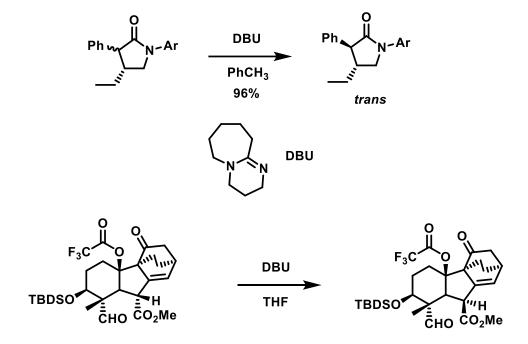
# **Organic synthesis**

**Kamil Paruch** 

Masaryk University, Brno

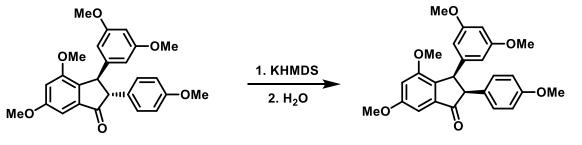
# epimerization of carbonyl compounds

• thermodynamic equilibration (deprotonation-protonation)



J. Am. Chem. Soc. 1985, 107, 5574.

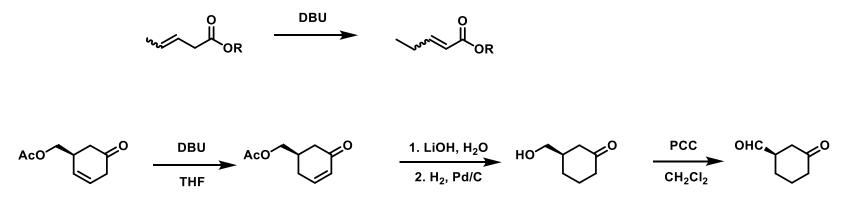
• *kinetic* epimerization is also possible



• C-protonation of enolate from less hindered site J. Am. Chem. Soc. 2009, 131, 1753.

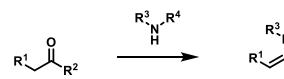
### migration of double bond in carbonyl compounds

• deprotonation-protonation: anion is in conjugation with double bond



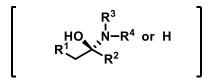
Tetrahedron Lett. 1983, 24, 4883.

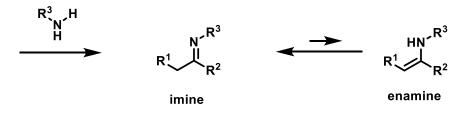
## tautomerism imine-enamine





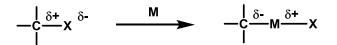
enamine



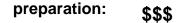


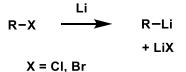
(more stable tautomer)





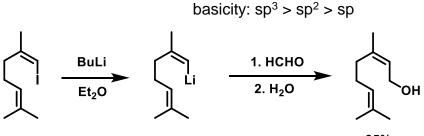
RLi (organolithium reagents)





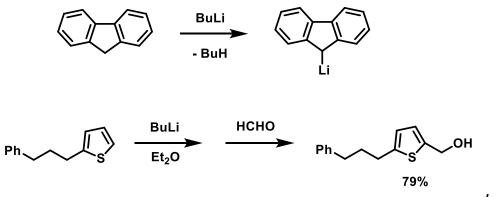
BuLi as a source of Li

- lithium-halogen exchange (typically fast even at low temperature)
- generated RLi is a weaker base than BuLi



#### BuLi as a source of Li

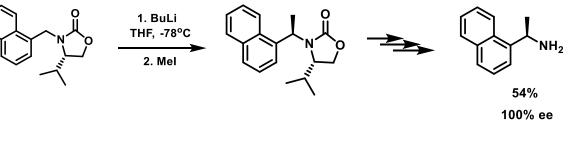
• deprotonation of compounds that are more acidic than BuH



J. Med. Chem. 1992, 35, 3156.



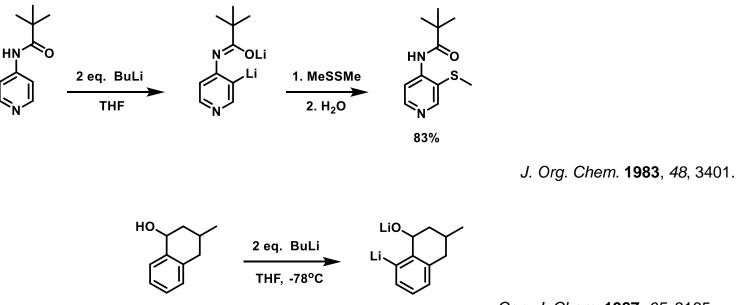
# organometallics



J. Org. Chem. 1989, 54, 3002.

#### ortho-lithiation

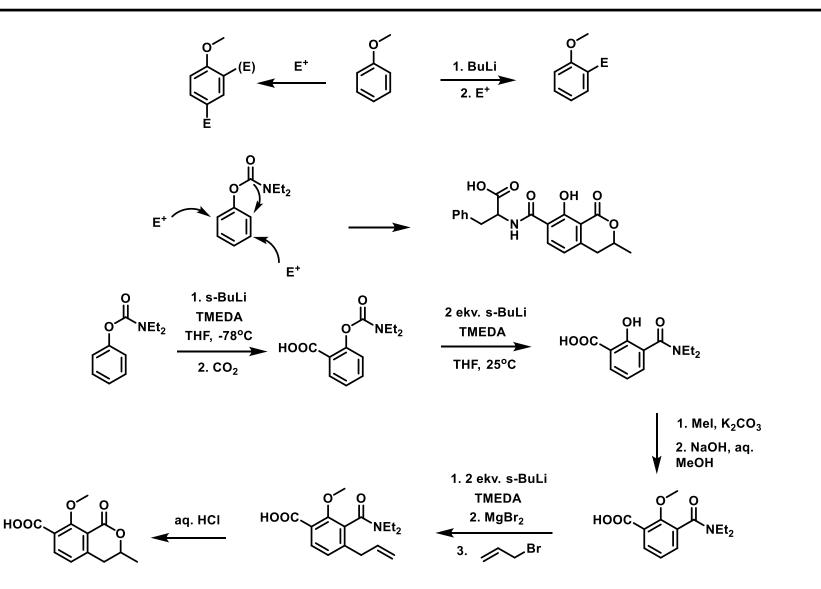
• regioselectivity of lithiation is directed by coordination of Li to the directing group (CONHR, CH<sub>2</sub>OH, NHCOtBu, SO<sub>2</sub>NHR, OMOM)



Can. J. Chem. 1987, 65, 3165.

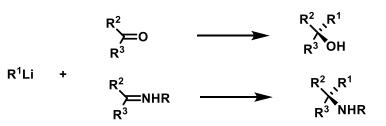
# organometallics

Organic Synthesis C4450

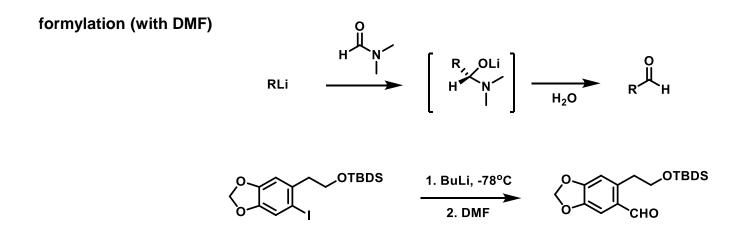


J. Am. Chem. Soc. 1985, 107, 6312.

nucleophilic attack of C=O, C=N bonds



competition: deprotonation

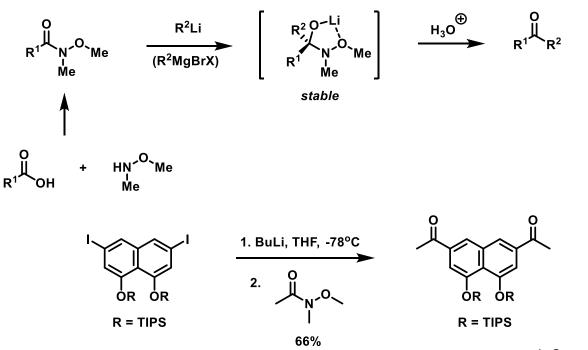


J. Am. Chem. Soc. 1994, 116, 9791.

#### Weinreb amides

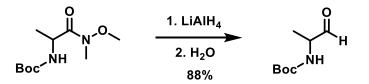
• reliable preparation of ketones

R<sup>1</sup> can be also H (formylation): *Tetrahedron Lett.* **1999**, *40*, 7889.



J. Org. Chem. 2000, 65, 8774.

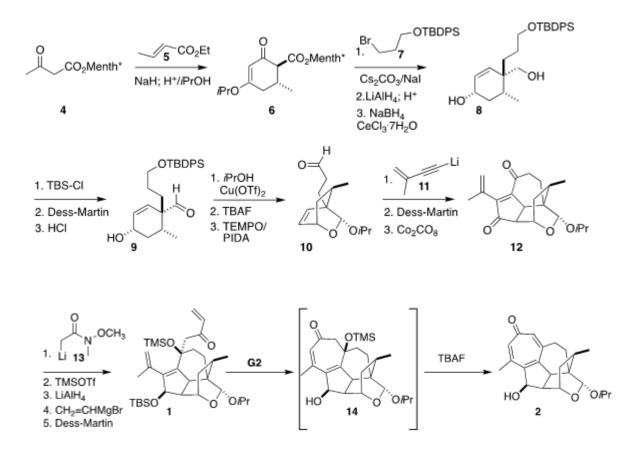
• similarly: reduction to aldehydes



J. Med. Chem. 1985, 28, 1874.

#### Weinreb amides

• can be part of organometallic reagent (e.g. reagent 13)

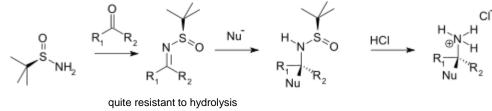


Angew. Chem. Int. Ed. **2021**, 60, 18572. Eur. J. Org. Chem. **2022**, e202101430.

#### Ellman's reagent (sulfinamide)

- · both enantiomeric forms are commercially available
- used in asymmetric synthesis as chiral auxiliaries, often as chiral ammonia equivalents for the synthesis of amines

Chem. Rev. 2010, 110, 3600.



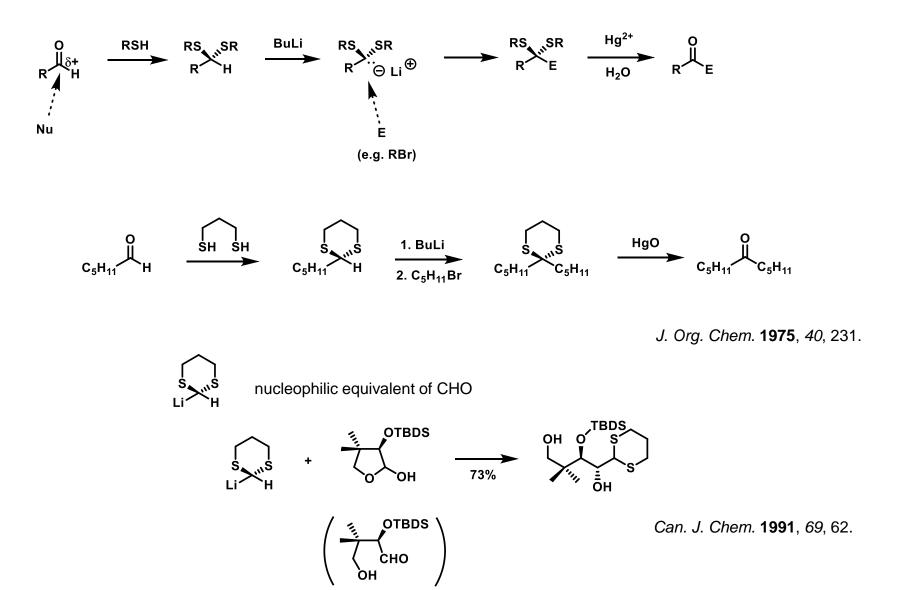
but react with nucleophiles

• typical nucleophiles: RMgX, RZnX, RLi, enolates

13

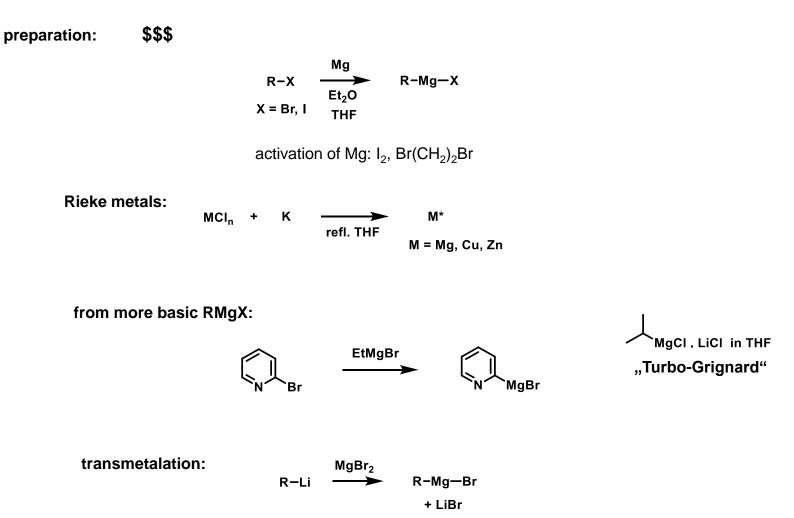
#### deprotonation of 1,3-dithianes

• generation of nucleophilic acyl equivalents



Organic Synthesis C4450

### **RMgX (Grignard reagents)**

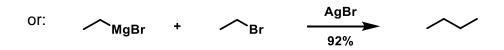


#### formation of C-C bonds

reactions with s alkyl halides typically do not give good yields

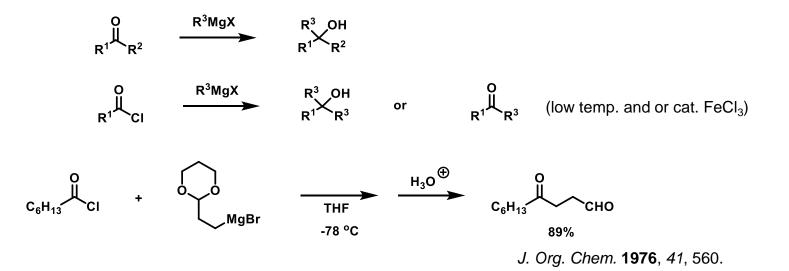
exception:

reactive RMgX (e.g. CH<sub>2</sub>=CH-CH<sub>2</sub>MgX) + reactive RX (e.g. MeI, BnBr, allylBr)



J. Am. Chem. Soc. 1971, 93, 1483.

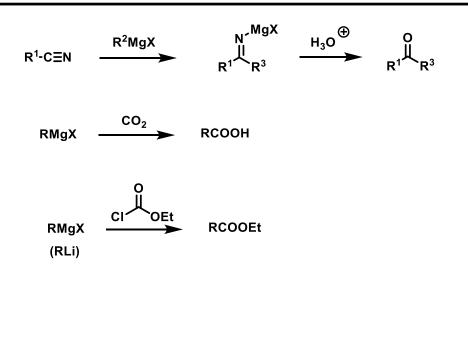
nucleophilic attack on C=O bond

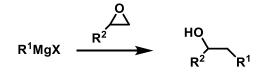


Kamil Paruch

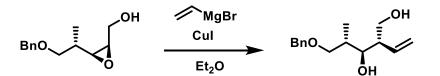
# organometallics

Organic Synthesis C4450



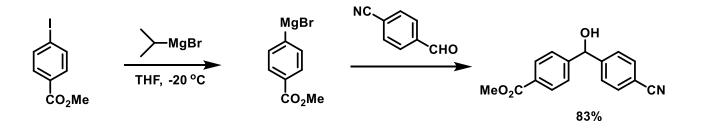


directed attack on epoxide



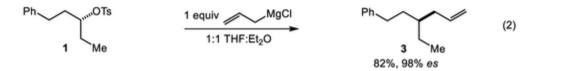
Tetrahedron Lett. 1979, 4343.

under proper conditions, in situ prepared RMgX reagents can be compatible with reactive functional groups

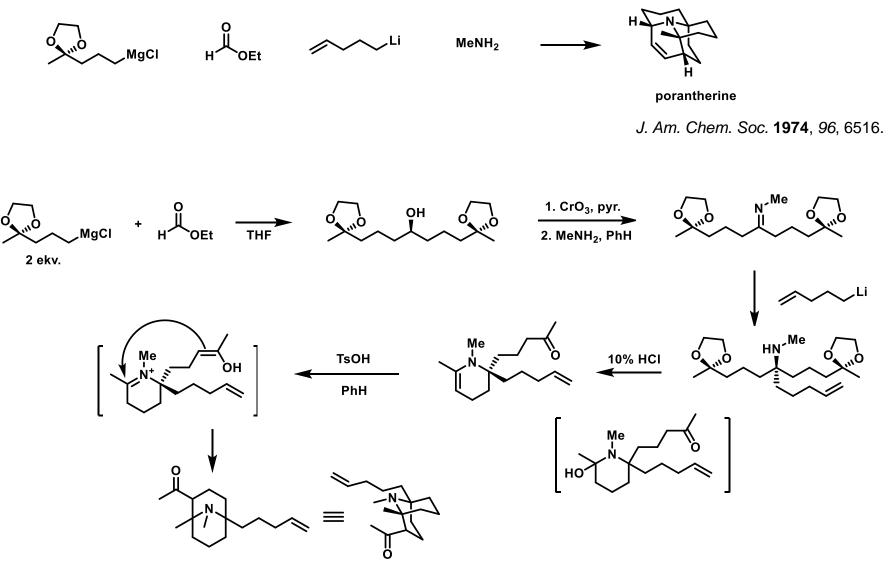


Angew. Chem. Int. Ed. 2003, 42, 4302.

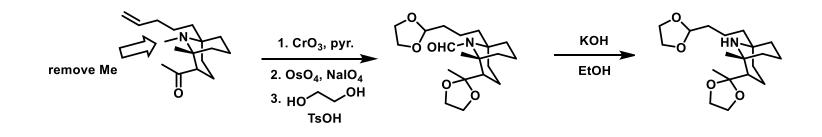
allyl magnesium chloride can undergo nucleophilic substitution with tosylates:

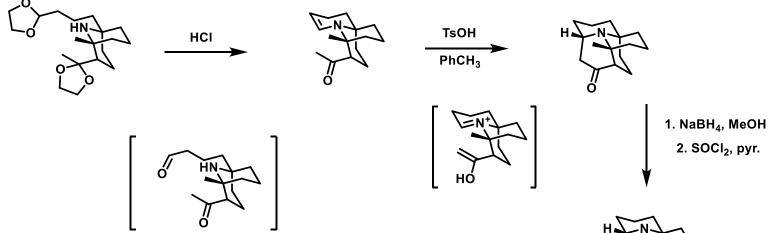


Org. Lett. 2021, 23, 7215.



1st step: oxidation of N-Me to N-CHO

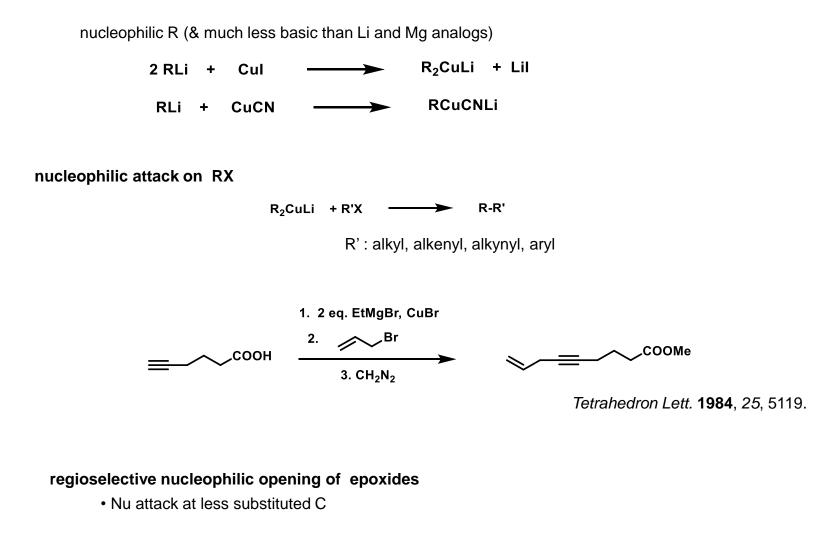




н

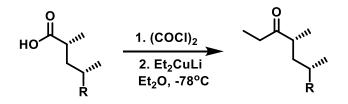
porantherine

# Organocuprates: RCu; R<sub>2</sub>CuLi; R<sub>3</sub>CuLi<sub>2</sub>; RCuCNLi



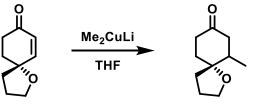


preparation of ketones from acid chlorides



J. Am. Chem. Soc. 1981, 103, 1568.

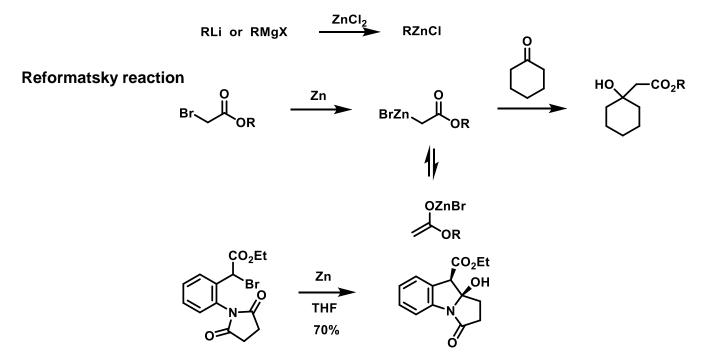
1,4-addition to  $\alpha$ , $\beta$ -unsaturated carbonyl compounds



Tetrahedron Lett. 1985, 26, 6015.

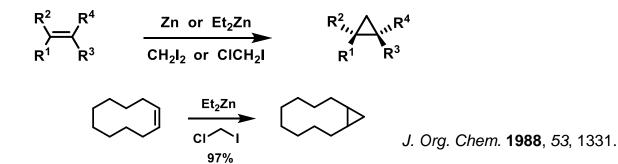
# Organozinc reagents: RZnX; R<sub>2</sub>Zn

· also much less basic than Li and Mg analogs



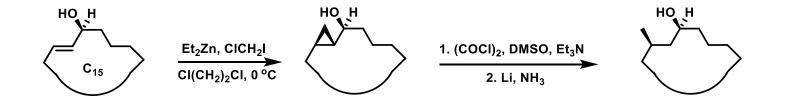
#### Cyclopropanation (Simmons-Smith rxn)

stereoselective: syn-addition (cyclic transition state)



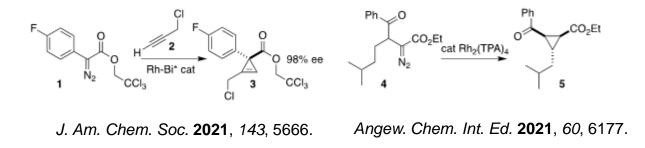
cyclopropanation can be stereoselectively directed using neighboring OH group

• cyclopropane ring is reactive and can udergo subsequent transformations

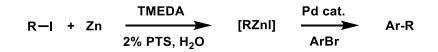


J. Am. Chem. Soc. 1993, 115, 1593.

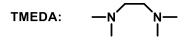
alternative cyclopropanation: generation of carbene from diazo compounds



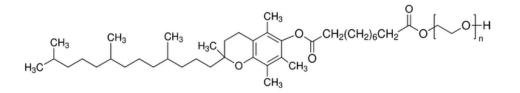
## Generation and use of organozinc reagents in water



J. Am. Chem. Soc. 2009, 131, 15592.



PTS: polyoxyethanyl-α-tocopheryl sebacate



- micelar catalysis
- "nanoreactors" from PTS in water

Organic Synthesis C4450

