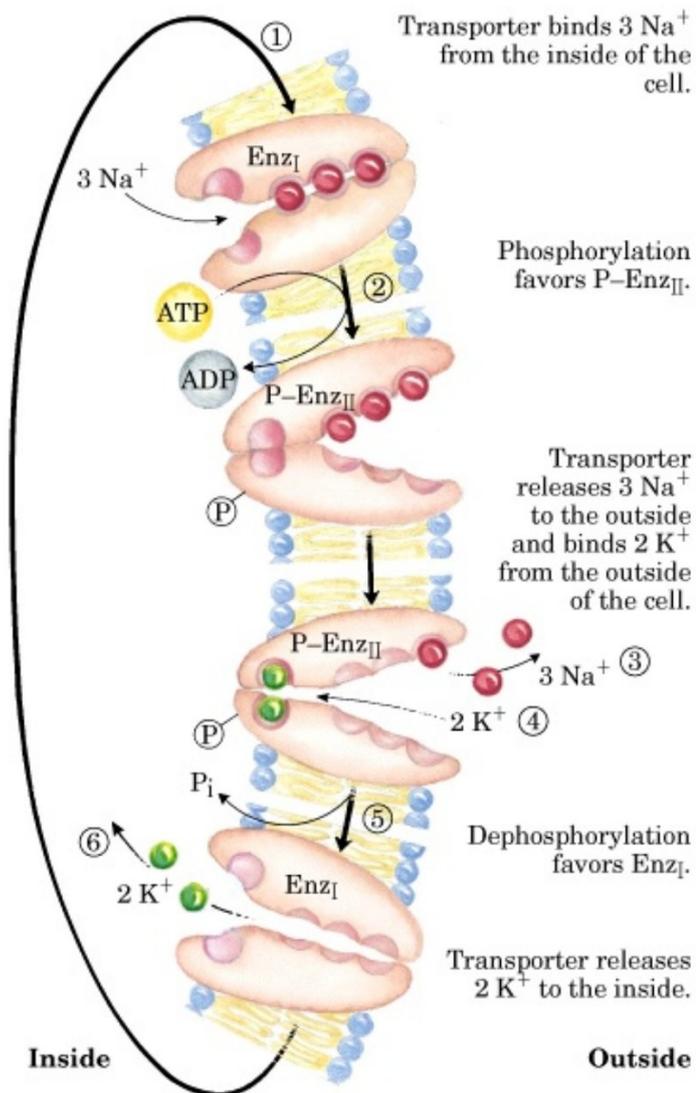


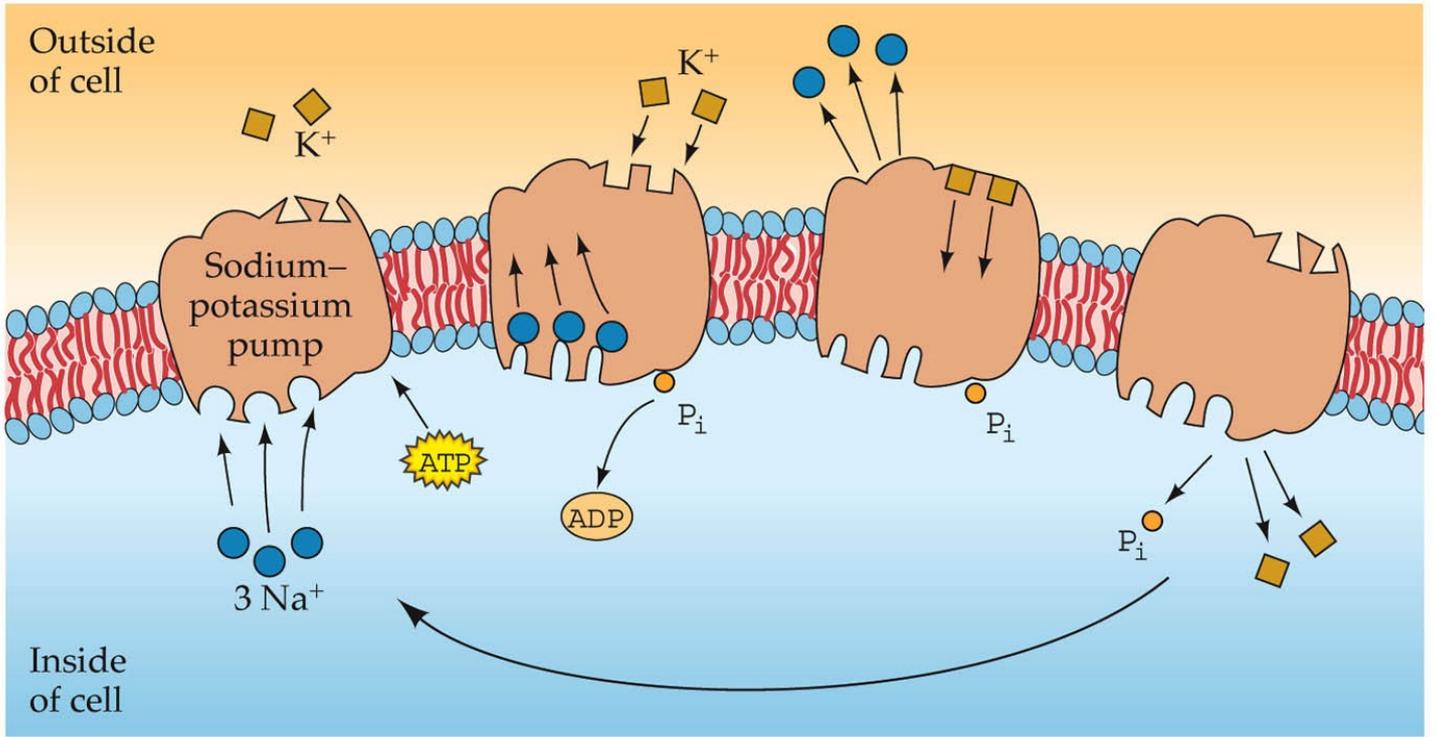
29. Regulace metabolismu a membránový transport

Významné transportní systémy

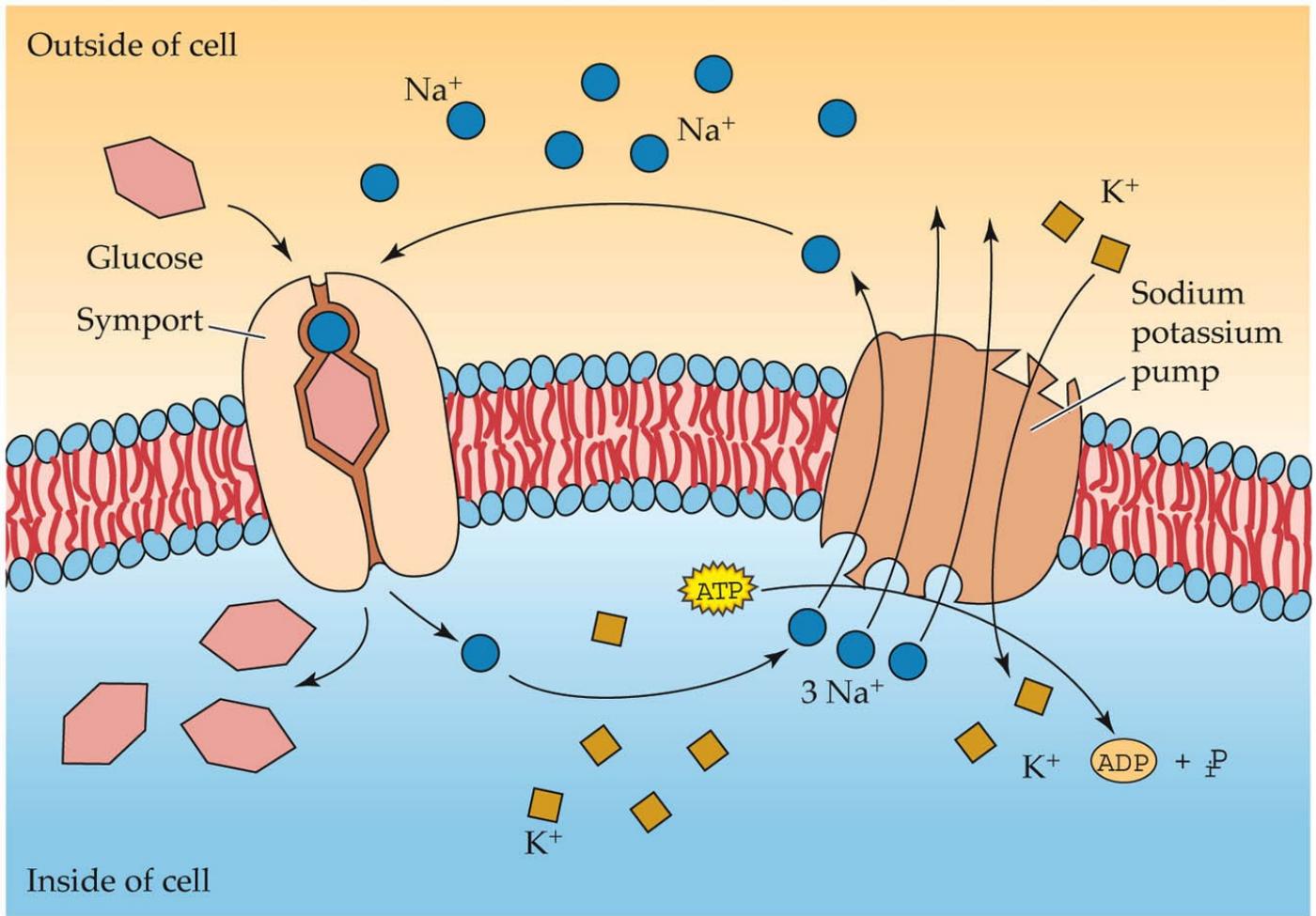
NaK - ATPasa

Mechanism of the Na-K transporter





LIFE: THE SCIENCE OF BIOLOGY, Seventh Edition, Figure 5.13 Primary Active Transport: The Sodium-Potassium Pump
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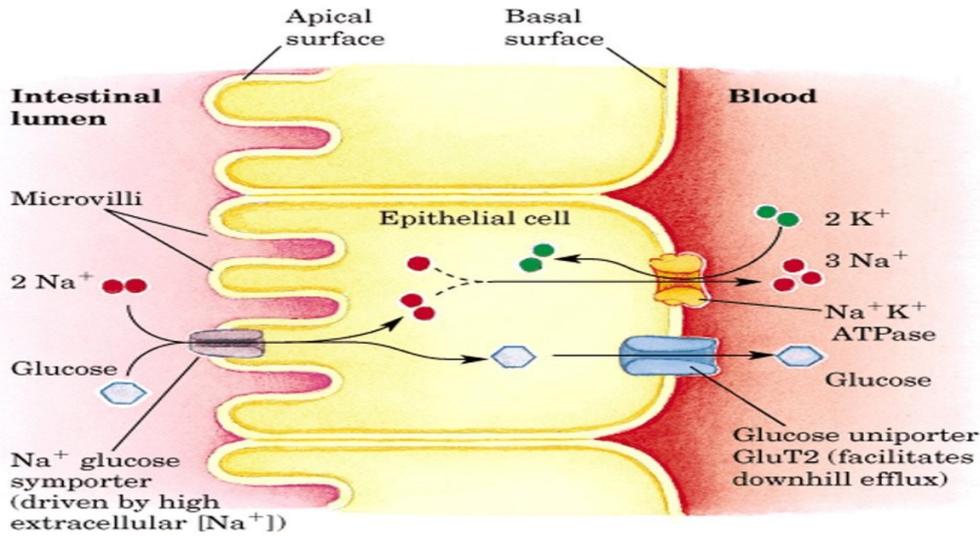


LIFE: THE SCIENCE OF BIOLOGY, Seventh Edition, Figure 5.14 Secondary Active Transport

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Transport glukosy

Na⁺/glucose symporter



For charged ion transport: $\Delta G = RT \ln \frac{[Na^+]_{in}}{[Na^+]_{out}} + n F \Delta E$

$$= 5.7 \text{ KJ} \log \frac{[Na^+]_{in}}{[Na^+]_{out}} + n \times 96.5 \Delta E$$

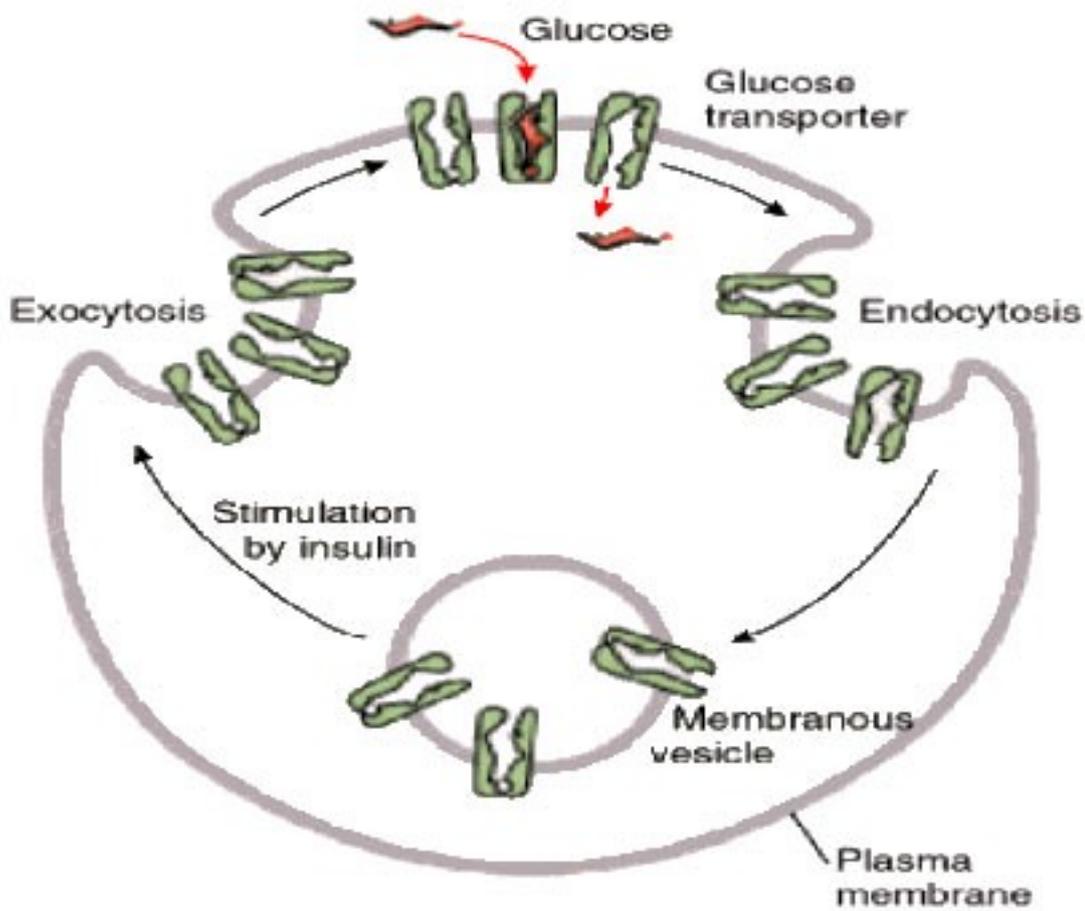
$$\text{Moving a Na}^+ \text{ ion into the cell releases } = 5.7 \log(12/145) + 1 \times 96.5 \times (-0.05) \\ = -6.2 \text{ KJ/mol} - 4.8 \text{ KJ/mol} = -11 \text{ KJ/mol}$$

If two Na⁺ ions move, energy available to pump glucose is -22 KJ/mol.

This energy could transport glucose against a concentration gradient; it's magnitude would be governed by the available energy:

$$\Delta G = 22 \text{ KJ/mol} = 5.7 \text{ kJ/mol} \times \log \frac{[\text{glucose}]_{in}}{[\text{glucose}]_{out}}$$

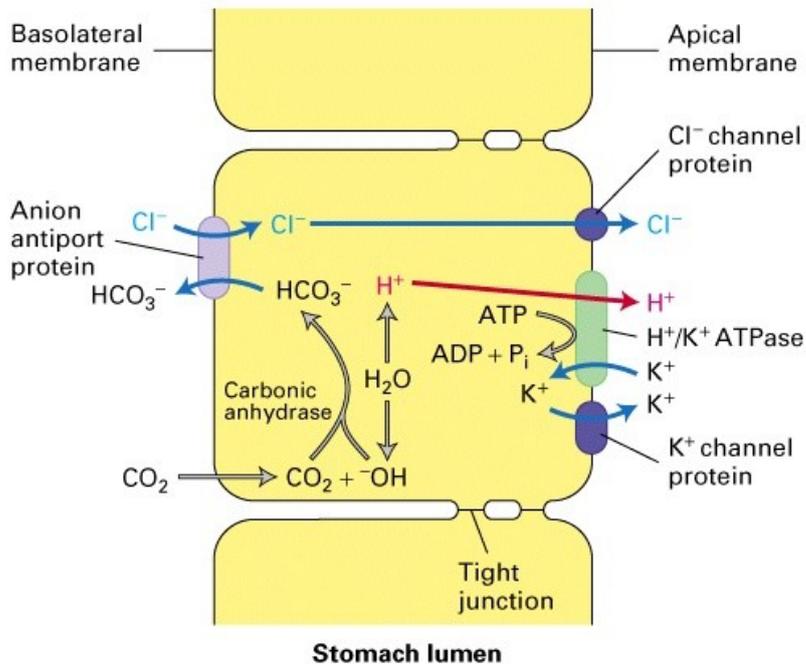
$$3.86 = \log \frac{[\text{glucose}]_{in}}{[\text{glucose}]_{out}} \text{ therefore } \frac{[\text{glucose}]_{in}}{[\text{glucose}]_{out}} = 7000.$$



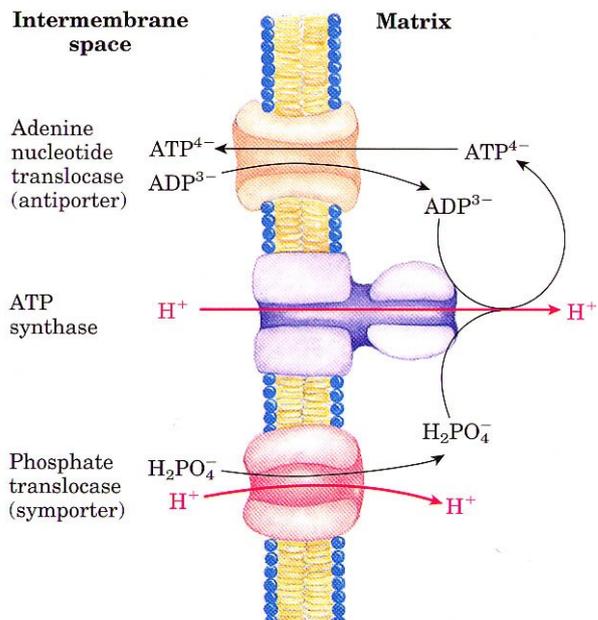
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REGULACE příjmu glukosy insulinem

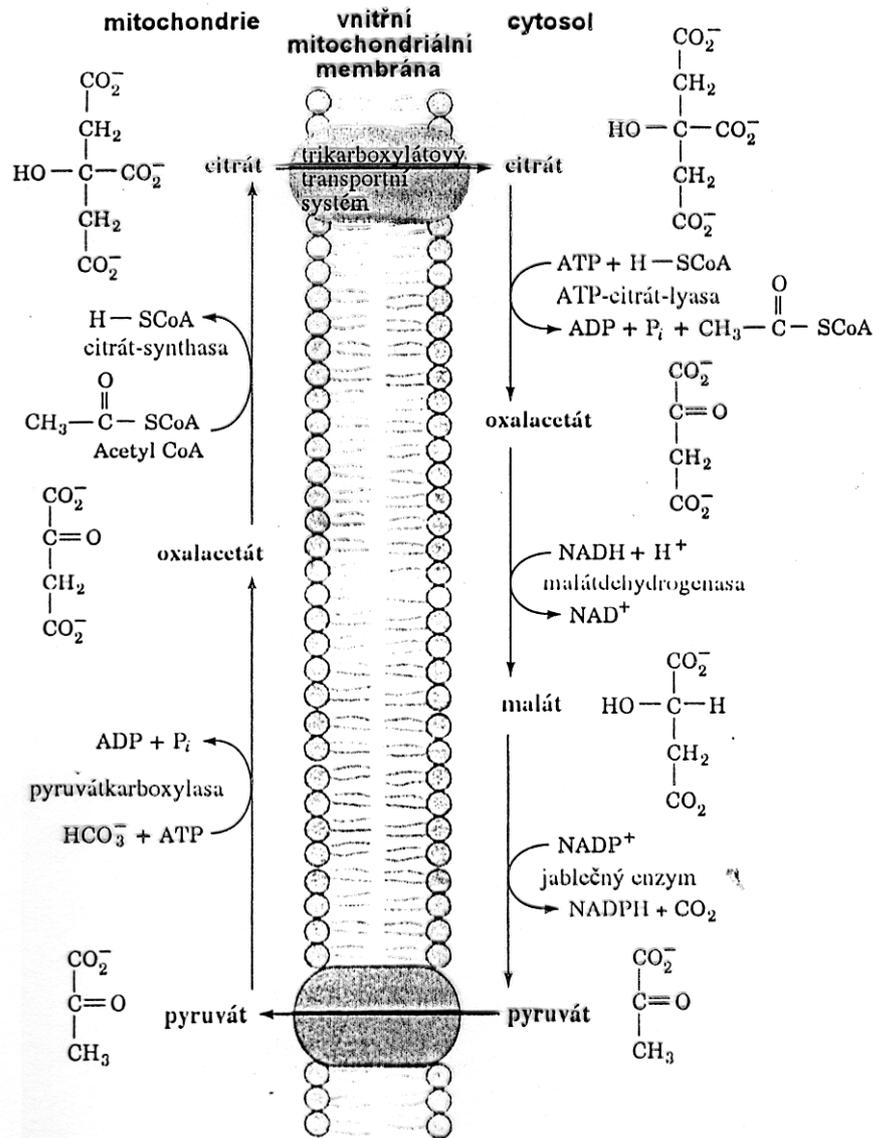
Transport HCl



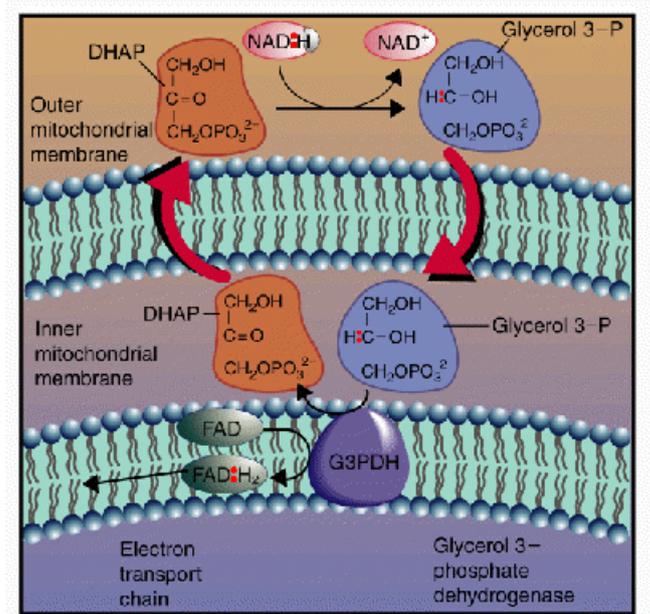
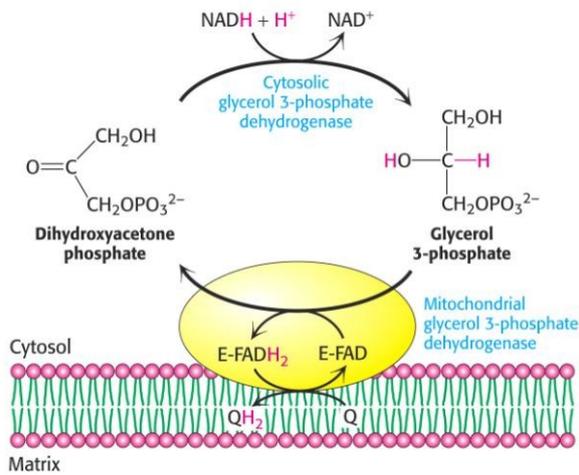
Mitochondriální transporty



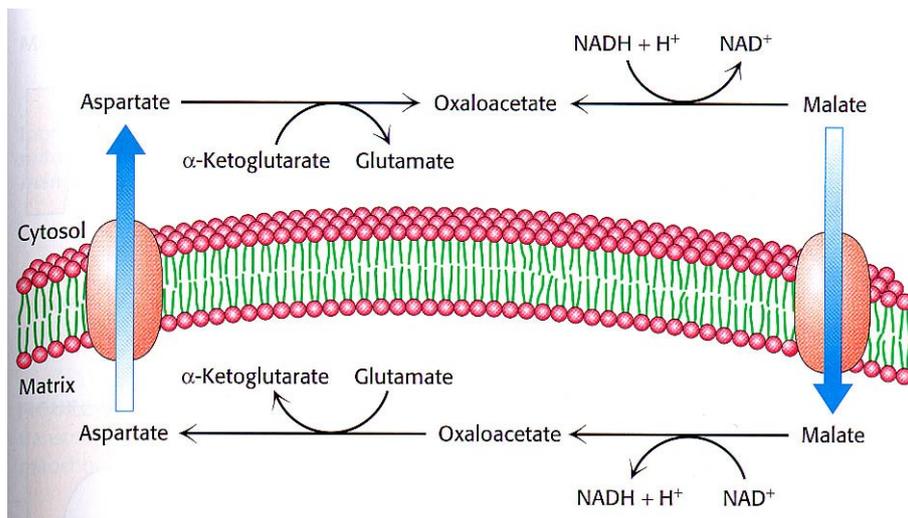
Přenos AcetCoA vně mitochondrie



Oxidoredukční člunky – problém oxidace cytosolického NADH



GLYCEROLFOSFÁTOVÝ člunek



Malate-aspartate shuttle

