## **Homework problems #1**

1. Helmholtz free energy of liquids can by approximated by

$$F(T,V,N) = -kT \ln \left\{ \left( \frac{h^2}{2\pi mkT} \right)^{-3N/2} \left[ \frac{4}{3}\pi \left( \left( \sqrt{2}\frac{V}{N} \right)^{1/3} - \sigma \right)^3 \right]^N \right\},$$

where h, m, k and  $\sigma$  are constants. Determine the equation of state of this gas p = p(T, V), the difference of specific heats  $c_p - c_V$ , and show that both formulae correspond to ideal gas for  $V/N \gg \sigma$ .

2. *Computer problem*: The number of microstates of the system of *N* classical noninteracting particles with energy lower than *E* in a volume *V* is given by

$$\Omega(E) = \left(\frac{2\pi mE}{h^2}\right)^{\frac{3}{2}N} \frac{V^N}{N! \, \Gamma(\frac{3}{2}N+1)}.$$

Let us study two such systems in a thermal interaction with total energy 2E. We shall study fluctuations in these systems, within which one of the system has energy  $E+\Delta E$  and the second one  $E-\Delta E$ . Plot the number of microstates of combined system as a function of  $\Delta E$  for  $N=10, 10^2, 10^4$ . (It is advisable to express the energy in units of  $\varepsilon=h^2/(2\pi m)$  an number of states in units of  $\gamma=V^N/(N!\Gamma(\frac{3}{2}N+1))$ .) Let us assume that  $\Delta E$  changes discountinuously by the value of  $\varepsilon/100$  and determine an error introduced by assuming that the entropy is calculated only from the states corresponding to the equilibrium. (calculate for  $N=10, 10^2, 10^4, 10^5$ ).

The solution should be submitted not later than on March 16th.