#### Composites

#### Chemically bonded mixture of organic matrix and inorganic filler

• Organic matrix

Bowen monomer Bis GMA UDMA Oligomer – more flowable (thinning) TEGMA



Filler Milled quartz Aluminium silicate glass Silica Prepolymer Nanoparticles

#### Coupling agent

Silan

**Binding between organic and inorganic part Homogenous distribution of** *filler* in composite



**Iniciators and accelerators** 

Other components

- Pigments
- UV absorbers

# Classification acc.to size of filler particles

Macrofiller composites (size micrometers)

Microfiller composites (size 0,02 -0,04 micrometers)

Homogenous

Non homogenous

Hybride composites (combination of filler)

#### **Classification acc. type of filler**



# Classification according to filler particles

- Macrofiller (macrofilled composites)
- Microfiller (microfiled composites)
- -homogenous
- -inhomogenous
- Nanofiller (nanocomposites)
- Hybride (hybrid composits)

# **Filler particles**



Filler size (medium particle size)	Composite category
<10 μm	hybrid composite
< 5 μm	fine particle hybrid
< 3 μm	ultrafine particle hybrid
< 1 µm	submicron hybrid, nano composites

#### Filler - example





More filler → Less amount of resin → Lower polymerization shrinkage

#### 11 Polymerization reaction

Accelerator

Iniciator

Double bonds – split

Polymer network





• Light curing composites (Light activated). Light activation is accomplished with blue light (470 nm)Initiator is camphorquinon Chemically cured composites Iniciator is organic peroxide, accelerator tertiary amine

# Classification acc to mode of curing

• Light curing

• Self curing (chemically curing)

• Dual curing (cementzs)

• Heat curing (for dental lab)

#### **Polymerization units**

Quartz halogen units

• LED units

Blue light, 400 – 500 nm. Camphorchinon 470 nm (maximum absorption)



# Polymer network

**Polymerization shrinkage** 

Pre –gel phase Gel-point Post –gel phase

#### Indications

Frontal area:Class III., IV, V.

Posterior area Small – moderate cavities I. and II. class

#### **Contraindications**

Bad level of oral hygiene

• Dry operation field is impossible

• Large cavities I. and II. class

#### Principle of retention - micromechanical



#### **Retention in enamel**

Acid etching of enamel Irregular surface – retentive pattern Speces between enamel rods or in enamel rods are created The material flows into these spaces





#### **Aprismatic enamel**

• On the surface of enamel there is aprismatic enamel. It is difficult to be etched in order to achieve the retentive pattern. It must be removed with diamond (red coded, fine)



#### **Retention in dentin**

- o More water and organic components
- o Tubular liquor
- Smear layer (layer after preparation composed of collagen fibers, crystals of hydroxyapatite that have been destroyed during preparation and microbs)



With acid etching smear layer is removed, dentin tubuls are opened and the surface of collagen network decalcified

#### Smear layer



The material can flow into dentin tubules and penetrates into spaces in collagen network. This is also micromechanical retention

S



35% - 37% phosphoric acid silica particles blue dye

Whra-Etch

#### Adhesive system

• We need adhesive systém for penetration into microscopic spaces after acid etching.

It consist of primer and bond

### Bond

 Flows into spaces in enamel after etching, into dentin tubules and to spaces in collagen network. Dentin must be pretreated with primer.



Open the collagen network in dentin and keep it open: Withiout it the bond wolud not be able to penetrate inside.

# Making filling

- Preparation
- Enamel is beveled in most cases retentive border
- Acid etching
- Washing
- Bonding
- Placement of filling material in portions
- Curing with light
- Finishing and polishing (extra and ultrafine dimonds and rubber instruments)









#### Glassionomers

#### <u>Composition</u> <u>Powder:</u> Aluminiumsilicate glass(SiO<sub>2</sub>, Al<sub>2</sub>O<sub>5</sub>, CaO, N<sub>2</sub>O,P<sub>2</sub>O<sub>5</sub>, F)

Liquid: Polyacid (polyacrylic, polymaleic) Tartaric acid, Water

#### Glassionomers

#### Principle of setting – acid base reaction



#### Glassionomers

Chemical bonding to hard dental tissues
Thermal expansion similar to dentin
Realease fluoride ions

Mechanical not strong enough Aesthetics acceptable

## Glassionomers acc to curing

• <u>Acid – base reaction</u>

• Dual cured glassionomers (resin modified)







# resinmodified

# high strength

## **Glassionomers - indications**

• Fillings

Class V., III., I., II

Sealants

Protection of tooth surface



#### Hand

#### Power driven - capsulated



## Making filling

- Preparation
- Smoth bordes
- Limited on caries lesion only
- Conditioning
- Washing
- Filling in one block
- Varnish after setting
- Polishing in next appointment

