

Aerobic metabolism

6.1) Citric Acid Cycle

6.2) Electron transport

6.3) Oxidative phosphorylation

6.4) Oxidative stress

- Aerobic oxidation of glucose- greater amount of energy than does fermentation
- Oxygen highly reactive
- Oxidative cell damage: enzymes, antioxidant molecules

The Citric Acid Cycle

The citric acid cycle is the final common pathway for the oxidation of fuel molecules: amino acids, fatty acids, & carbohydrates.

- Most fuel molecules enter the cycle as acetyl coenzyme A
- This cycle is the central metabolic hub of the cell
- It is the gateway to aerobic metabolism for any molecule that can be transformed into an acetyl group or dicarboxylic acid,
- It is also an important source of precursors for building blocks
- Also known as, Krebs Cycle, & Tricarboxylic Acid Cycle (TCA)

Chapter 17: Outline

17.1 The citric acid cycle oxidizes two-carbon units

17.2 Entry to the cycle and metabolism through it are controlled

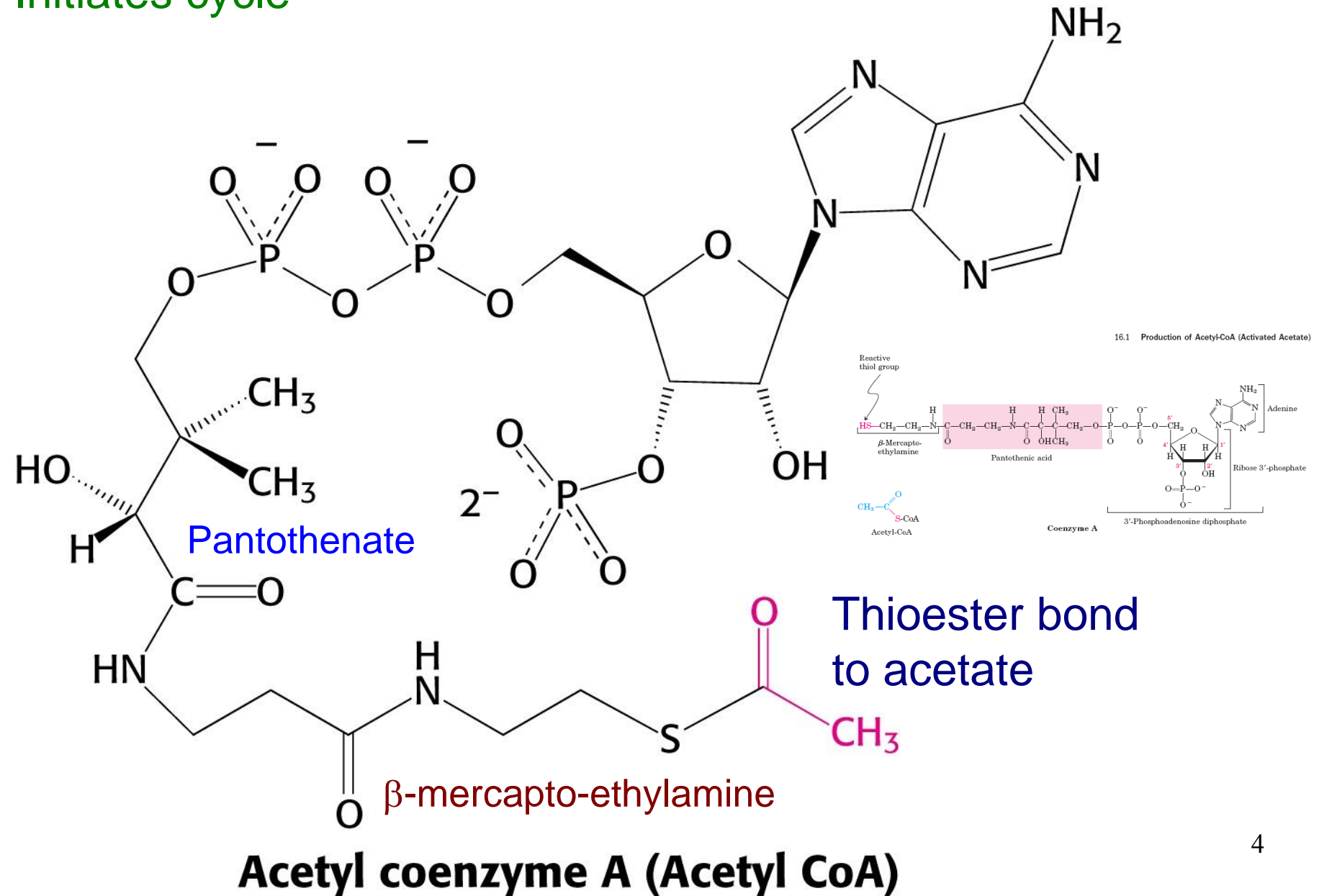
17.3 The cycle is a source of biosynthetic precursors

Overview of citric acid cycle

1. The function of the cycle is the harvesting of high-energy electrons from carbon fuels
2. The cycle itself neither generates ATP nor includes O_2 as a reactant
3. Instead, it removes electrons from acetyl CoA & uses them to form NADH & $FADH_2$ (high-energy electron carriers)
4. In oxidative phosphorylation, electrons from reoxidation of NADH & $FADH_2$ flow through a series of membrane proteins (electron transport chain) to generate a proton gradient
5. These protons then flow back through ATP synthase to generate ATP from ADP & inorganic phosphate
6. O_2 is the final electron acceptor at the end of the electron transport chain
7. The cytric acid cycle + oxidative phosphorylation provide > 95% of energy used in human aerobic cells

Fuel for the Citric Acid Cycle

Initiates cycle

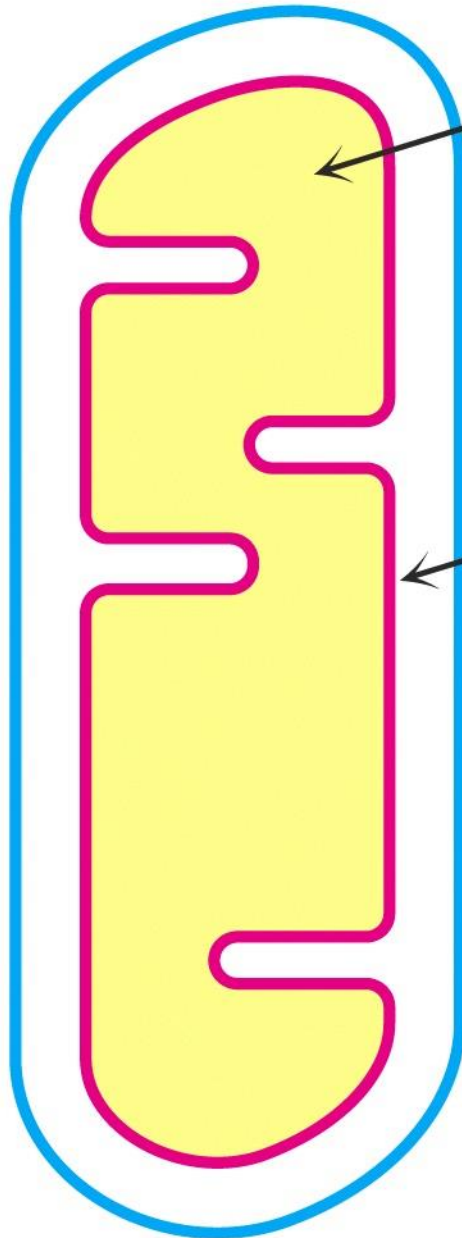


Mitochondrion



Double membrane, & cristae: invaginations of inner membrane

Mitochondrion



Matrix Oxidative decarboxilation of pyruvate, & citric acid cycle take place in matrix, along with fatty acid oxidation

Inner mitochondrial membrane

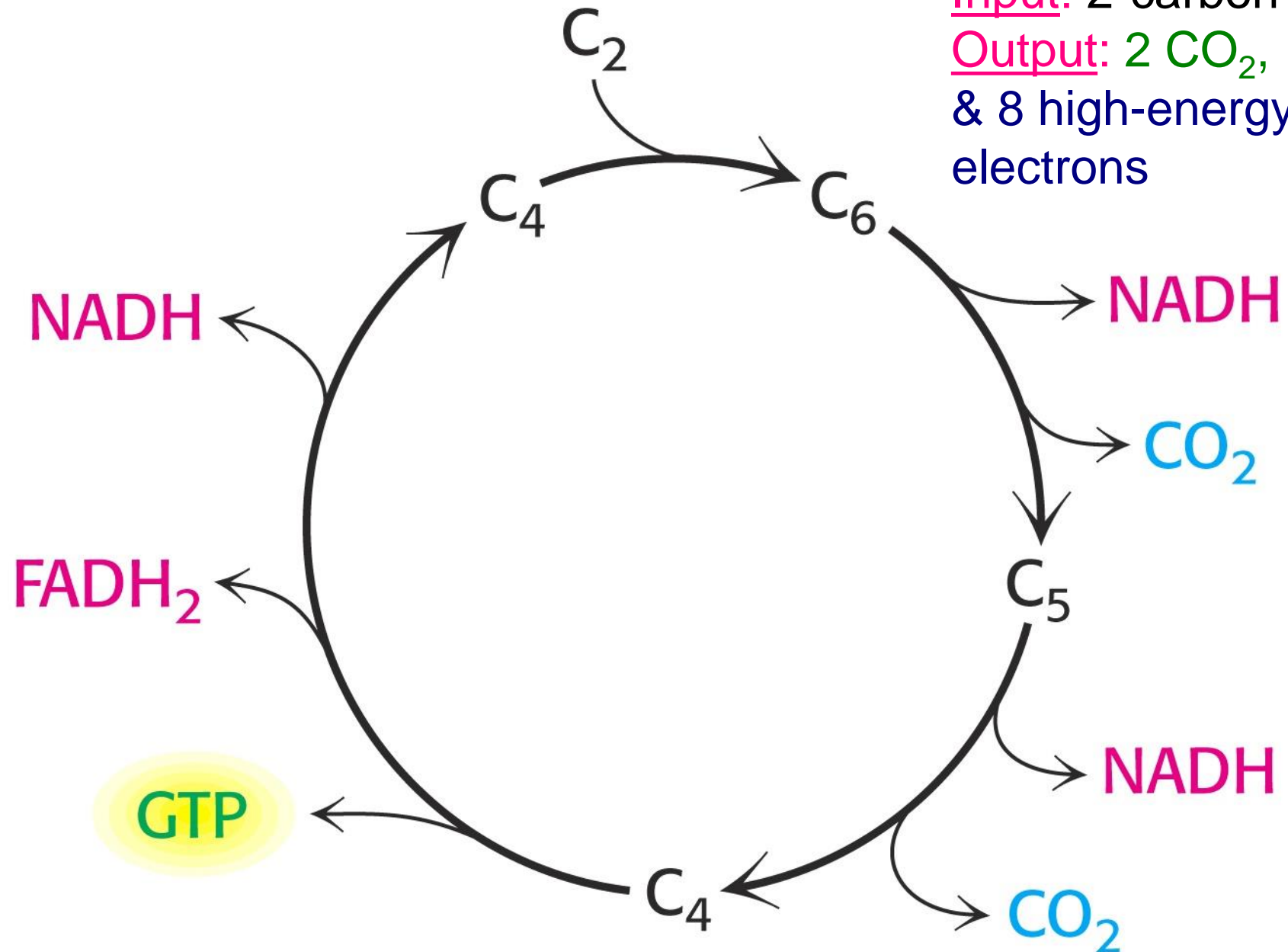
Site of oxidative phosphorylation

Outer mitochondrial membrane

Permeable

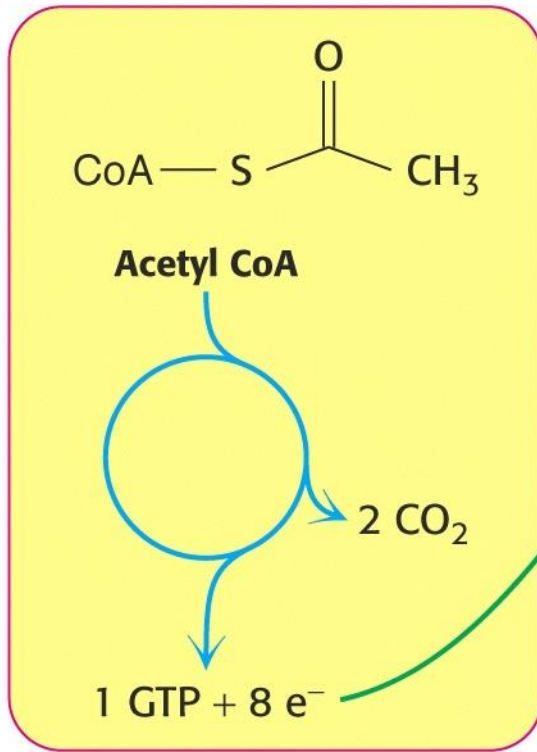
Citric Acid Cycle: Overview

Input: 2-carbon units
Output: 2 CO₂, 1 GTP,
& 8 high-energy
electrons



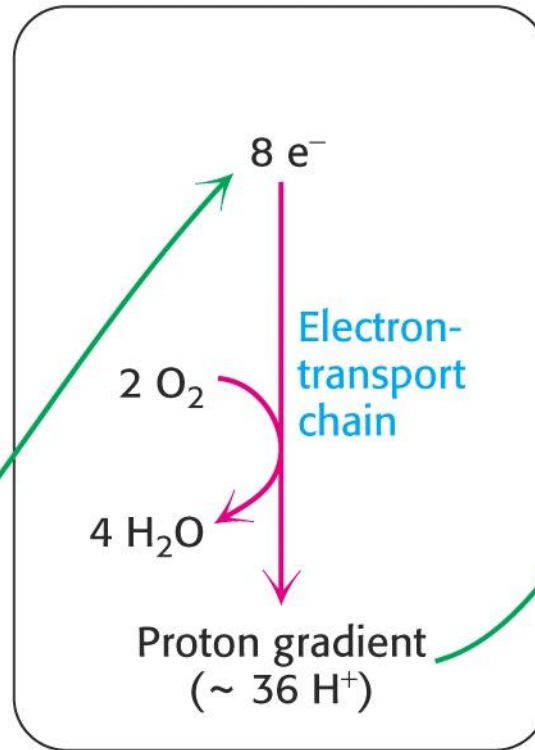
Cellular Respiration

CITRIC ACID CYCLE

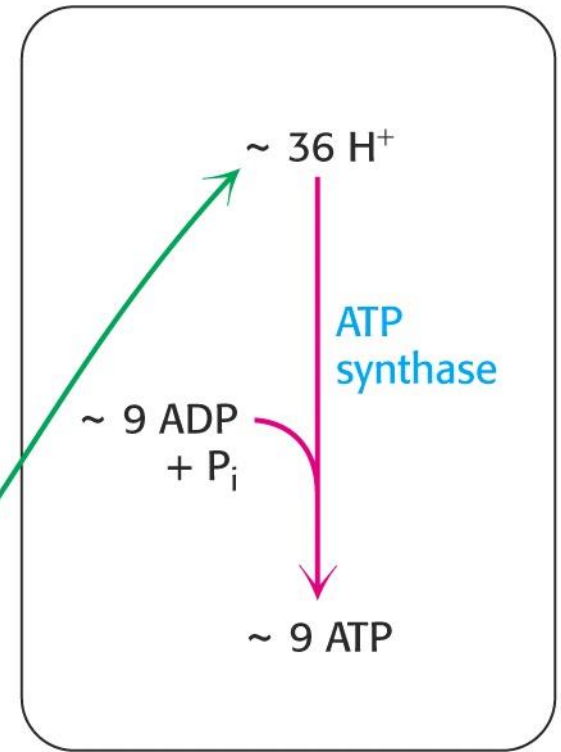


8 high-energy electrons from carbon fuels

OXIDATIVE PHOSPHORYLATION

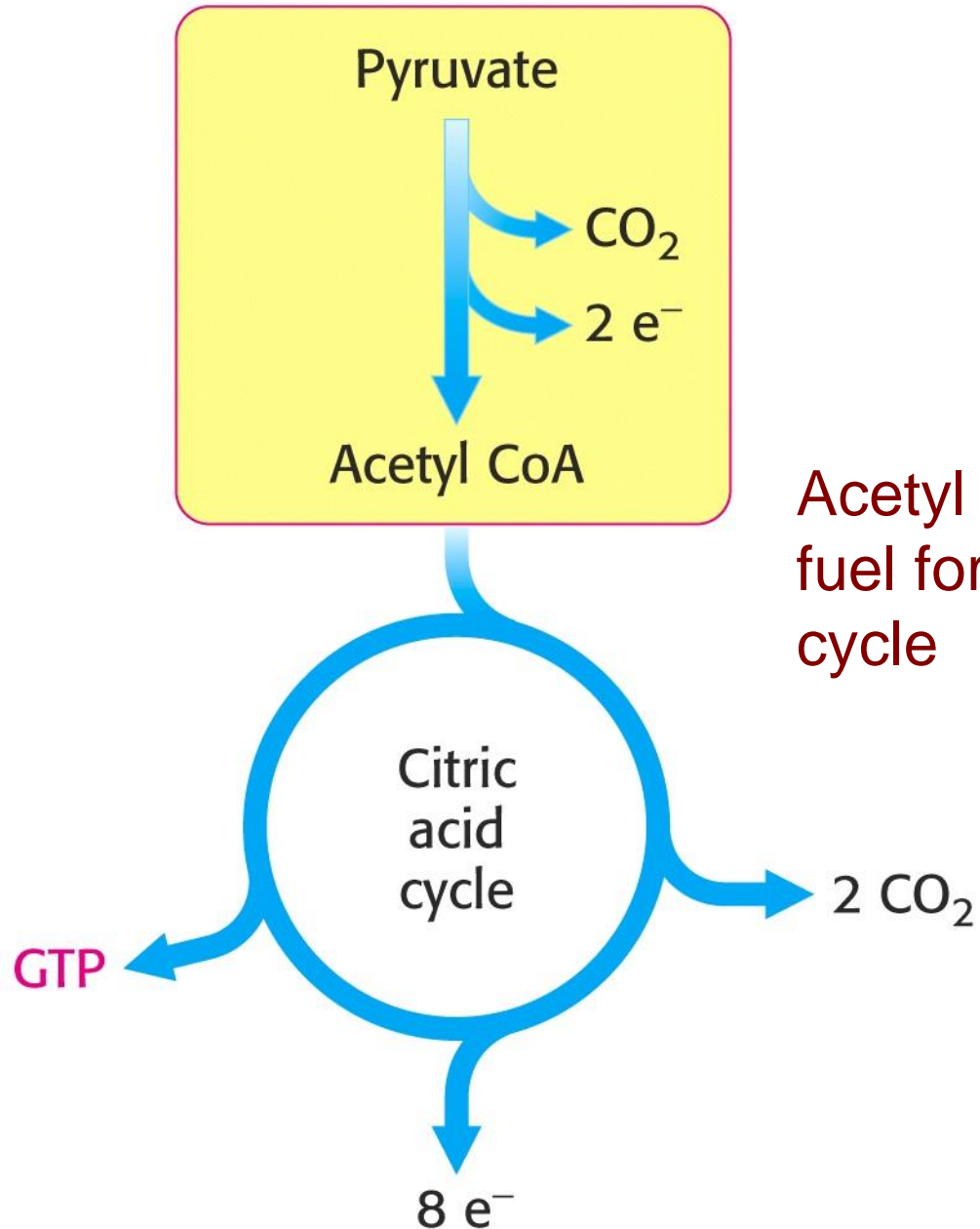


Electrons reduce O₂ to generate a proton gradient



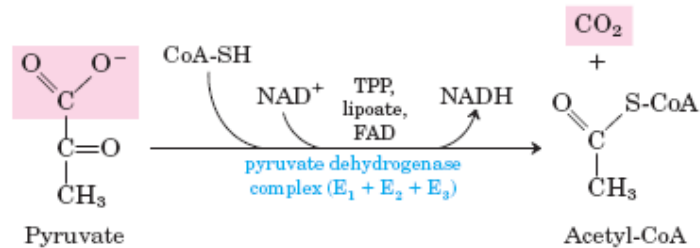
ATP synthesized from proton gradient

Glycolysis to citric acid cycle link



Oxidative decarboxylation

A large, highly integrated complex of three kinds of enzymes



$$\Delta G'^{\circ} = -33.4 \text{ kJ/mol}$$

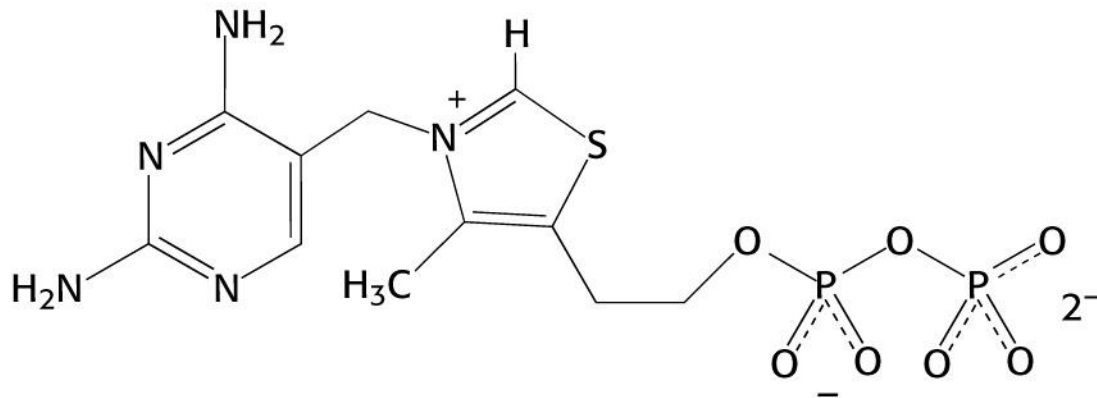
TABLE 17.1 Pyruvate dehydrogenase complex

Enzyme	Abbreviation	Number of chains	Prosthetic group	Reaction catalyzed
Pyruvate dehydrogenase component	E ₁	24	TPP	Oxidative decarboxylation of pyruvate
Dihydrolipoyl transacetylase	E ₂	24	Lipoamide	Transfer of the acetyl group to CoA
Dihydrolipoyl dehydrogenase	E ₃	12	FAD	Regeneration of the oxidized form of lipoamide

Groups travel from one active site to another, connected by tethers to the core of the structure

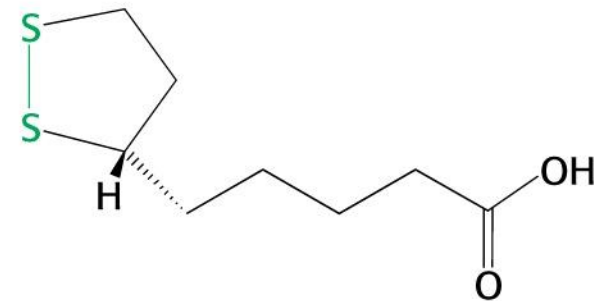
-irreversible oxidation, carboxyl group is removed from pyruvate as CO₂ + AcCoA

3 enzymes 5 Coenzymes:



Thiamine pyrophosphate (TPP)

B₁ vitamin



Lipoic acid

Vitamines: thiamine, riboflavin (FAD), niacin (NAD), pantothenate (CoA)

CoA

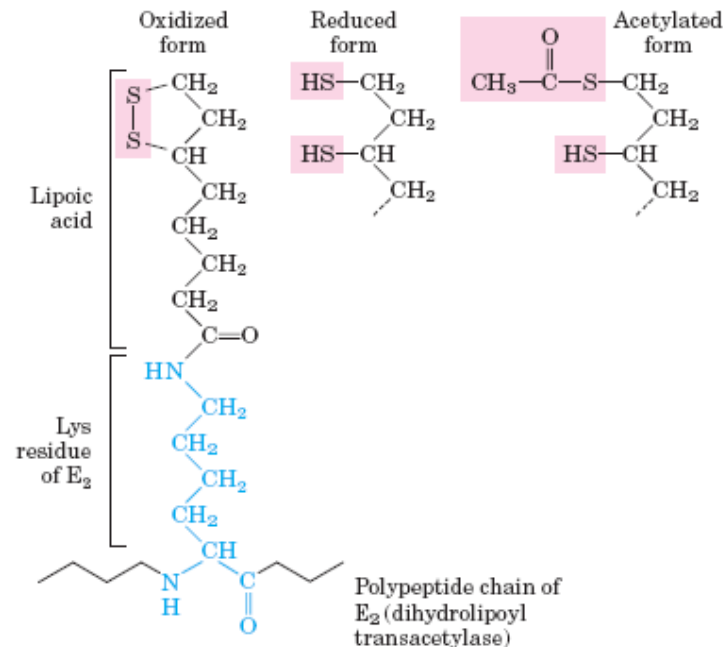
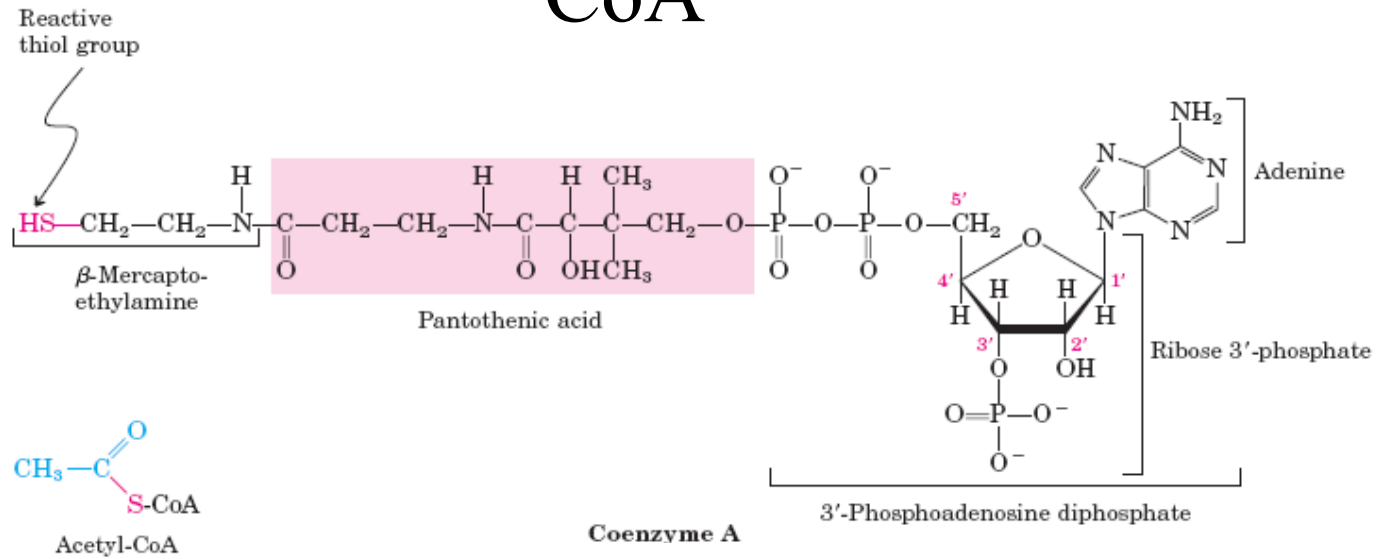
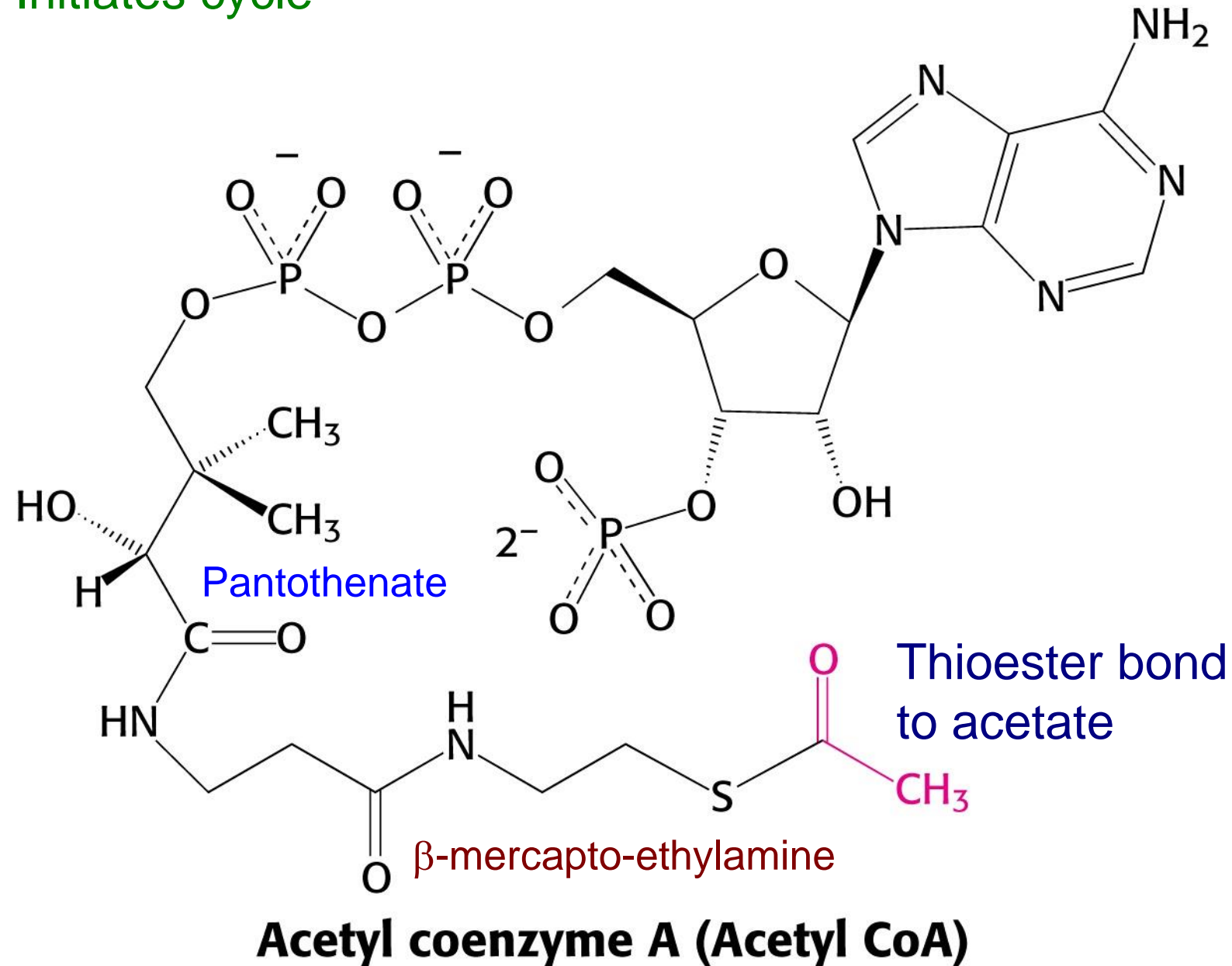


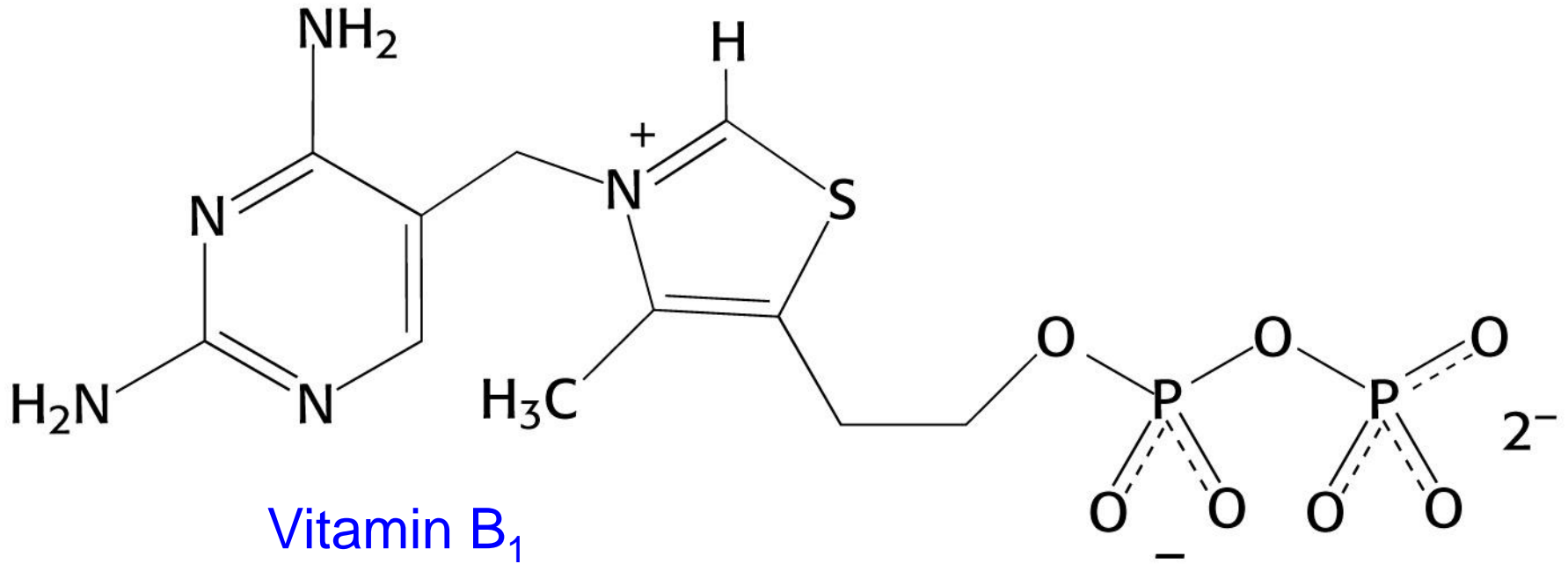
FIGURE 16-4 Lipoic acid (lipoate) in amide linkage with a Lys residue. The lipoyllysyl moiety is the prosthetic group of dihydrolipoyl transacetylase (E_2 of the PDH complex). The lipoyl group occurs in

Fuel for the Citric Acid Cycle

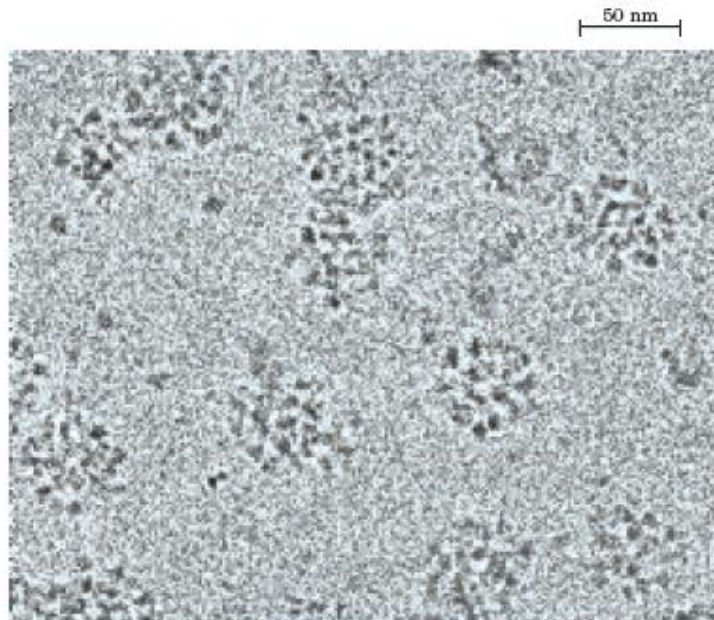
Initiates cycle



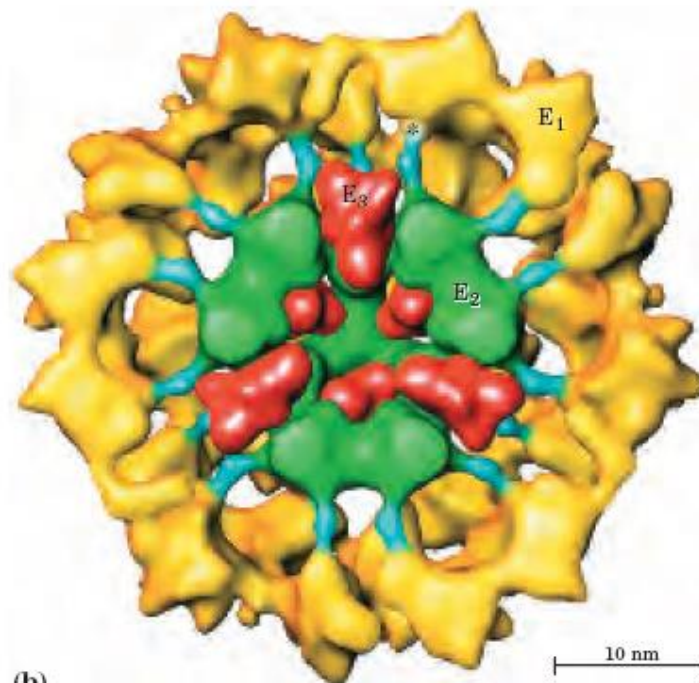
TPP



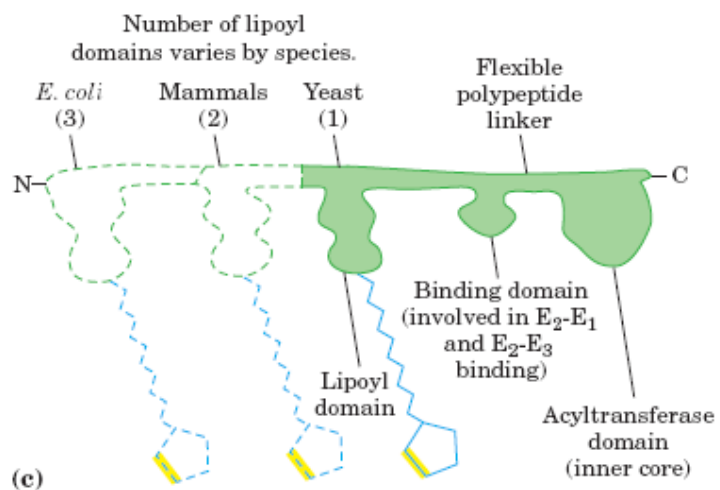
Thiamine pyrophosphate (TPP)



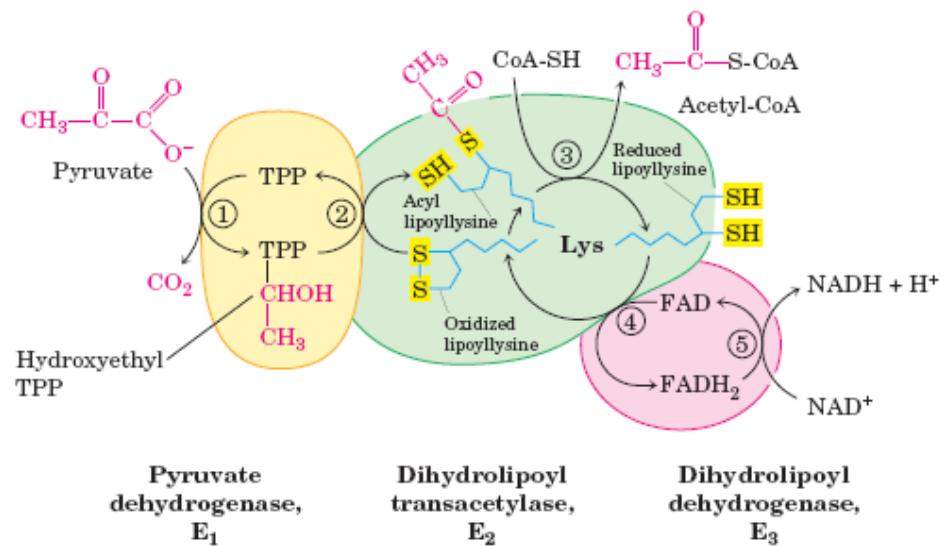
(a)

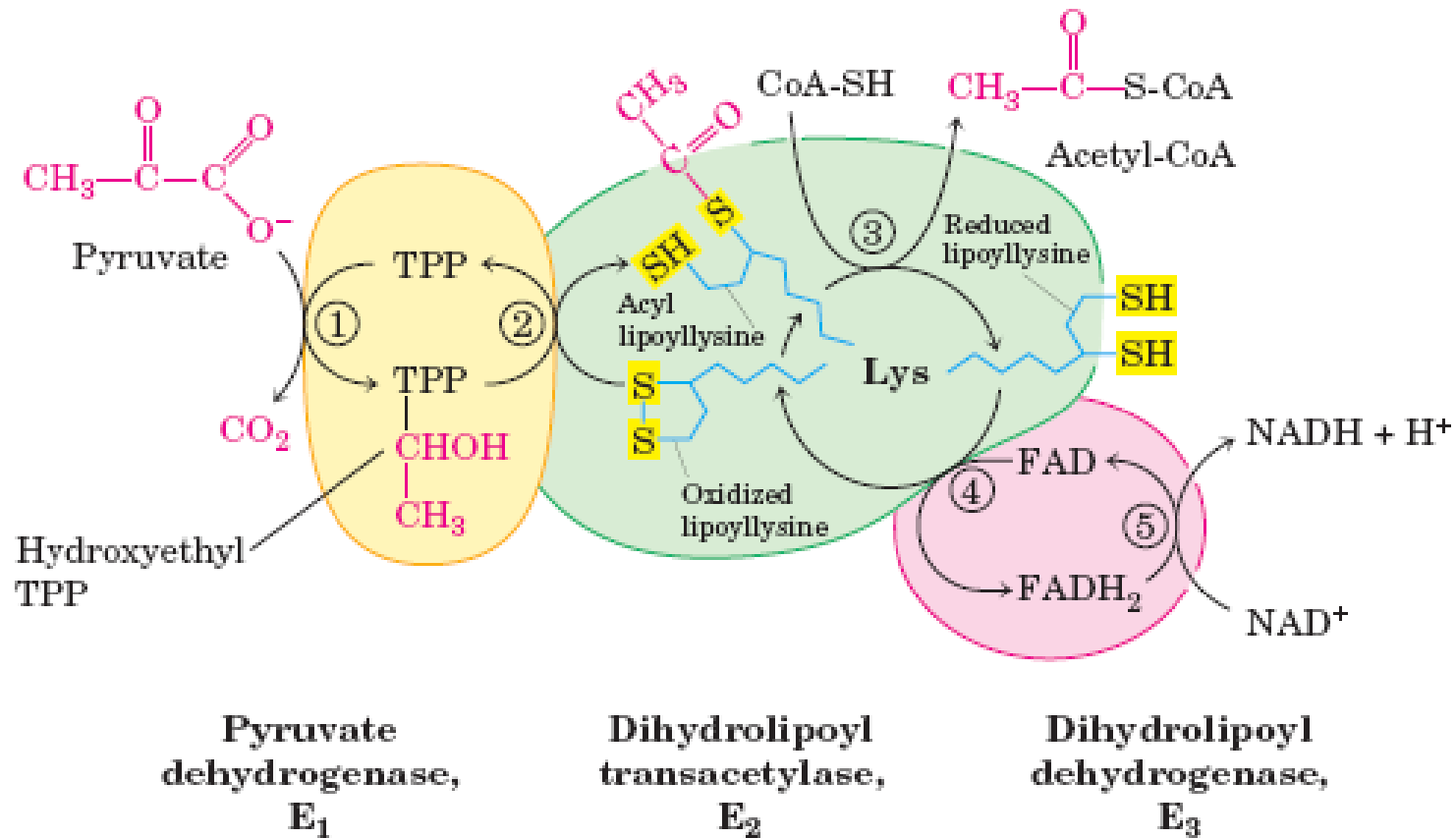


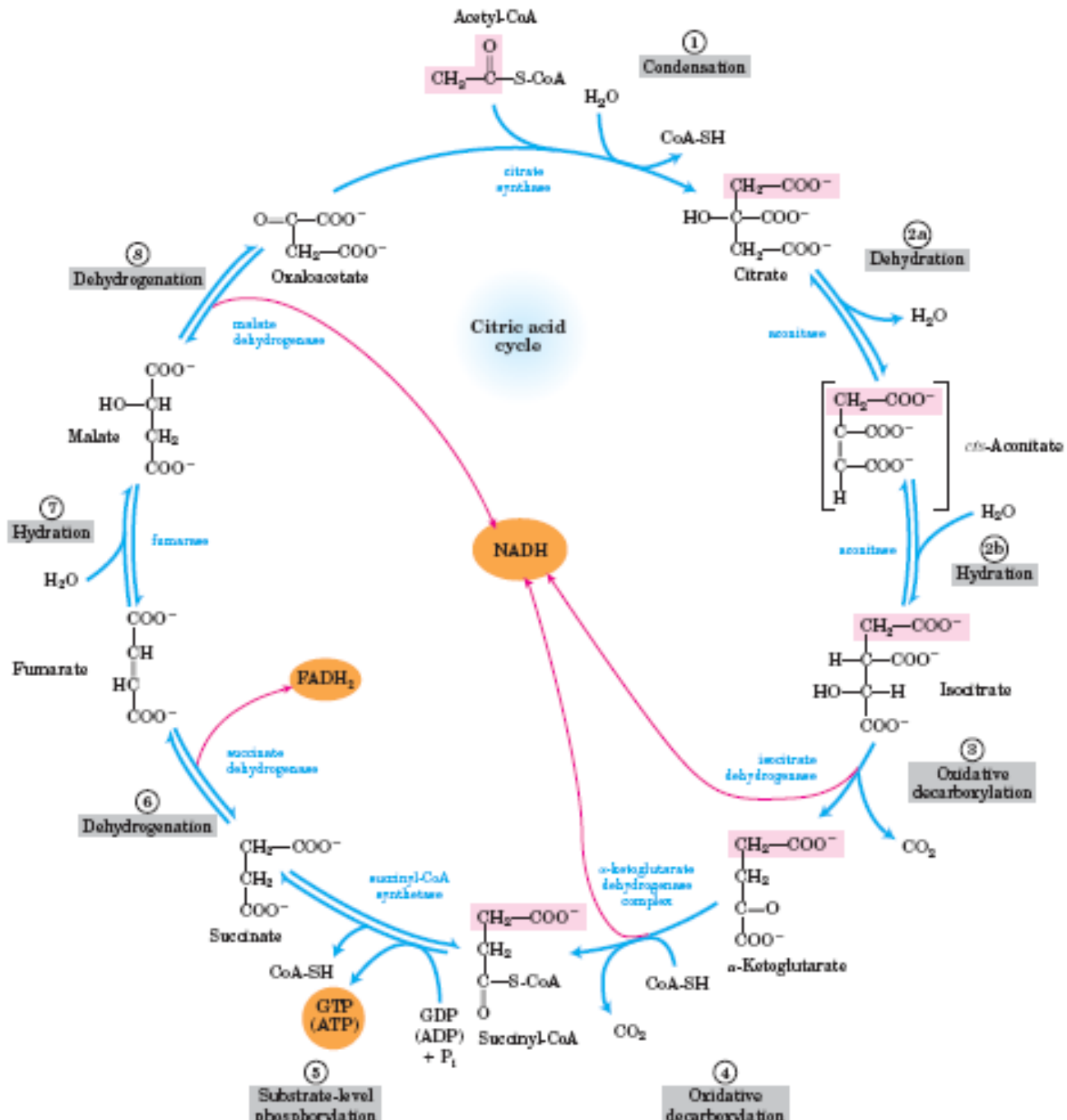
(b)

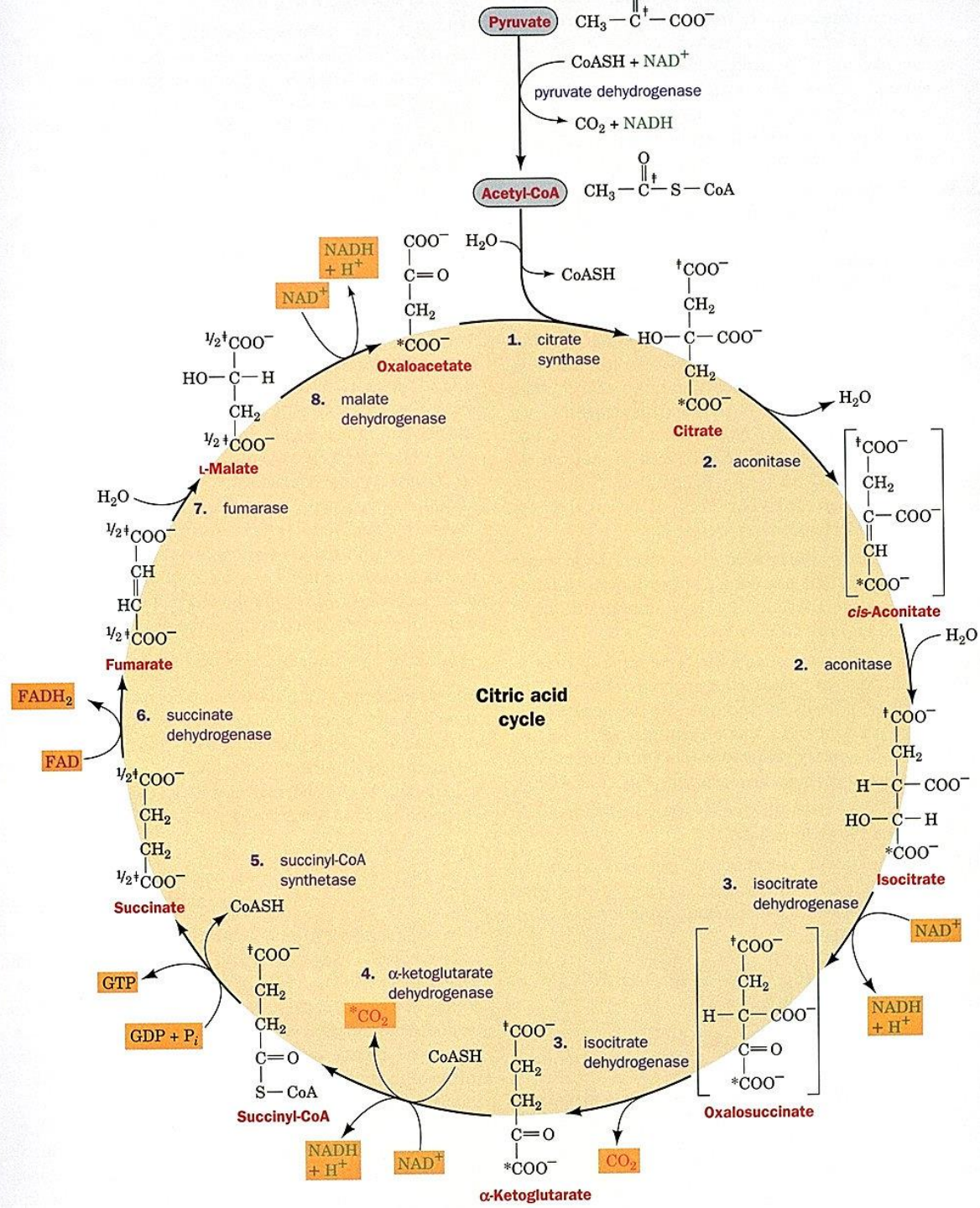


(c)



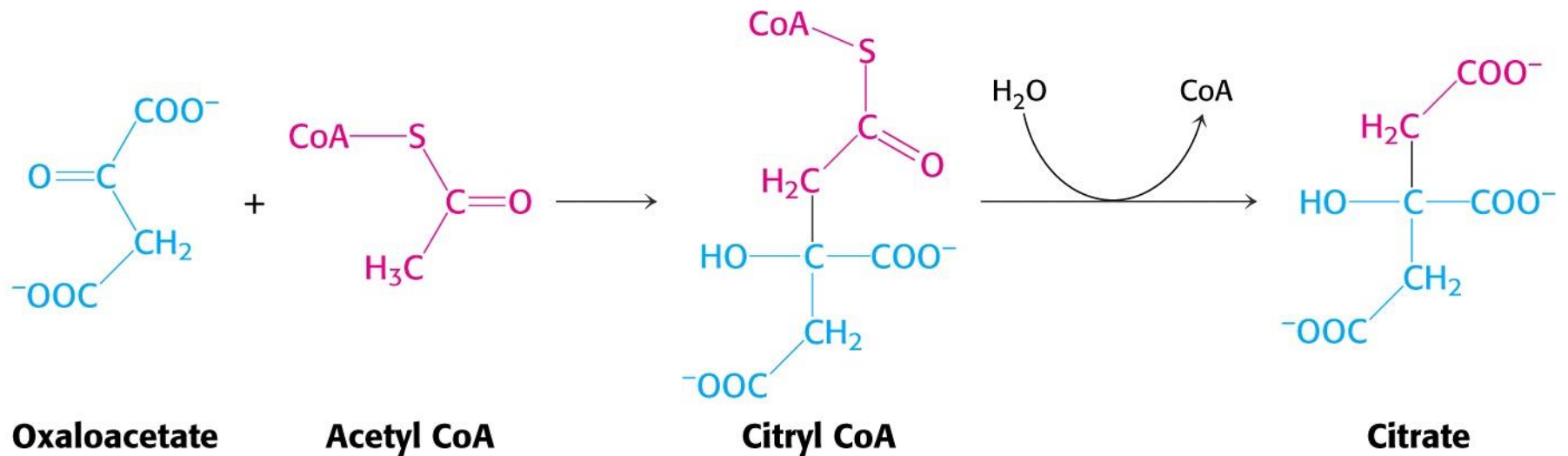






Citrate Cycle: step 1 (citrate formation)

Enzyme: Citrate synthase

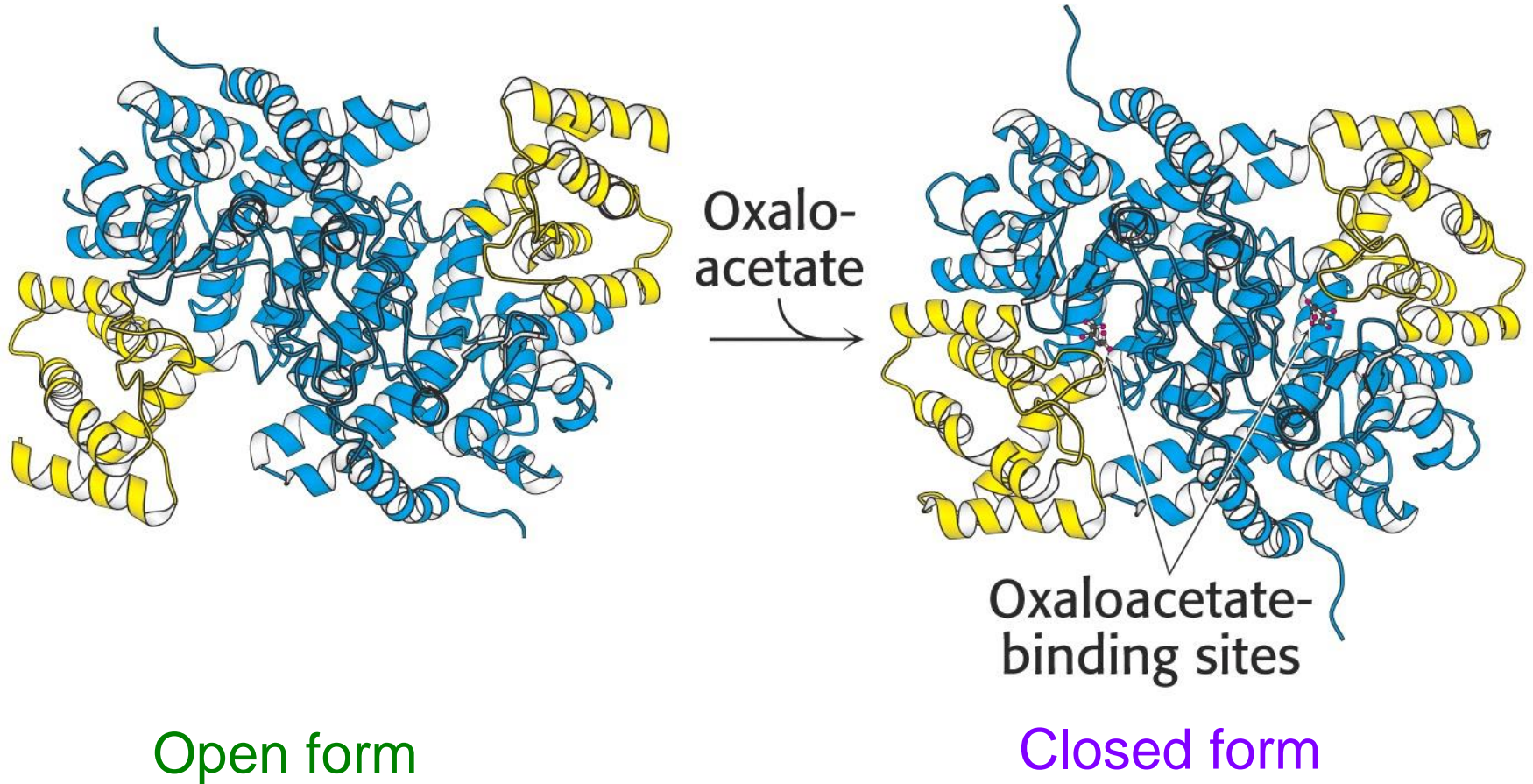


Condensation reaction

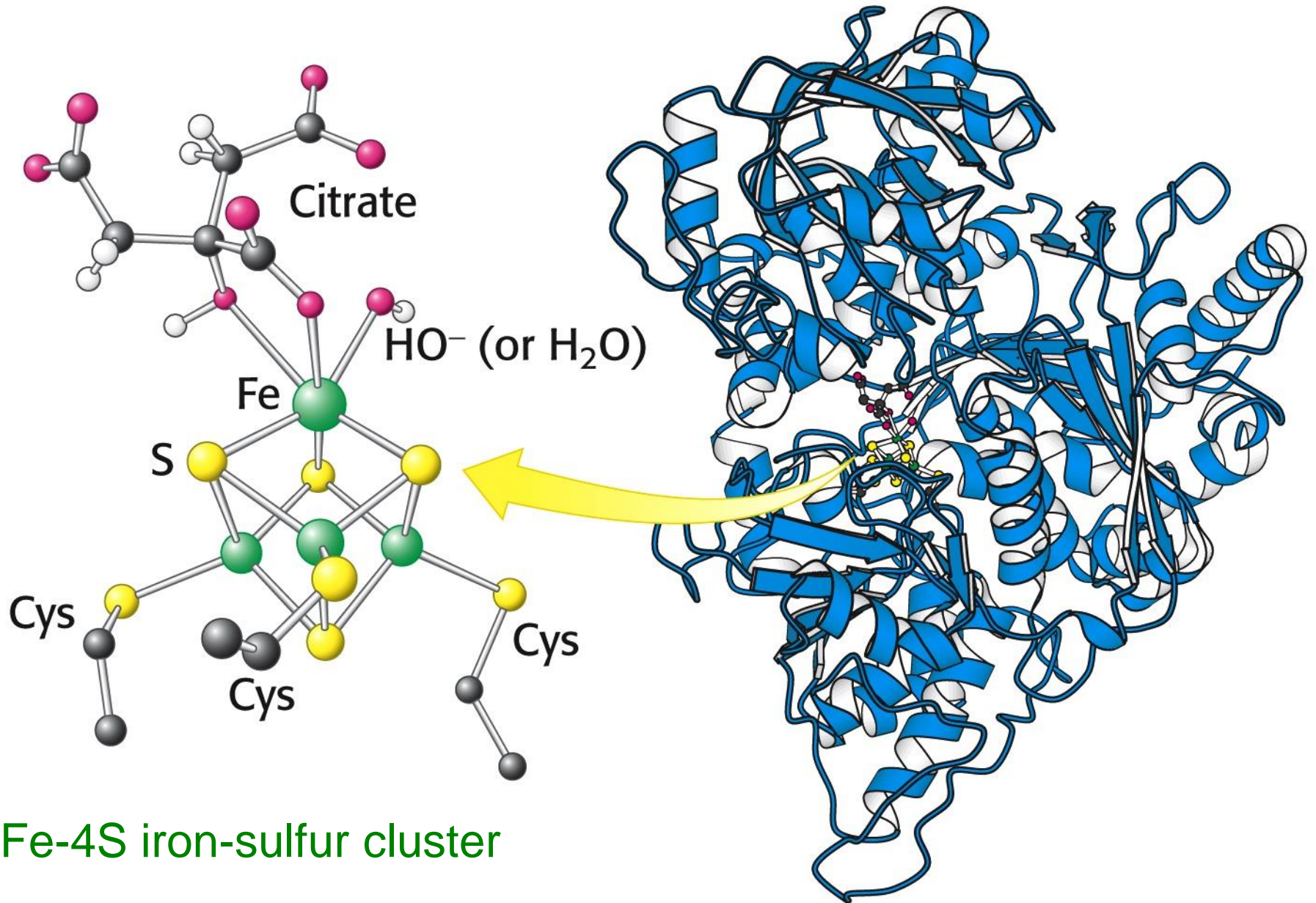
Hydrolysis reaction

Conformational changes in citrate synthase

Homodimer with large (blue) & small (yellow) domains

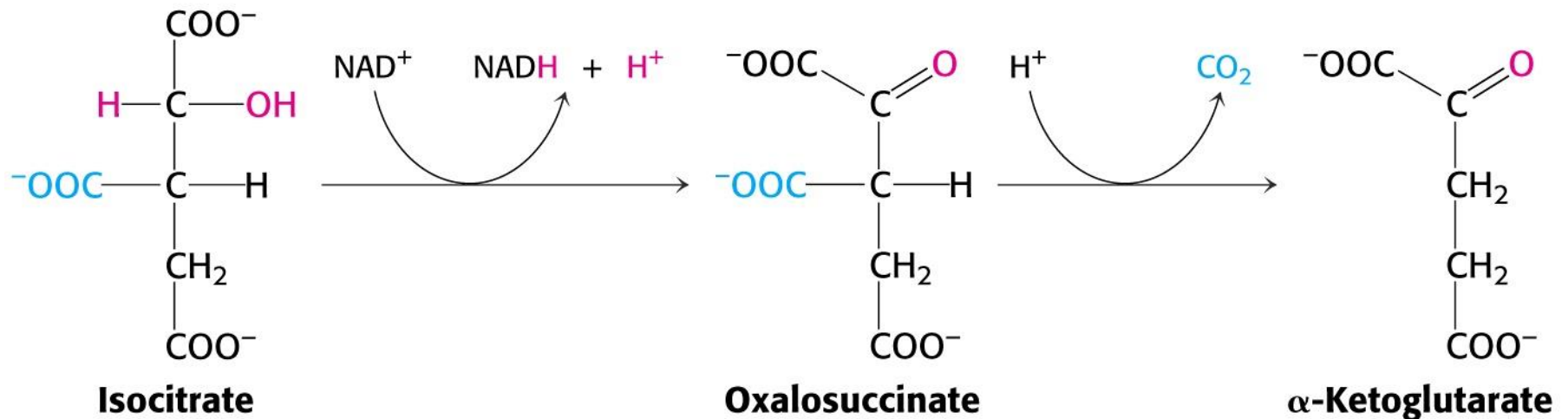


Aconitase: citrate binding to iron-sulfur cluster



Isocitrate to α -ketoglutarate: step 3

Enzyme: isocitrate dehydrogenase



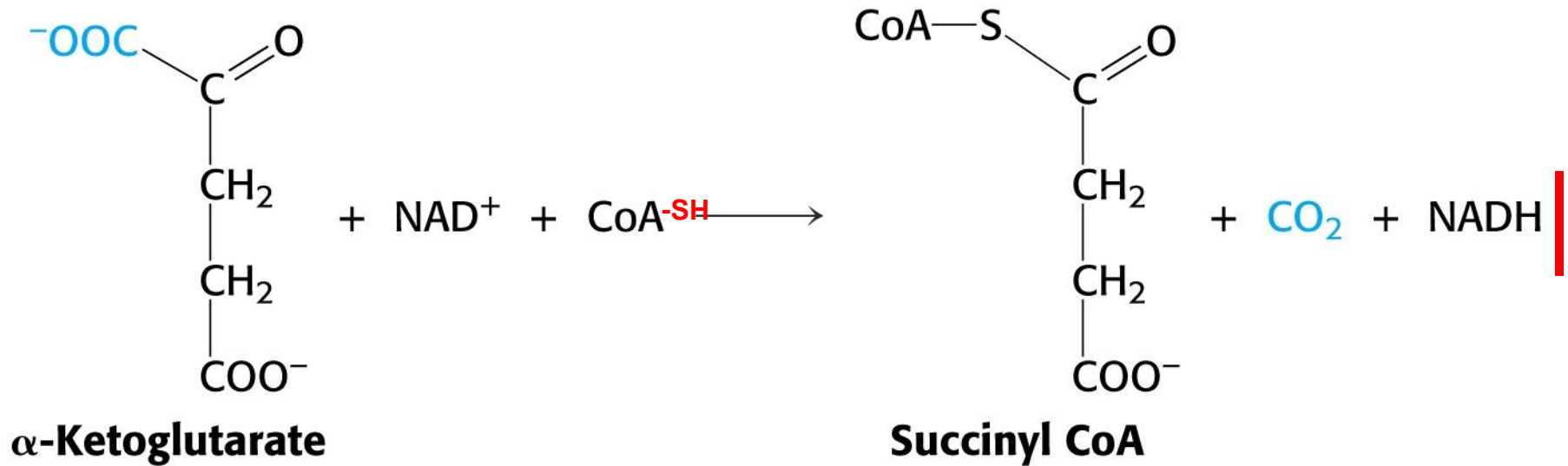
1st NADH produced

1st CO_2 removed

Oxidation of IC to aKG and CO_2 , Mn^{2+} in active site

Succinyl CoA formation: step4

Enzyme: α -ketoglutarate dehydrogenase



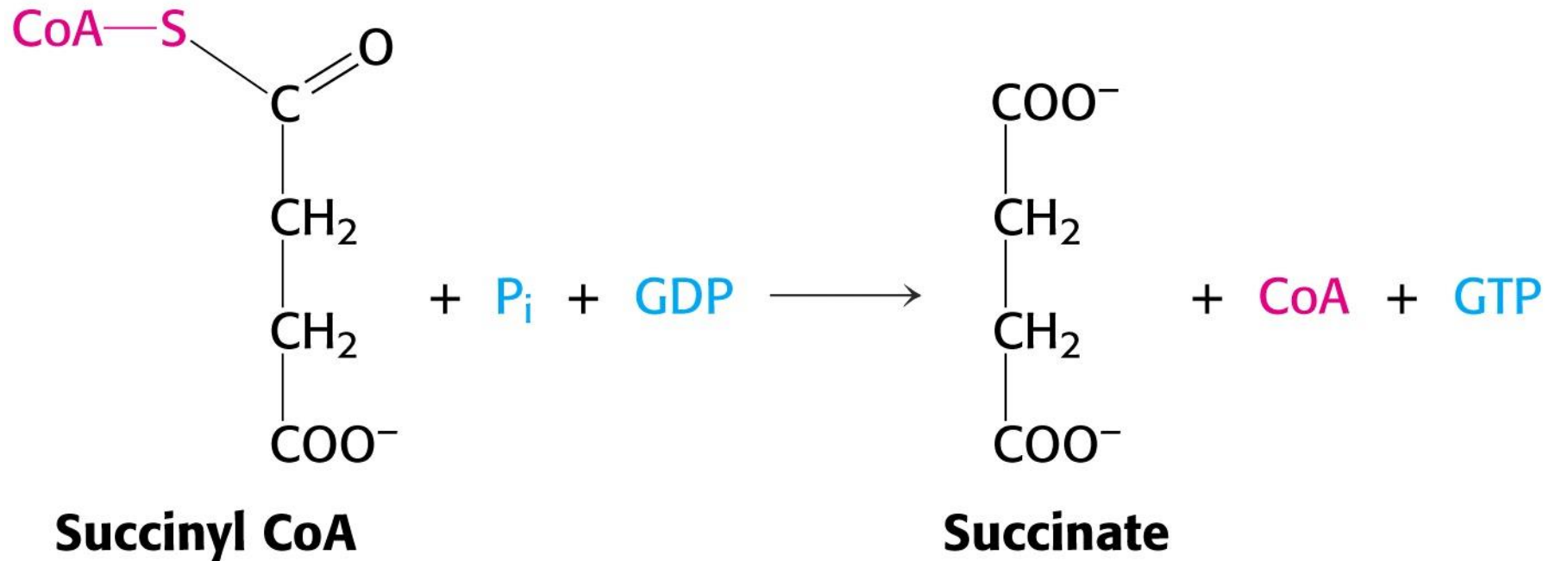
2nd NADH produced

2nd CO₂ removed

oxidation

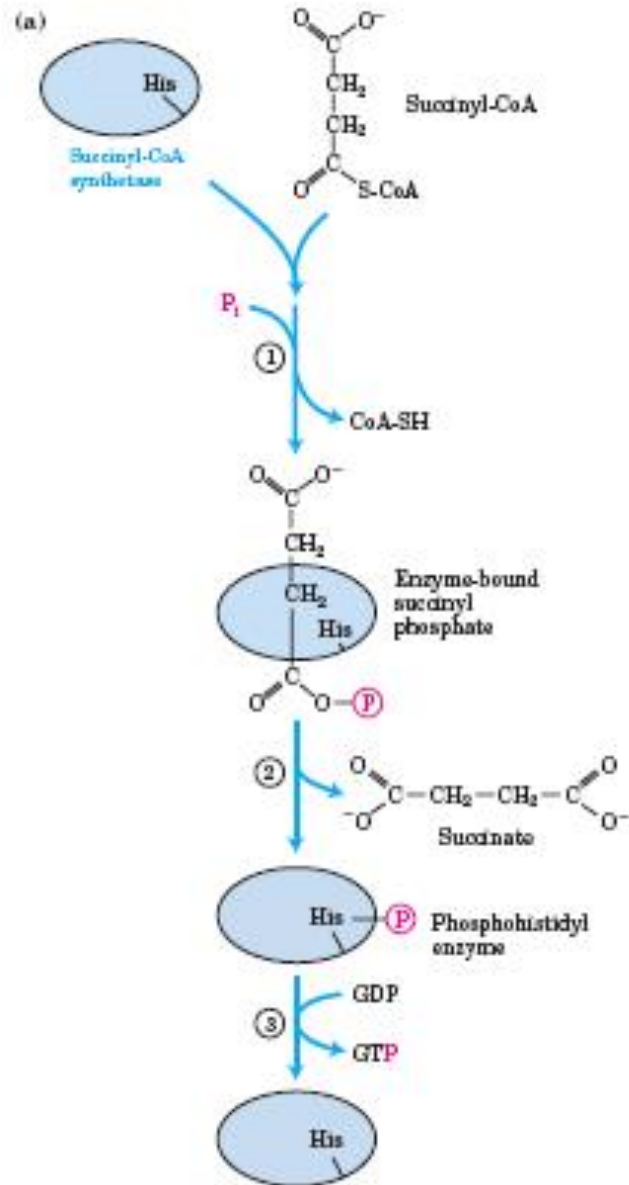
Succinate formation: step5

Enzyme: succinyl CoA synthetase

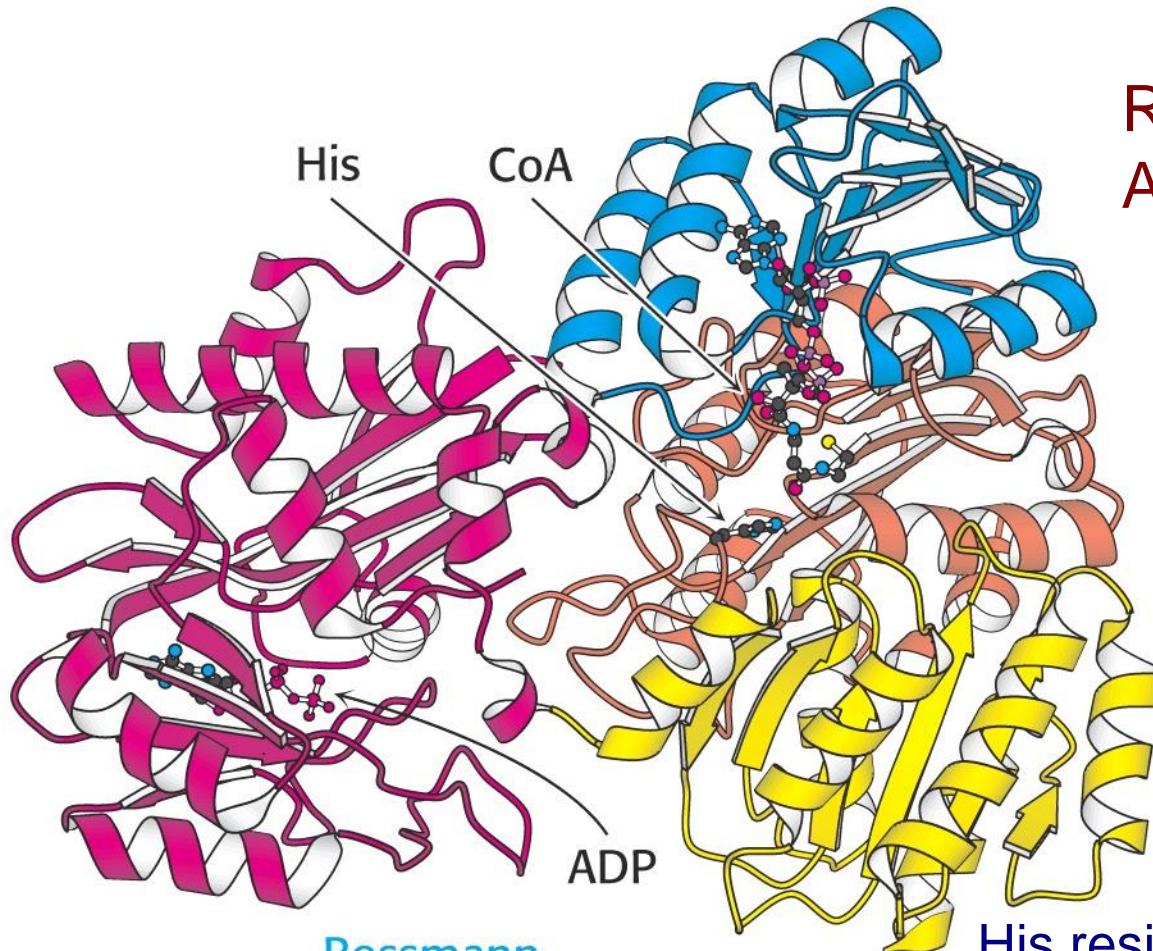


GTP produced



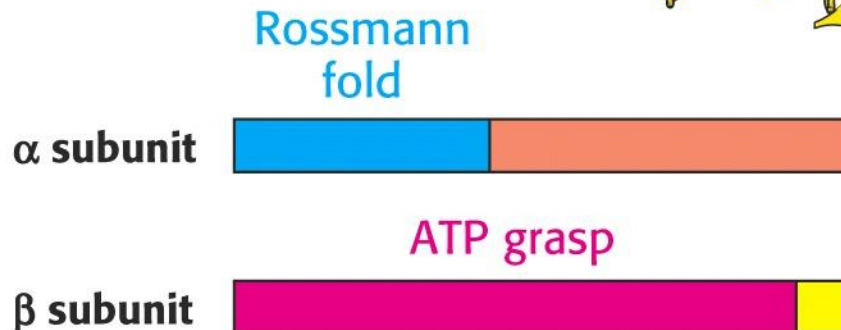


Succinyl CoA synthetase



Rossmann fold binds ADP component of CoA

ATP-grasp domain is a nucleotide-activating domain, shown binding ADP.

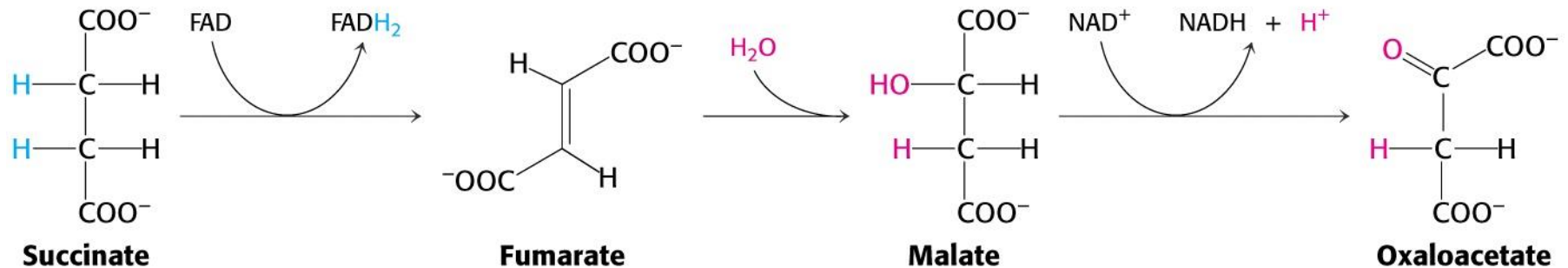


His residue picks up phosphoryl group from near CoA, & swings over to transfer it to the nucleotide bound in the ATP-grasp domain

Oxaloacetate regenerated by oxidation of succinate:

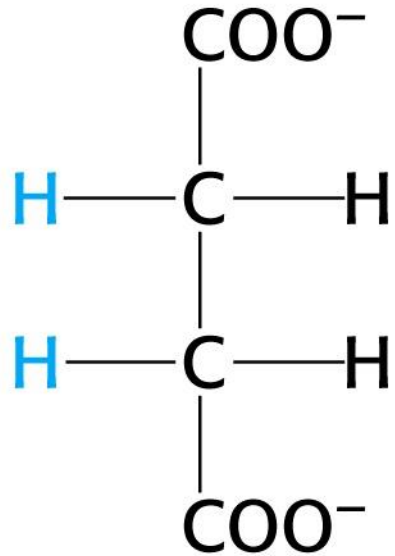
Steps 6 - 8

Oxidation, hydration, and oxidation

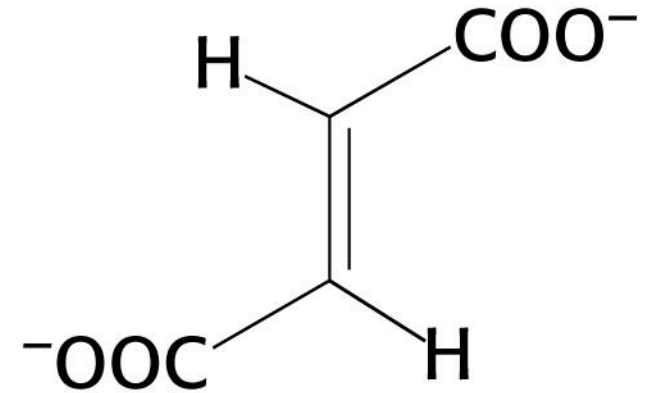
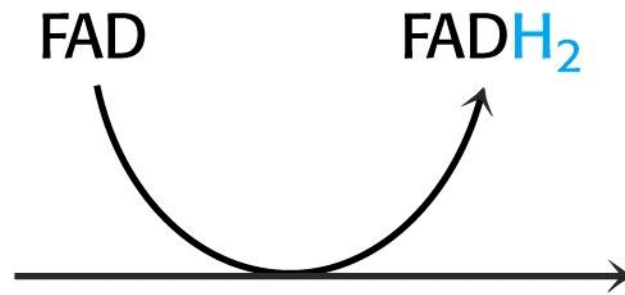


Succinate to Fumarate: step 6

Enzyme: succinate dehydrogenase



Succinate

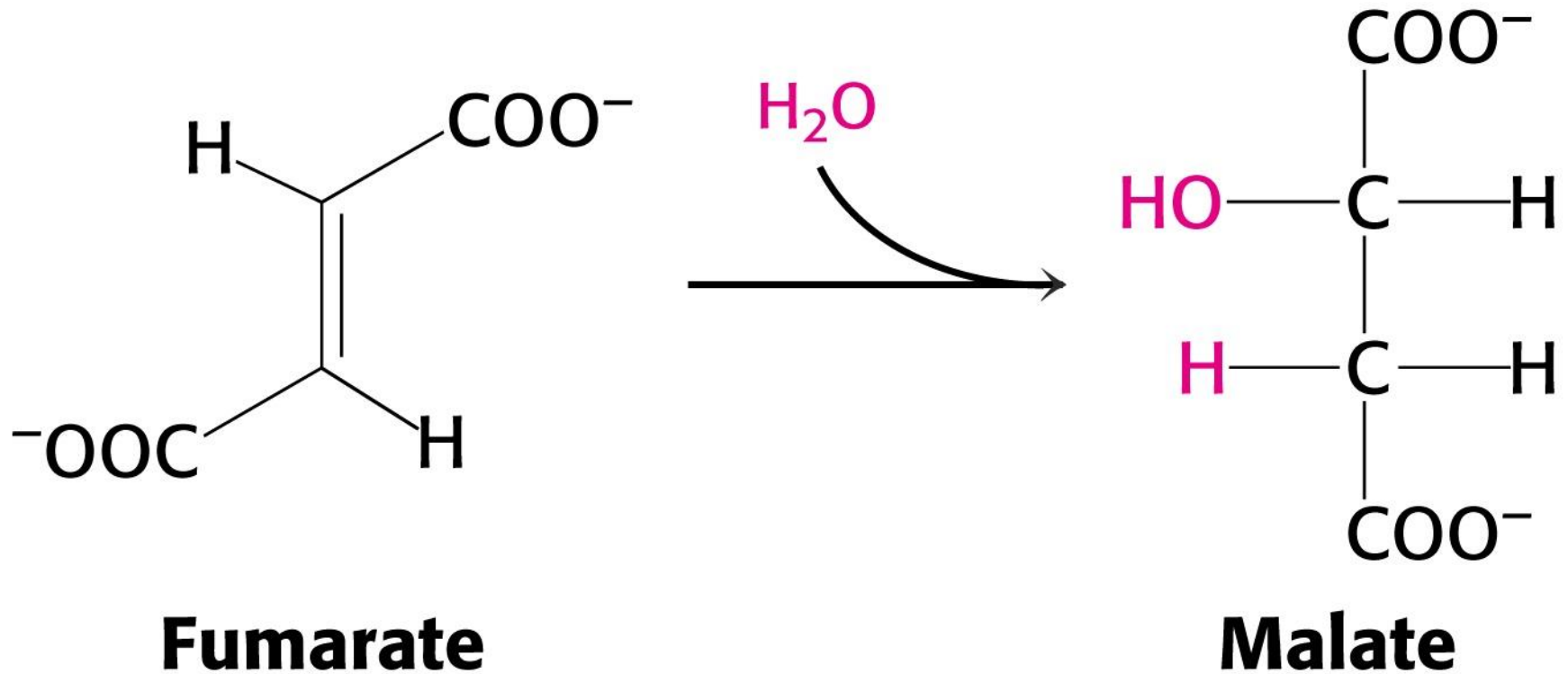


Fumarate

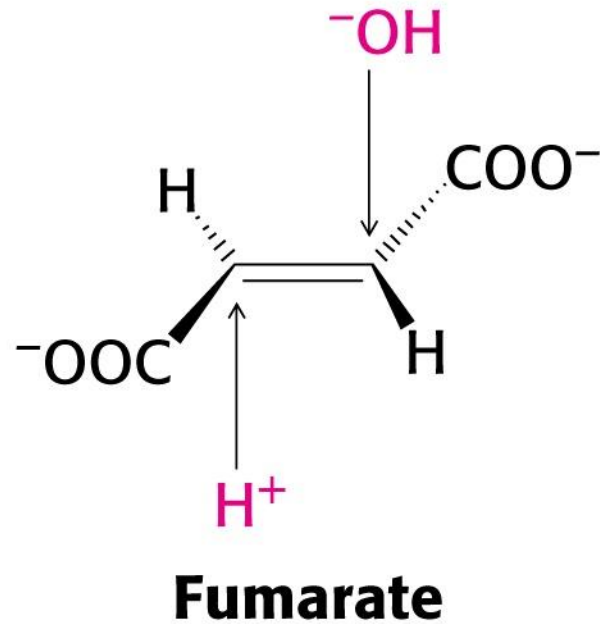
FADH₂ produced

Fumarate to Malate: step 7

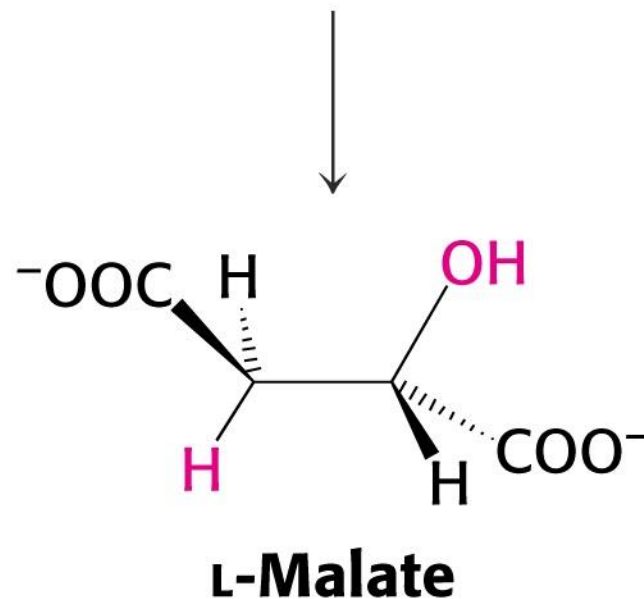
Enzyme: fumarase



Fumarate to L-Malate

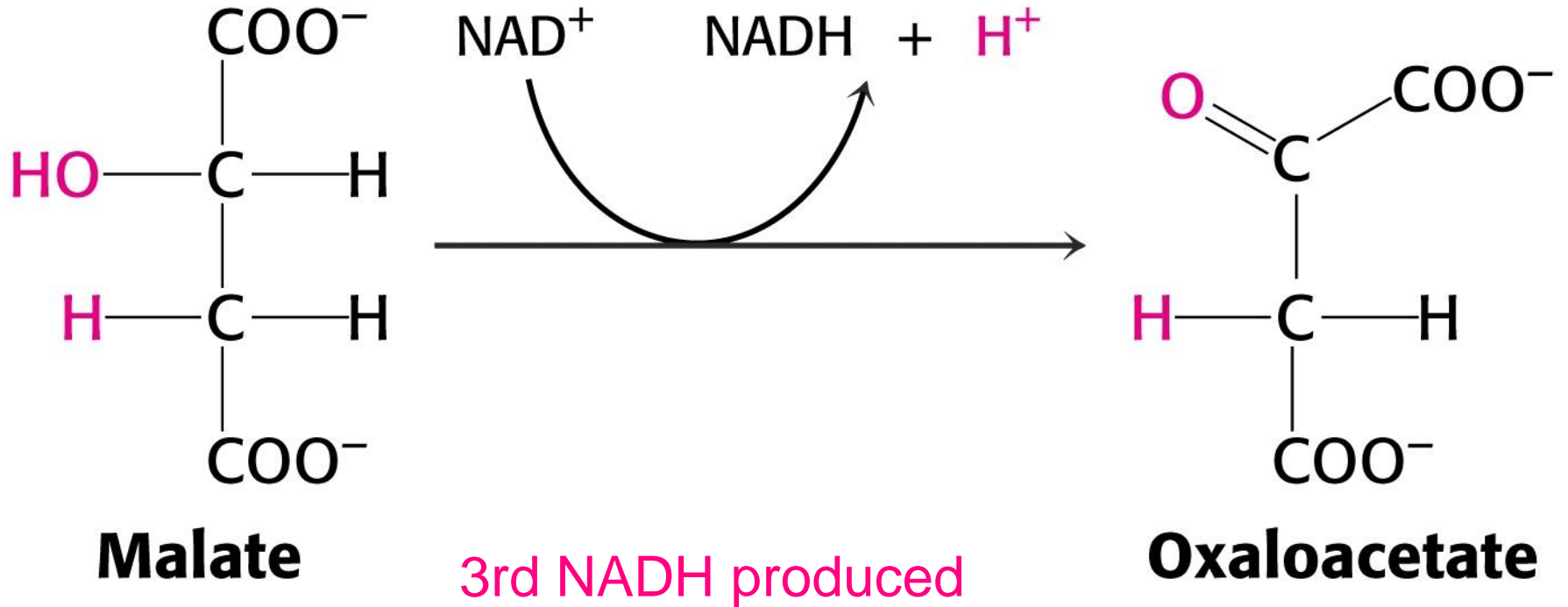


Hydroxyl group to one side only of fumarate double bond; hence, only L isomer of malate formed

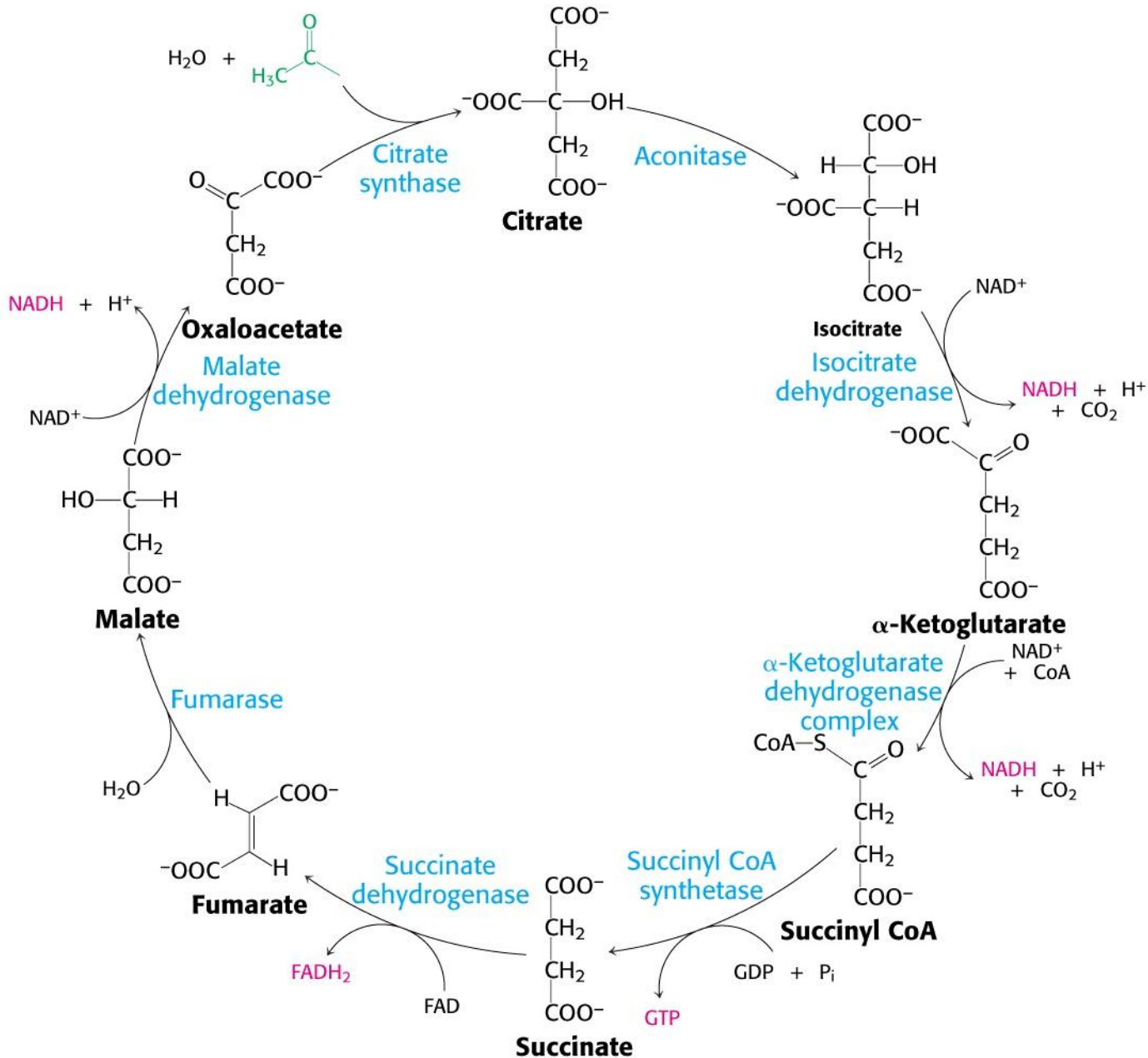


Malate to Oxalate: step 8

Enzyme: malate dehydrogenase



The citric acid cycle



Summary of 8 steps

TABLE 17.2 Citric acid cycle

Step	Reaction	Enzyme	Prosthetic group	Type*	$\Delta G^{\circ'}$	
					kcal mol ⁻¹	kJ mol ⁻¹
1	Acetyl CoA + oxaloacetate + H ₂ O \longrightarrow citrate + CoA + H ⁺	Citrate synthase		a	-7.5	-31.4
2a	Citrate \rightleftharpoons <i>cis</i> -aconitate + H ₂ O	Aconitase	Fe-S	b	+2.0	+8.4
2b	<i>cis</i> -Aconitate + H ₂ O \rightleftharpoons isocitrate	Aconitase	Fe-S	c	-0.5	-2.1
3	Isocitrate + NAD ⁺ \rightleftharpoons α -ketoglutarate + CO ₂ + NADH	Isocitrate dehydrogenase		d + e	-2.0	-8.4
4	α -Ketoglutarate + NAD ⁺ + CoA \rightleftharpoons succinyl CoA + CO ₂ + NADH	α -Ketoglutarate dehydrogenase complex	Lipoic acid, FAD, TPP	d + e	-7.2	-30.1
5	Succinyl CoA + P _i + GDP \rightleftharpoons succinate + GTP + CoA	Succinyl CoA synthetase		f	-0.8	-3.3
6	Succinate + FAD (enzyme-bound) \rightleftharpoons fumarate + FADH ₂ (enzyme-bound)	Succinate dehydrogenase	FAD, Fe-S	e	~0	0
7	Fumarate + H ₂ O \rightleftharpoons L-malate	Fumarase		c	-0.9	-3.8
8	L-Malate + NAD ⁺ \rightleftharpoons oxaloacetate + NADH + H ⁺	Malate dehydrogenase		e	+7.1	+29.7

*Reaction type: (a) condensation; (b) dehydration; (c) hydration; (d) decarboxylation; (e) oxidation; (f) substrate-level phosphorylation.

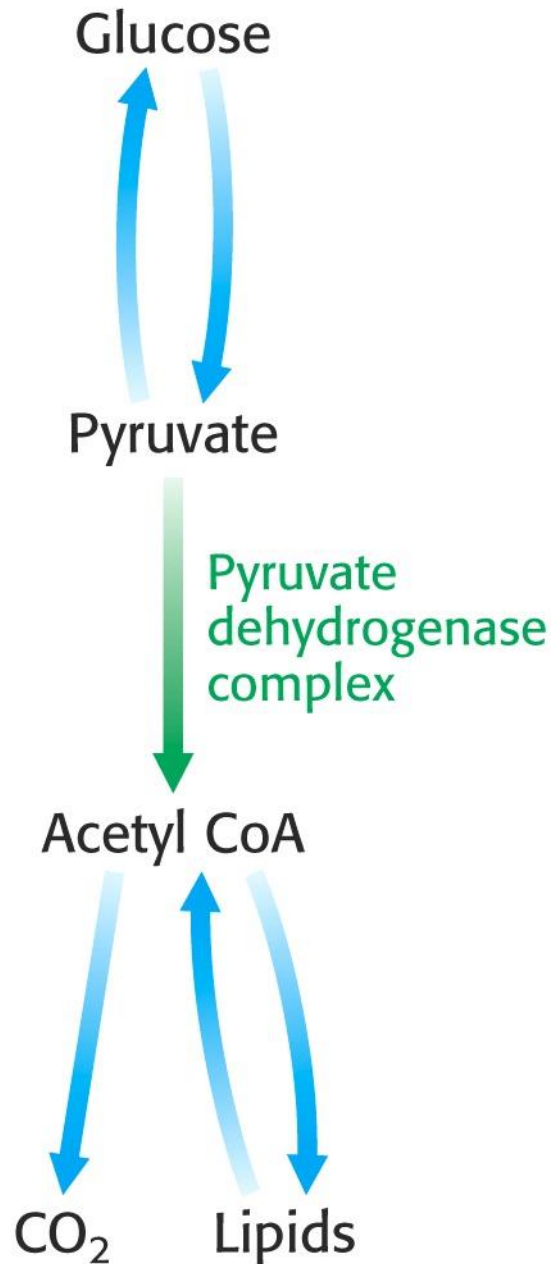
Proton gradient generates 2.5 ATP per NADH, & 1.5 per FADH₂

9 ATP from 3 NADH + 1 FADH₂. Also, 1 GTP

Thus, 1 acetate unit generates equivalent of 10 ATP molecules.

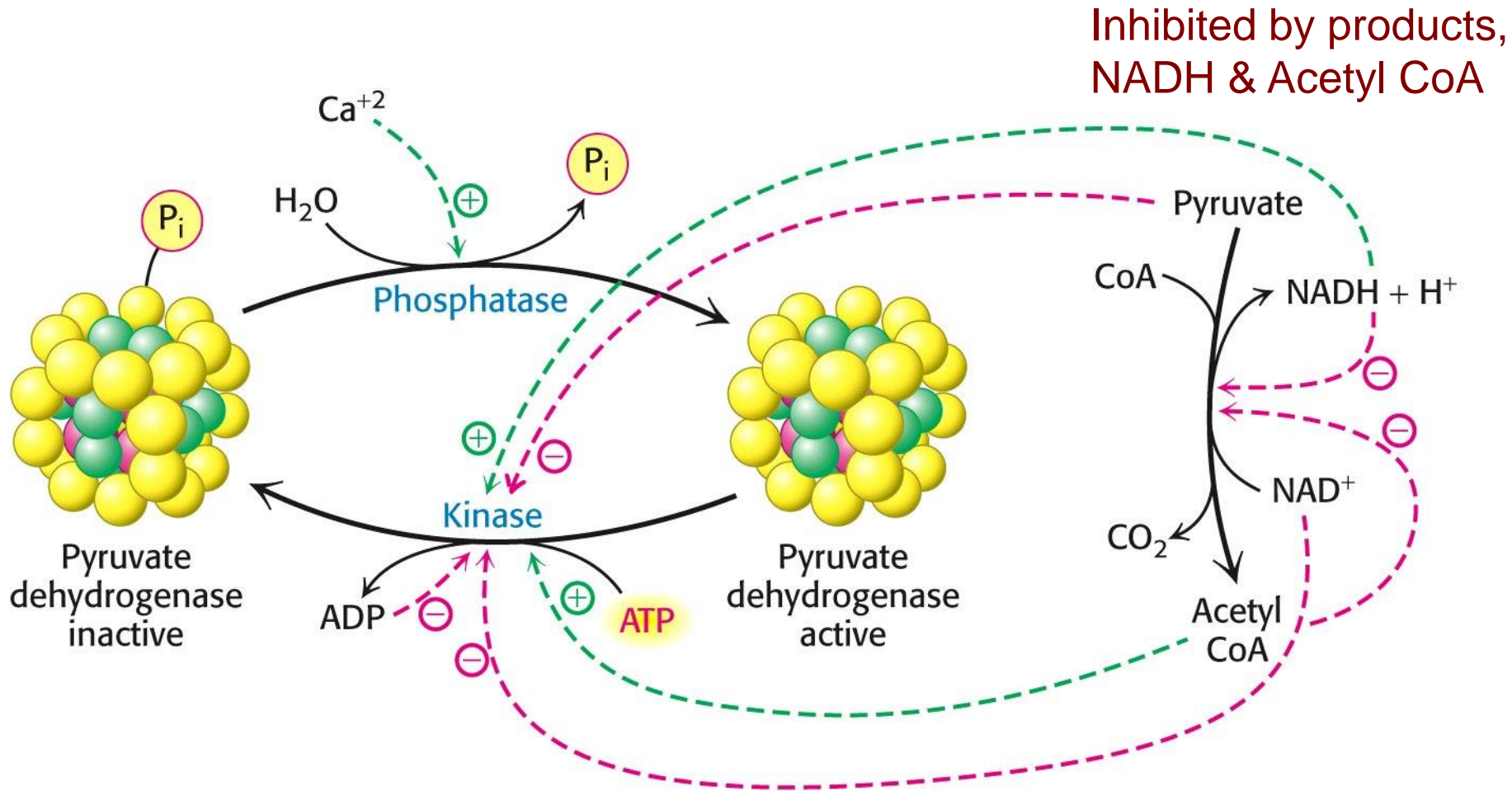
In contrast, 2 ATP per glucose molecule in anaerobic glycolysis

Pyruvate to Acetyl CoA, irreversible



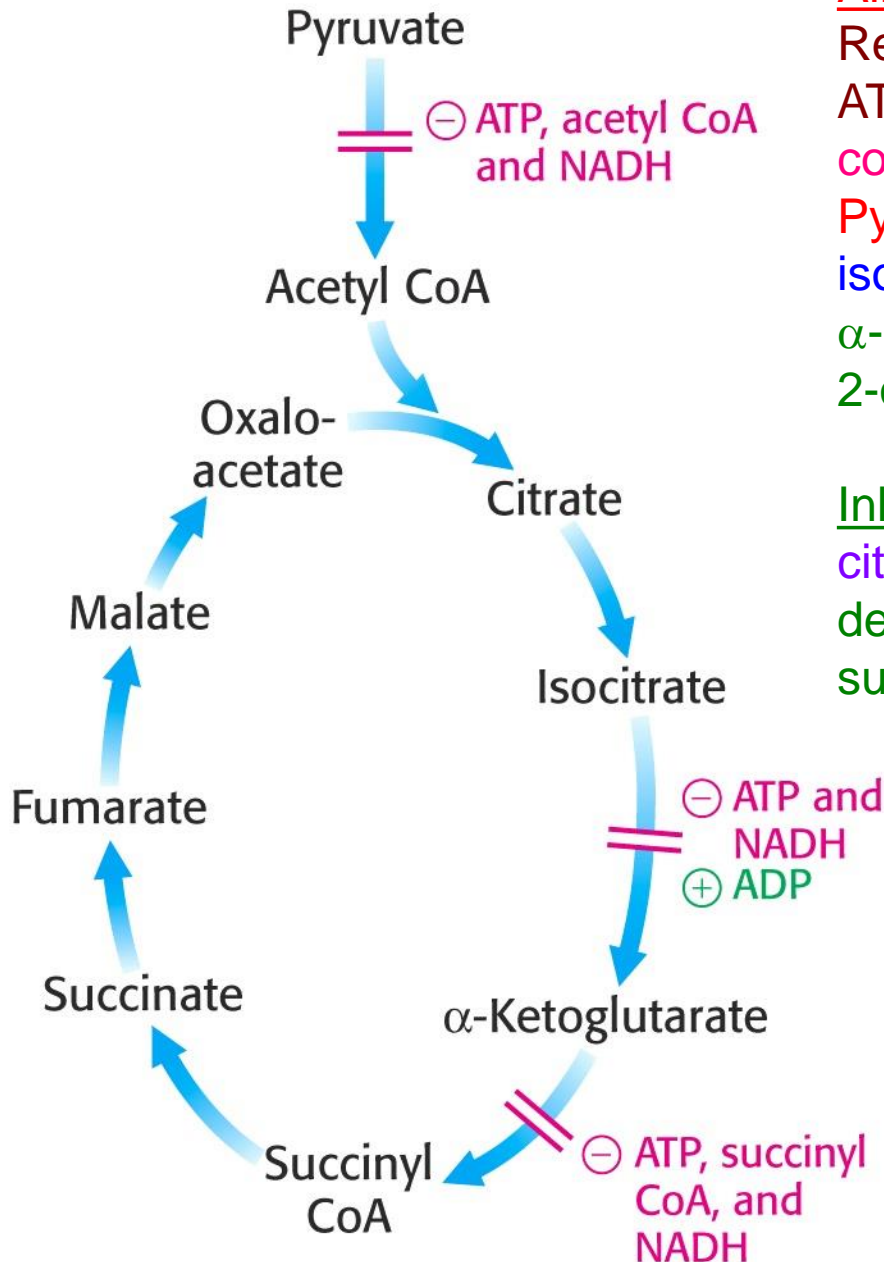
Key irreversible step
in the metabolism of
glucose

Regulation of pyruvate dehydrogenase



Also regulated by covalent modification,
the kinase & phosphatase also regulated

Control of citric acid cycle



Allosteric regulation

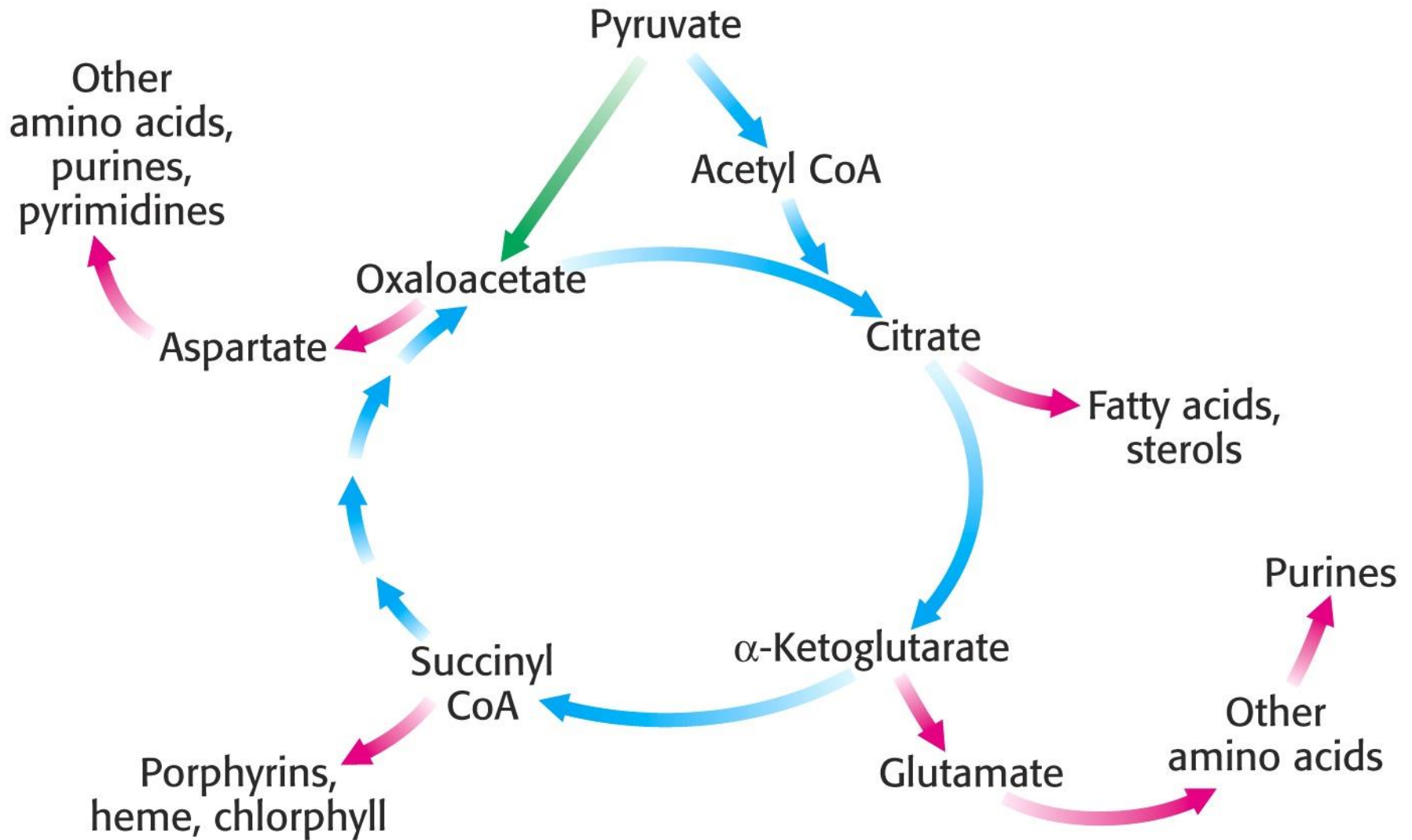
Regulated primarily by ATP & NADH concentrations, control points:

Pyruvate dehydrogenase
isocitrate dehydrogenase &
 α -ketoglutarate dehydrogenase
(2-oxo-glutarate dehydrogenase)

Inhibition by product:

citrate synthase- citrate- 2-oxo-glutarate
dehydrogenase- succinyl CoA

Biosynthetic roles of the citric acid cycle



Key Enzymes for regulation of CAC, inhibitors and activators

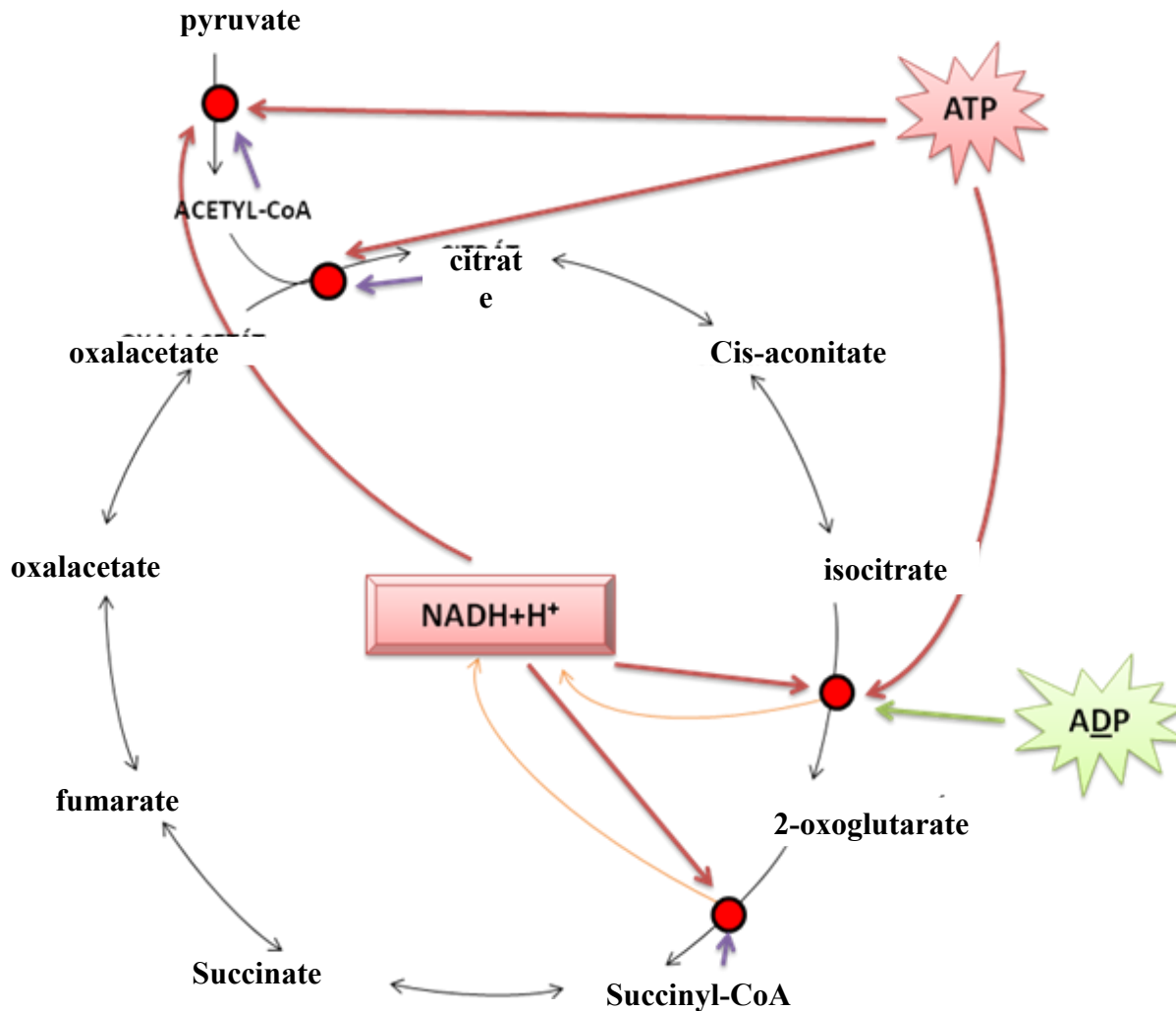
Enzyme	ATP ^a	NADH ^a	different
Pyruvate dehydrogenase	-	-	- acetyl-CoA (inh. prod.)
Citrate syntethase	-		- citrate (inhibition by product)
Isocitrate dehydrogenase	-	-	+ ADP (allosteric activation)
2-OG-dehydrogenase		-	- sukcinyl-CoA (inh. prod.)

^a allosteric inhibitor

^b feedback inhibitor (inhibition by reaction product)

^c allosteric activator

Regulation of CAC



NOVÁK, Jan. *Biochemie I.* Brno: Muni, 2009, s. 238.

