

# HPLC detectors

# HPLC detectors

- Corresponding sensitivity
- Stability and reproducibility
- Wide dynamic linear range
- Fast response
- Minimal volume of detection cell
  
- detectors can be universal or selective
  - A universal detector depends on the characteristic of a system as a complex (refractive index)
  - A selective detector depends on the certain characteristic of the sample (fluorescent)
- detectors can be destructive (AAS, AES) or non-destructive (UV/Vis)
- detectors can with combined benefits (multiple detection)
- detectors can work in several domains of one characteristic (diode array detector)
- some detectors need a special interface to the LC system

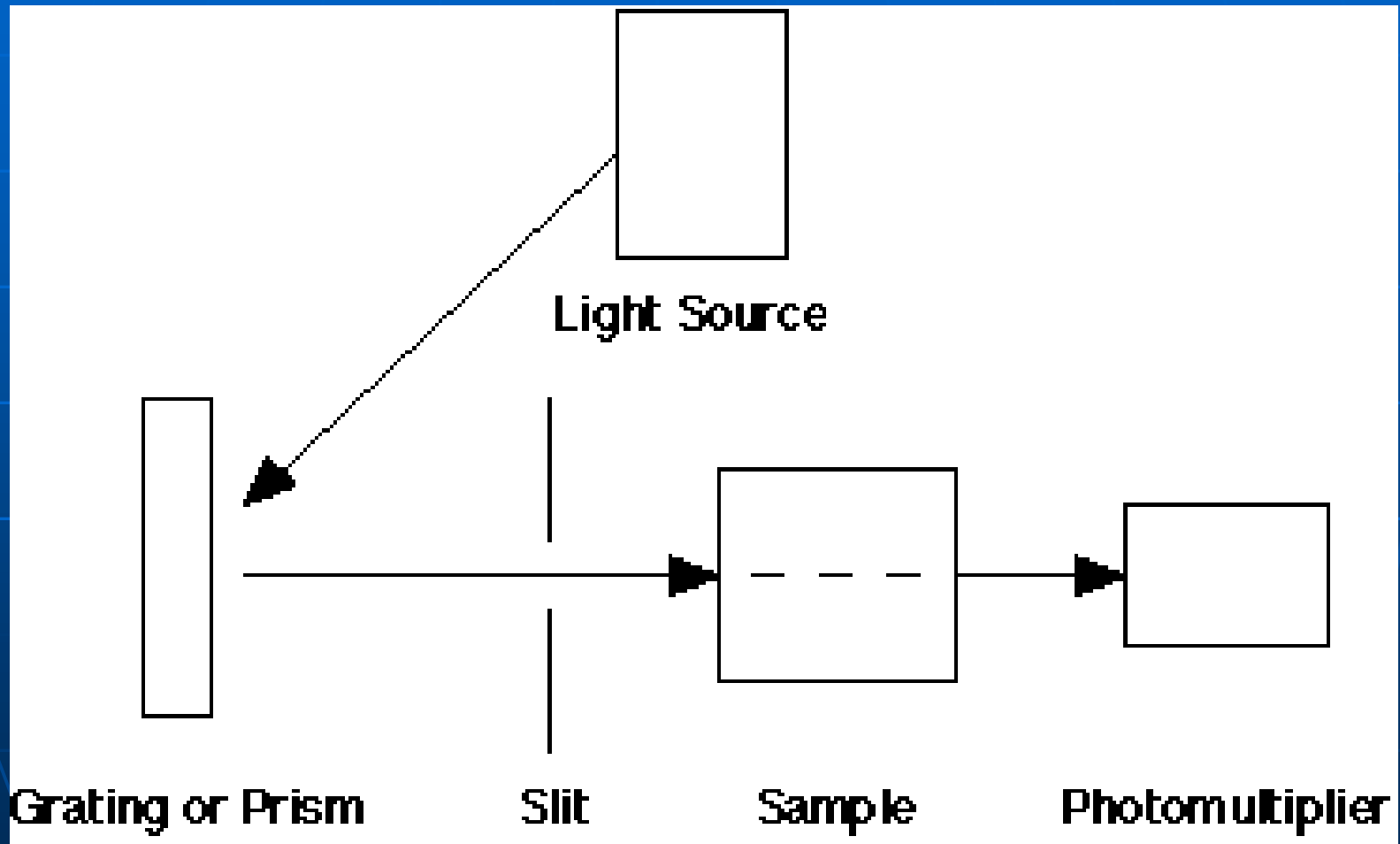
- UV/Vis detection
- Fluorescent detectors
- Radiochemical detectors
- Electrochemical detectors
- Nuclear Magnetic Resonance (NMR) detectors
- Light-Scattering (LS) detectors
- Refractometric detectors
- Polarometric detectors
- IR detectors

# HPLC detectors

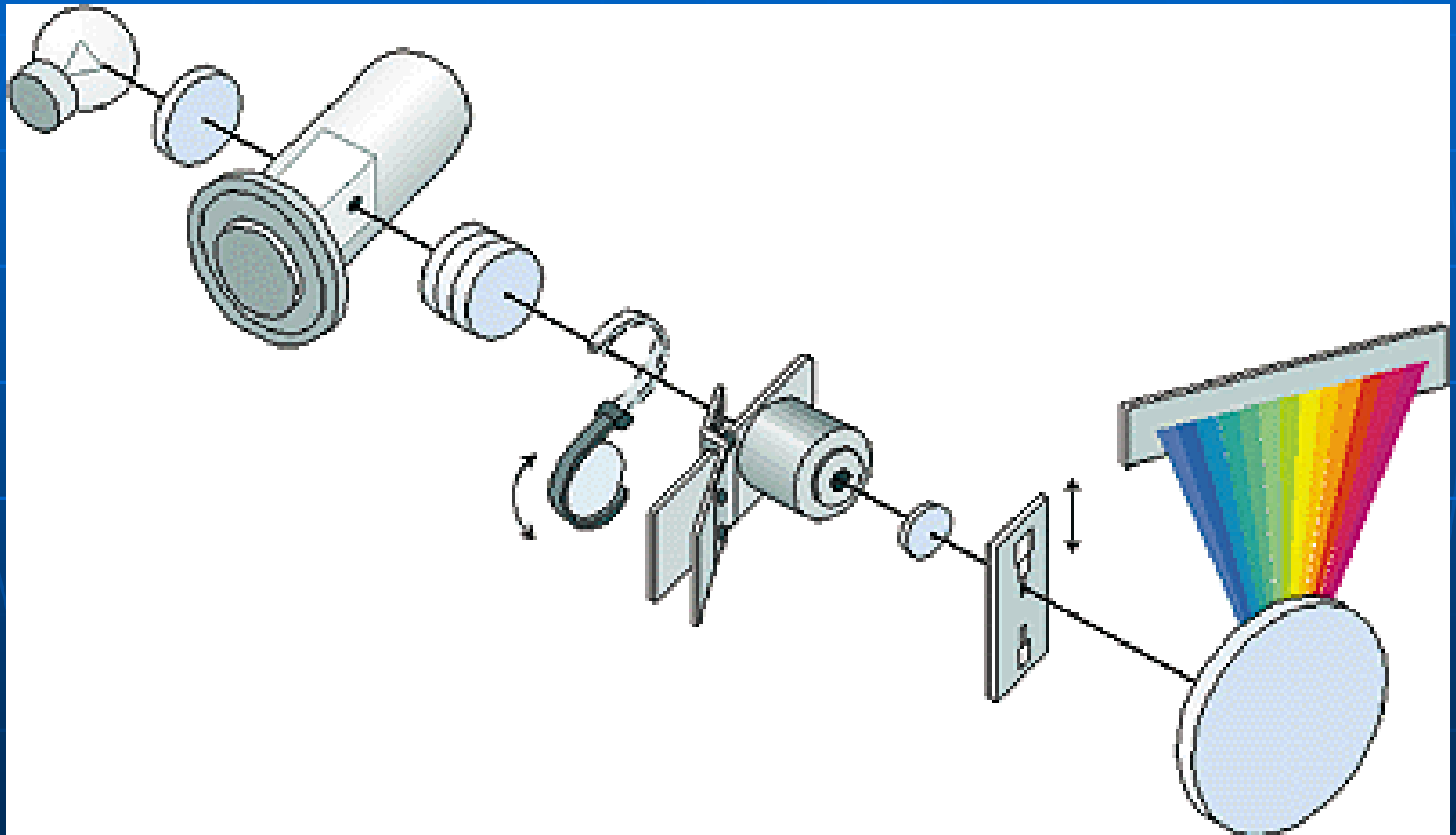
- UV/Vis detection
  - Fixed wavelength
  - Variable wavelength
  - DAD
- standard monochromatic spectrophotometer, deuterium lamp, wavelength is changed by the turning of a prism, at any moment one wavelength only

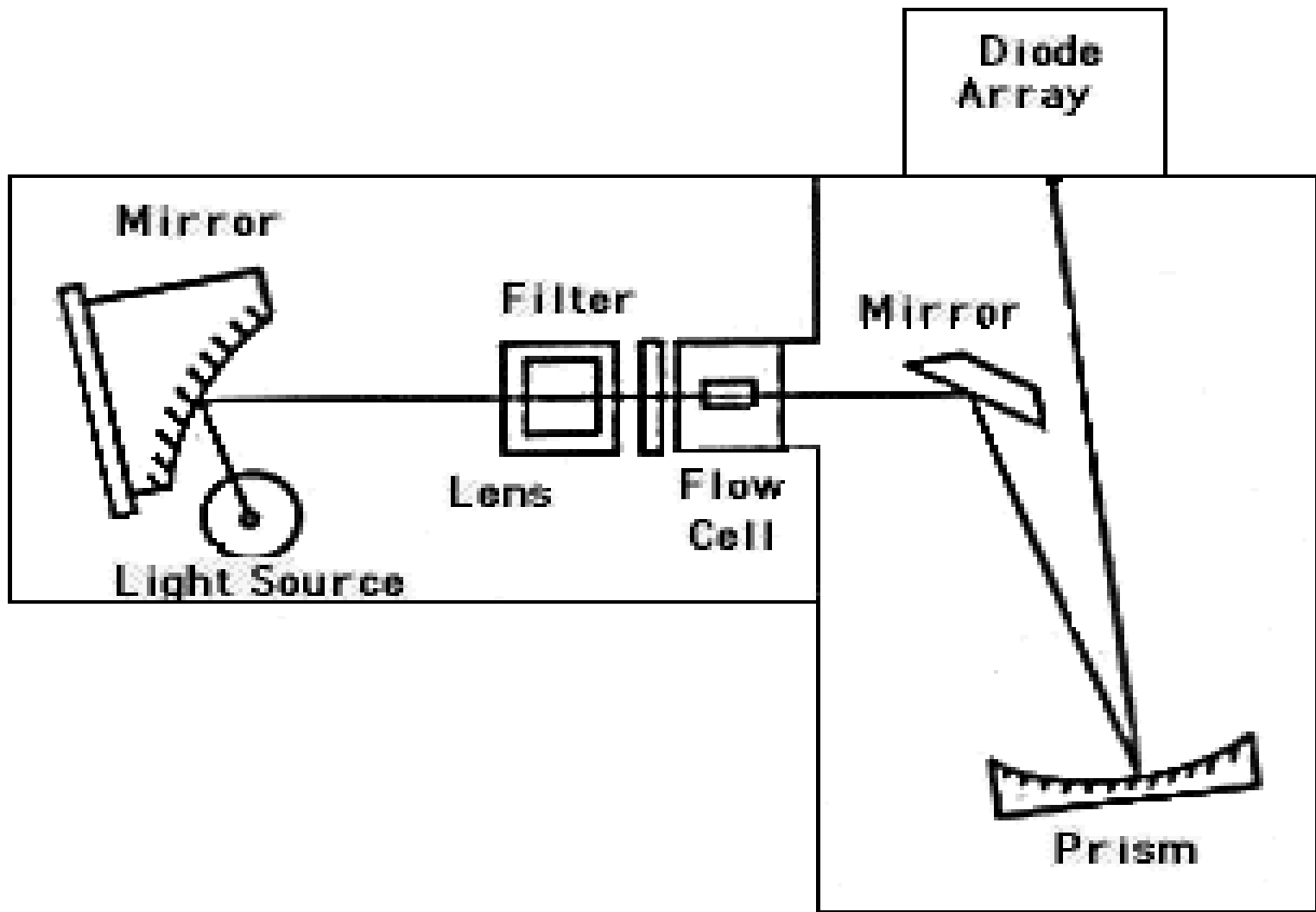
- continuous monitoring of the complete UV/Vis spectrum:
- a) several standard spectrophotometers on-line in line – several UV/Vis regions
- b) DAD - Diode Array Detector – simultaneous measurement of the whole area, data available immediately, possibility of creation of 3D chromatogram

# Standard monochromatic spectrophotometer

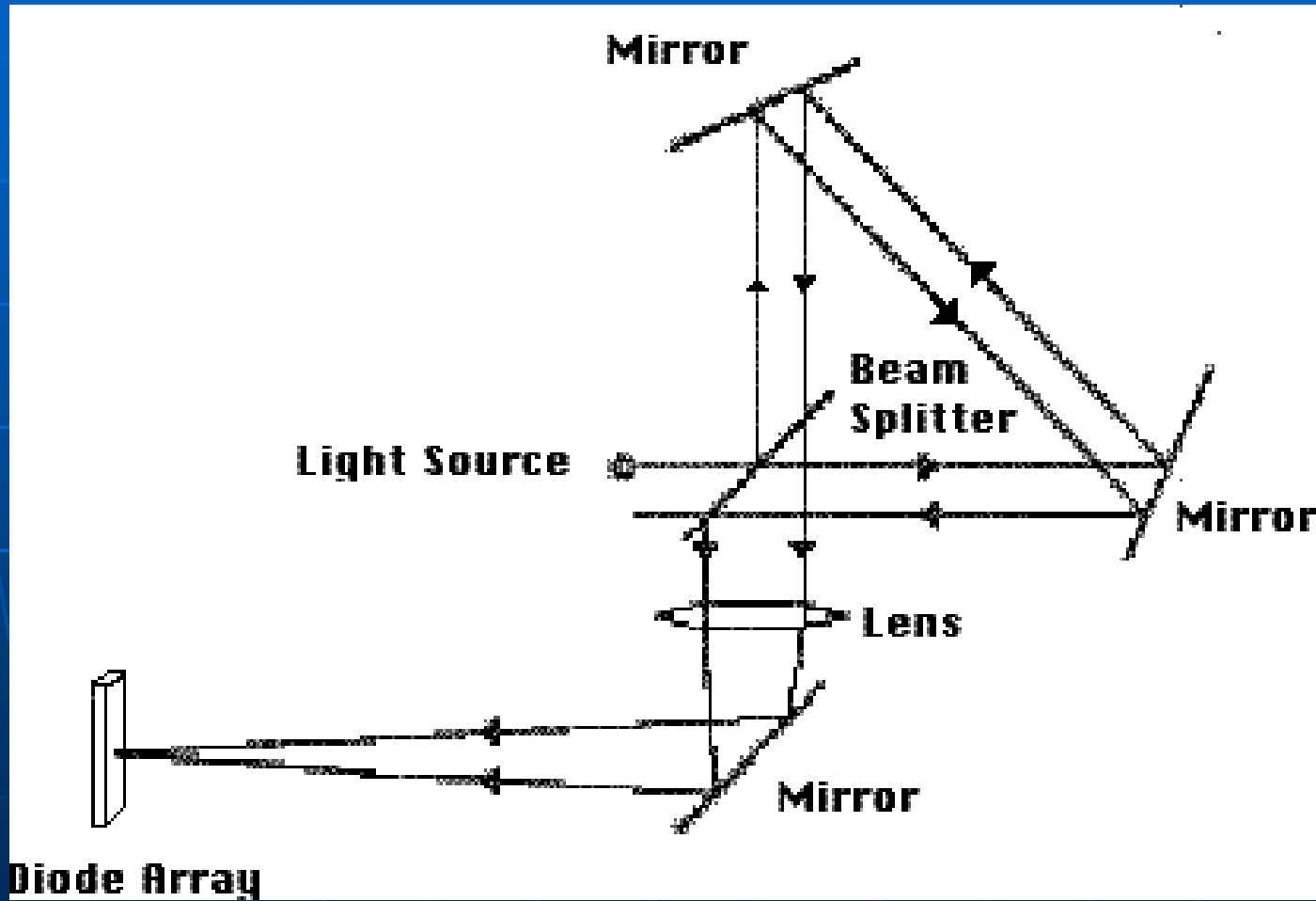


# Scheme of DAD

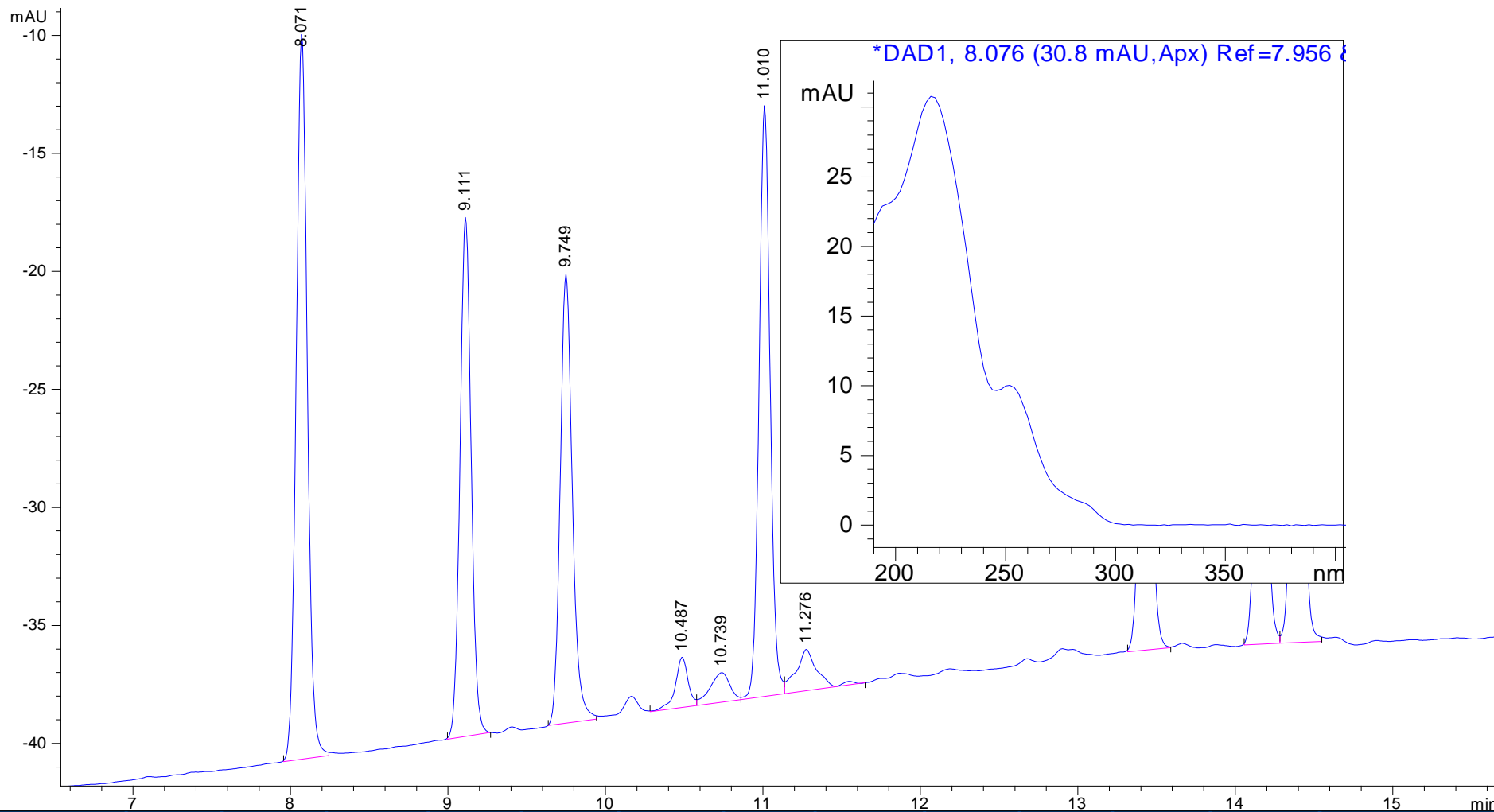








DAD1 C, Sig=215,8 Ref=700,100 (C:\HPCHEM1\DATA\SCHIZAND\I\Z050217\ST000026.D)



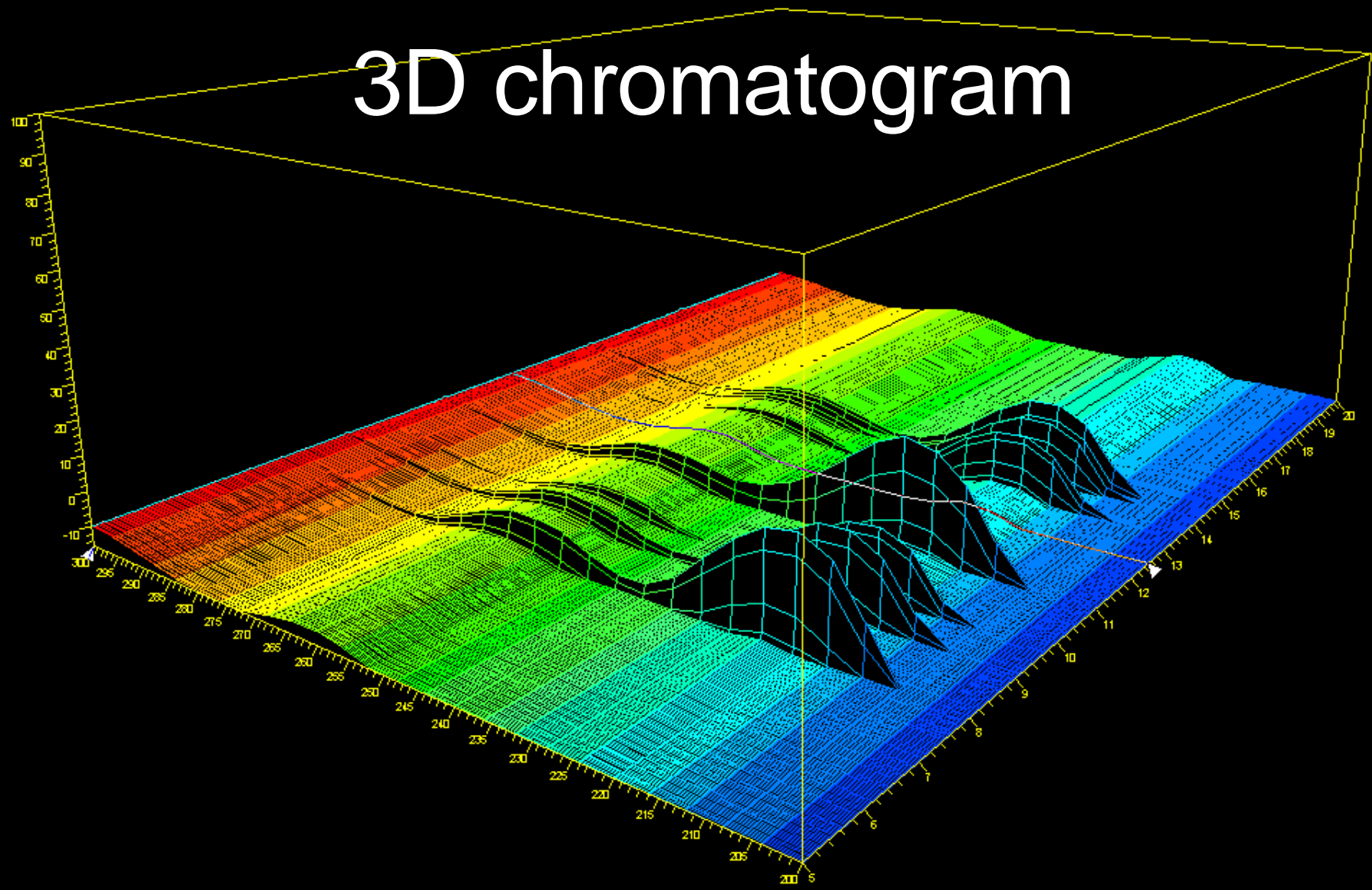
x: 12.50 min y: 300.00 nm z: -9.41 mAU  Cube

x  y  z

-140 -120 -100 -80 -60 -40 -20 0 20 40 60 80 100 120 140 160 180 200 220 240

Redraw

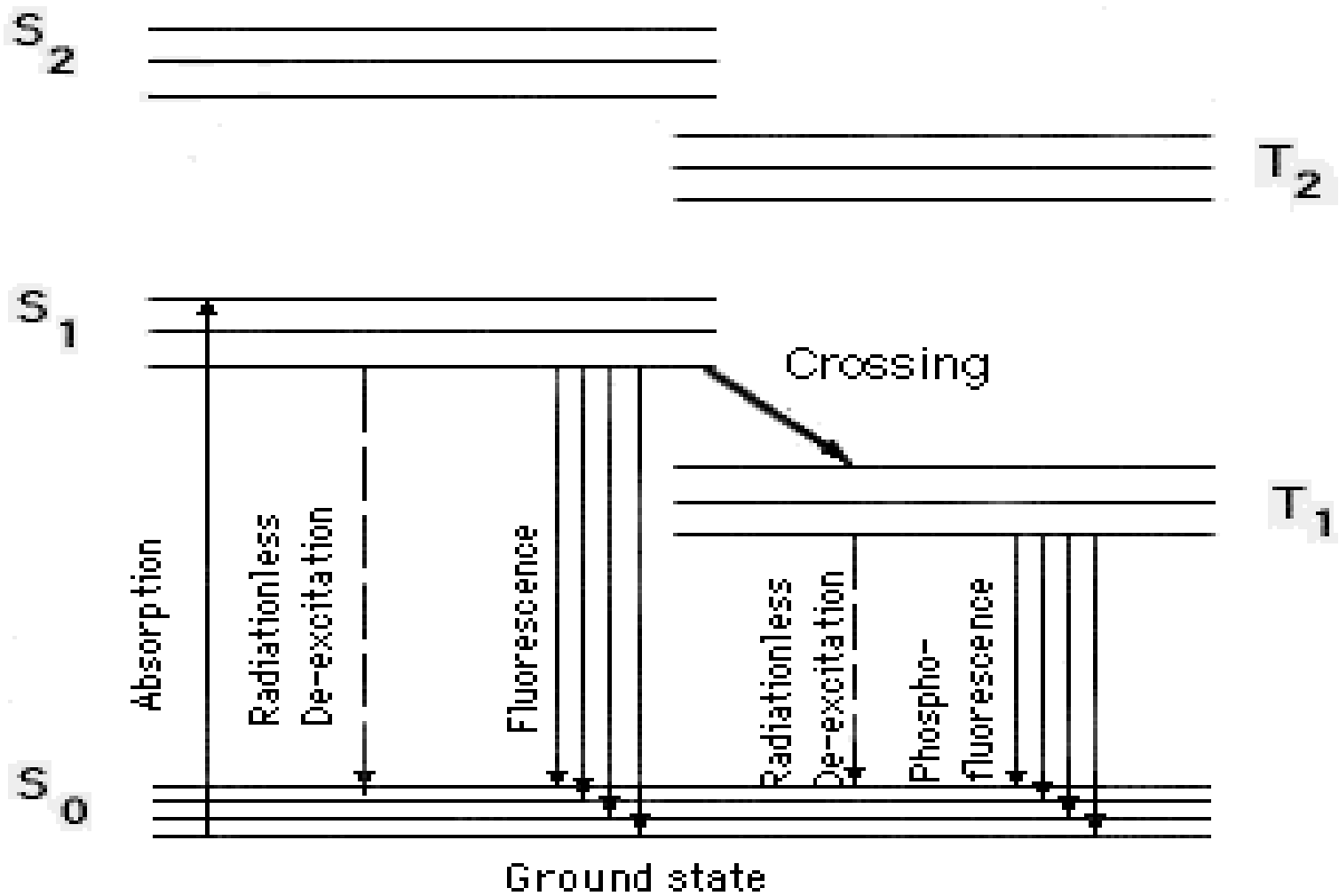
# 3D chromatogram



Expand z Data view Projection Print Close

# Fluorescent detectors

- Measurement of the ability to absorb and emit a quantum of light at a certain wavelength
- Compounds can characteristically fluoresce. Source of excitation goes through the flow cell into the photo detector, the intensity of emitted light at a certain wavelength is measured
- Sensitivity  $10^{-9}$  to  $10^{-11}$  g/ml.



# Usage of fluorescent detection

- 1) DNA/RNA
- 2) Enzymatic methods
- 3) Protein-Ligand Interactions
- 4) Fluorescent compounds detections

# Refractive index detectors – refractometers

- Reliable
- Affected by the flow rate of the mobile phase
- Usable for all types of solvents
- Low sensitivity to impurities and air bubbles
- Sensitive to temperature
- Expensive

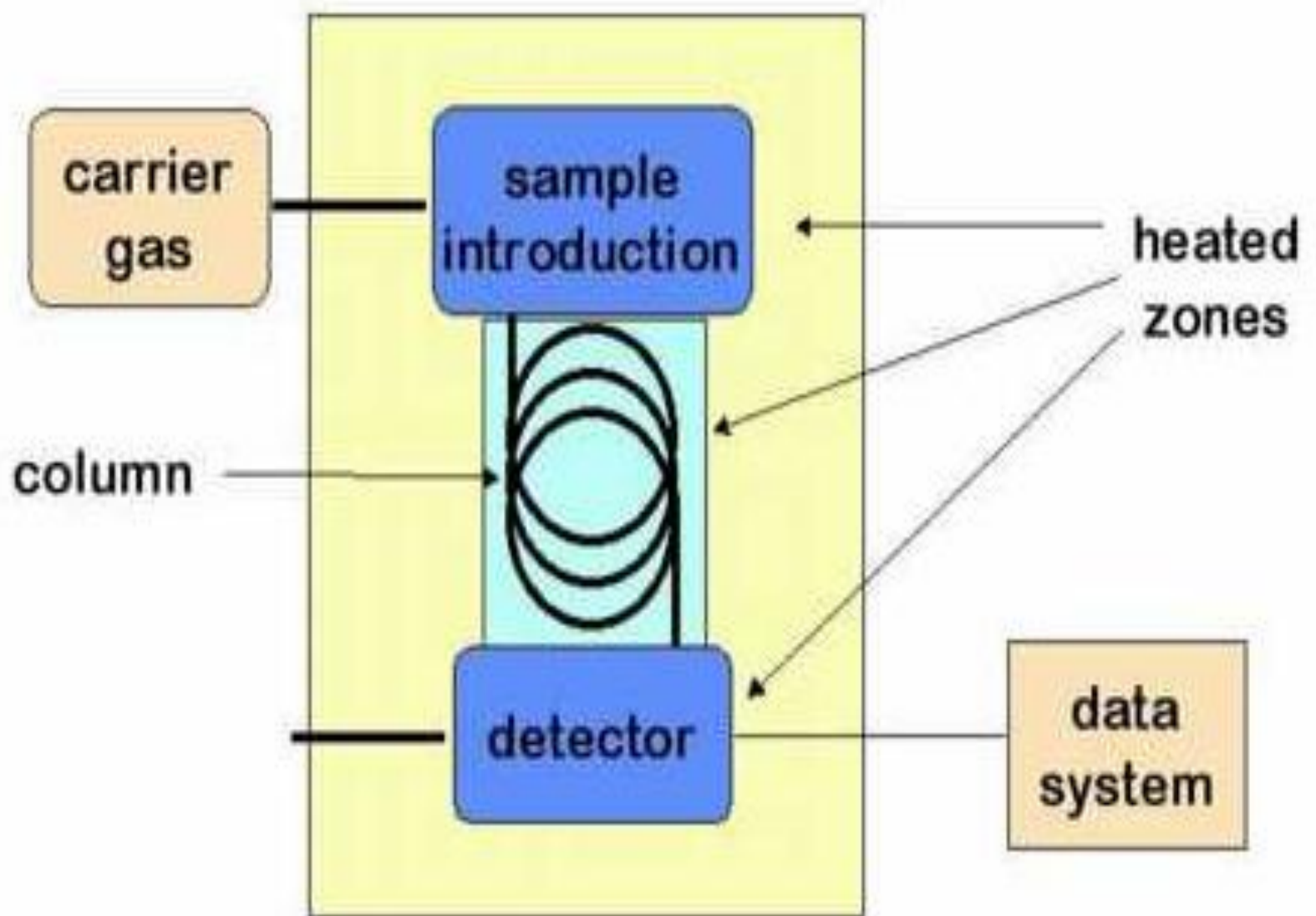
# GC - gas chromatography



- Analytical separation technique useful for separation of volatile organic substances
- Separation based on different distribution and selective retardation in column
- mobile phase - inert gas, only transportation of the analyte through the column, no interaction between the analyte and the mobile phase

# Equipment is composed of:

- Reservoir of the mobile phase - inert gas (He, Ar)
- Injector – maintained at stable temperature
- Separation column containing the stable stationary phase – tempered to controlled stable temperature
- Detector



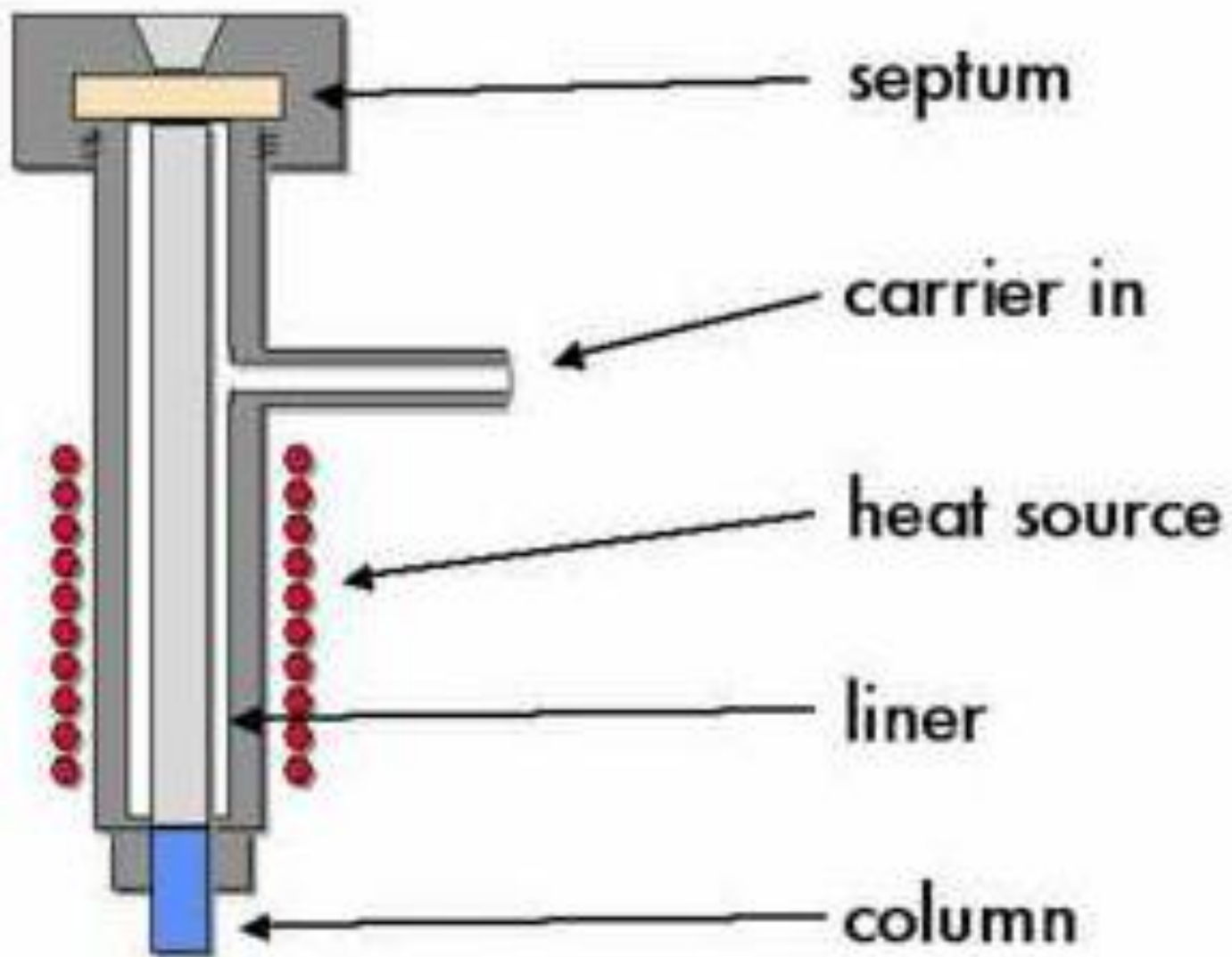


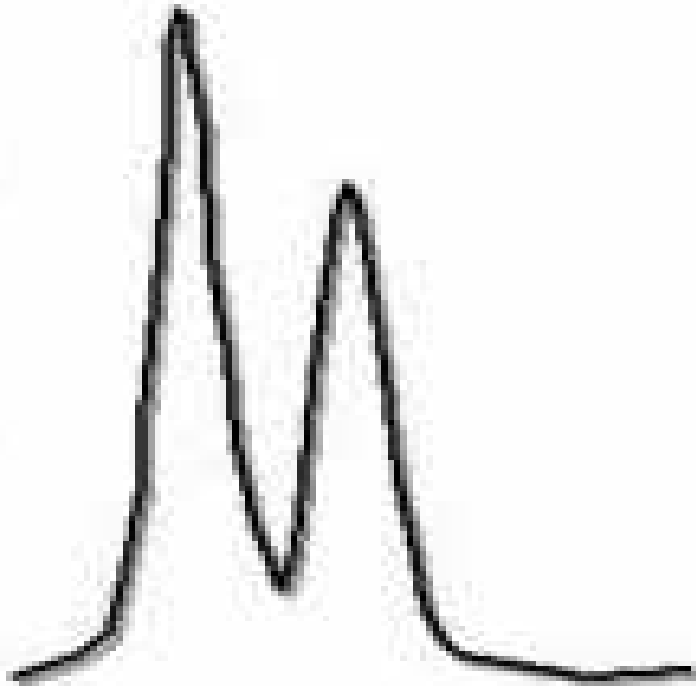
# Mobile phase

- Carrier gas
- The task is the transport of the analyte
- Inert both to analyte and stationary phase
- $N_2$ ,  $H_2$ , He in the case of ionization detectors with radioactive Ar

# Injector

- At the entrance into the column
- Small space for the gasification of sample
- Volume of sample as low as possible (0.1 - 1  $\mu\text{L}$ )
- Temperature of chamber higher than boiling point of the least volatile component in the sample





Slow



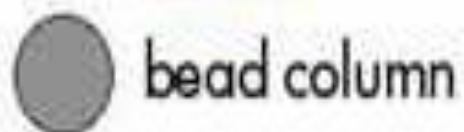
Fast



# Columns

- Packed x capillary
- Different materials
- Shape according to the type of chromatograph
  - direct
  - U-shaped
  - helix
- Necessary to maintain the stable set temperature – velocity of movement compounds dependent on temperature

Packed



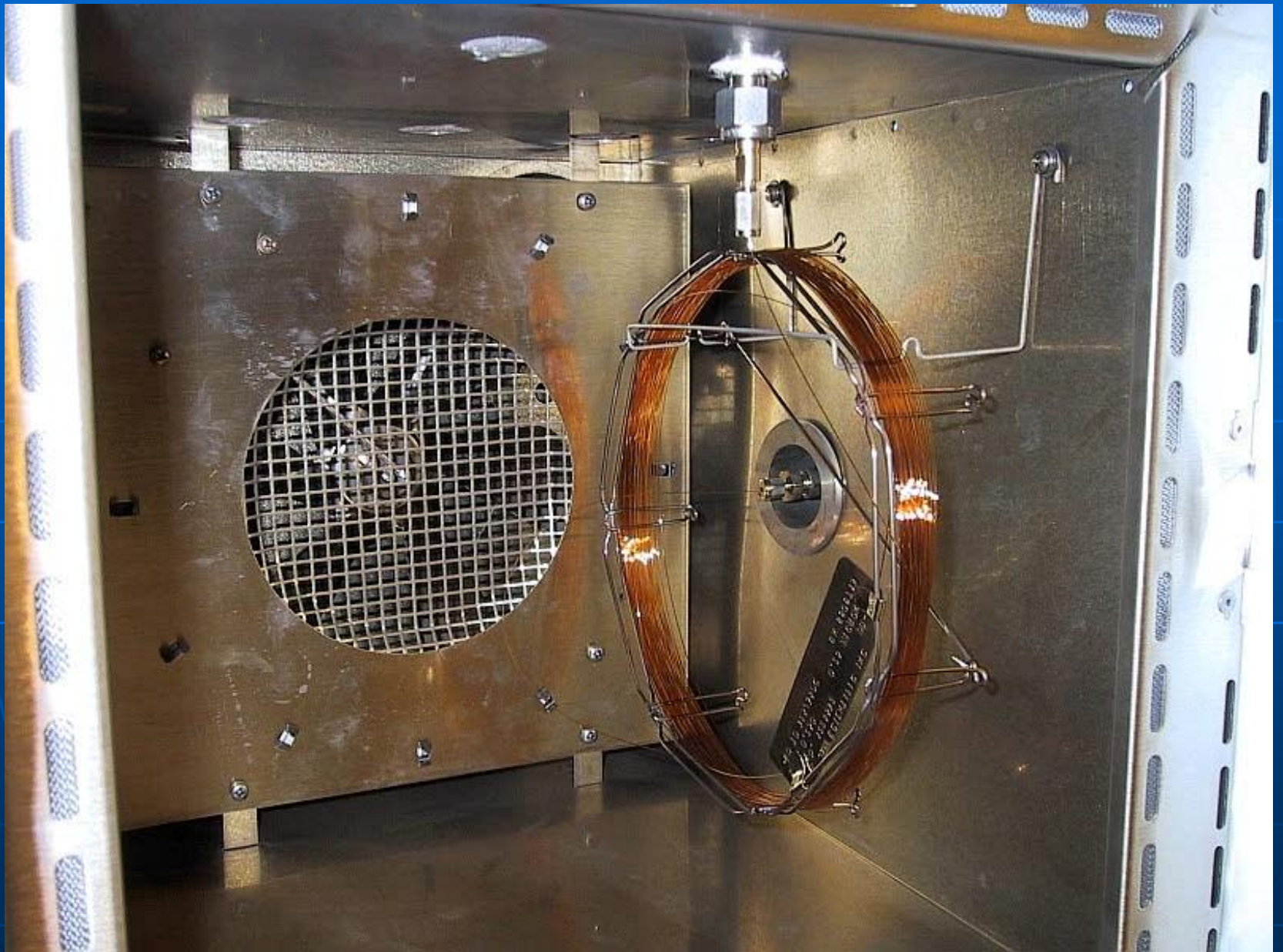
open (capillary)



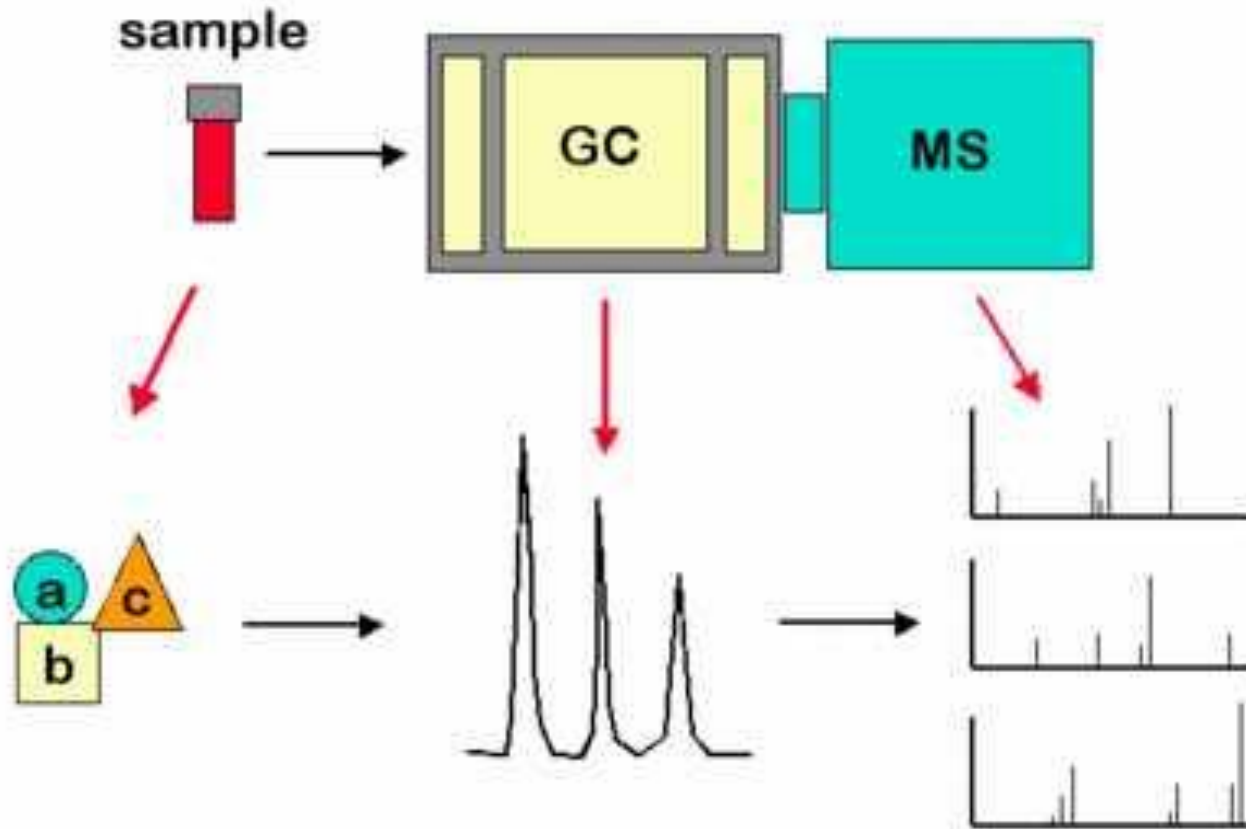
Porous  
Layer  
Open  
Tube



Wall  
Coated  
Open  
Tube



# GC/MS process



# Mass detectors

## Mass spectroscopy MS

- combination of ionization, fragmentation and separation process
- formation of ions by different methods
- fragmentation of compounds
- after trapping – mass spectrum
- interface between GC/MS
  - high vacuum
  - molecular separators

# Mass Spectroscopy (MS) – mass detectors

- a) Electron Impact (EI)- ionization by impact of electron beam or strong electric field
- b) Chemical Ionization- ionization via ionized gas
- c) Fast Atom Bombardment (FAB) – bombarding by Xenon atoms
- sensitivity  $10^{-8}$  to  $10^{-10}$  g/l.

## A simple mass spectrum

The masses in this example should look familiar.  
This is a spectrum for air - a mixture.

