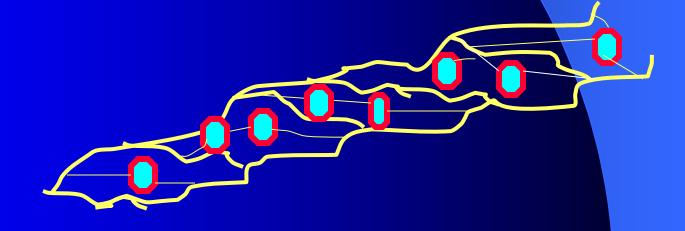
Composites

Chemically bonded mixture of organic matrix and inorganic filler



Organic matrix

Bowen monomer Bis GMA

UDMA

Oligomer – more flowable (thinning)

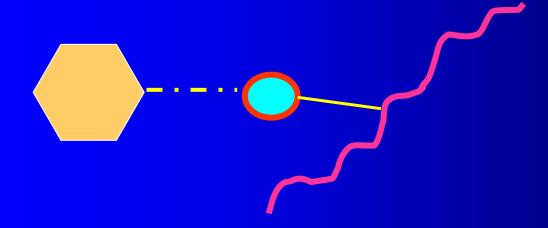
TEGMA

Filler
Milled quartz
Aluminium silicate glass
Silica
Prepolymer
Nanoparticles

Coupling agent

Silan

Binding between organic and inorganic part Homogenous distribution of *filler* in composite



Iniciators and accelerators

Other components

- Pigments
- UV absorbers

Classification acc.to size of filler particles

Macrofiller composites (size micrometers)

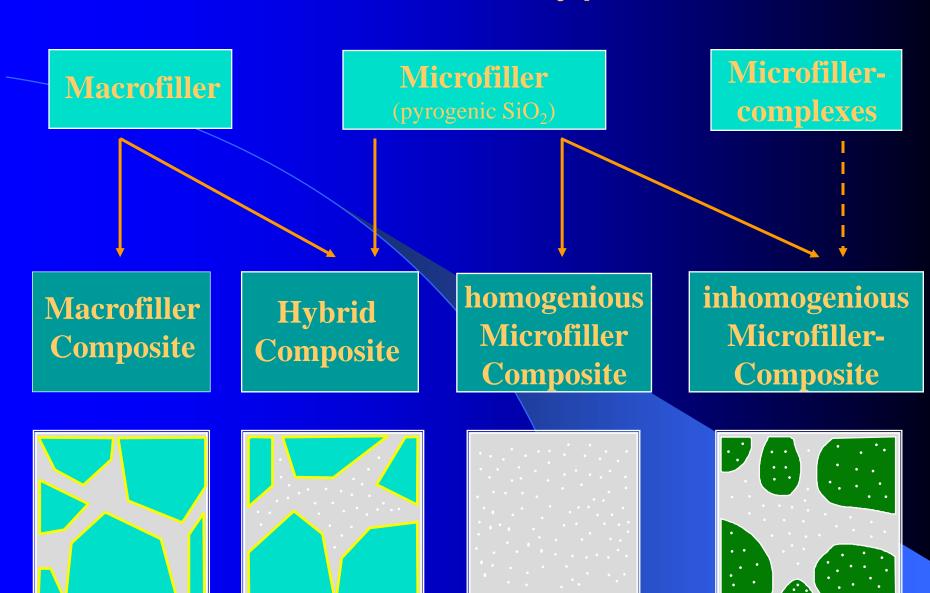
Microfiller composites (size 0,02 -0,04 micrometers)

Homogenous

Non homogenous

Hybride composites (combination of filler)

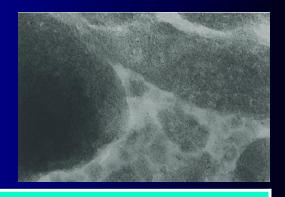
Classification acc. type of filler



Classification according to filler particles

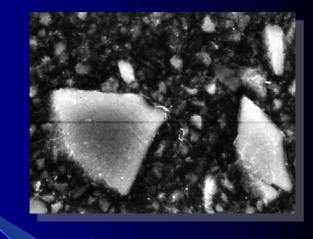
- Macrofiller (macrofilled composites)
- Microfiller (microfiled composites)
- -homogenous
- -inhomogenous
- Nanofiller (nanocomposites)
- Hybride (hybrid composits)

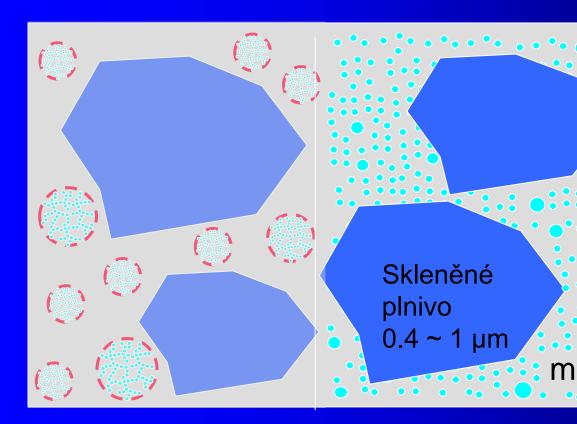
Filler particles

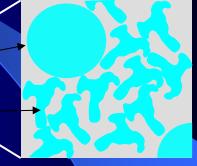


Filler size (medium particle size)	Composite category
< 10 μm	hybrid composite
< 5 μm	fine particle hybrid
< 3 μm	ultrafine particle hybrid
< 1 μm	submicron hybrid, nano composites

Filler - example







More filler

→
Less amount of resin

→
Lower polymerization shrinkage

11

Polymerization reaction

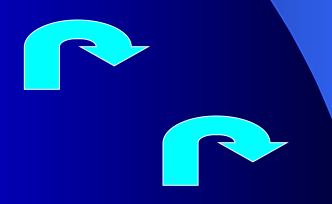
Accelerator



Iniciator

Double bonds – split

Polymer network



Curing

- Light curing composites
- (Light activated).

Light activation is accomplished with blue light (470 nm)

Initiator is camphorquinon

Chemically cured composites

Iniciator is organic peroxide, accelerator tertiary amine

Classification acc to mode of curing

Light curing

Self curing (chemically curing)

Dual curing (cementzs)

Heat curing (for dental lab)

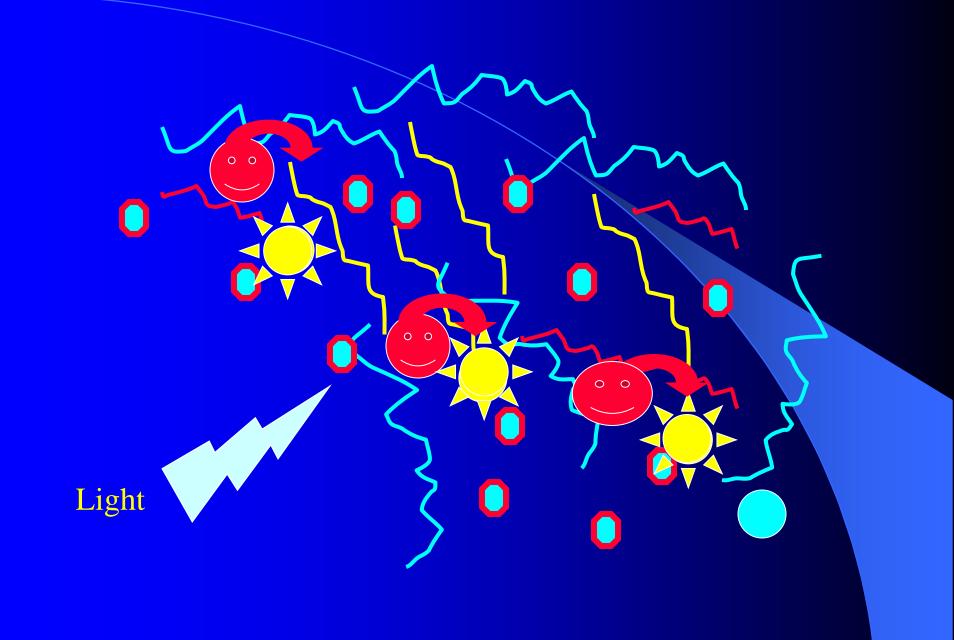
Polymerization units

Quartz halogen units

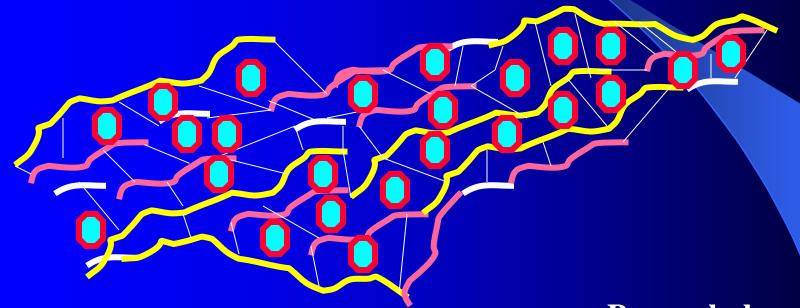
LED units

Blue light, 400 - 500 nm.

Camphorchinon 470 nm (maximum absorption)



Polymer network



Polymerization shrinkage

Pre –gel phase Gel-point Post –gel phase

Indications

Frontal area:

Class III., IV, V.

Posterior area

Small – moderate cavities I. and II. class

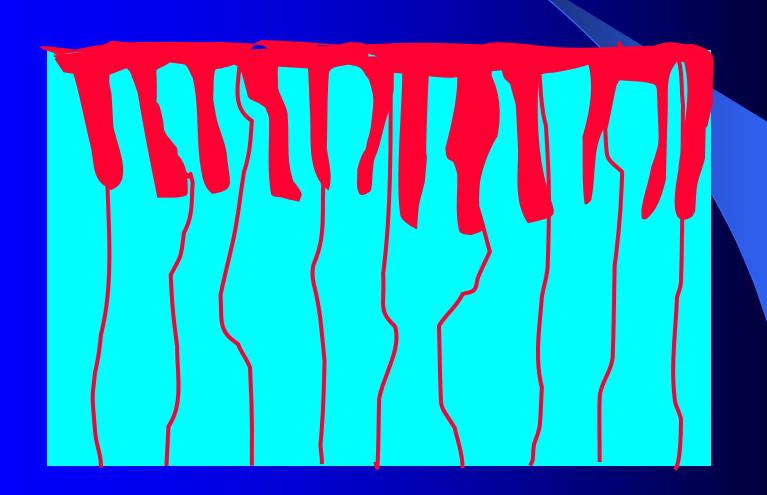
Contraindications

Bad level of oral hygiene

Dry operation field is impossible

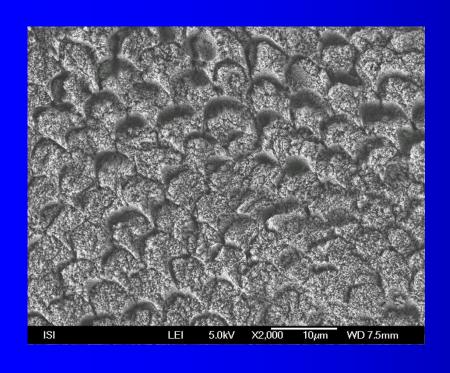
Large cavities I. and II. class

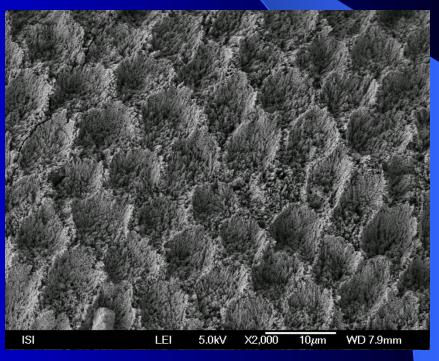
Principle of retention - micromechanical



Retention in enamel

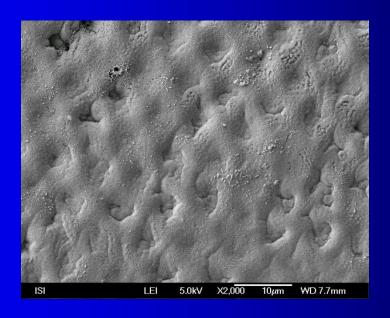
Acid etching of enamel
Irregular surface – retentive pattern
Speces between enamel rods or in enamel rods are created
The material flows into these spaces





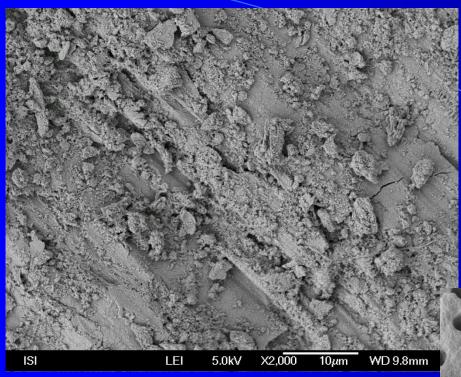
Aprismatic enamel

• On the surface of enamel there is aprismatic enamel. It is difficult to be etched in order to achieve the retentive pattern. It must be removed with diamond (red coded, fine)



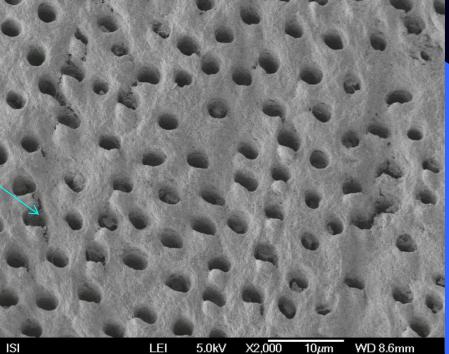
Retention in dentin

- o More water and organic components
- o Tubular liquor
- o Smear layer (layer after preparation composed of collagen fibers, crystals of hydroxyapatite that have been destroyed during preparation and microbs)



Smear layer

With acid etching smear layer is removed, dentin tubuls are opened and the surface of collagen network decalcified



The material can flow into dentin tubules and penetrates into spaces in collagen network. This is also micromechanical retention

WD 9.0mm



Adhesive system

 We need adhesive systém for penetration into microscopic spaces after acid etching.

It consist of primer and bond

Bond

• Flows into spaces in enamel after etching, into dentin tubules and to spaces in collagen network. Dentin must be pretreated with primer.

Primer

Open the collagen network in dentin and keep it open: Withiout it the bond wolud not be able to penetrate inside.

Making filling

- Preparation
- Enamel is beveled in most cases retentive border
- Acid etching
- Washing
- Bonding
- Placement of filling material in portions
- Curing with light
- Finishing and polishing (extra and ultrafine dimonds and rubber instruments)









Glassionomers

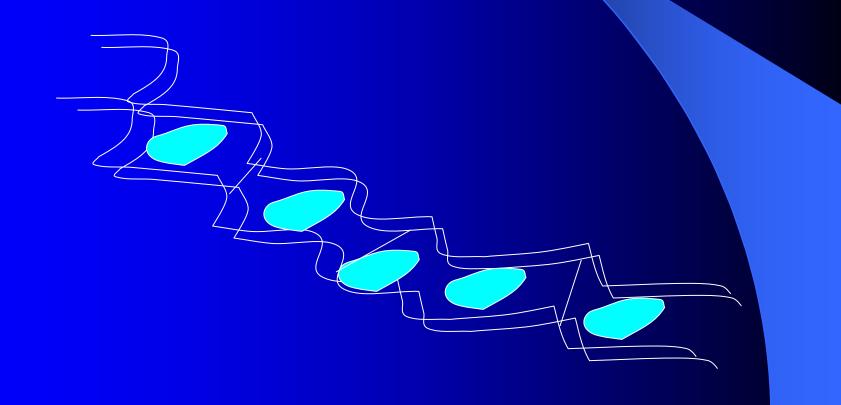
Composition

Powder: Aluminiumsilicate glass(SiO₂, Al₂O₅, CaO, N₂O,P₂O₅, F)

Liquid: Polyacid
(polyacrylic, polymaleic)
Tartaric acid,
Water

Glassionomers

Principle of setting – acid base reaction



Glassionomers

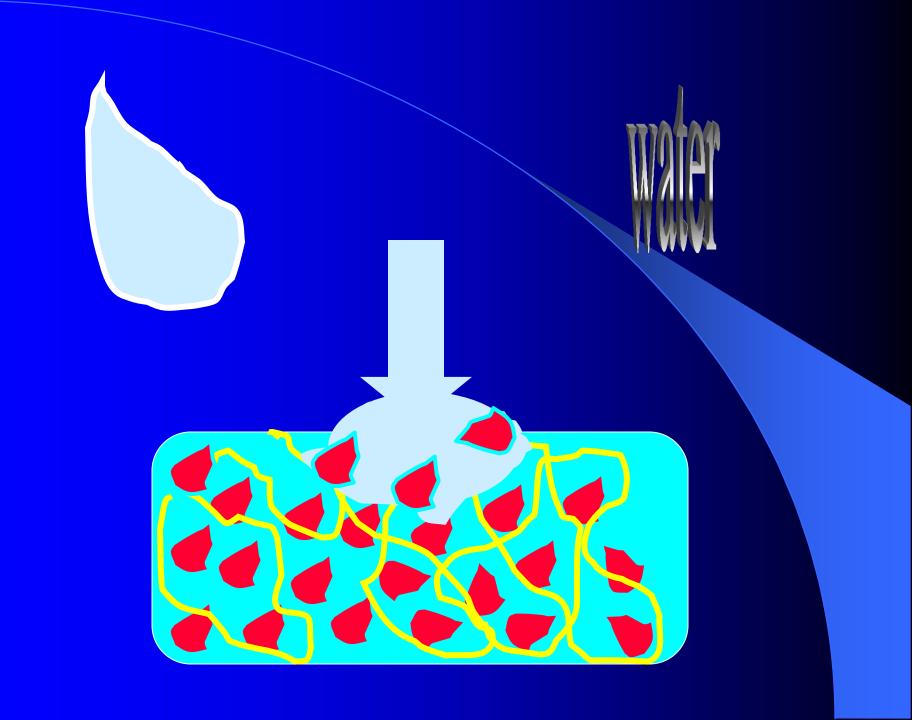
- Chemical bonding to hard dental tissues
- Thermal expansion similar to dentin
- Realease fluoride ions

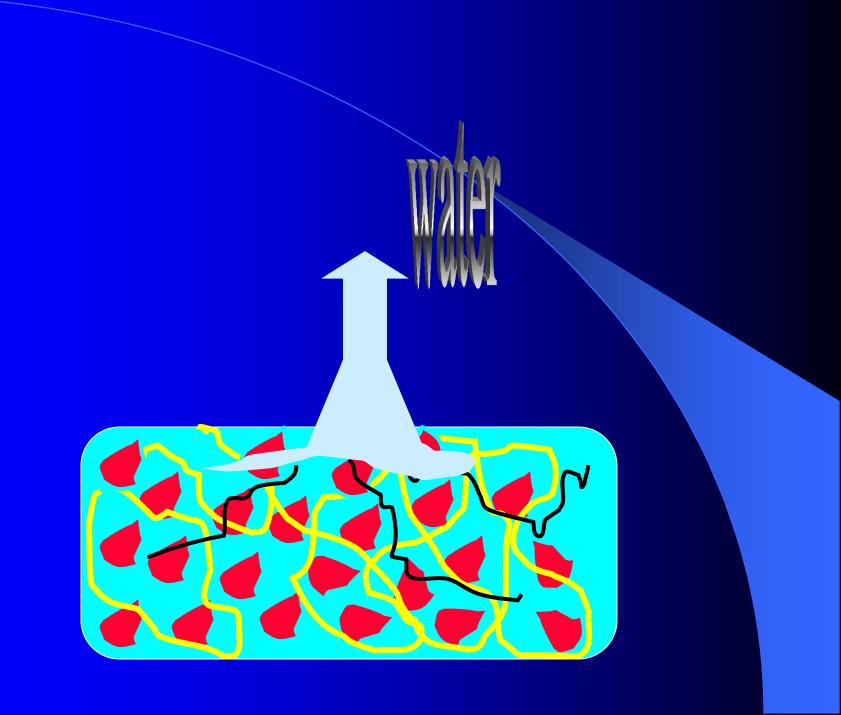
Mechanical not strong enough Aesthetics acceptable

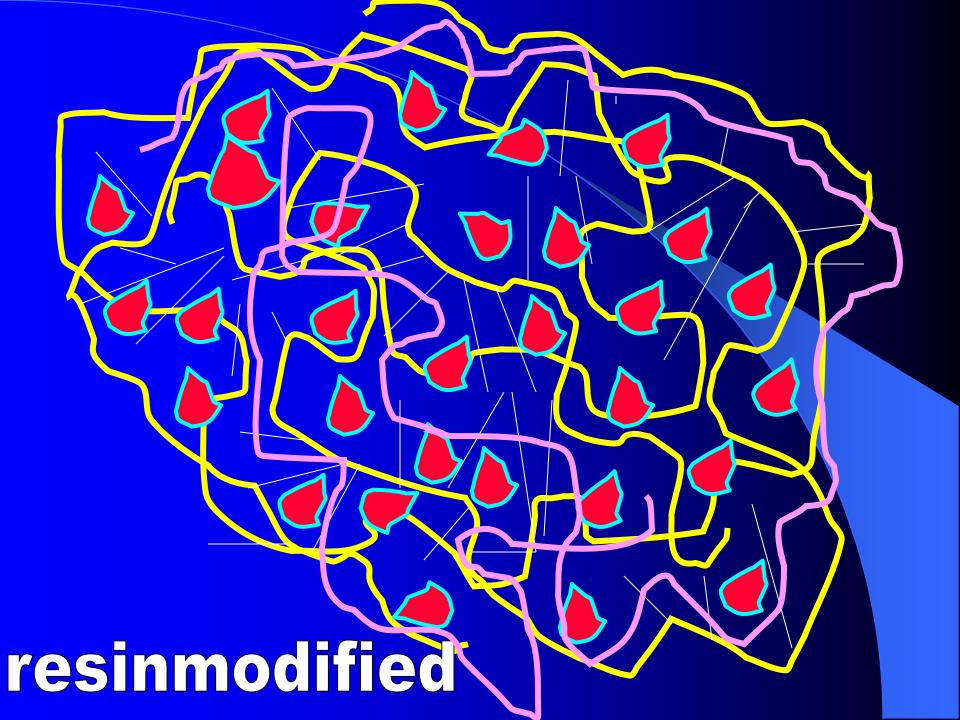
Glassionomers acc to curing

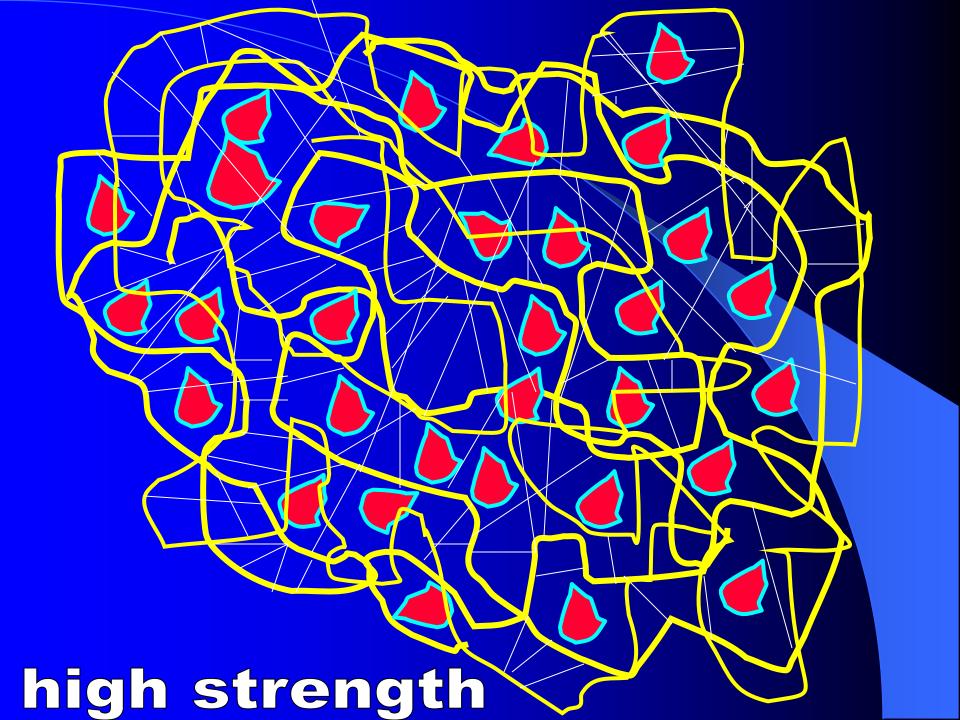
- Acid base reaction
- Dual cured glassionomers (resin modified)











Glassionomers - indications

Fillings

Class V., III., I., II

Sealants

Protection of tooth surface

Mixing

Hand

Power driven - capsulated



Making filling

- Preparation
- Smoth bordes
- Limited on caries lesion only
- Conditioning
- Washing
- Filling in one block
- Varnish after setting
- Polishing in next appointment

