## MEAN ACTIVITY COEFFICIENTS OF ELECTROLYTES AS A FUNCTION OF CONCENTRATION

The mean activity coefficient  $\gamma$  of an electrolyte  $X_a Y_b$  is defined

values refer to a temperature of 25  $^{\circ}$ C. Substances are arranged in alphabetical order by formula.

$$\gamma = (\gamma_+^a \gamma_-^b)^{1/(a+b)}$$

## where $\gamma_+$ and $\gamma_-$ are activity coefficients of the individual ions (which cannot be directly measured). This table gives the mean activity coefficients of about 100 electrolytes in aqueous solution as a function of concentration, expressed in molality terms. All

## References

- 1. Hamer, W. J., and Wu, Y. C., J. Phys. Chem. Ref. Data, 1, 1047, 1972.
- Staples, B. R., J. Phys. Chem. Ref. Data, 6, 385, 1977; 10, 767, 1981; 10, 779, 1981.
- 3. Goldberg, R. N. et al., *I. Phys. Chem. Ref. Data*, 7, 263, 1978; 8, 923, 1979; 8, 1005, 1979; 10, 1, 1981; 10, 671, 1981.

Mean Activity Coefficient at 25 °C								
$m/\text{mol kg}^{-1}$	$AgNO_{_3}$	BaBr <sub>2</sub>	BaCl	BaI	CaBr <sub>2</sub>	CaCl	Cal	
0.001	0.964	0.881	0.887	0.890	0.890	0.888	0.890	
0.002	0.950	0.850	0.849	0.853	0.853	0.851	0.853	
0.005	0.924	0.785	0.782	0.792	0.791	0.787	0.791	
0.010	0.896	0.727	0.721	0.737	0.735	0.727	0.736	
0.020	0.859	0.661	0.653	0.678	0.674	0.664	0.677	
0.050	0.794	0.573	0.559	0.600	0.594	0.577	0.600	
0.100	0.732	0.517	0.492	0.551	0.540	0.517	0.552	
0.200	0.656	0.463	0.436	0.520	0.502	0.469	0.524	
0.500	0.536	0.435	0.391	0.536	0.500	0.444	0.554	
1.000	0.430	0.470	0.393	0.664	0.604	0.495	0.729	
2.000	0.316	0.654		1.242	1.125	0.784		
5.000	0.181				18.7	5.907		
10.000	0.108					43.1		
15.000	0.085							
m/mol kg <sup>-1</sup>	Cd(NO <sub>2</sub> ) <sub>2</sub>	Cd(NO <sub>3</sub> ) <sub>2</sub>	CoBr <sub>2</sub>	CoCl <sub>2</sub>	CoI <sub>2</sub>	Co(NO <sub>3</sub> ) <sub>2</sub>	CsBr	
0.001	0.881	0.888	0.890	0.889	0.887	0.888	0.965	
0.002	0.837	0.851	0.854	0.852	0.849	0.850	0.951	
0.005	0.759	0.787	0.794	0.789	0.783	0.786	0.925	
0.010	0.681	0.728	0.740	0.732	0.724	0.728	0.898	
0.020	0.589	0.664	0.681	0.670	0.661	0.663	0.864	
0.050	0.451	0.576	0.605	0.586	0.582	0.576	0.806	
0.100	0.344	0.515	0.556	0.528	0.540	0.516	0.752	
0.200	0.247	0.465	0.523	0.483	0.527	0.469	0.691	
0.500	0.148	0.428	0.538	0.465	0.596	0.446	0.605	
1.000	0.098	0.437	0.685	0.532	0.845	0.492	0.540	
2.000	0.069	0.517	1.421	0.864	2.287	0.722	0.485	
5.000	0.054		13.9		55.3	3.338	0.454	
10.000					196			
m/mol kg <sup>-1</sup>	CsCl	CsF	CsI	CsNO <sub>3</sub>	CsOH	Cs <sub>2</sub> SO <sub>4</sub>	CuBr <sub>2</sub>	
0.001	0.965	0.965	0.965	0.964	0.966	0.885	0.889	
0.002	0.951	0.952	0.951	0.951	0.953	0.845	0.853	
0.005	0.925	0.929	0.925	0.924	0.930	0.775	0.791	
0.010	0.898	0.905	0.898	0.897	0.906	0.709	0.735	
0.020	0.864	0.876	0.863	0.860	0.878	0.634	0.674	
0.050	0.805	0.830	0.804	0.796	0.836	0.526	0.594	
0.100	0.751	0.792	0.749	0.733	0.802	0.444	0.541	
0.200	0.691	0.755	0.688	0.655	0.772	0.369	0.504	
0.500	0.607	0.721	0.601	0.529	0.755	0.285	0.503	
1.000	0.546	0.726	0.534	0.421	0.782	0.233	0.591	
2.000	0.496	0.803	0.470				0.859	
5.000	0.474							
10.000	0.508							

$m/\mathrm{mol}~\mathrm{kg}^{-1}$	CuCl <sub>2</sub>	Cu(ClO <sub>4</sub> ) <sub>2</sub>	Cu(NO <sub>3</sub> ) <sub>2</sub>	FeCl <sub>2</sub>	HBr	HCl	HClO <sub>4</sub>
0.001	0.887	0.890	0.888	0.888	0.966	0.965	0.966
0.002	0.849	0.854	0.851	0.850	0.953	0.952	0.953
0.005	0.783	0.795	0.787	0.785	0.930	0.929	0.929
0.010	0.722	0.741	0.729	0.725	0.907	0.905	0.906
0.020	0.654	0.685	0.664	0.659	0.879	0.876	0.878
0.050	0.561	0.613	0.577	0.570	0.837	0.832	0.836
0.100	0.495	0.572	0.516	0.509	0.806	0.797	0.803
0.200	0.441	0.553	0.466	0.462	0.783	0.768	0.776
0.500	0.401	0.617	0.431	0.443	0.790	0.759	0.769
1.000	0.405	0.892	0.456	0.500	0.872	0.811	0.826
2.000	0.453	2.445	0.615	0.782	1.167	1.009	1.055
5.000	0.601		2.083		3.800	2.380	3.100
10.000					33.4	10.4	30.8
15.000							323
/ 11 -1	HE	111	IDIO	H.CO	I/D	VCNC	I/Cl
m/mol kg <sup>-1</sup>	HF	HI	HNO <sub>3</sub>	H <sub>2</sub> SO <sub>4</sub>	KBr	KCNS	KCl
0.001	0.551	0.966	0.965	0.804	0.965	0.965	0.965
0.002	0.429	0.953	0.952	0.740	0.952	0.951	0.951
0.005	0.302	0.931	0.929	0.634	0.927	0.927	0.927
0.010	0.225	0.909	0.905	0.542	0.902	0.901	0.901
0.020	0.163	0.884	0.875	0.445	0.870	0.869	0.869
0.050	0.106	0.847	0.829	0.325	0.817	0.815	0.816
0.100	0.0766	0.823	0.792	0.251	0.771	0.768	0.768
0.200	0.0550	0.811	0.756	0.195	0.722	0.716	0.717
0.500	0.0352	0.845	0.725	0.146	0.658	0.647	0.649
1.000	0.0249	0.969	0.730	0.125	0.617	0.598	0.604
2.000	0.0175	1.363	0.788	0.119	0.593	0.556	0.573
5.000	0.0110	4.760	1.063	0.197	0.626	0.525	0.593
10.000	0.0085	49.100	1.644	0.527			
15.000	0.0077		2.212	1.077			
20.000	0.0075		2.607	1.701			
m/mol kg <sup>-1</sup>	KClO <sub>3</sub>	K <sub>2</sub> CrO <sub>4</sub>	KF	KH <sub>2</sub> PO <sub>4</sub> *	K <sub>2</sub> HPO <sub>4</sub> **	KI	KNO <sub>3</sub>
0.001	0.965	0.886	0.965	0.964	0.886	0.965	0.964
0.002	0.951	0.847	0.952	0.950	0.847	0.952	0.950
0.005	0.926	0.779	0.927	0.924	0.779	0.927	0.924
0.010	0.899	0.715	0.902	0.896	0.715	0.902	0.896
0.020	0.865	0.643	0.870	0.859	0.643	0.871	0.860
0.050	0.805	0.539	0.818	0.793	0.538	0.820	0.797
0.100	0.749	0.460	0.773	0.730	0.457	0.776	0.735
0.200	0.681	0.385	0.726	0.652	0.379	0.731	0.662
0.500	0.569	0.296	0.670	0.529	0.283	0.676	0.546
1.000	0.507	0.239	0.645	0.422	0.200	0.646	0.444
2.000		0.199	0.658	0.122		0.638	0.332
5.000		0.177	0.871			0.050	0.552
10.000			1.715				
15.000			3.120				
$m/\mathrm{mol}\ \mathrm{kg}^{\scriptscriptstyle{-1}}$	КОН	$K_2SO_4$	LiBr	LiCl	$LiClO_4$	LiI	LiNO <sub>3</sub>
0.001	0.965	0.885	0.965	0.965	0.966	0.966	0.965
0.002	0.952	0.844	0.952	0.952	0.953	0.953	0.952
0.005	0.927	0.772	0.929	0.928	0.931	0.930	0.928
0.010	0.902	0.704	0.905	0.904	0.908	0.908	0.904
0.020	0.871	0.625	0.877	0.874	0.882	0.882	0.874
0.050	0.821	0.511	0.832	0.827	0.843	0.843	0.827
0.100	0.779	0.424	0.797	0.789	0.815	0.817	0.788
0.200	0.740	0.343	0.767	0.756	0.795	0.802	0.753
0.500	0.710	0.251	0.754	0.739	0.806	0.824	0.726
1.000	0.733		0.803	0.775	0.887	0.912	0.743
2.000	0.860		1.012	0.924	1.161	1.197	0.837
5.000	1.697		2.696	2.000			1.298

m/mol kg <sup>-1</sup>	КОН	$K_2SO_4$	LiBr	LiCl	LiClO <sub>4</sub>	LiI	LiNO <sub>3</sub>
10.000	6.110		20.0	9.600			2.500
15.000	19.9		147	30.9			3.960
20.000	46.4		486				4.970
m/mol kg <sup>-1</sup>	LiOH	$\text{Li}_{2}\text{SO}_{4}$	$MgBr_2$	$\mathbf{MgCl}_{2}$	$MgI_2$	MnBr <sub>2</sub>	MnCl <sub>2</sub>
0.001	0.964	0.887	0.889	0.889	0.889	0.889	0.888
0.002	0.950	0.847	0.852	0.852	0.853	0.853	0.850
0.005	0.923	0.780	0.790	0.790	0.791	0.791	0.786
0.010	0.895	0.716	0.733	0.734	0.736	0.735	0.727
0.020	0.858	0.645	0.672	0.672	0.677	0.674	0.662
0.050	0.794	0.544	0.593	0.590	0.602	0.595	0.574
0.100	0.735	0.469	0.543	0.535	0.556	0.543	0.513
0.200	0.668	0.400	0.512	0.493	0.535	0.508	0.464
0.500	0.579	0.325	0.540	0.485	0.594	0.519	0.437
1.000	0.522	0.284	0.715	0.577	0.858	0.650	0.477
2.000	0.484	0.270	1.590	1.065	2.326	1.224	0.661
5.000	0.493		36.1	14.40	109.8	6.697	1.539
m/mol kg <sup>-1</sup>	Mn(ClO <sub>4</sub> ) <sub>2</sub>	NH <sub>4</sub> Cl	NH <sub>4</sub> ClO <sub>4</sub>	(NH <sub>4</sub> ) <sub>2</sub> HPO <sub>4</sub> **	NH <sub>4</sub> NO <sub>3</sub>	NaBr	NaBrO <sub>3</sub>
0.001	0.892	0.965	0.964	0.882	0.964	0.965	0.965
0.002	0.858	0.952	0.950	0.839	0.951	0.952	0.951
0.005	0.801	0.927	0.924	0.763	0.925	0.928	0.926
0.010	0.752	0.901	0.895	0.688	0.897	0.903	0.900
0.020	0.700	0.869	0.859	0.600	0.862	0.873	0.867
0.050	0.637	0.816	0.794	0.469	0.801	0.824	0.811
0.100	0.604	0.769	0.734	0.367	0.744	0.783	0.759
0.200	0.596	0.718	0.663	0.273	0.678	0.742	0.698
0.500	0.686	0.649	0.560	0.171	0.582	0.697	0.605
1.000	1.030	0.603	0.479	0.114	0.502	0.687	0.528
2.000	3.072	0.569	0.399	0.074	0.419	0.730	0.449
5.000		0.563			0.303	1.083	
10.000					0.220		
15.000					0.179		
20.000					0.154		
m/mol kg <sup>-1</sup>	Na <sub>2</sub> CO <sub>3</sub>	NaCl	NaClO <sub>3</sub>	NaClO <sub>4</sub>	Na <sub>2</sub> CrO <sub>4</sub>	NaF	Na <sub>2</sub> HPO <sub>4</sub> *
0.001	0.887	0.965	0.965	0.965	0.887	0.965	0.887
0.002	0.847	0.952	0.952	0.952	0.849	0.951	0.848
0.005	0.780	0.928	0.927	0.928	0.783	0.926	0.780
0.010	0.716	0.903	0.902	0.903	0.722	0.901	0.717
0.020	0.644	0.872	0.870	0.872	0.653	0.868	0.644
0.050	0.541	0.822	0.818	0.821	0.554	0.813	0.539
0.100	0.462	0.779	0.771	0.777	0.479	0.764	0.456
0.200	0.385	0.734	0.719	0.729	0.406	0.710	0.373
0.500	0.292	0.681	0.646	0.668	0.318	0.633	0.266
1.000	0.229	0.657	0.590	0.630	0.261	0.573	0.191
2.000	0.182	0.668	0.537	0.608	0.231		0.133
5.000		0.874		0.648			
m/mol kg <sup>-1</sup>	NaI	NaNO <sub>3</sub>	NaOH	Na <sub>2</sub> SO <sub>3</sub>	Na <sub>2</sub> SO <sub>4</sub>	Na <sub>2</sub> WO <sub>4</sub>	NiBr <sub>2</sub>
0.001	0.965	0.965	0.965	0.887	0.886	0.886	0.889
0.002	0.952	0.951	0.952	0.847	0.846	0.846	0.853
0.005	0.928	0.926	0.927	0.779	0.777	0.777	0.791
0.010	0.904	0.900	0.902	0.716	0.712	0.712	0.735
0.020	0.874	0.866	0.870	0.644	0.637	0.638	0.675
0.050	0.827	0.810	0.819	0.540	0.529	0.534	0.596
0.100	0.789	0.759	0.775	0.462	0.446	0.457	0.546
0.200	0.753	0.701	0.731	0.386	0.366	0.388	0.514
0.500	0.722	0.617	0.685	0.296	0.268	0.320	0.535
1.000	0.734	0.550	0.674	0.237	0.204	0.291	0.692
2.000	0.823	0.480	0.714	0.196	0.155	0.291	1.476

<i>m</i> /mol kg <sup>-1</sup> 5.000	<b>NaI</b> 1.402	<b>NaNO</b> <sub>3</sub> 0.388	<b>NaOH</b> 1.076	$Na_2SO_3$	$\mathrm{Na_{2}SO_{4}}$	${\bf Na_2WO_4}$	$\mathbf{NiBr}_{_{2}}$
10.000	4.011	0.329	3.258				
15.000	1.011	0.527	9.796				
20.000			19.410				
20.000			17.110				
m/mol kg <sup>-1</sup>	NiCl <sub>2</sub>	Ni(ClO <sub>4</sub> ) <sub>2</sub>	Ni(NO <sub>3</sub> ) <sub>2</sub>	Pb(ClO <sub>4</sub> ) <sub>2</sub>	Pb(NO <sub>3</sub> ) <sub>2</sub>	RbBr	RbCl
0.001	0.889	0.891	0.889	0.889	0.882	0.965	0.965
0.002	0.852	0.855	0.851	0.851	0.840	0.951	0.951
0.005	0.789	0.797	0.787	0.787	0.764	0.926	0.926
0.010	0.732	0.745	0.730	0.729	0.690	0.900	0.900
0.020	0.669	0.690	0.666	0.666	0.604	0.866	0.867
0.050	0.584	0.621	0.581	0.580	0.476	0.811	0.811
0.100	0.527	0.582	0.524	0.522	0.379	0.760	0.761
0.200	0.482	0.567	0.481	0.476	0.291	0.705	0.707
0.500	0.465	0.639	0.467	0.458	0.195	0.630	0.633
1.000	0.538	0.946	0.528	0.516	0.136	0.578	0.583
2.000	0.915	2.812	0.797	0.799		0.535	0.546
5.000	4.785			4.043		0.514	0.544
10.000				33.8			
m/mol kg <sup>-1</sup>	RbF	RbI	RbNO <sub>3</sub>	$Rb_2SO_4$	$\mathbf{SrBr}_2$	SrCl,	SrI <sub>2</sub>
0.001	0.965	0.965	0.964	0.886	0.889	0.888	0.890
0.002	0.952	0.951	0.950	0.845	0.852	0.850	0.854
0.005	0.927	0.926	0.924	0.776	0.790	0.785	0.793
0.010	0.902	0.900	0.896	0.710	0.734	0.725	0.740
0.020	0.871	0.866	0.859	0.635	0.673	0.659	0.681
0.050	0.821	0.810	0.795	0.526	0.591	0.569	0.606
0.100	0.780	0.759	0.733	0.443	0.535	0.506	0.557
0.200	0.739	0.703	0.657	0.365	0.492	0.455	0.526
0.500	0.701	0.627	0.536	0.274	0.476	0.421	0.542
1.000	0.697	0.574	0.430	0.217	0.545	0.451	0.686
2.000	0.724	0.532	0.320		0.921	0.650	
5.000		0.517					
	$m/\mathrm{mol}\ \mathrm{kg}^{-1}$	UO,Cl,	UO <sub>2</sub> (NO <sub>3</sub> ) <sub>2</sub>	$\mathbf{ZnBr}_{2}$	ZnCl	ZnI,	
	0.001	0.888	0.888	0.890	0.887	0.893	
	0.002	0.851	0.849	0.854	0.847	0.859	
	0.005	0.787	0.784	0.794	0.781	0.804	
	0.010	0.729	0.726	0.741	0.719	0.757	
	0.020	0.666	0.663	0.683	0.652	0.708	
	0.050	0.583	0.583	0.606	0.561	0.644	
	0.100	0.529	0.535	0.553	0.499	0.601	
	0.200	0.493	0.509	0.515	0.447	0.574	
	0.500	0.501	0.532	0.516	0.384	0.635	
	1.000	0.601	0.673	0.558	0.330	0.836	
	2.000	0.948	1.223	0.578	0.283	1.062	
	5.000		3.020	0.788	0.342	1.546	
	10.000			2.317	0.876	4.698	
	15.000			5.381	1.914		
	20.000			7.965	2.968		
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<sup>\*</sup> The anion is  ${\rm H_2PO_4^{-1}}$ . \*\* The anion is  ${\rm HPO_4^{-2}}$ .