

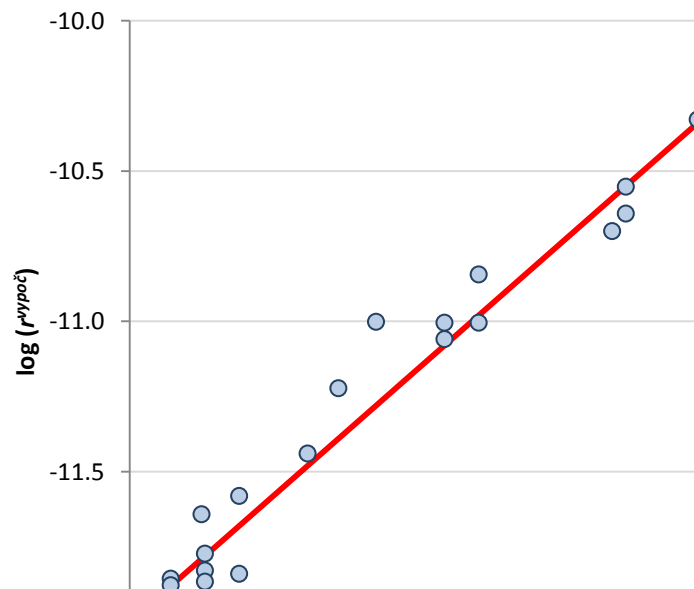
$$r = k_{\text{H}^+} a_{\text{H}^+}^n + k_{\text{OH}^-} a_{\text{OH}^-}^m$$

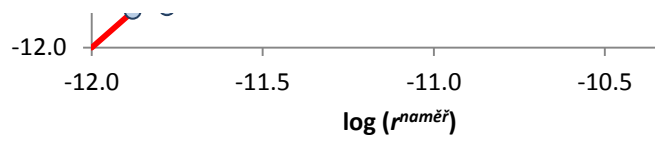
kH =	137.2	1.372E-10
n	0.3898	0.3898
kOH =	156.0	1.560E-10
m	0.2981	0.2981

Brantley 214

pH	log r	r	r*10 <sup>-12</sup>	D2	
1.10	-10.30	5.01E-11	50.12	51.14366	1.050495
2.15	-10.59	2.57E-11	25.70	19.96564	32.92829
2.93	-11.28	5.25E-12	5.25	9.968802	22.28527
3.08	-11.08	8.32E-12	8.32	8.731362	0.171168
3.51	-11.39	4.07E-12	4.07	5.993017	3.683382
4.10	-11.48	3.31E-12	3.31	3.63499	0.104768
5.17	-11.78	1.66E-12	1.66	1.688275	0.000823
5.43	-11.78	1.66E-12	1.66	1.483708	0.030933
5.49	-11.68	2.09E-12	2.09	1.44701	0.412531
5.59	-11.88	1.32E-12	1.32	1.393904	0.005722
5.66	-11.78	1.66E-12	1.66	1.362478	0.088274
5.76	-11.88	1.32E-12	1.32	1.325431	5.15E-05
7.76	-11.79	1.62E-12	1.62	2.282557	0.436586
7.99	-11.68	2.09E-12	2.09	2.626598	0.288694
9.98	-11.08	8.32E-12	8.32	9.899349	2.501811
9.98	-10.98	1.05E-11	10.47	9.899349	0.327111
10.52	-10.98	1.05E-11	10.47	14.32633	14.86137
11.20	-10.55	2.82E-11	28.18	22.83628	28.59627
11.50	-10.55	2.82E-11	28.18	28.05517	0.016554
12.25	-10.34	4.57E-11	45.71	46.93957	1.514755
					109.3049

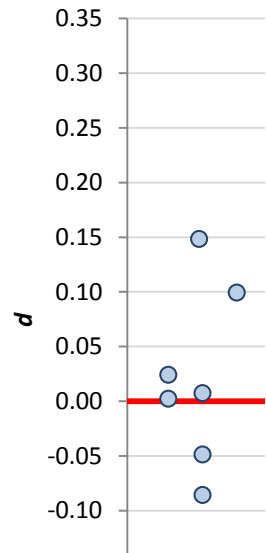
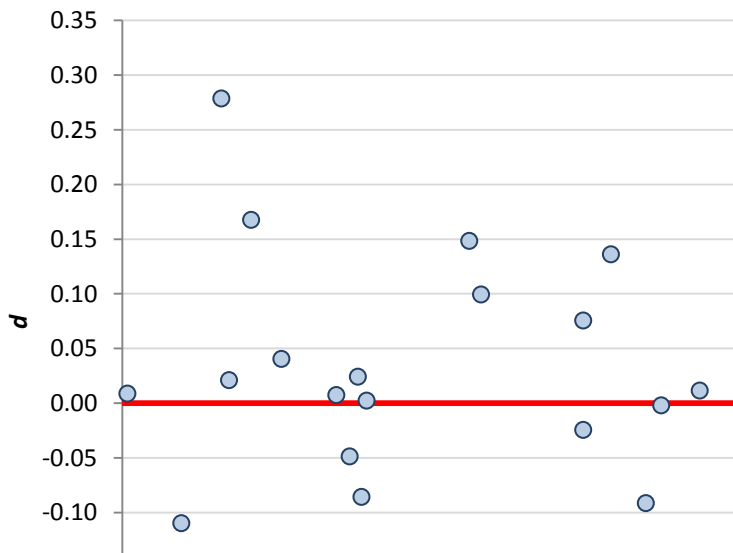
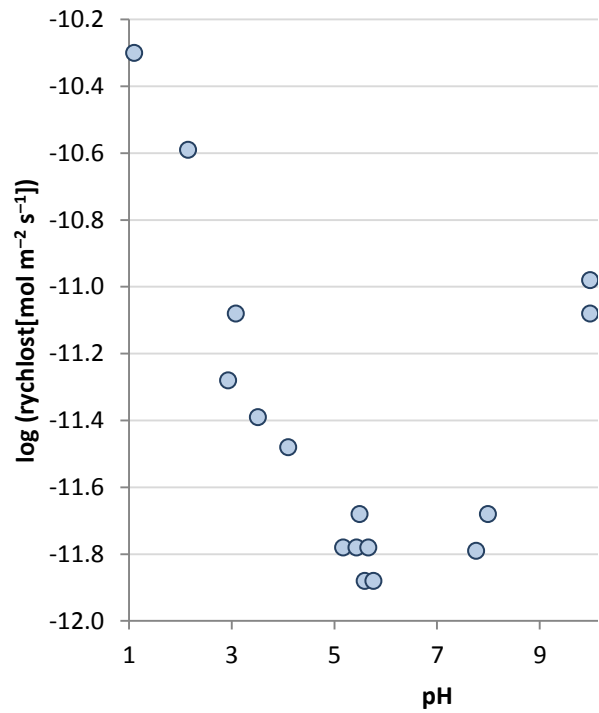
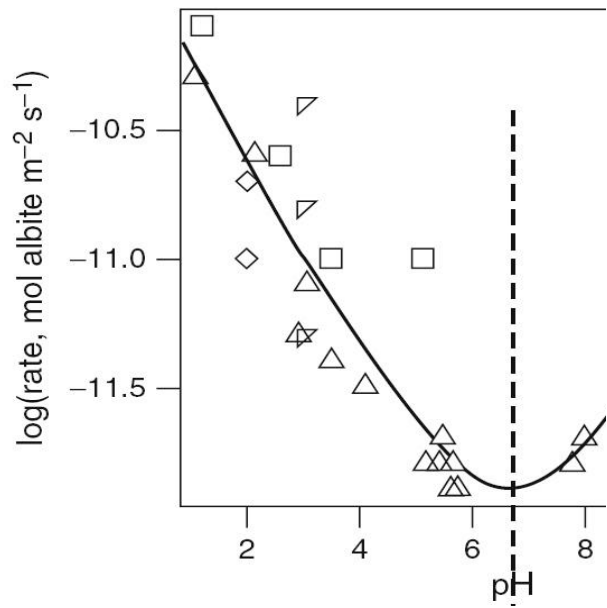
nulová čára		
1	0	
13	0	
nulová čára		
-12	0	
-10.2	0	
linie		
-12	-12	
-10.2	-10.2	





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log r vyp	D	pH	linie log r
-10.29	0.01	1.0	-10.25
-10.70	-0.11	1.5	-10.45
-11.00	0.28	2.0	-10.64
-11.06	0.02	2.5	-10.84
-11.22	0.17	3.0	-11.03
-11.44	0.04	3.5	-11.22
-11.77	0.01	4.0	-11.40
-11.83	-0.05	4.5	-11.58
-11.84	-0.16	5.0	-11.73
-11.86	0.02	5.5	-11.84
-11.87	-0.09	6.0	-11.90
-11.88	0.00	6.5	-11.88
-11.64	0.15	7.0	-11.81
-11.58	0.10	7.5	-11.71
-11.00	0.08	8.0	-11.58
-11.00	-0.02	8.5	-11.44
-10.84	0.14	9.0	-11.29
-10.64	-0.09	9.5	-11.15
-10.55	0.00	10.0	-11.00
-10.33	0.01	10.5	-10.85
		11.0	-10.70
		11.5	-10.55
		12.0	-10.40
		12.5	-10.25



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