# Modeling of geochemical processesGlobal Systems

J. Faimon

### Global systems

Reservoirs and fluxes



Flux into reservoir,  $j_{in}$ , flux out from reservoir,  $j_{out}$ , reservoir content **n**. **n** [ton, kg, mol, mol/l ...],  $\mathbf{j}$  [ton/year, mol/day ...]

Assumption: the flux from the reservoir is directly proportional to concentration or reservoir content. k is a constant.

Example: the flux of sulfates from ocean to sediments is proportional tosulfate content in ocean. **Example:** a photosynthesis rate is proportional to  $CO_2$ -content in atmosphere

If  $\mathbf{j}_{\text{in}} \sim \mathbf{a} = \text{const.}$ , it si valid for the reservoir content  $\mathbf{n}: \quad +\frac{d\mathbf{n}}{dt} = \mathbf{a} - \mathbf{k}\mathbf{n}$ dn $+\,-$ =−Steady state: reservoir content is constant  $+ \frac{d}{dx} = 0$ Then  $\mathbf{a} - \mathbf{k} \mathbf{n} = \mathbf{0}$  and  $\mathbf{n} = \mathbf{n}_{ss} = \frac{1}{k}$ dtdn $+ \longrightarrow =$ a $n = n_{ss} = =n_{\rm \, ss}=$ The solution of the differential equation  $(t = 0, n = n_0):$   $n = \frac{a}{k} - \left(\frac{a}{k} - n_0\right)e^{-kt}$ ne $k \qquad \qquad$ a $k$ a $\mathsf{n}=$  - $\int$   $\setminus$  $=\frac{a}{a}$ reorganizing gives:  $\ln n_0 e^{-kt} + \frac{a}{k} (1 - e^{-kt})$ ( $\bigg)$ 1e $k^{\vee}$ a $n = n_0 e^{-\alpha}$ ne $=$   $\ln 6$  +  $\ln -c$ +

Interpretation:

(1) Initial content of element  $n_0$  is transformed into steady state content  $n_{ss} = a/k$ , with decrease of the exponential term  $e^{-kt}$  with time. In time t = 0 is <sup>-kt</sup> with time. In time  $t = 0$  is  $e^{-kt} = 1$ <sup>−⊾</sup>ี (2) Initial content of element  $n_0$  decays in t =  $\infty$ is time!  $\frac{a}{k}$   $\left(1-e^{-kt}\right)$  $\infty$   $(e^{-kt} = 0)$ <sup>−κι</sup> = ()e $k^{\vee}$ athe second term is  $a/k$  at this time!  $\frac{a}{k}$  (1-e<sup>-1</sup>)

Infinite time is needed for reaching the steady state.

However, significant decrease of the exponential term $m e^{-kt}$ e

is reached at 
$$
t = \frac{1}{k}
$$
 where  $e^{-kt} = \frac{1}{e} = 0.3679$ 

This time is a response time

The residence time is given by

$$
\tau_{res} = \frac{n_{ss}}{j_{in}} = \frac{n_{ss}}{j_{out}}
$$
 Substitution gives  $\tau_{res} = \frac{a/k}{a} = \frac{1}{k}$ 

For simple linear model, response time equals residence time

#### Two-reservoir model



The increments in reservoir contents are expressed by differential equations

$$
+\frac{dx_1}{dt} = -k_{12}x_1 + k_{21}x_2 + \frac{dx_2}{dt} = k_{12}x_1 - k_{21}x_2
$$

In matrix form, it is $\overline{\phantom{a}}$  $\overline{\phantom{a}}$  $\overline{\phantom{a}}$   $\overline{\phantom{a}}$   $\overline{\phantom{a}}$  $\sqrt{2}$ −− $\vert$  $\overline{\phantom{a}}$  $\overline{\phantom{a}}$  $\sqrt{2}$ 21 $12 \qquad \qquad \mathbf{\Lambda}_{21}$  $12 \qquad \qquad \mathbf{\Lambda} 21$ 21 $\rm X$  $\bf{X}$  $k_1$   $- k_2$ kk $\bf X$  $\bf{X}$  $dt$  : d

In vector form, it is K $\mathbf X$ x $- =$  dtd $\frac{\partial u}{\partial t} = \mathbf{K} \mathbf{x}$ , where **x** is a vector of variables **x**<sub>i</sub> and K is matrix of rate constants