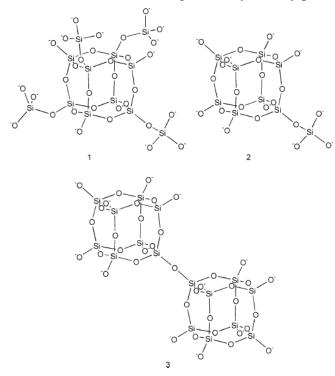
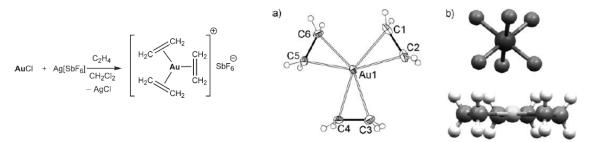
HW 2	Multinuclear NMR	Name:	
Points:	C6800	Date:	
Max. 100 points	Spring 2011	Version A	

1. Predict the number and relative intensity of the signals in the 29 Si NMR spectra of molecules 1–3. For 1 and 3 give the symmetry point groups.



2. Gold(I) tris(ethylene) complex $[Au(C_2H_4)_3][SbF_6]$ was prepared by the following reaction and was structurally characterized.



a) Draw clearly all symmetry elements present in the cation.

b) Give the symmetry point groups of the cation and the anion.

c) Is this complex diamagnetic or paramagnetic?

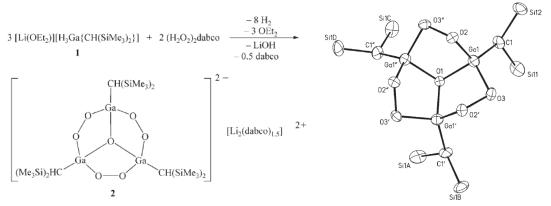
d) How many resonances (signals) do you expect in the ¹H and ¹³C NMR spectra.

e) These resonances are significantly upfield relative to the free ethylene (5.28 / 123.2 ppm). What causes this shielding effect?

	$\Delta\delta(^{13}C) = \delta(^{13}C)_{\text{complex}} - \delta(^{13}C)_{\text{ethylene}}$
$[Ag(C_2H_4)_3] [Al\{OC(CF_3)_3\}_4]$	-7.0 ppm
$[Au(C_2H_4)_3][SbF_6]$	-30.7 ppm

f) What are the ¹³C NMR chemical shifts of the two complexes?

3. Alkyl gallium peroxide anion was synthesized by the following procedure:

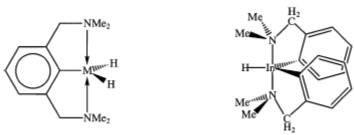


a) Draw clearly all symmetry elements present in the anion.

b) Give the symmetry point group of the anion.

c) Are there any geminal groups? Assign them as Homo-, enantio-, or diastereotopic.
d) How many resonances (signals) do you expect in the ¹H, ¹³C, and ²⁹Si NMR spectra.

4. How many signals of the NMe₂ and CH₂ groups do you expect in the following compounds:



5. How many signals do you expect in ¹⁹F a ³¹P NMR spectra of the following compound:

