

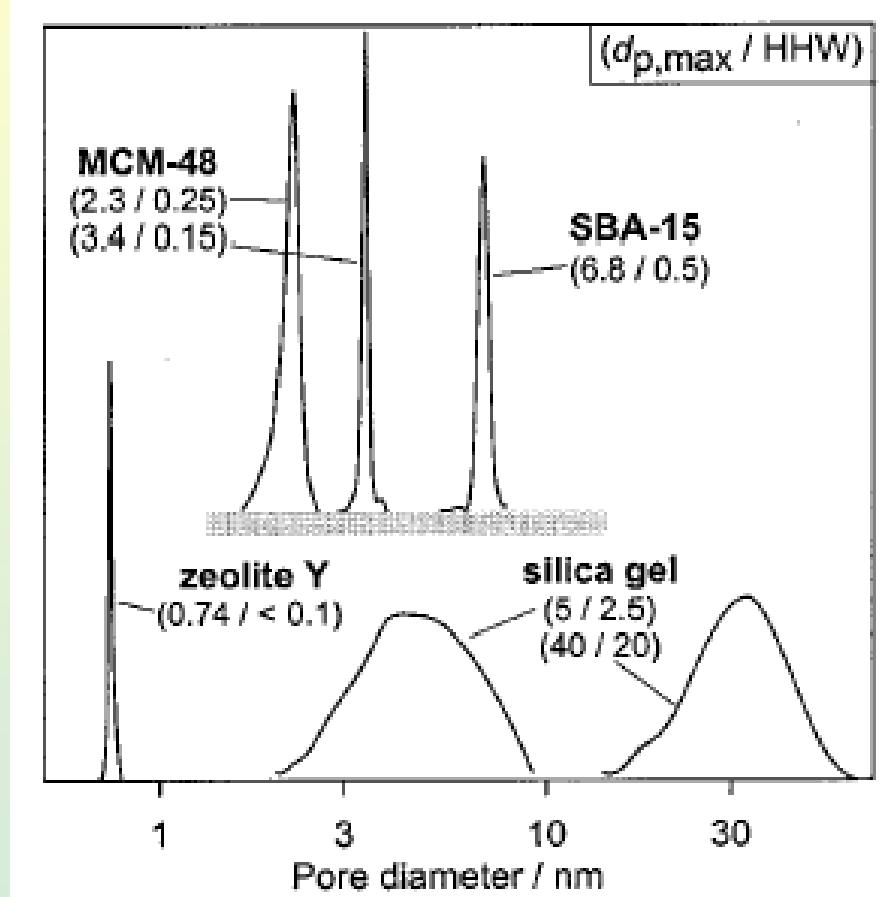
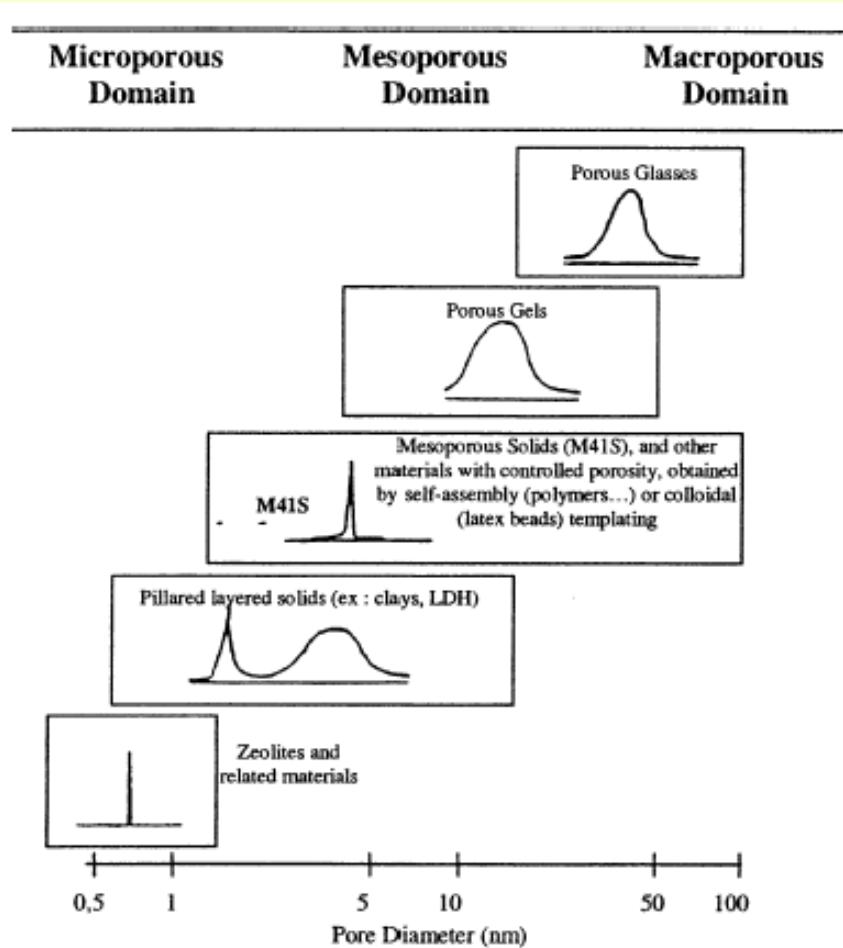
Mesoporous Materials

Amorphous, disordered - silica xerogels

Ordered, amorphous walls

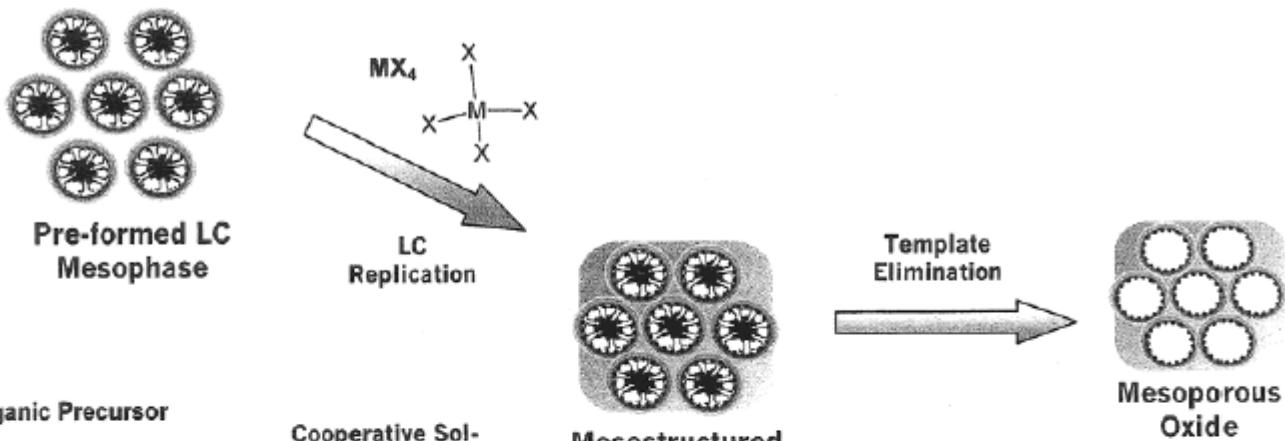
Pore diameter, d [nm]	Material	Example
$d > 50$	Macroporous	Aerogels
$2 < d < 50$	Mesoporous	Xerogels
$d < 2$	Microporous	Zeolites

Pore size distribution

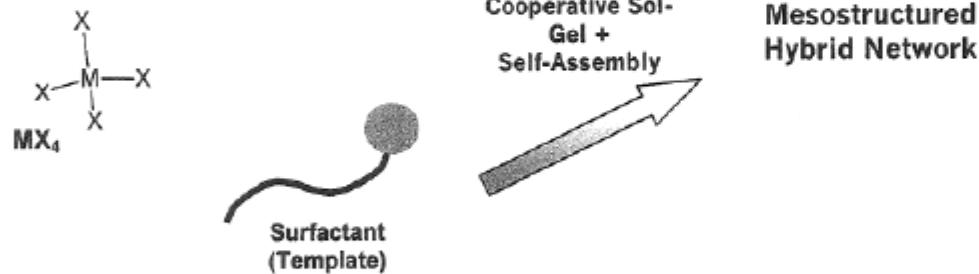


Mesostucture Assembly

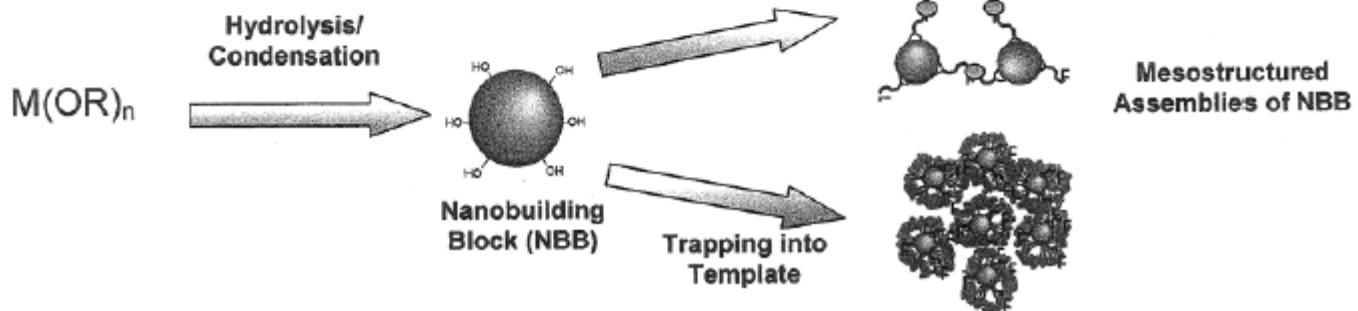
A



B



C



Mesoporous Materials

MMS mesoporous molecular sieves

MCM-n Mobil Composition of Matter

M41S

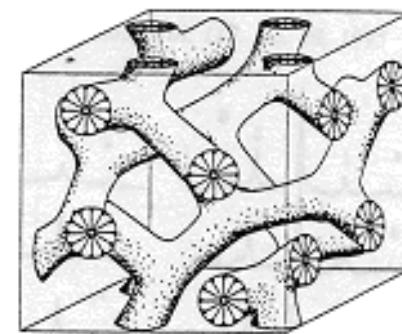
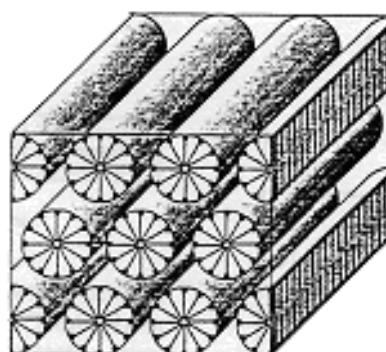
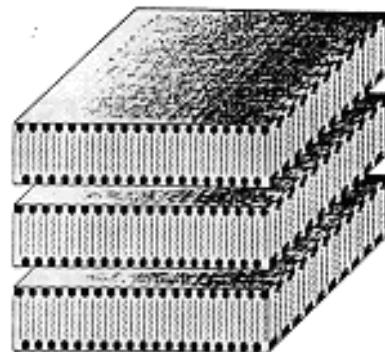
Discovered 1992

A - lamellar MCM-50

B - hexagonal MCM-41

C - cubic MCM-48

Inverse hexagonal

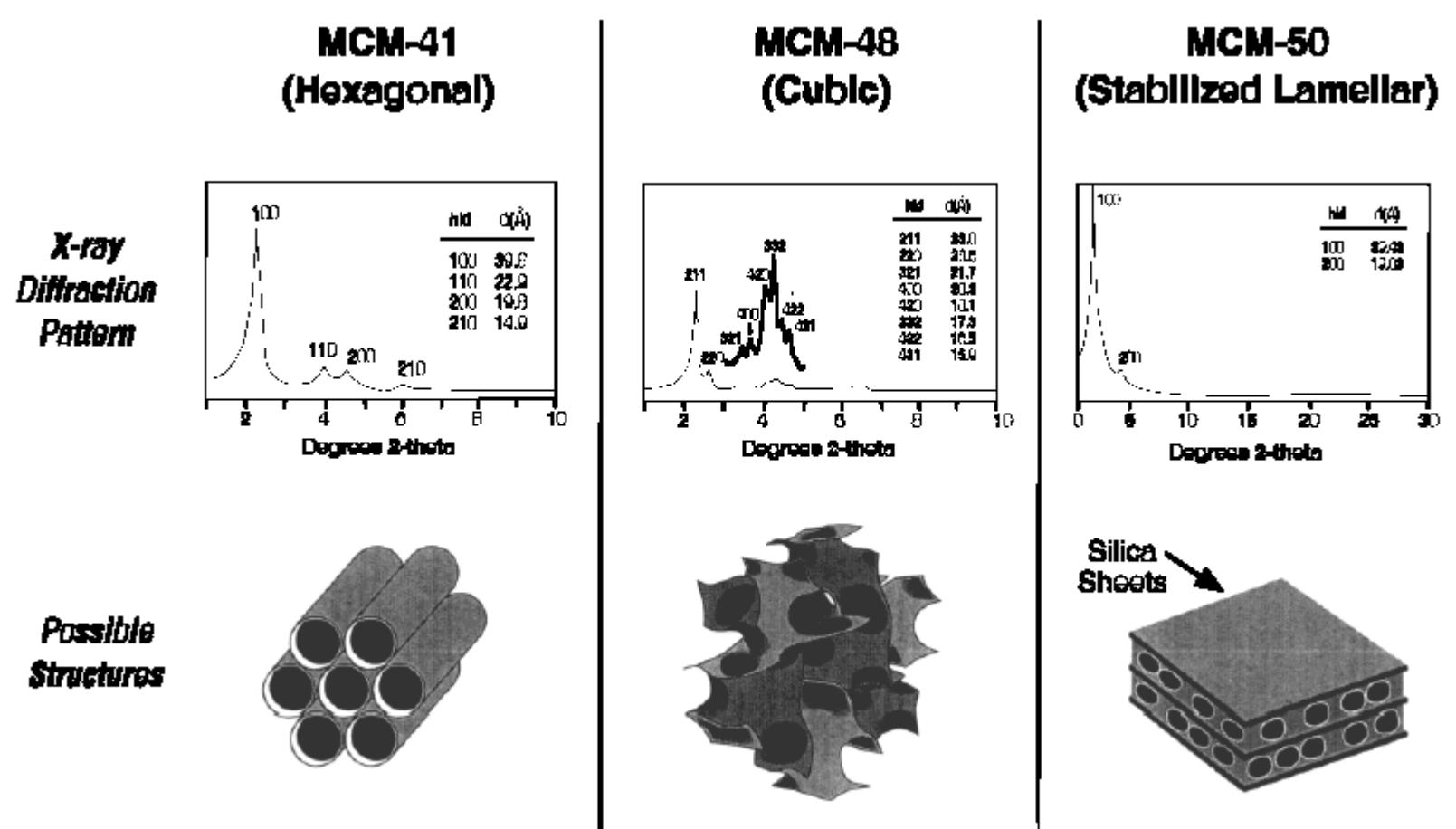


A

B

C

Mesoporous Materials



Supramolecular Templating

Surfactants - amphiphilic molecules, polar (head group) and nonpolar (chain, tail) part
lyophilic, lyophobic

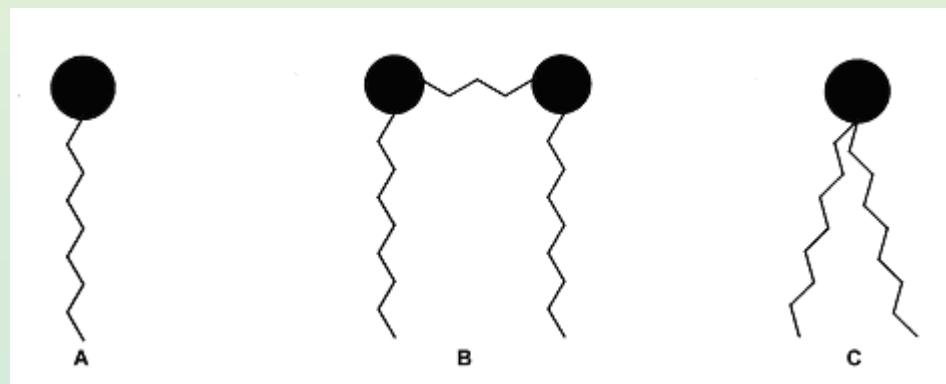
Ionic surfactants, cationic, anionic, zwitterionic

Nonionic amines, polyethylenoxides

A - normal surfactant molecule

B - gemini

C - swallow tail



Surfactants

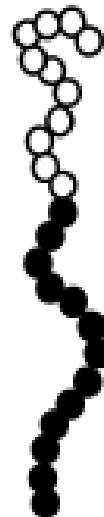
Hydrotrope
(flexible surfactant)



Classical surfactant
(rigid surfactant)



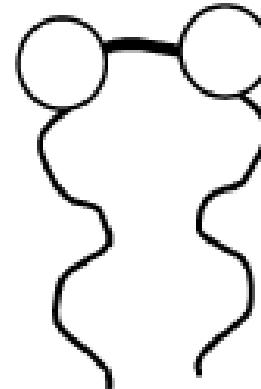
Diblock copolymer
surfactant



Bolaform surfactant



Gemini surfactant
(dimeric)



Polymeric surfactant



Surfactants

Anionic

- *sulfates:* $C_nH_{2n+1}OSO_3^-Na^+$
- *sulfonates:* $C_nH_{2n+1}SO_3H$
- *phosphates:* $C_nH_{2n+1}OPO_3H_2$
- *carboxylates:* $C_nH_{2n+1}COOH$

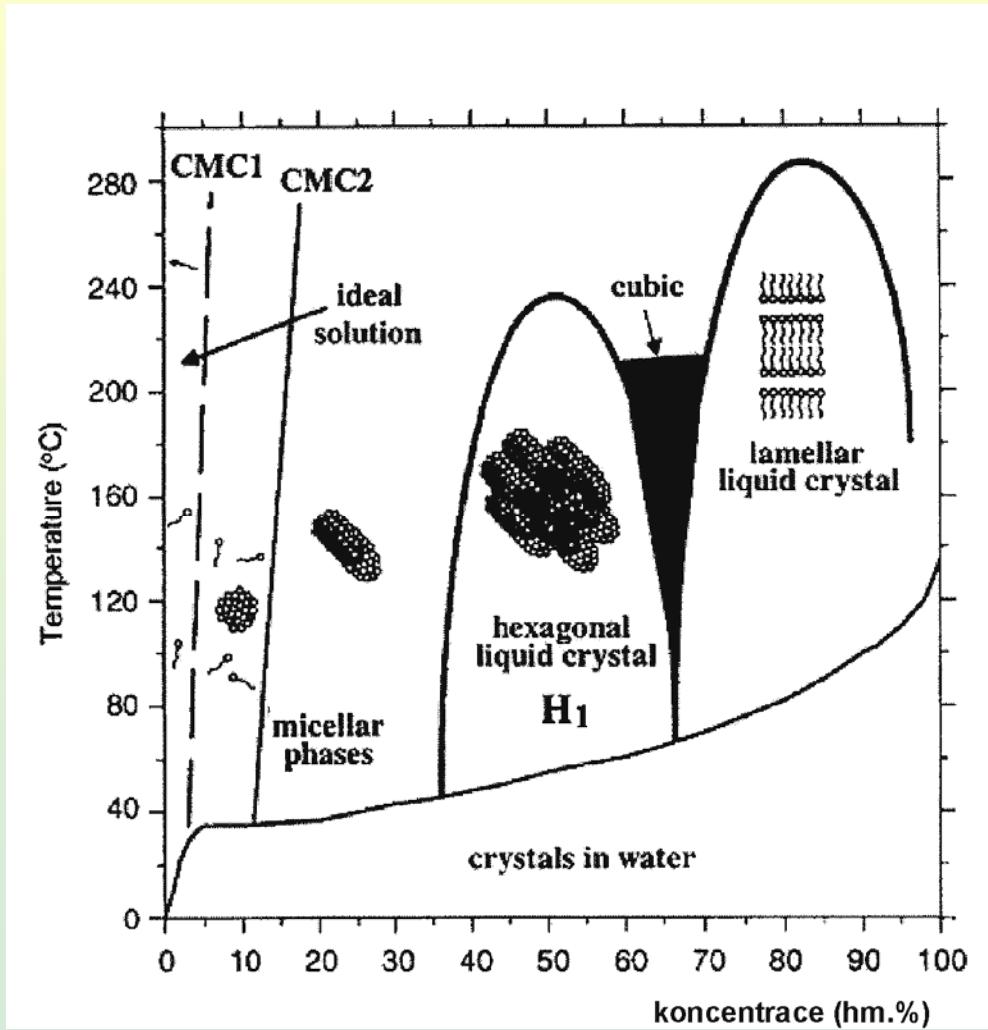
Cationic

- *alkylammonium salts:* $C_nH_{2n+1}(CH_3)_3NX \quad X = OH, Cl, Br, HSO_4$
- *dialkylammonium salts:* $(C_{16}H_{33})_2(CH_3)_2N^+Br^-$

Noionic

- *primary amines:* $C_nH_{2n+1}NH_2$
- *polyethyleneoxides:* $HO(CH_2CH_2O)_nH$

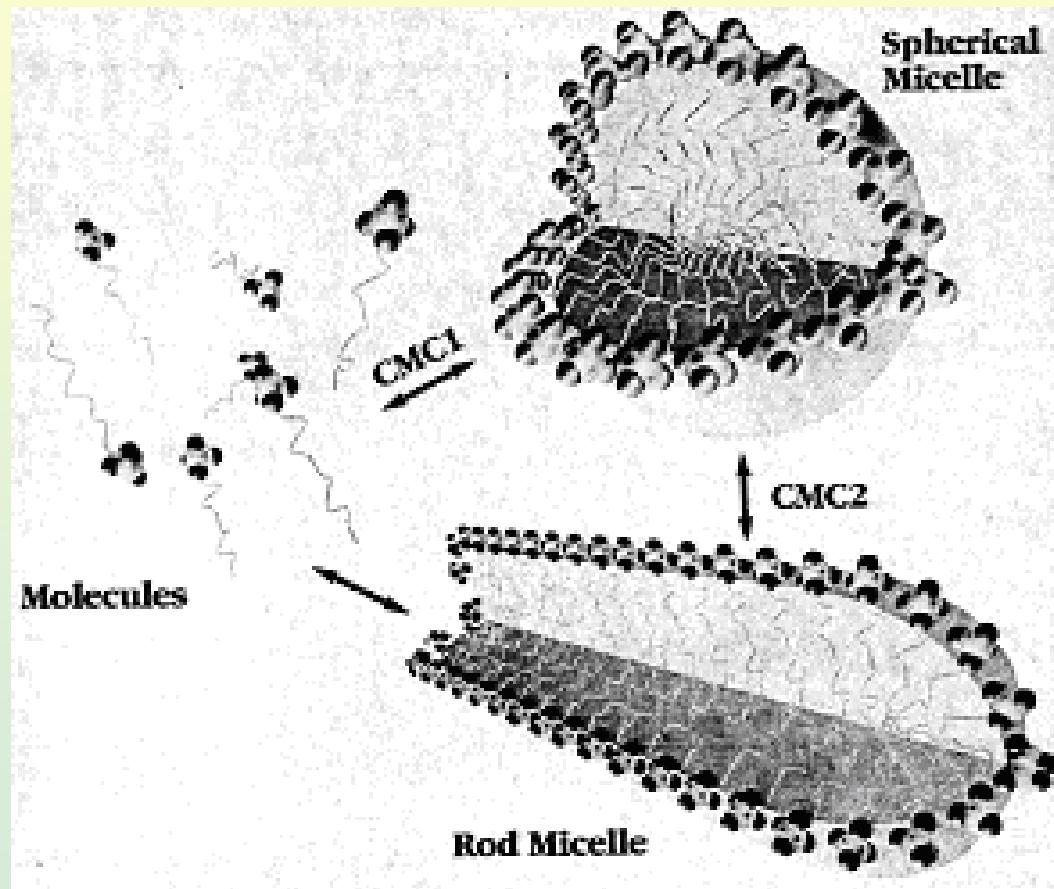
Supramolecular templating



Phase diagram of $\text{C}_{16}\text{TMABr}$

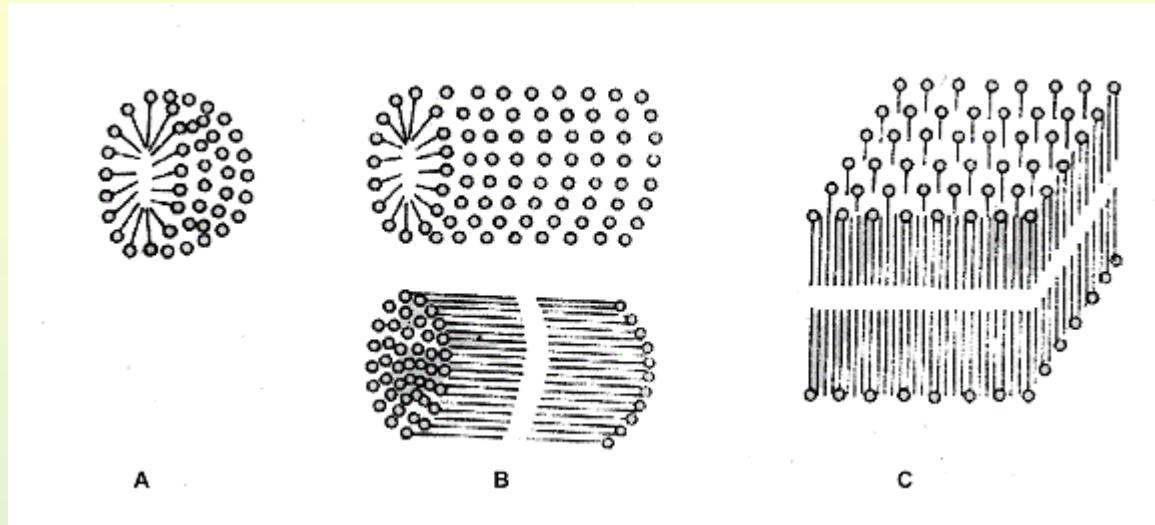
CMC = critical micelle conc.

Micelles - Supramolecular Templates



Micellar shapes

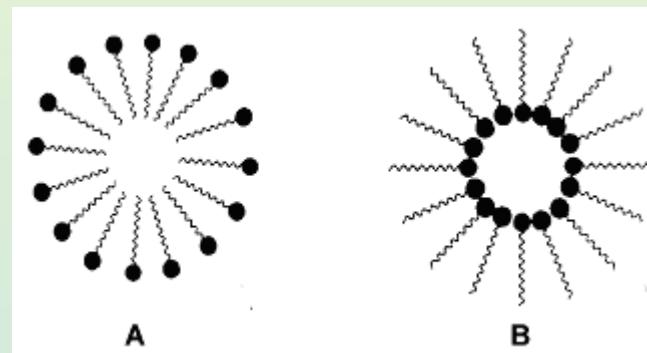
A -spherical, B - rod-like, C - lamellar



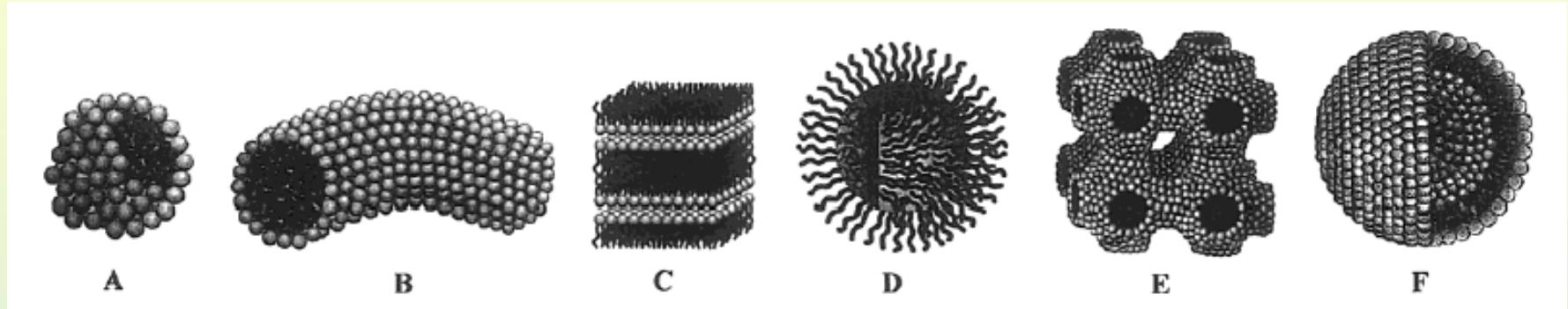
Micelles in media

A - normal, in polar solvent, H_2O

B - inverse, in nonpolar solvent, organics



Micellar shapes



Micellar structures

A) sphere, B) cylinder, C) planar bilayer,
D) reverse micelles, E) bicontinuous phase, F) liposomes).

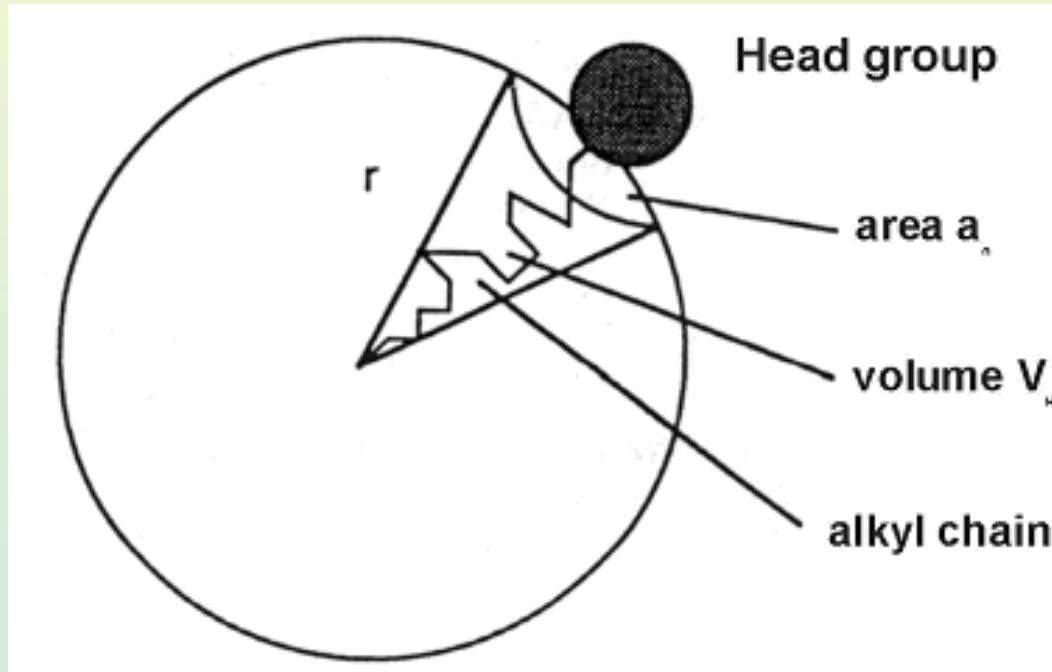
Critical packing parameter – CPP

$$\text{CPP} = V_H / a_0 l_c$$

V_H volume of the hydrophobic part, a_0 surface area of the hydrophilic part, l_c critical chain length:

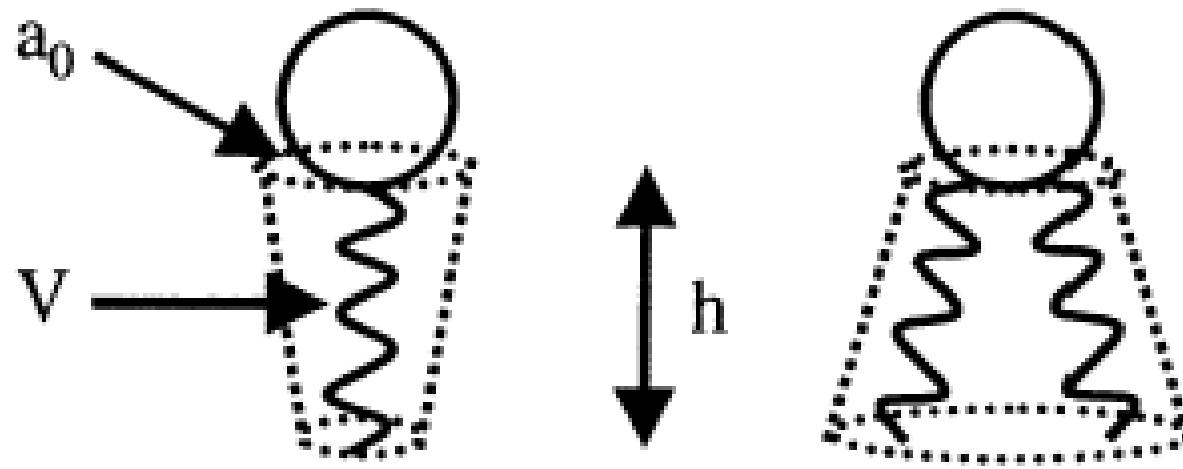
$$l_c \leq 1.5 + 1.265 n \quad [\text{\AA}]$$

n number of carbon atoms. l_c depends on the chain shape.



CPP	surfactant	micelle shape
< 0.33	linear chain, large head	spherical
0.33 - 0.5	linear chain, small head	cylindrical
0.5 - 1.0	two chains, large head	bilayers

Surfactant Molecules



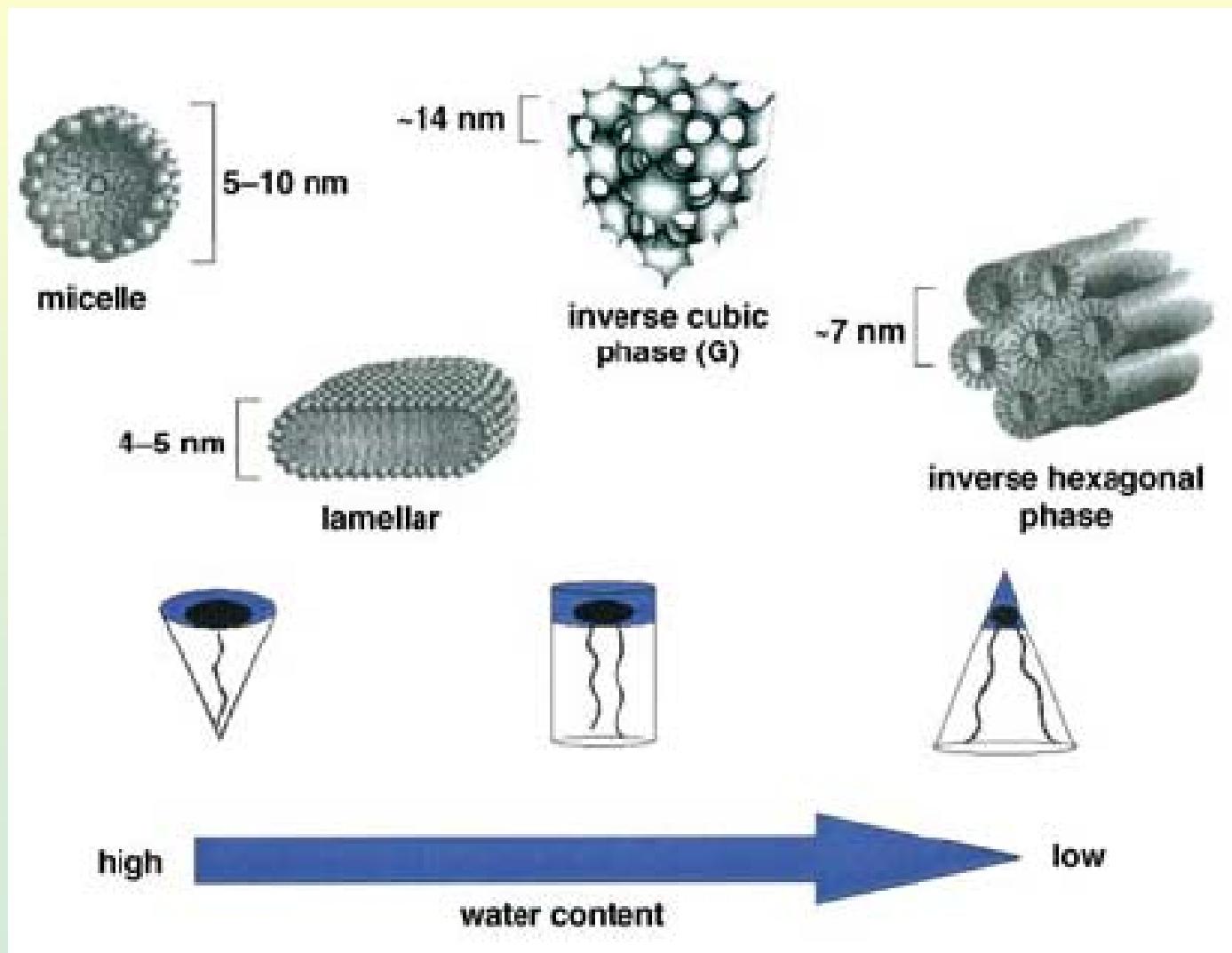
A

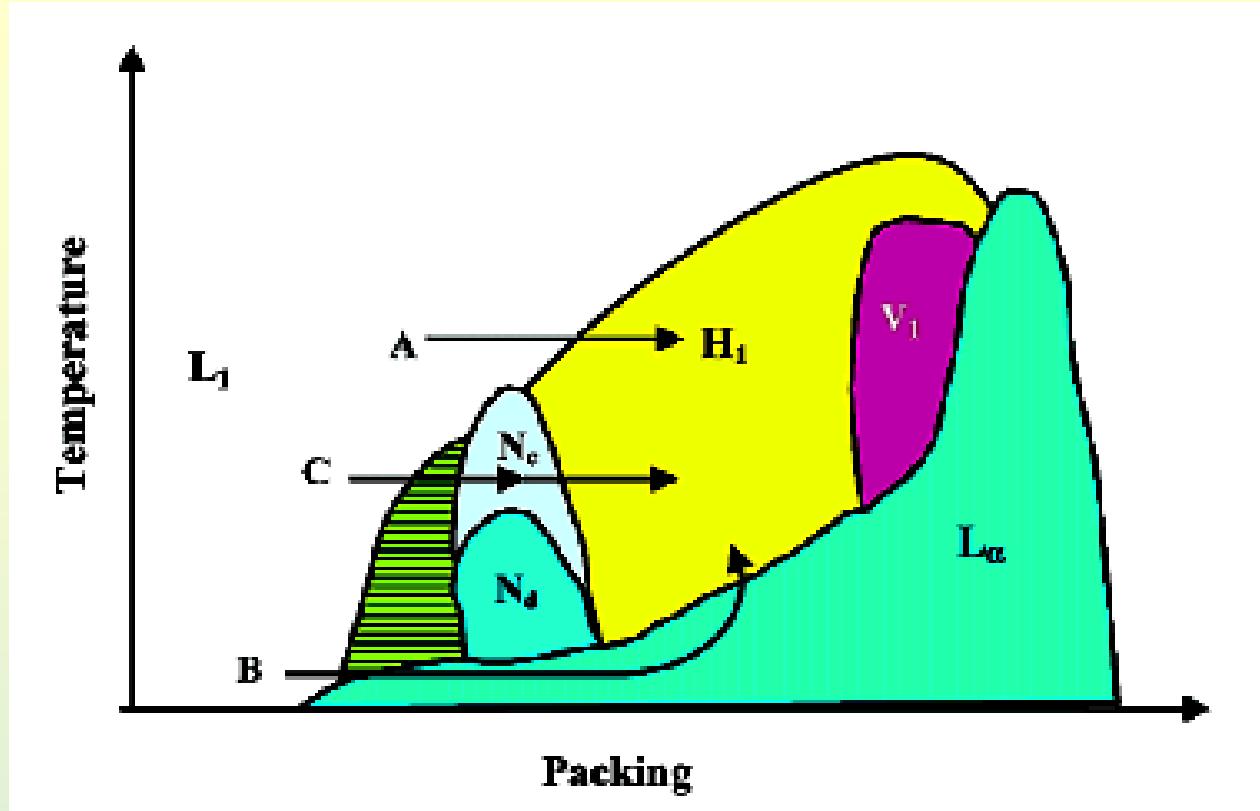
B

Conical (*icecream cone*, A)

Inverse conical (*champagne cork*, B)

Surfactant Molecules





L₁ = micellar solution; **N_c** = nematic phase; **H₁** = normal hexagonal phase (MCM-41; SBA-15);
V₁ = normal bicontinuous cubic phase (MCM-48); **L_α** = lamellar phase (MCM-50)

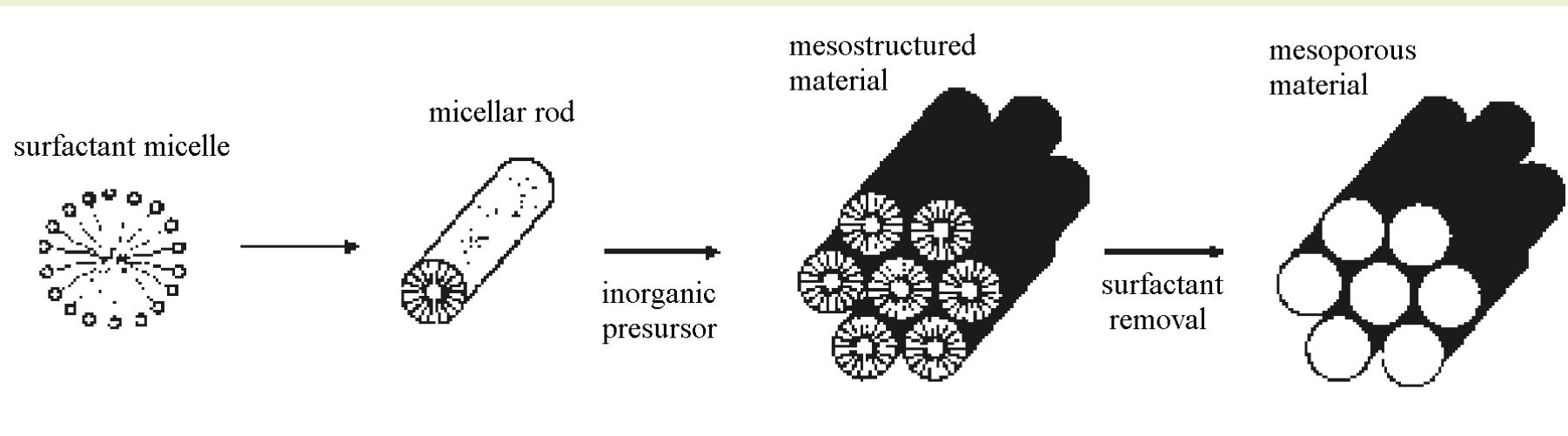
path A, the micellar solution route

path B, the lamellar phase route

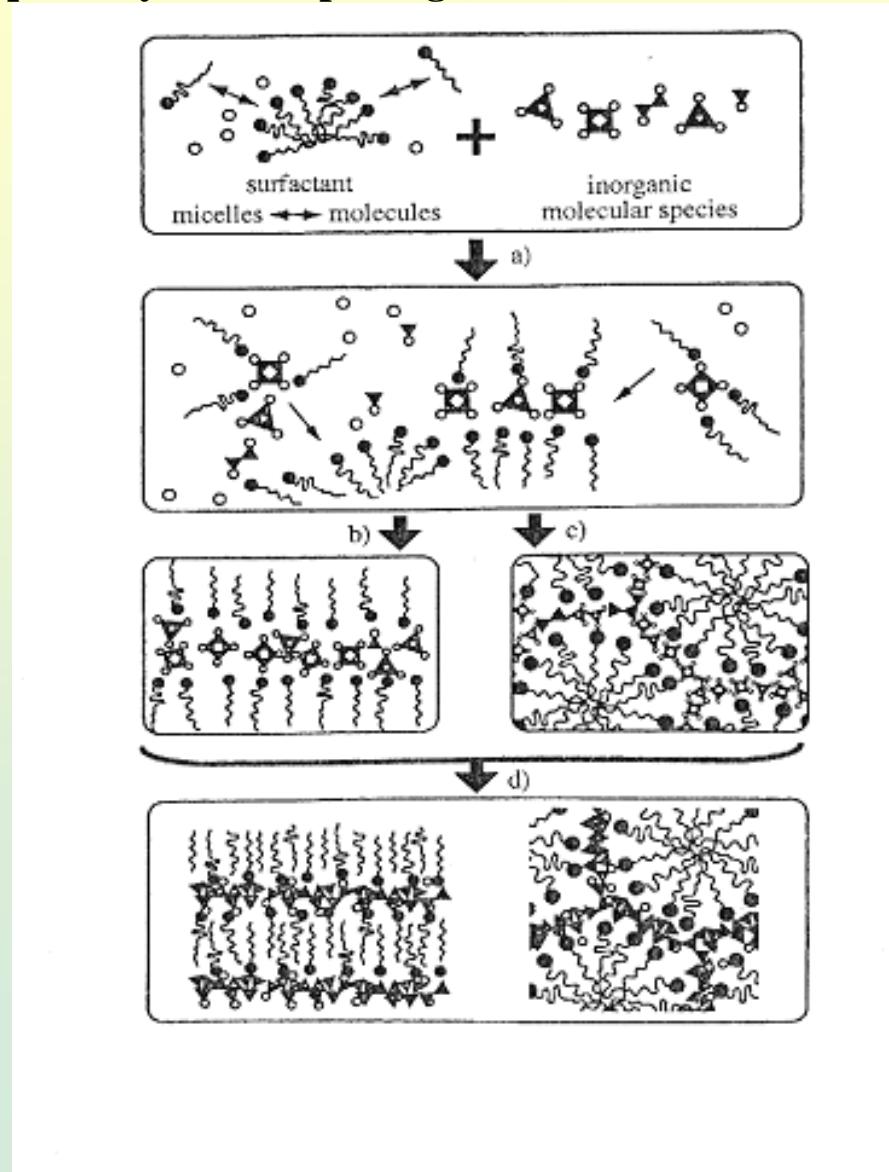
path C, the nematic phase route

Mechanism of the mesoporous material formation (hexagonal, MCM-41)

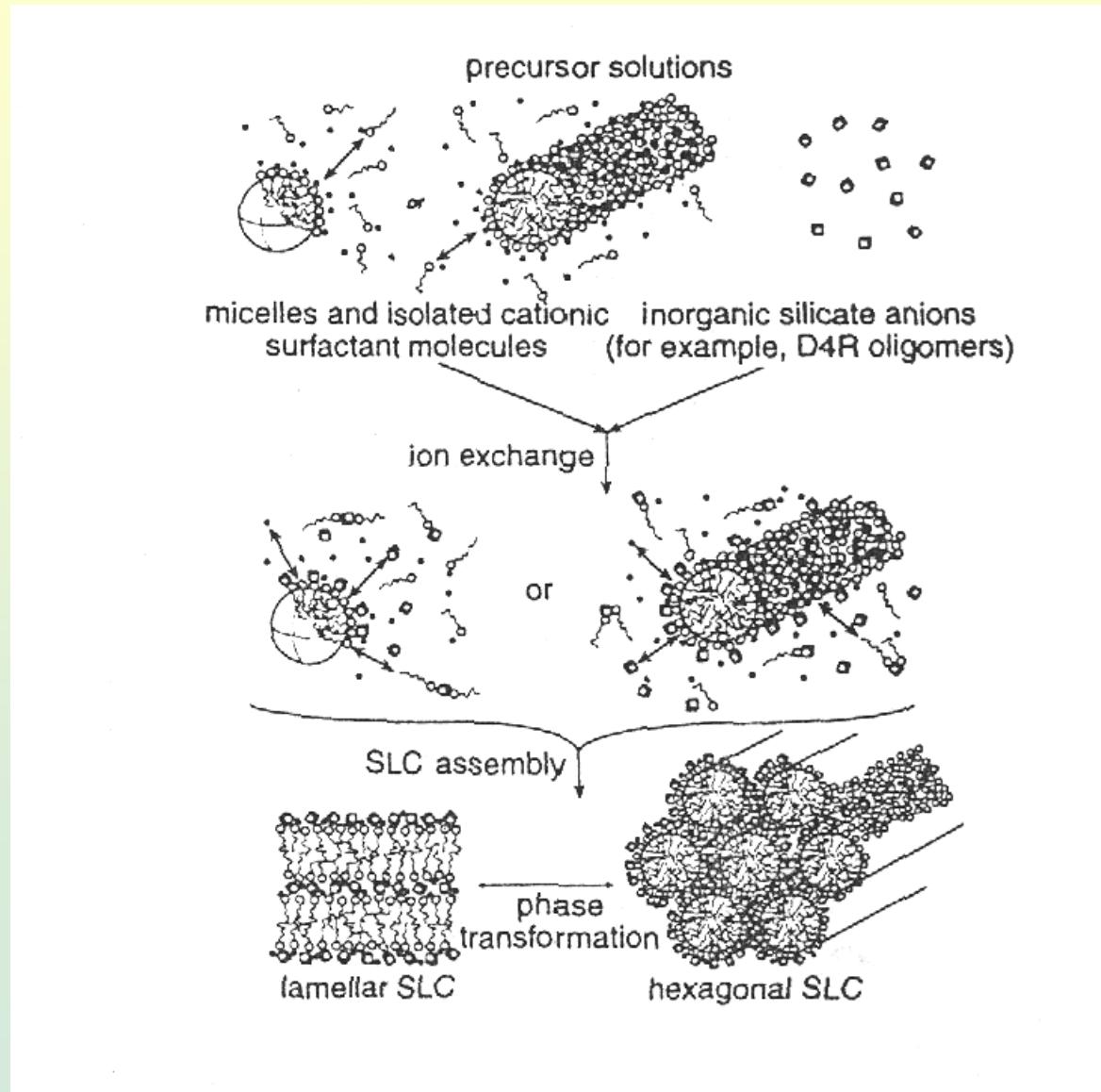
LCT Liquid Crystal Templating

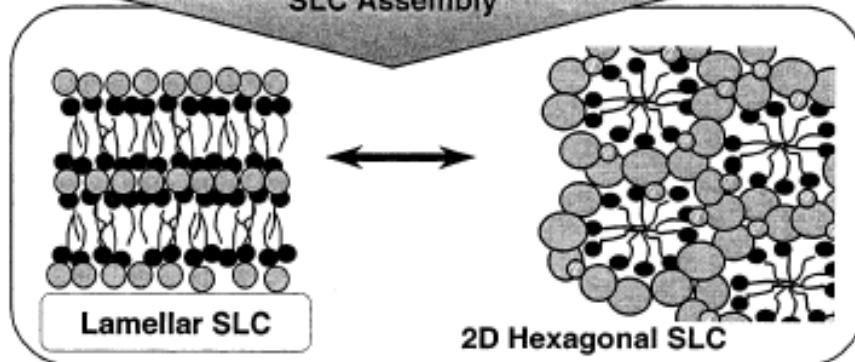
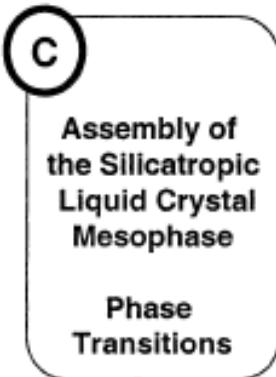
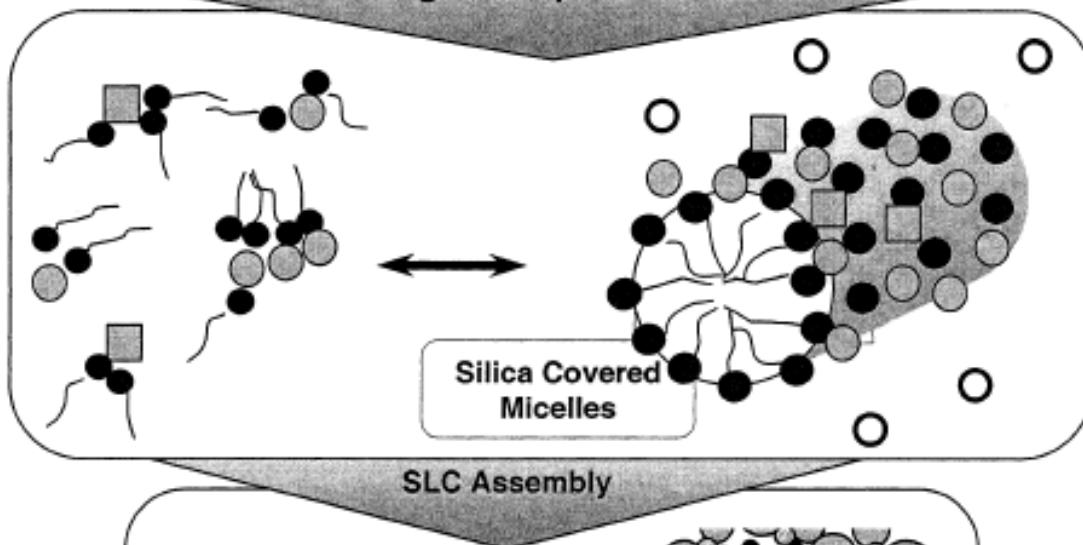
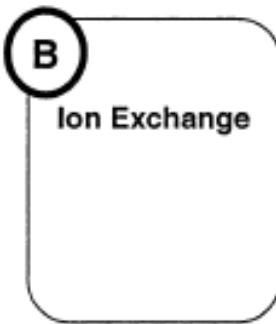
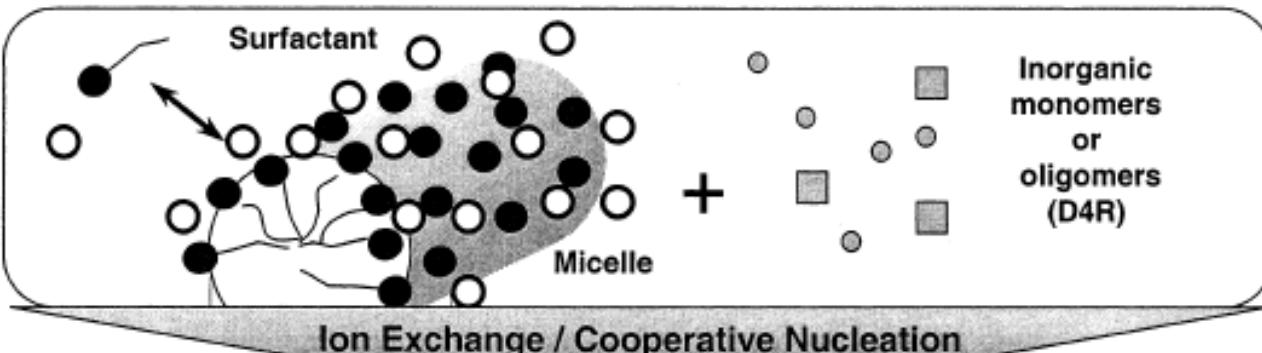
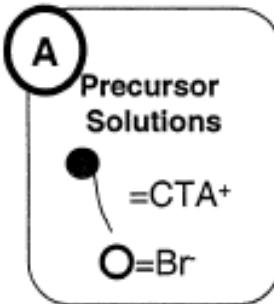


General Liquid Crystal Templating (LCT) Mechanism

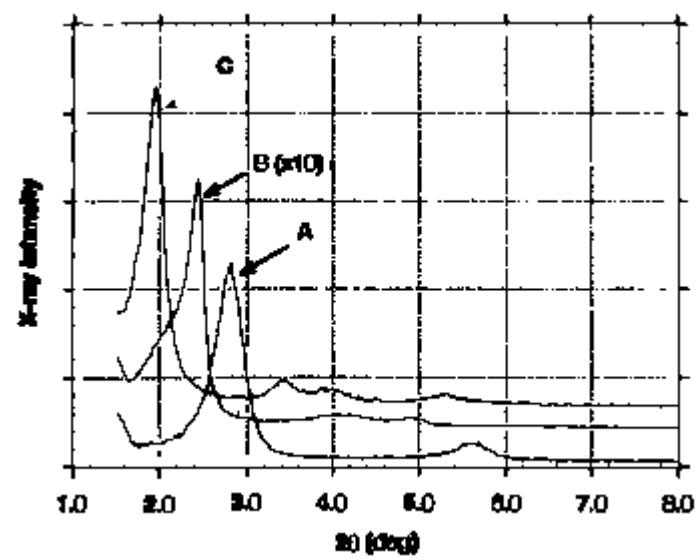
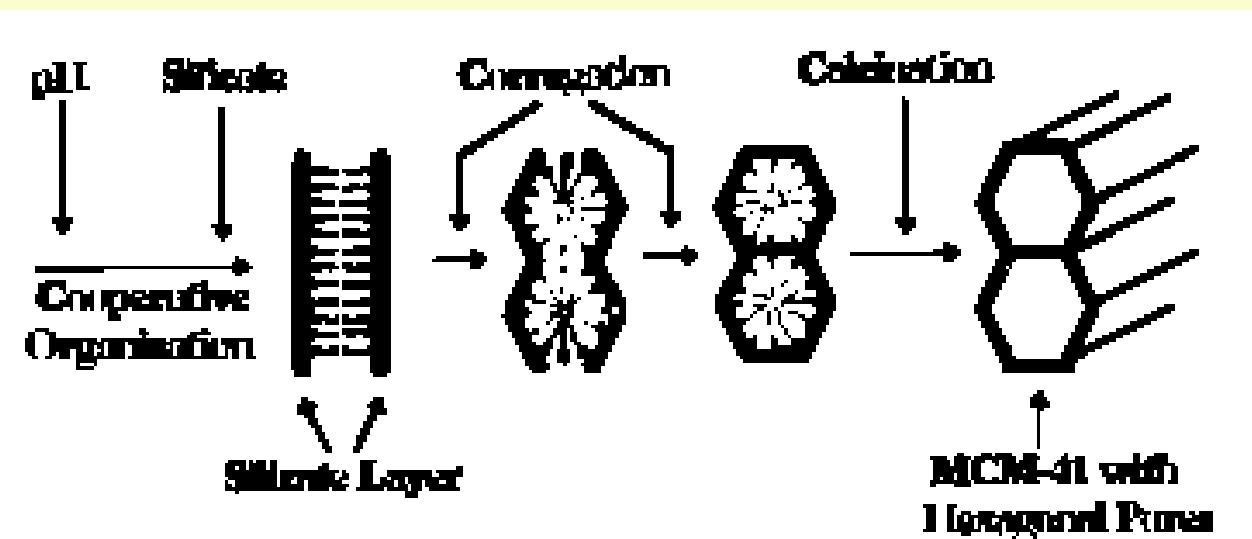


SLC Silicatropic Liquid Crystals

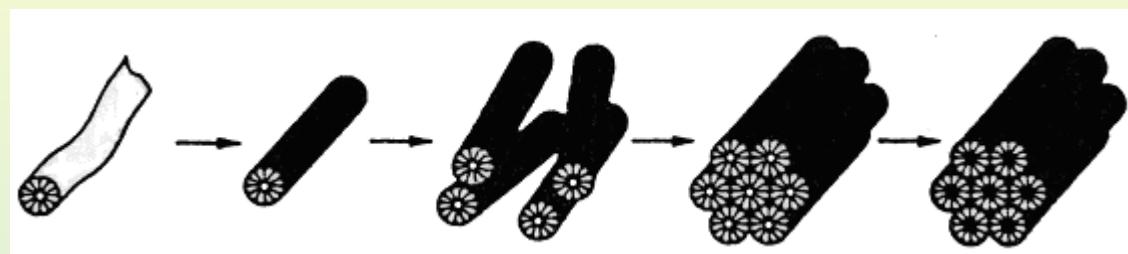




Lamellar to Hexagonal Transformation



Silicate Rod Assembly



- Electrostatic interactions

a) $\mathbf{S}^+\mathbf{I}^-$

\mathbf{I} = silicate

\mathbf{S} = trimethylammonium



b) $\mathbf{S}\mathbf{I}^+$

\mathbf{I} = Fe^{2+} , Fe^{3+} , Co^{2+} , Ni^{2+} ,
 Mg^{2+} , Mn^{2+} , Pb^{2+} , Al^{3+}

\mathbf{S} = sulfonane

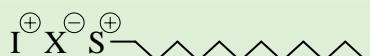


c) $\mathbf{S}^+\mathbf{X}\mathbf{I}^+$

\mathbf{I} = silicate – polyelectrolyte
positive charge

\mathbf{X} = Cl

\mathbf{S} = trimethylammonium



d) $\mathbf{S}\mathbf{M}^+\mathbf{I}^-$

\mathbf{I} = aluminate

\mathbf{M} = Na

\mathbf{S} = phosphate



- **Hydrogen Bond**

a) S^0I^0

I = silicate

S = ammine



b) N^0I^0

I = silicate

N = polyethylenoxide



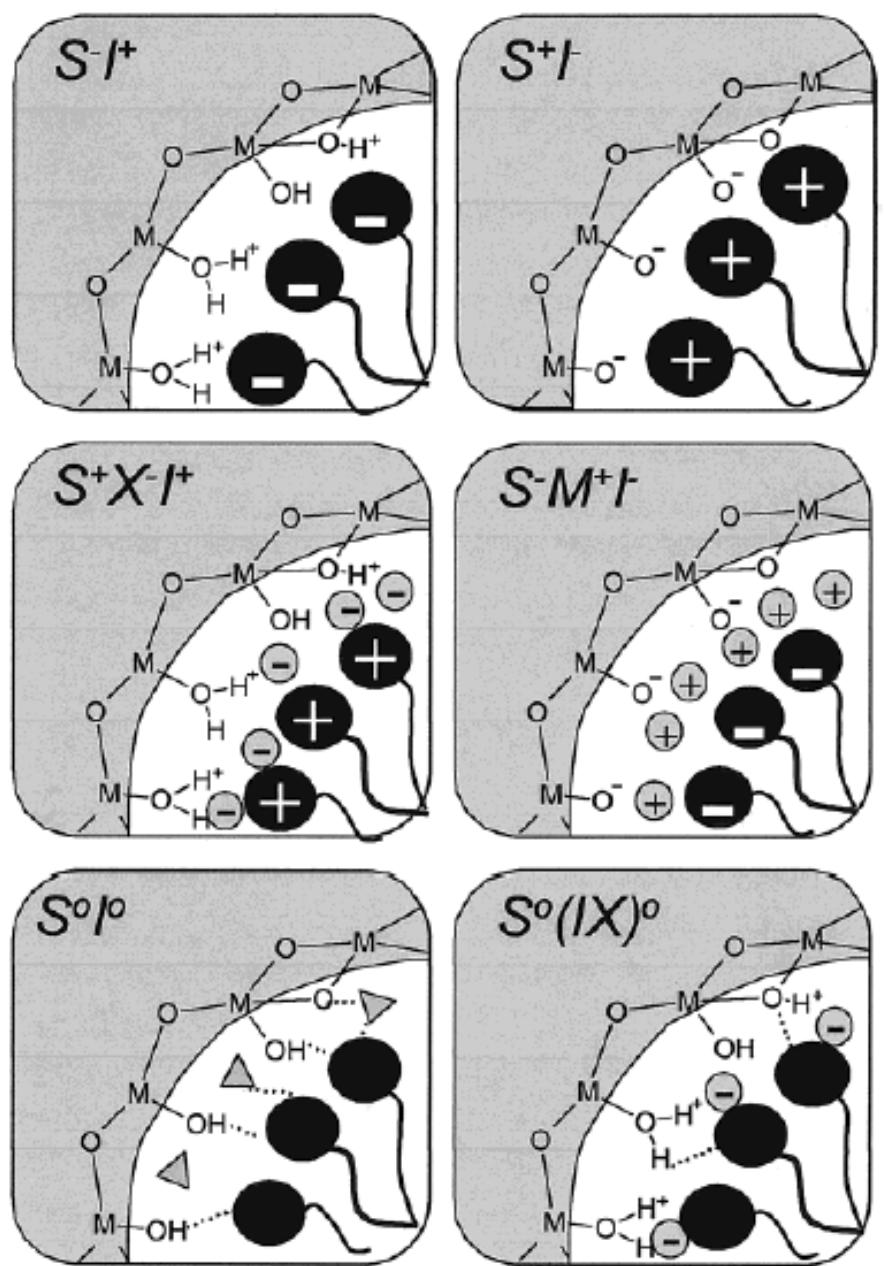
- **Covalent Bond**

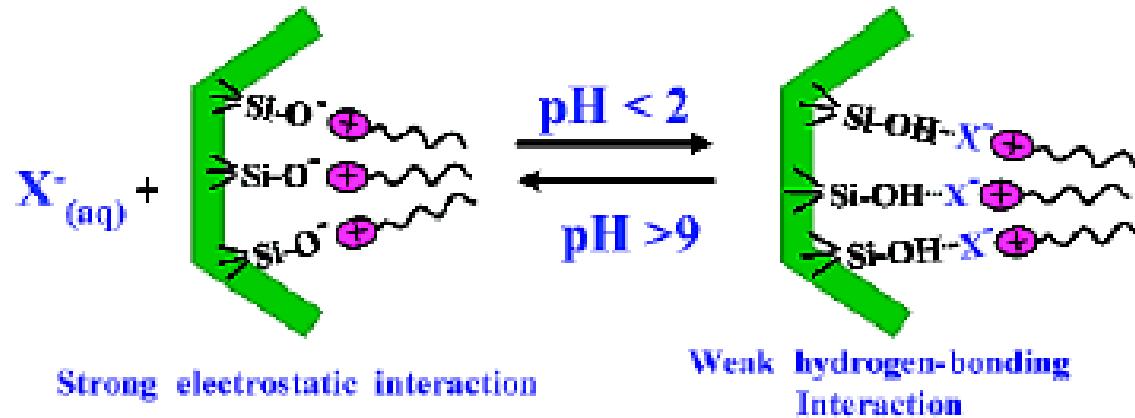
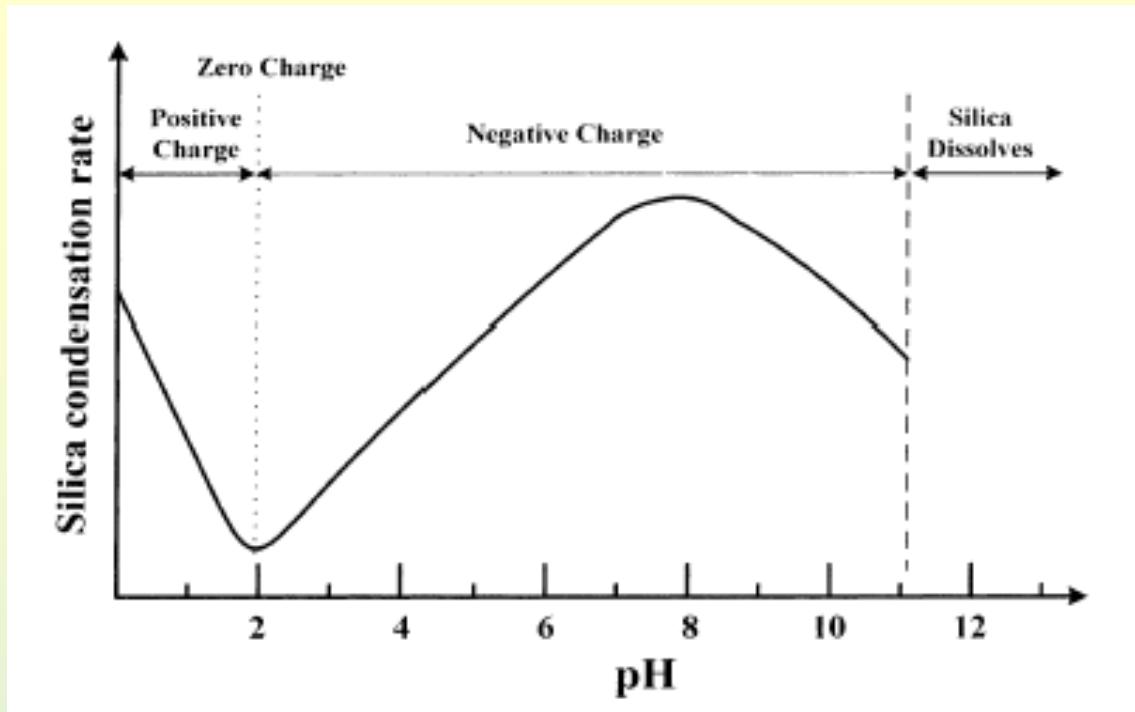
a) S-I

I = niobate, tantalate

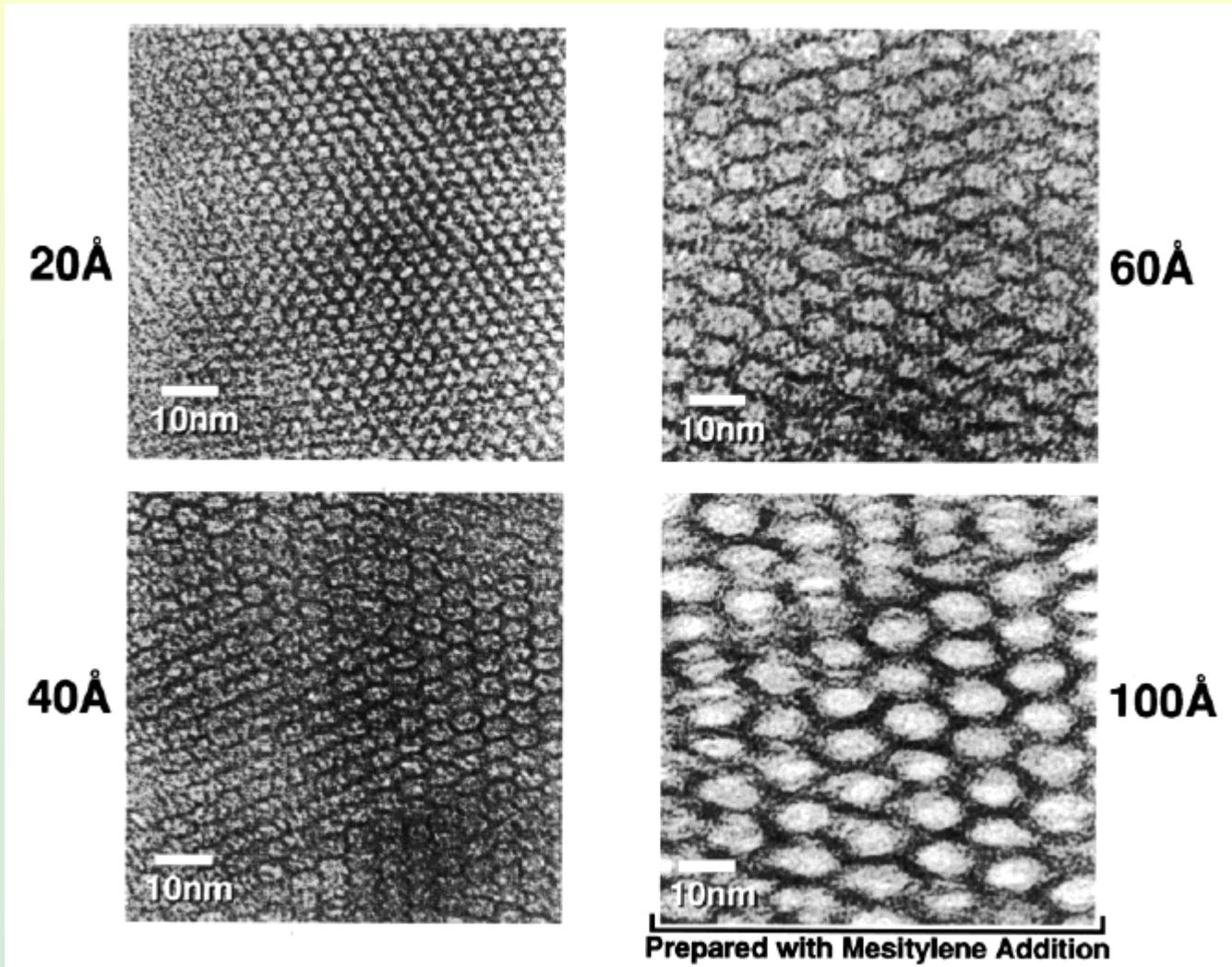
S = ammine



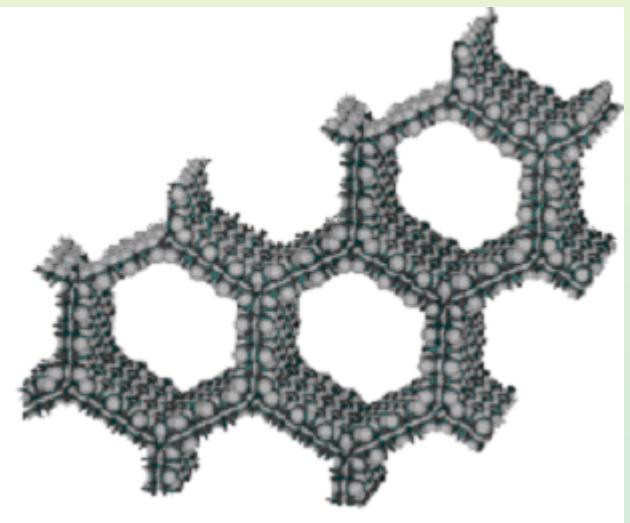
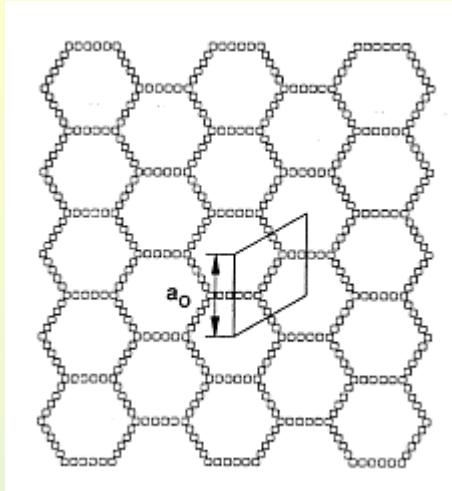
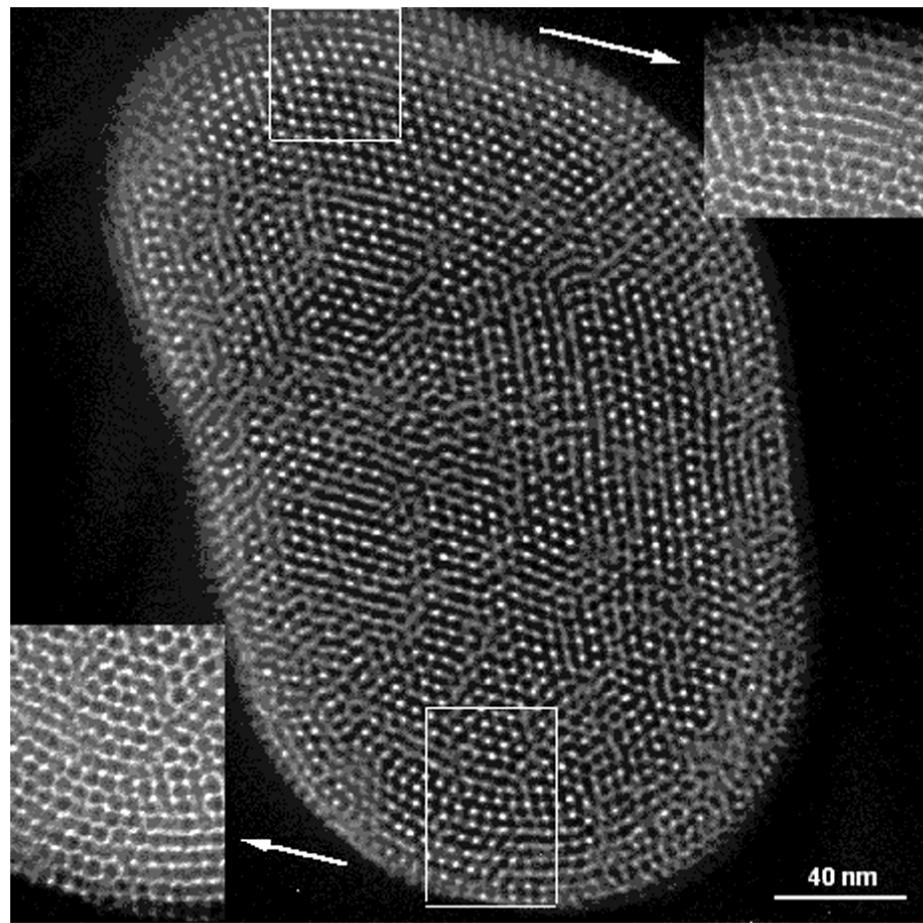




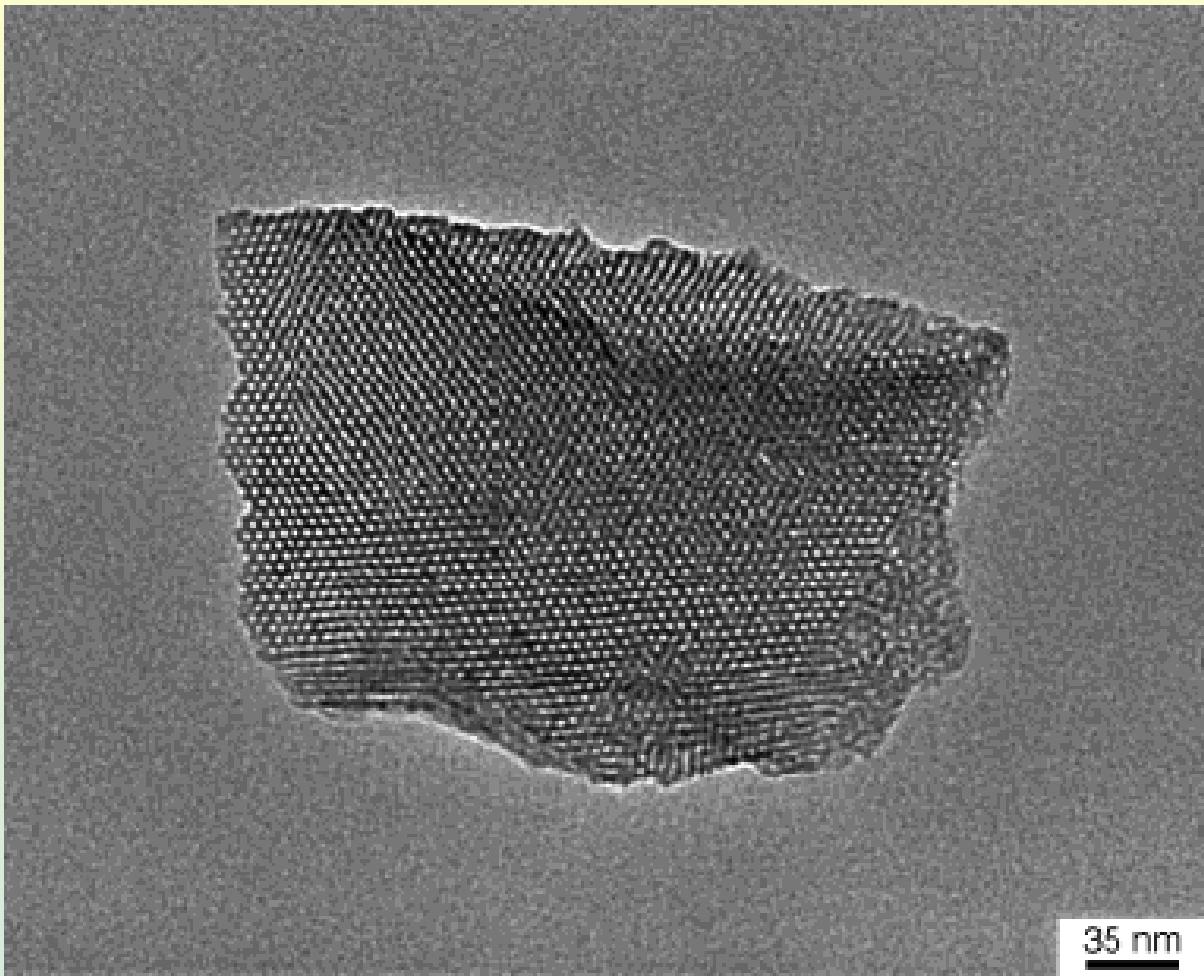
MCM-41



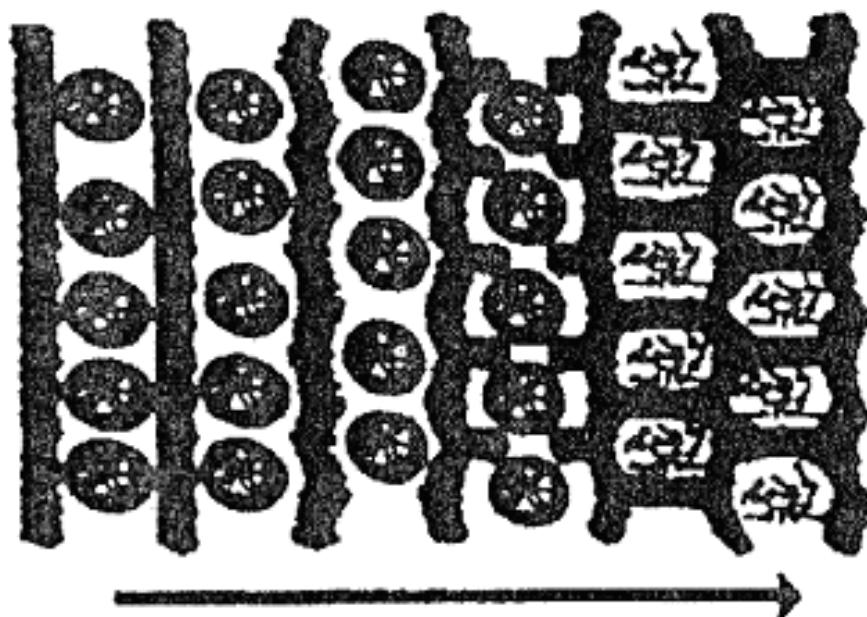
TEM micrograph of hexagonal molecular sieve



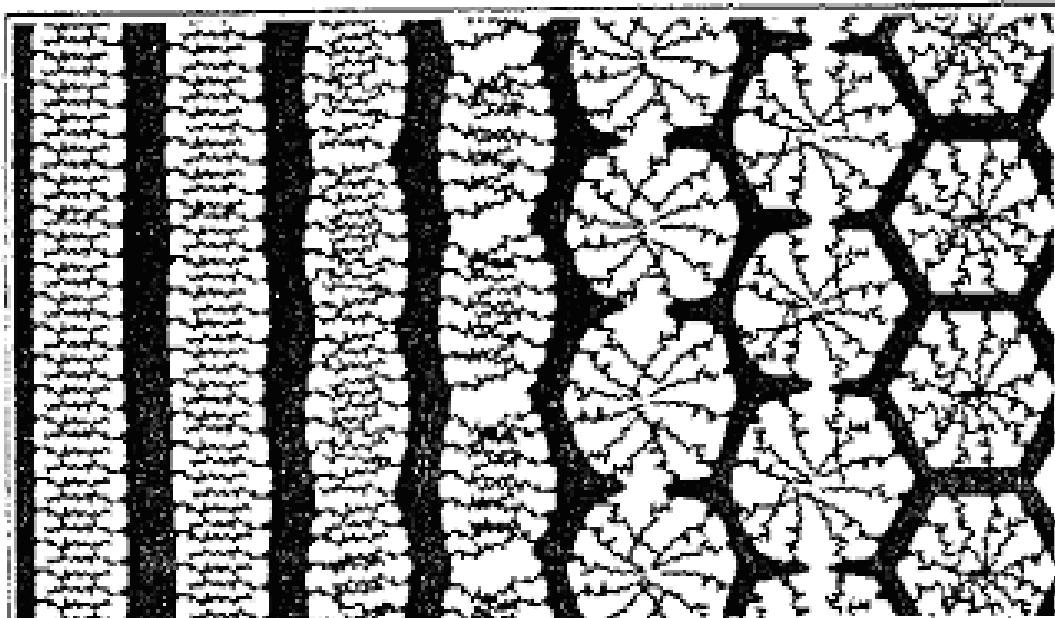
TEM image of the Pd-grafted mesoporous silicate material



Silicate Layer Puckering

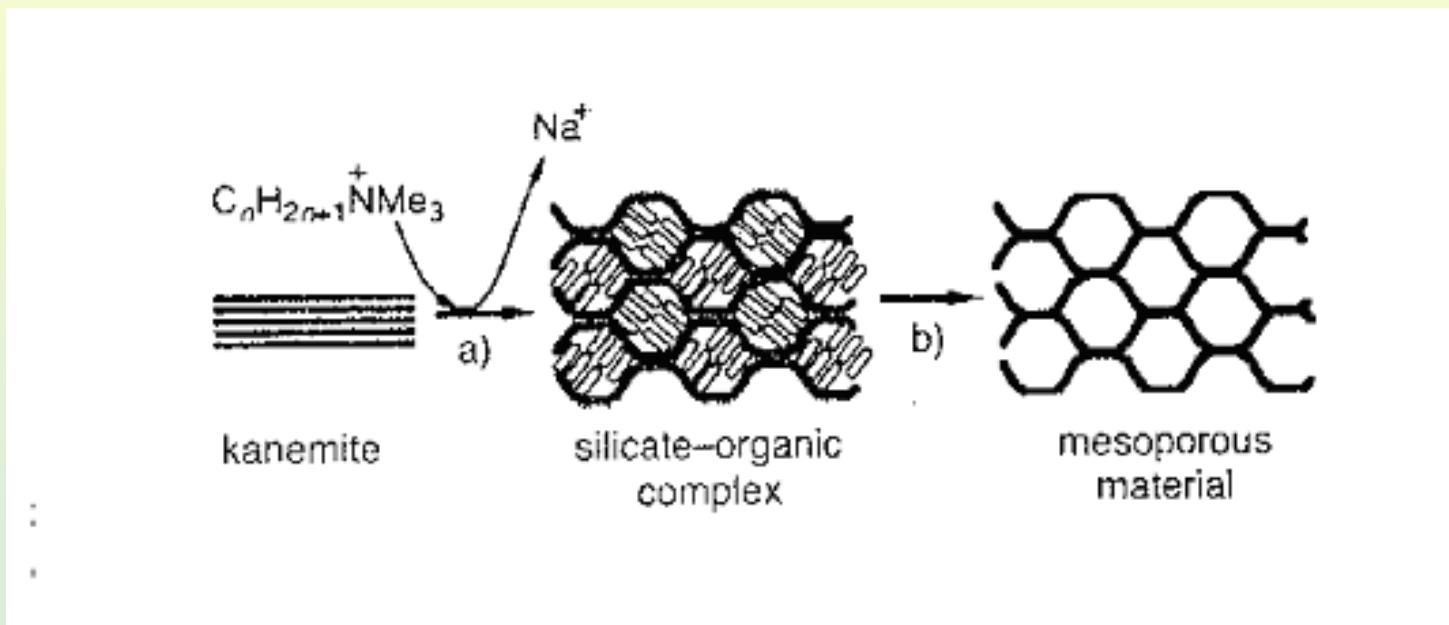


Charge Density Matching

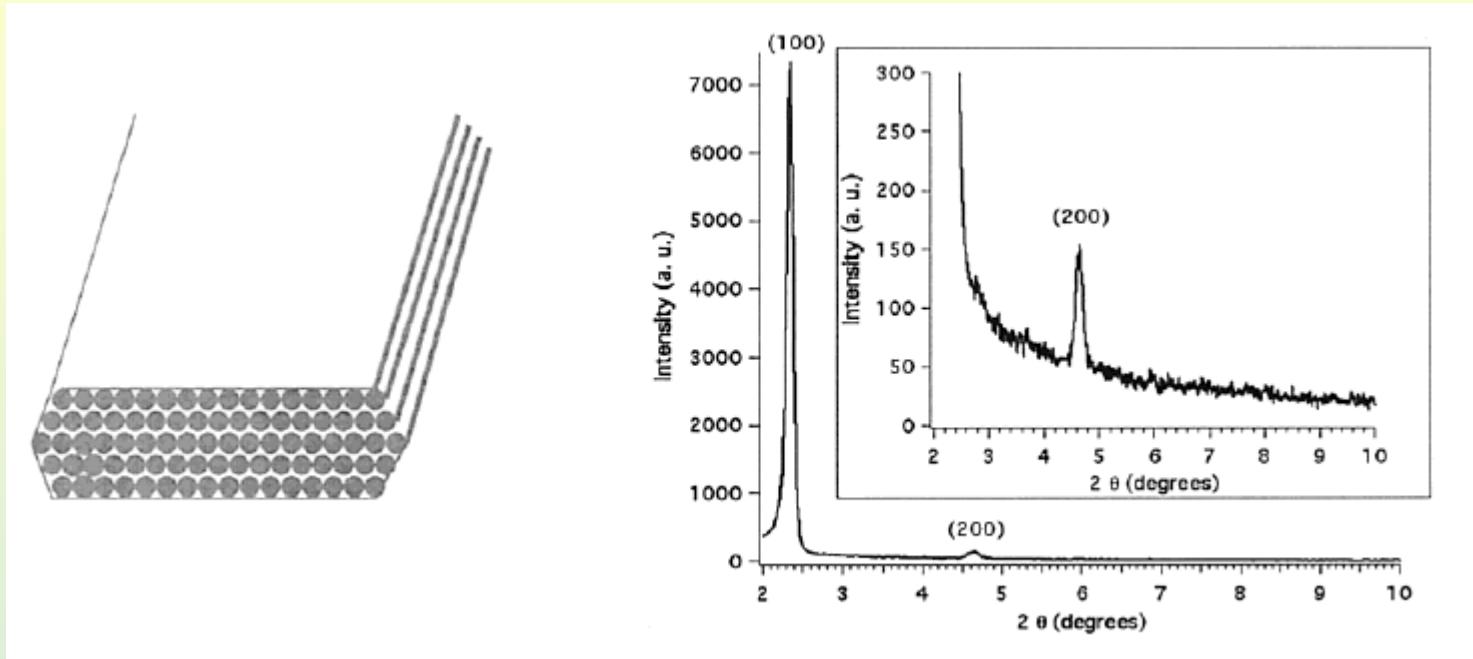


■ SiO_2 \rightarrow

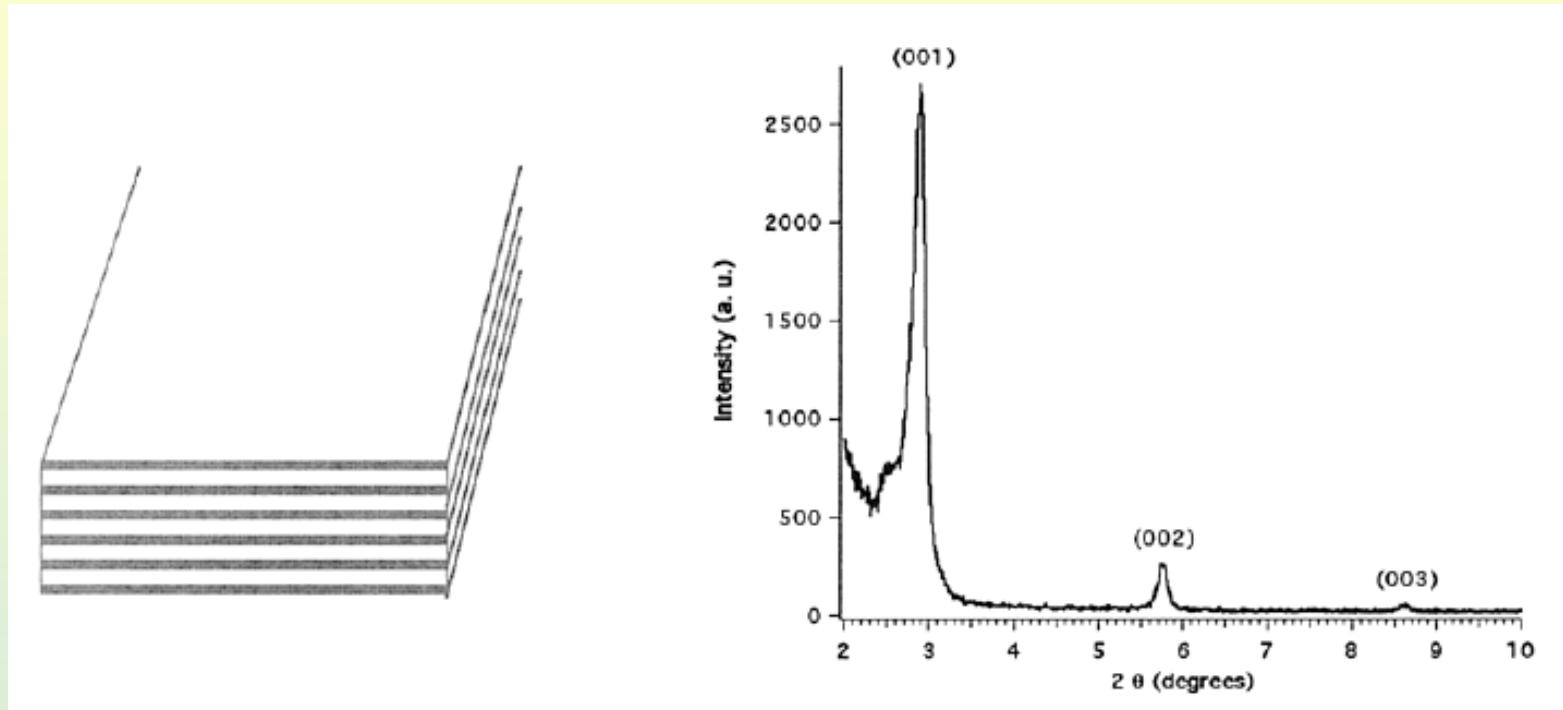
Folding Sheets

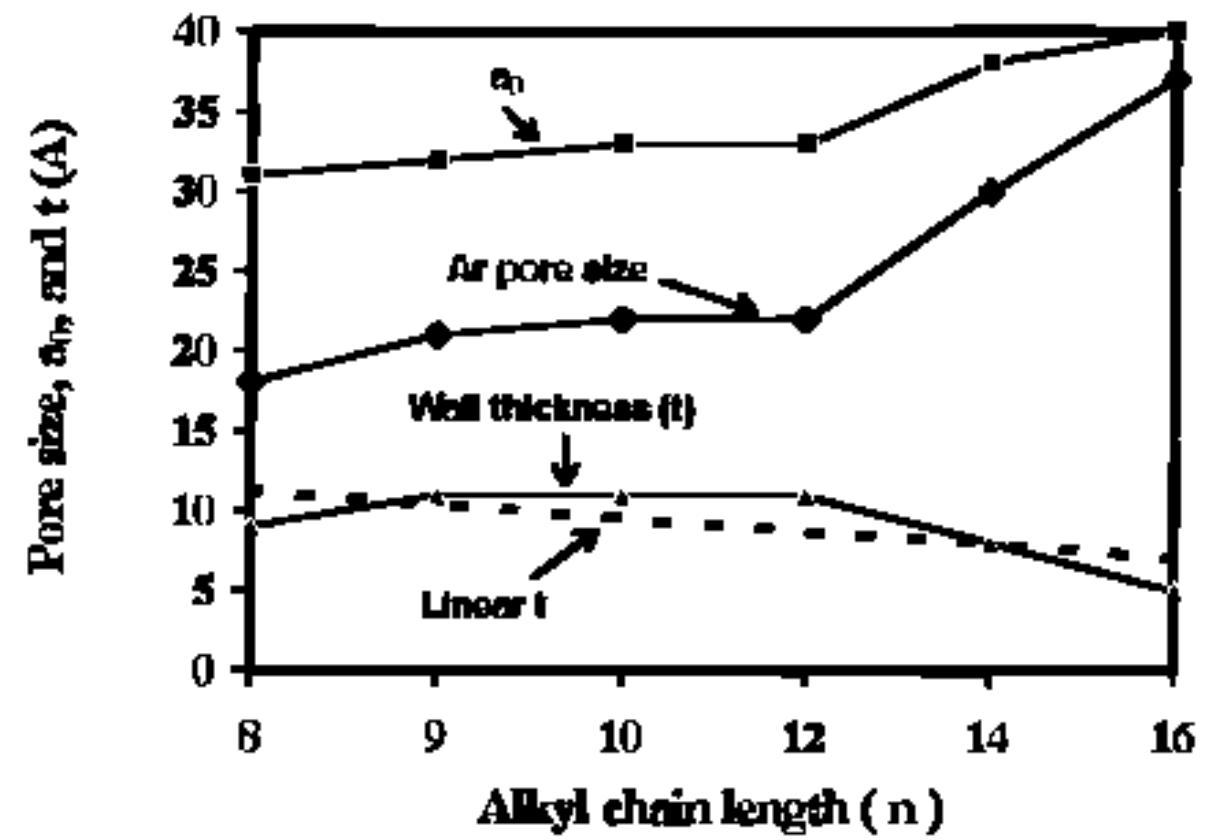


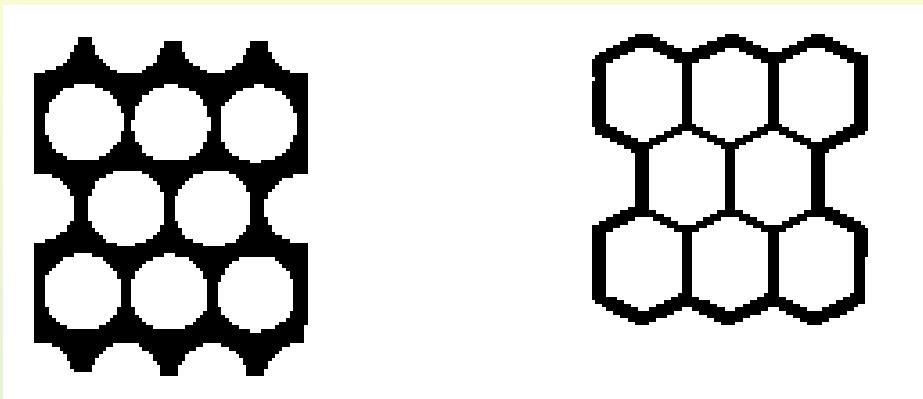
XRD of hexagonal MCM-41



XRD of lamellar MCM-50



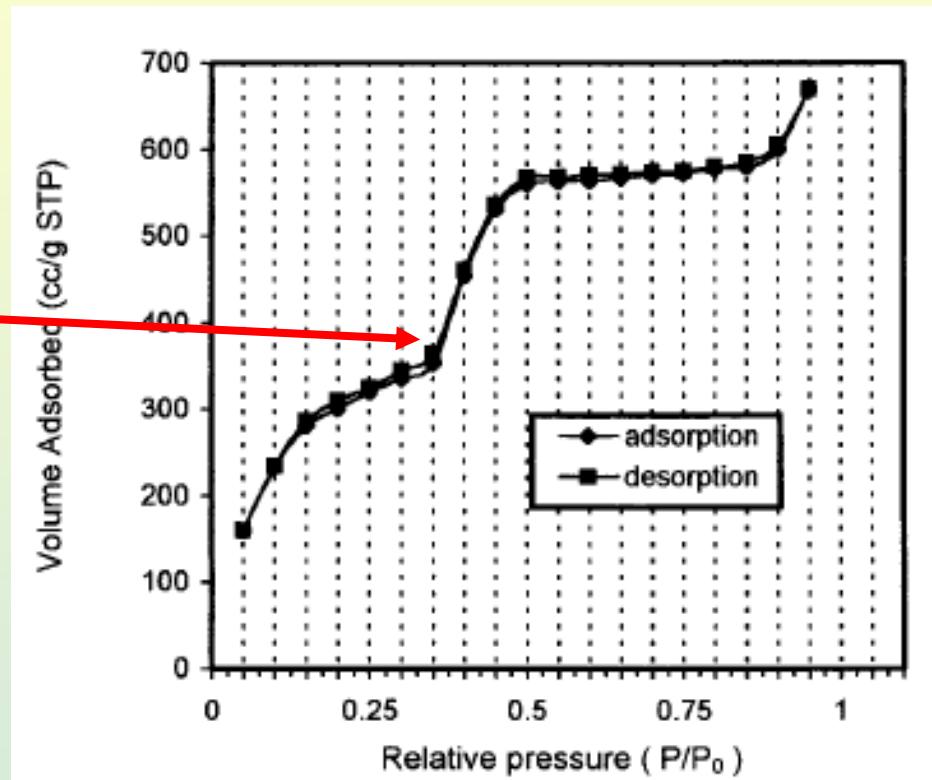




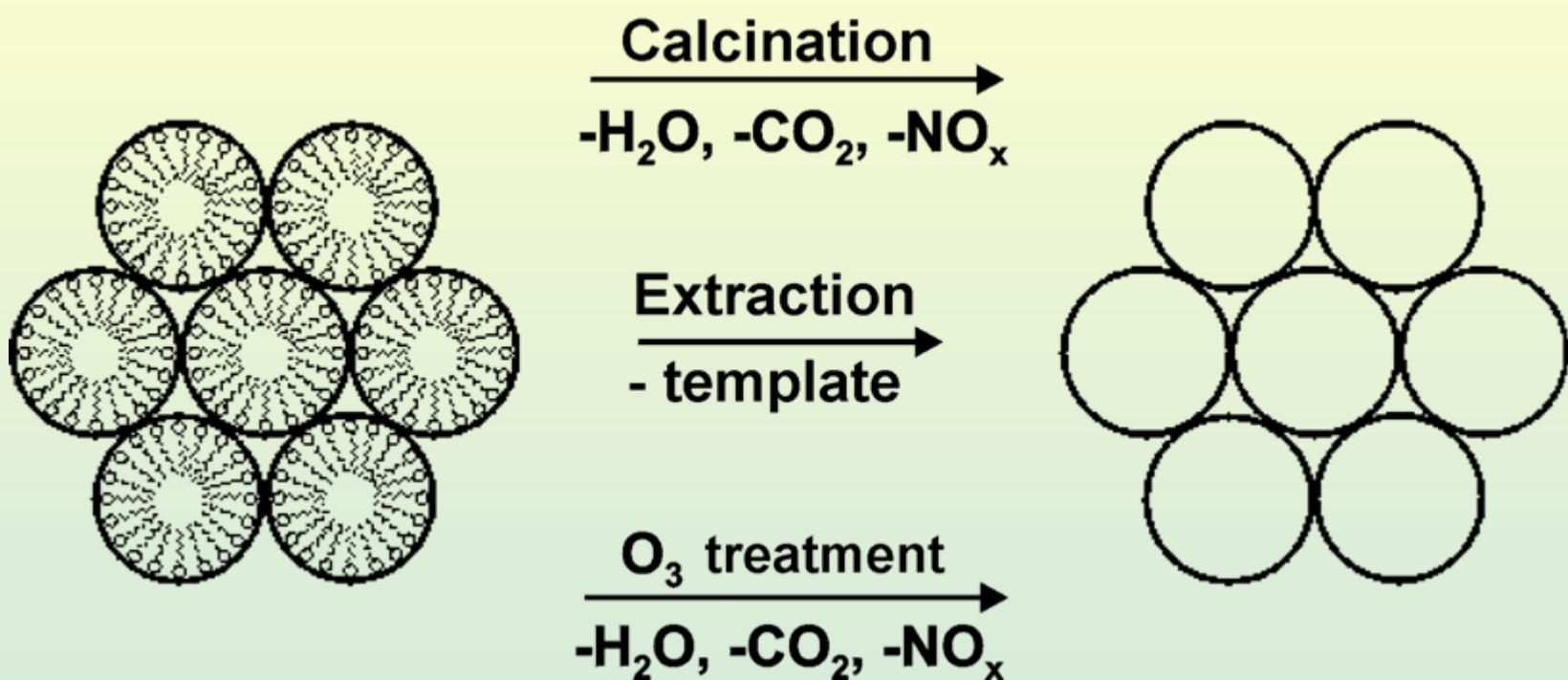
$$a_0 = \frac{2d_{100}}{\sqrt{3}}$$

Gas Adsorption Isotherms

Pore filling



Template Removal



Mesoporous Platinum Metal

$\text{H}_2[\text{PtCl}_6]$ or $(\text{NH}_4)_2[\text{PtCl}_6]$

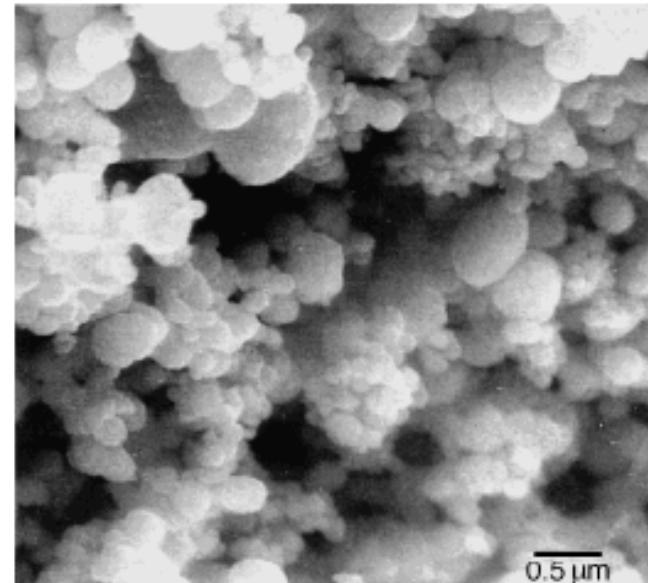
$\text{C}_{16}(\text{EO})_8$

Assembly of liquid crystalline phase

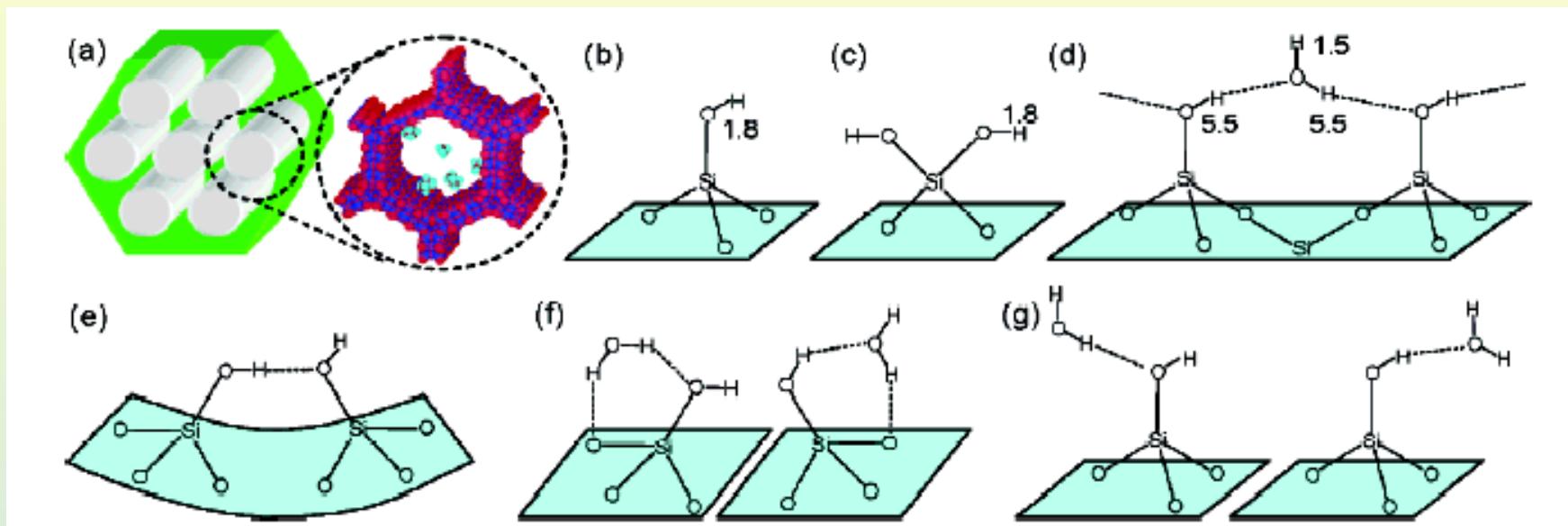
Reductants: Fe, Zn, Hg, NH_2NH_2

Washed with acetone, water, HCl

SEM (upper) and TEM (lower)
images of mesoporous Pt metal
show particles 90-500 nm in
diameter and a pore diameter of
30 Å and a pore wall thickness of
30 Å.



Surface Silanols in MCM-41 Pores



Chemistry inside the Pores

