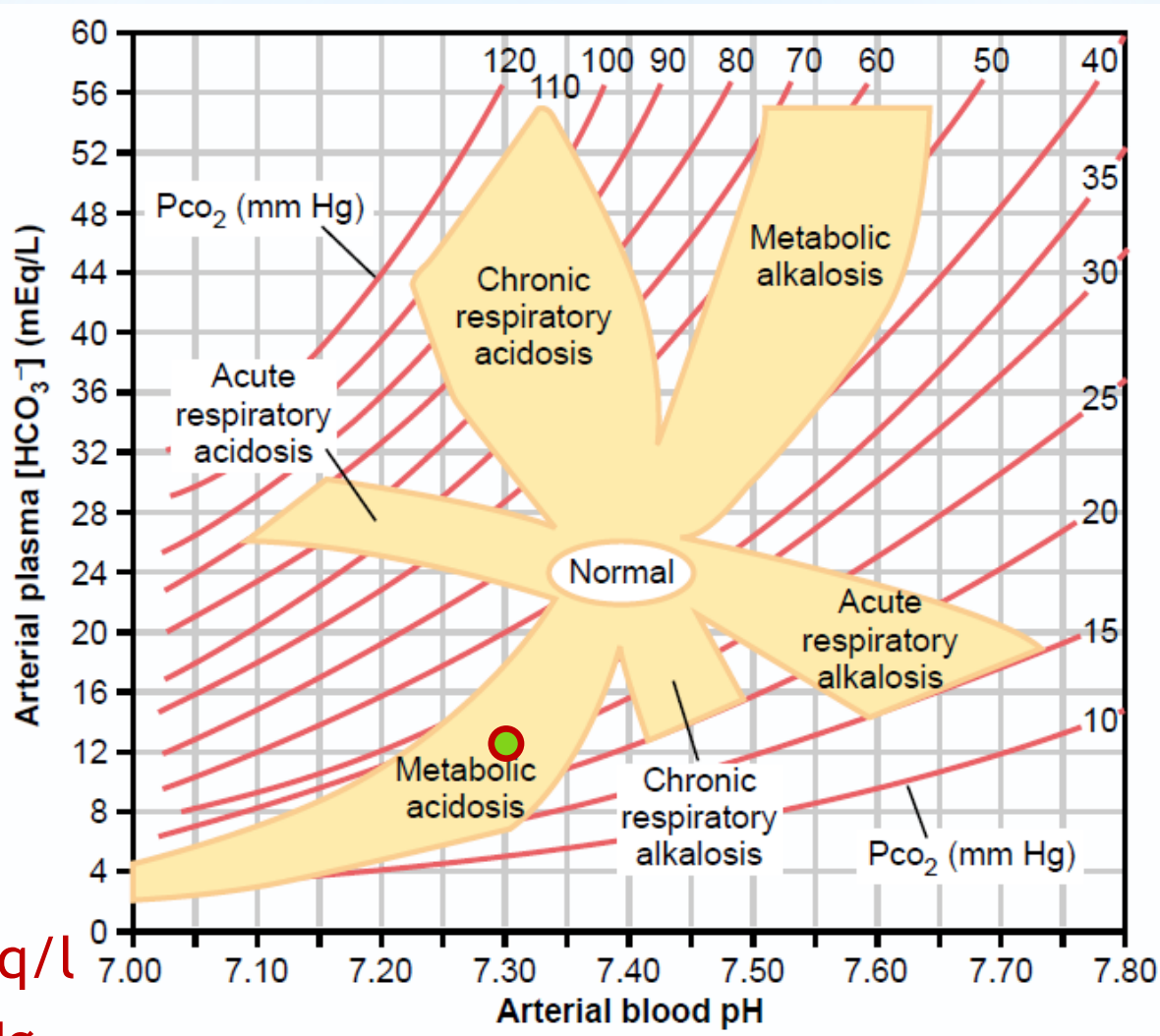
Acid-Base Balance and its Regulation

Diagnostics



Acid-Base Balance and its Regulation

Diagnosics



pH: 7,3

HCO_3^- : 12 mEq/l

P_{CO_2} : 25 mmHg