

C9930, 6. přednáška, 7. 4. 2021

### 3. EHT: dokončení

Literatura: John P. Löwe, Quantum Chemistry

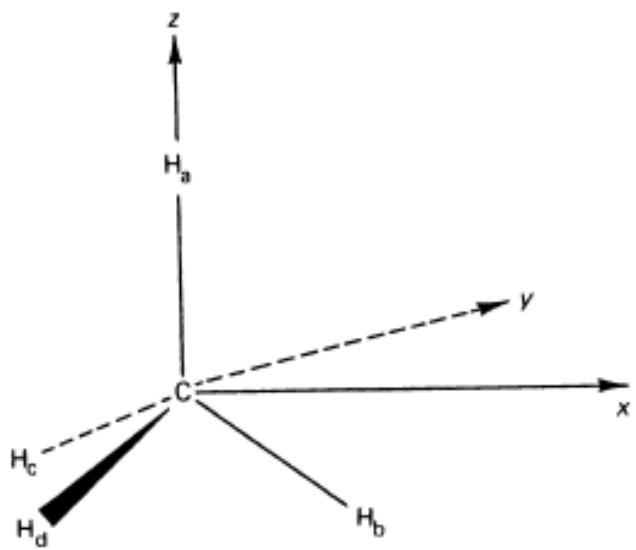
Kapitola 10

## 3. Rozšířená Hückelova metoda

3.1 Souřadnice jader (Lowe 10-1.A)

3.2 Báze pro MO-LCAO (Lowe 10-1.B)

3.3 Matice překryvu (Lowe 10-1.C)



**Figure 10-1 ►** Orientation of methane in a Cartesian axis system.

**TABLE 10-2 ►** Basis AOs for Methane

AO no.	Atom	Type	$n^a$
1	C	2s	2
2	C	2p <sub>z</sub>	2
3	C	2p <sub>x</sub>	2
4	C	2p <sub>y</sub>	2
5	H <sub>a</sub>	1s	1
6	H <sub>b</sub>	1s	1
7	H <sub>c</sub>	1s	1
8	H <sub>d</sub>	1s	1

**TABLE 10-3 ►** Overlap Matrix for STOs of Table 10-2

	1	2	3	4	5	6	7	8
1	1.0000	0.0	0.0	0.0	0.5133	0.5133	0.5133	0.5133
2	0.0	1.0000	0.0	0.0	0.4855	-0.1618	-0.1618	-0.1618
3	0.0	0.0	1.0000	0.0	0.0	0.4577	-0.2289	-0.2289
4	0.0	0.0	0.0	1.0000	0.0	0.0	0.3964	-0.3964
5	0.5133	0.4855	0.0	0.0	1.0000	0.1805	0.1805	0.1805
6	0.5133	-0.1618	0.4577	0.0	0.1805	1.0000	0.1805	0.1805
7	0.5133	-0.1618	-0.2289	0.3964	0.1805	0.1805	1.0000	0.1805
8	0.5133	-0.1618	-0.2289	-0.3964	0.1805	0.1805	0.1805	1.000

### 3.2 Báze pro MO-LCAO (Lowe 10-1.B)

### 3.3 Matice překryvu (Lowe 10-1.C)

3.4 Matice Hamiltoniánu (Lowe 10-1.D)

**TABLE 10-4** ► The Extended Hückel Hamiltonian Matrix for CH<sub>4</sub><sup>a</sup>

	1	2	3	4	5	6	7	8
1	-0.7144	0.0	0.0	0.0	-0.5454	-0.5454	-0.5454	-0.5454
2	0.0	-0.3921	0.0	0.0	-0.3790	0.1263	0.1263	0.1263
3	0.0	0.0	-0.3921	0.0	0.0	-0.3573	0.1787	0.1787
4	0.0	0.0	0.0	-0.3921	0.0	0.0	-0.3094	0.3094
5	-0.5454	-0.3790	0.0	0.0	-0.5000	-0.1579	-0.1579	-0.1579
6	-0.5454	0.1263	-0.3573	0.0	-0.1579	-0.5000	-0.1579	-0.1579
7	-0.5454	0.1263	0.1787	-0.3094	-0.1579	-0.1579	-0.5000	-0.1579
8	-0.5454	0.1263	0.1787	0.3094	-0.1579	-0.1579	-0.1579	-0.5000

<sup>a</sup>All energies in a.u.

3.3 Matice překryvu (Lowe 10-1.C)

3.4 Matice Hamiltoniánu (Lowe 10-1.D)

3.5 Vlastní funkce a vektory (Lowe 10-1.E)

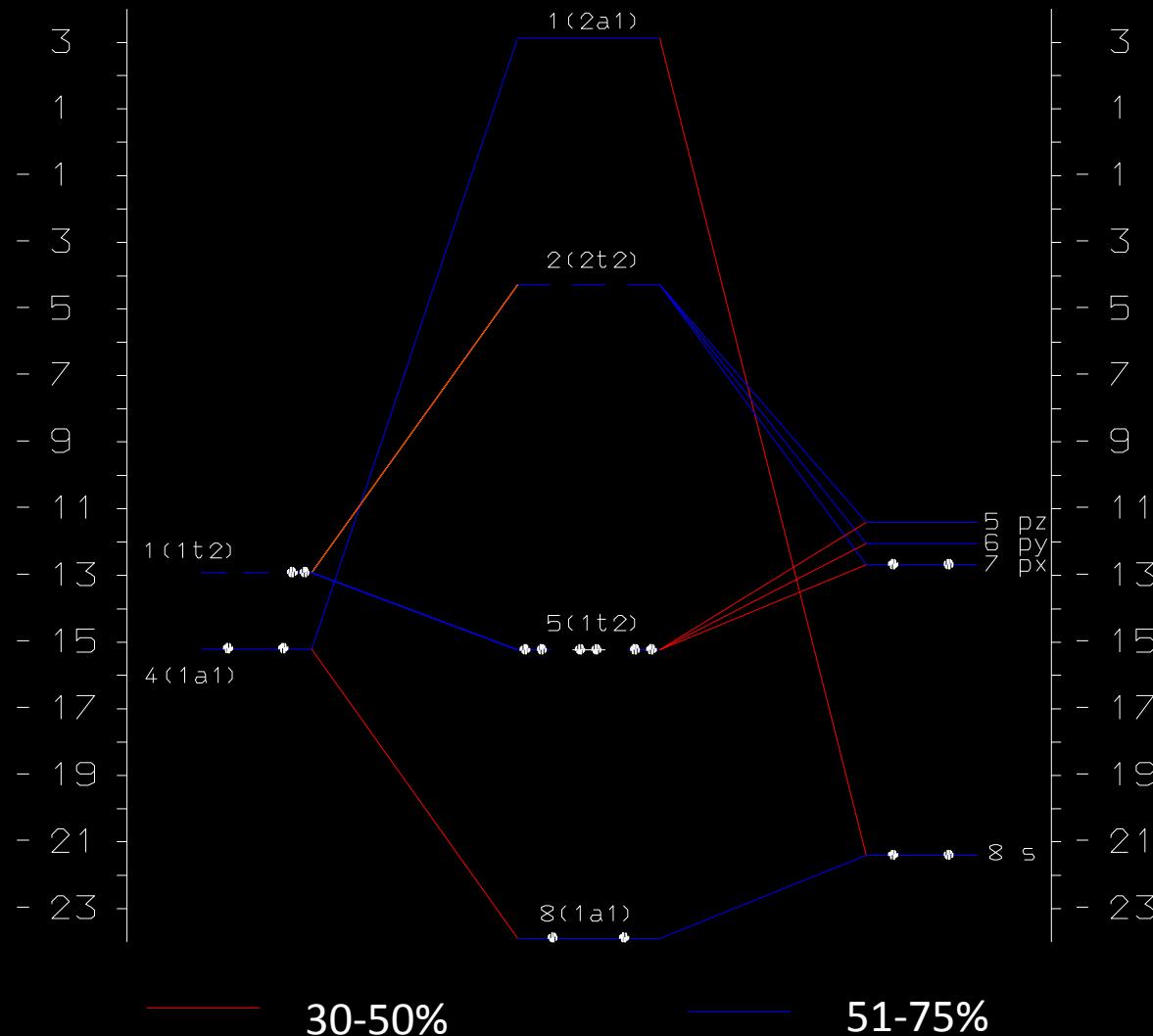
**TABLE 10-5 ► Energies for Methane by the Extended Hückel Method**

MO no.	Energy (a.u.)	Occ. no.
8	1.1904	0
7	0.2068	0
6	0.2068	0
5	0.2068	0
4	-0.5487	2
3	-0.5487	2
2	-0.5487	2
1	-0.8519	2

# Interakční diagram molekuly $\text{CH}_4$ vypočtený programem C.A.C.A.O.\*

Energie (eV)

Energie (eV)



\* (Computer Aided Composition of Atomic Orbitals). The package consists of an EHC program (SIMCON), an orbital analysis program (MOAN) and a display program for atomic orbitals (CACAO). Davide M. Proserpio, Universita di Milano.