Draw structural formulas.
Acetoacetic acid
Lactone derived from 4-hydroxypentanoic acid
Acetylcholine
Ethanolamine
Butyl sulfate
Glyceric acid

2 Write reaction using structural formulas and give the names of products.
Dehydrogenation and deamination of glutamate:

Names of products $\qquad$
Conversion of fumarate in citrate cycle:

Names of products


3 Draw the structure of valine prevailing at pH 5
4 Draw the structure of adenosine.
$\left(\mathrm{p} K_{\mathrm{A} 1}=2.3 ; \mathrm{p} K_{\mathrm{A} 2}=9.7\right)$.

8 Consider the reaction $\mathrm{A} \rightarrow \mathrm{P}$. Draw the dependence of reaction velocity $v$ on time $t$ for this reaction.
v

9 A strong acid was added to hydrogen carbonate buffer. Express the buffer response by chemical equation.

| Exam evaluation: | 50-57 excellent (A) |
| :---: | :---: |
|  | 47-49 outstanding(B) |
| 1. Part: ................ / 40 | 42-46 good (C) |
| 2. Part: ............... / 17 | 38-41 satisfactory (D) |
| Total: .............../ 57 | 34-37 acceptable (E) |
| Signatures: | 0-33 unacceptable (F) |

