



FACULTY
OF SCIENCE
Masaryk University

Nutrients

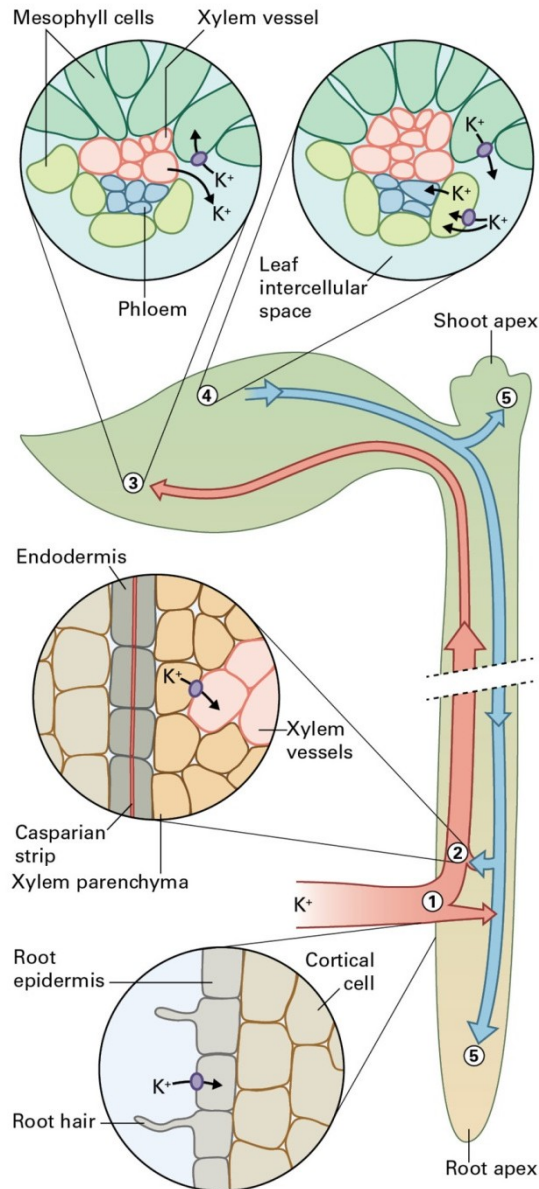


Essential mineral nutrients

Element	Chemical symbol	Concentration in dry material ($\mu\text{g g}^{-1}$)	Concentration in fresh tissue*
<i>Macronutrients</i>			
Nitrogen	N	15,000	71.4 mM
Potassium	K	10,000	17 mM
Calcium	Ca	5,000	8.3 mM
Magnesium	Mg	2,000	5.5 mM
Phosphorus	P	2,000	4.3 mM
Sulfur	S	1,000	2.1 mM
<i>Micronutrients</i>			
Chlorine	Cl	100	188 μM
Boron	B	20	123 μM
Iron	Fe	100	120 μM
Manganese	Mn	50	61 μM
Zinc	Zn	20	20.4 μM
Copper	Cu	6	6.2 μM
Molybdenum	Mo	0.1	0.07 μM
Nickel	Ni	0.005	0.006 μM

*Fresh weight concentrations were calculated by assuming a 15:1 fresh weight–dry weight ratio.

Potassium



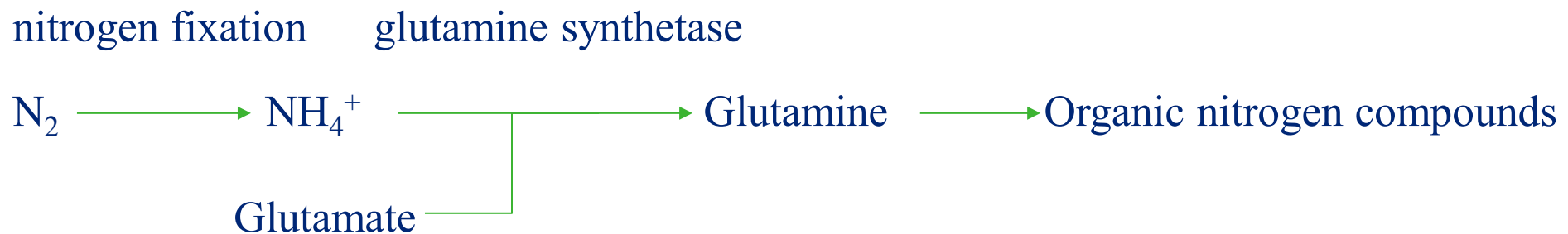
K⁺ is the most abundant cellular cation

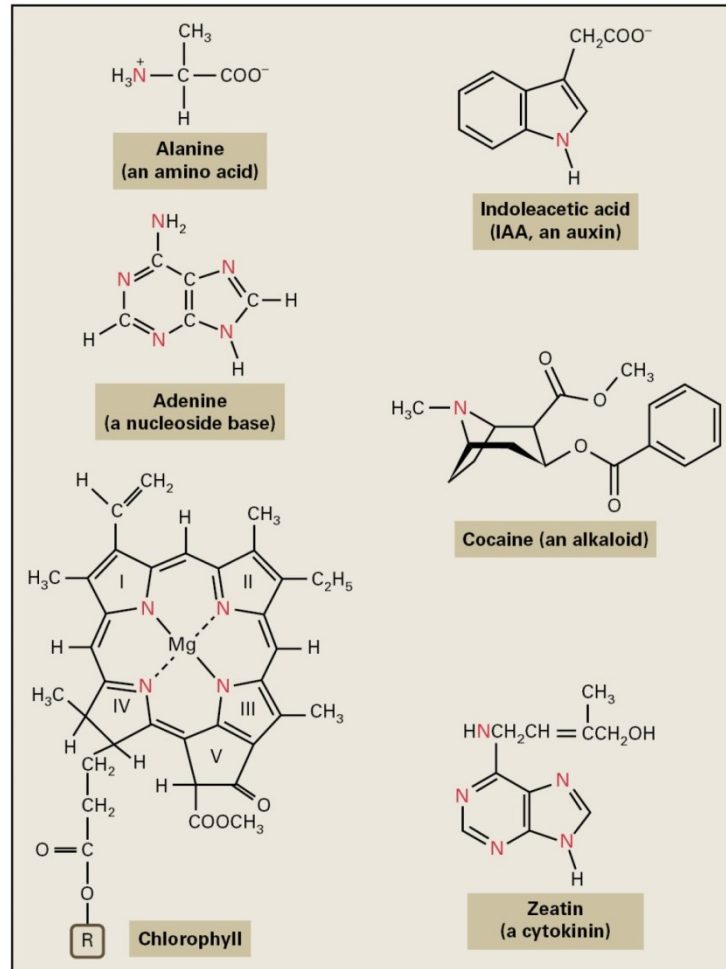
K⁺ functions:

- osmoticum
- charge balance
- enzyme activation

Nitrogen

Compound	Oxidation state of N	Name
N_2	0	Dinitrogen (nitrogen gas)
HN_3	-3	Ammonia
NH_4^+	-3	Ammonium ion
N_2O	+1	Nitrous oxide
NO	+2	Nitric oxide
NO_2^-	+3	Nitrite
NO_2	+4	Nitrogen dioxide
NO_3^-	+5	Nitrate





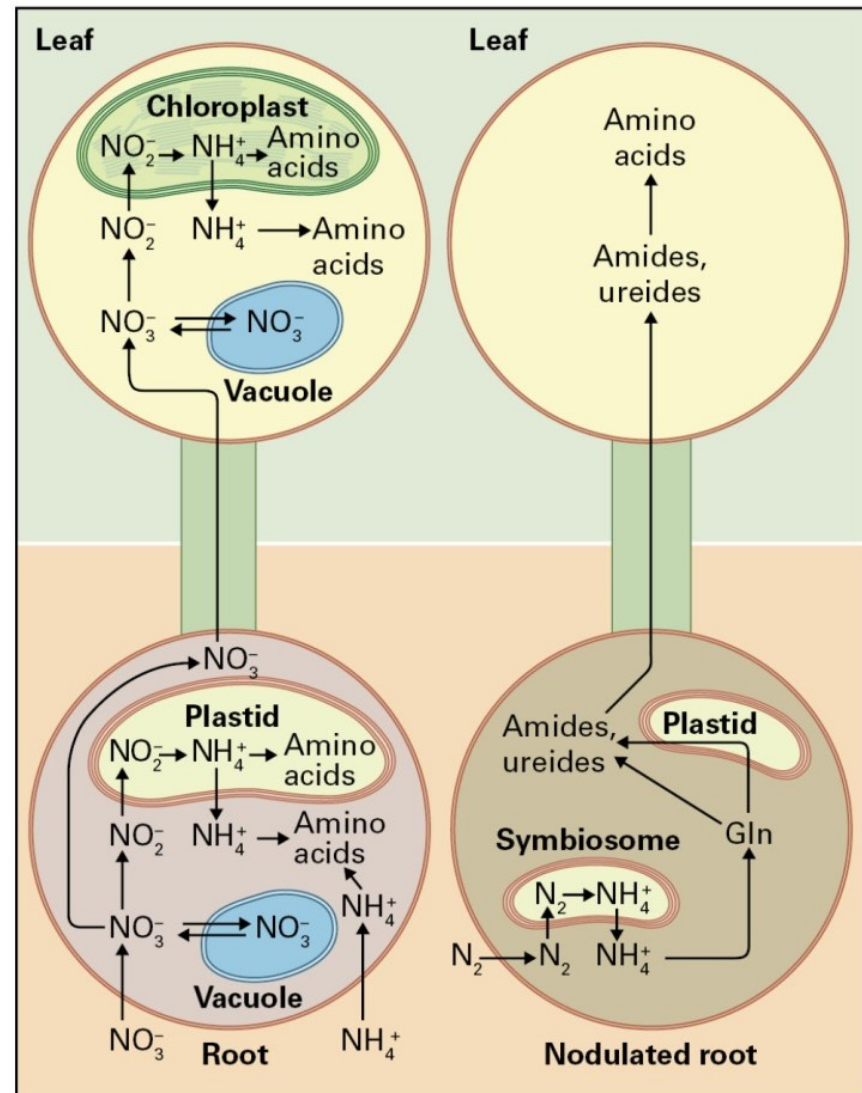
Nitrogen deficiency phenotype

Selected organic nitrogen compounds

Plants may acquire N as:

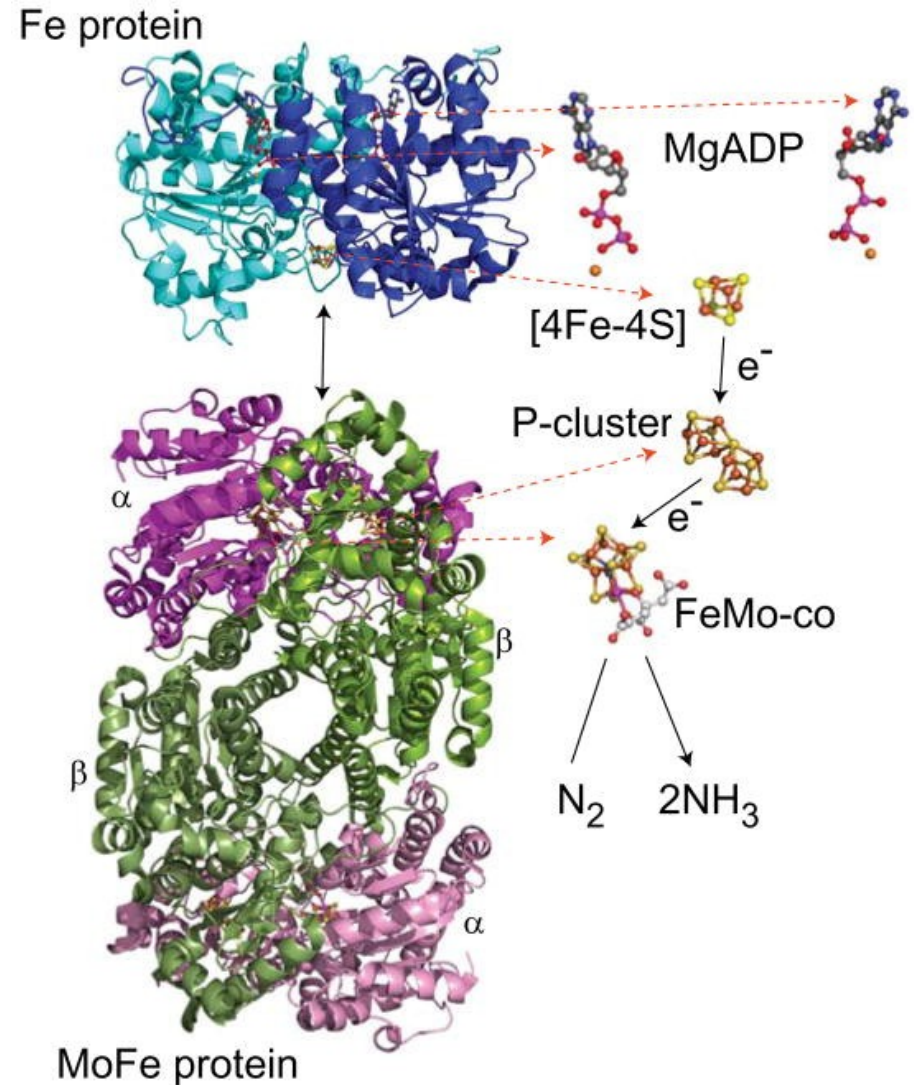
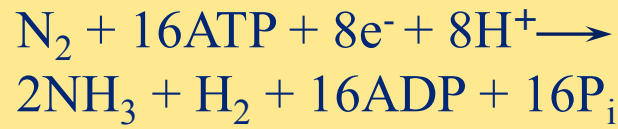
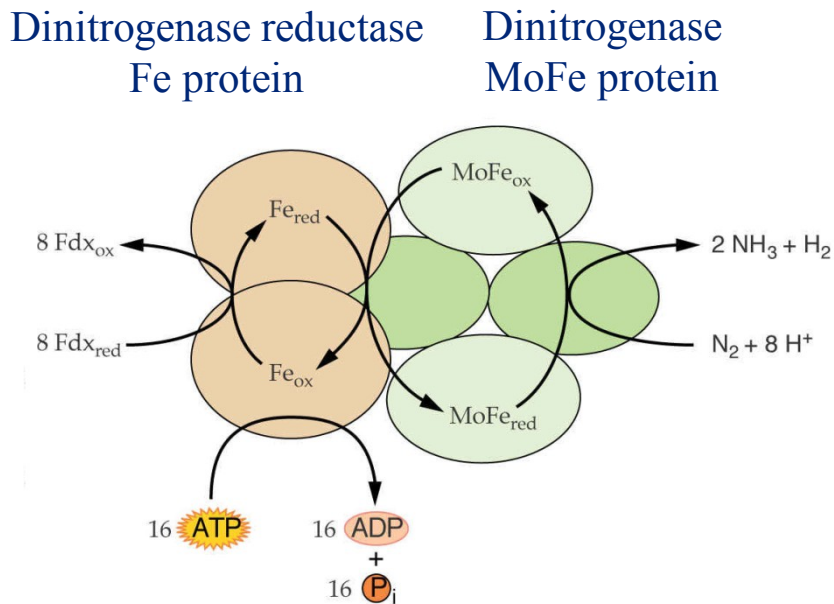
- ammonium ion
- nitrate
- dinitrogen, only in the case of plant species capable of endosymbiosis with nitrogen-fixing bacteria

Obtaining nitrogen through symbiosis consumes 12 to 17 g of carbohydrate per gram of N fixed



Nitrogen fixation

Nitrogenase complex

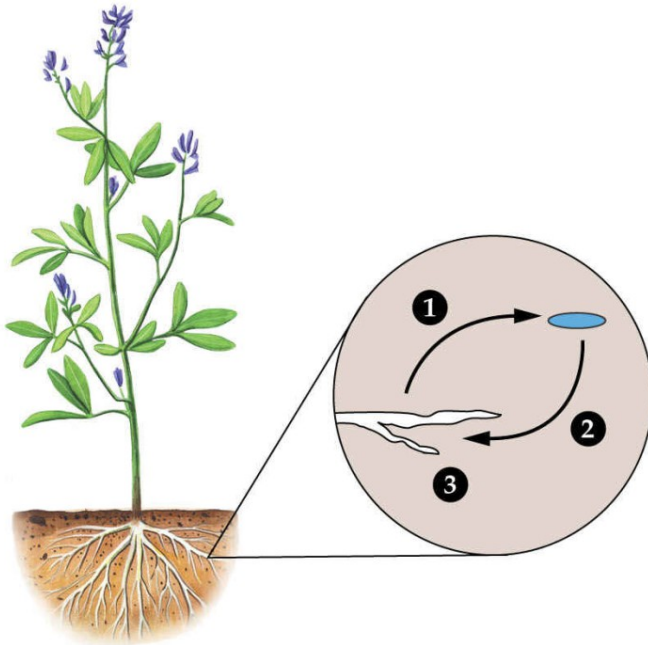


Legume-rhizobial symbiosis

The plant creates root nodules to ensure:

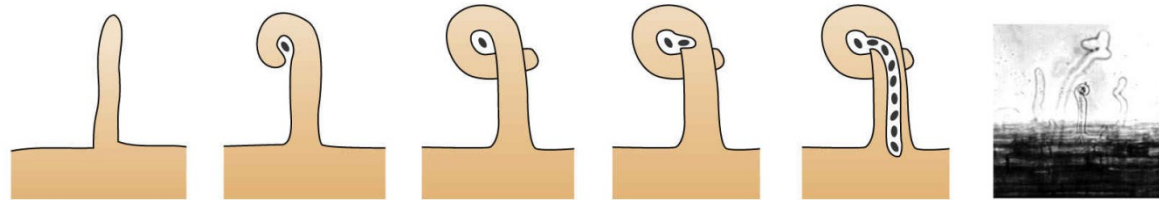
- microaerobic environment
- organic acids to feed the bacteria
- carbon skeletons to transport fixed nitrogen

Bacterial symbionts fix nitrogen and release the resulting ammonia

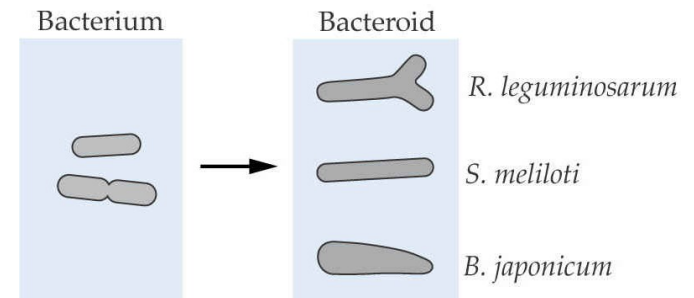
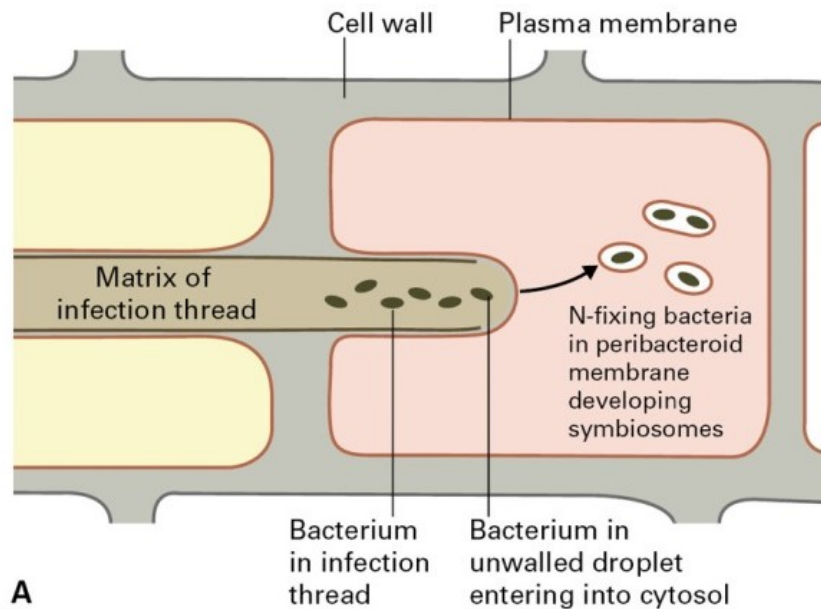


1. Plant signals
2. Nod factors
3. Nodulin proteins

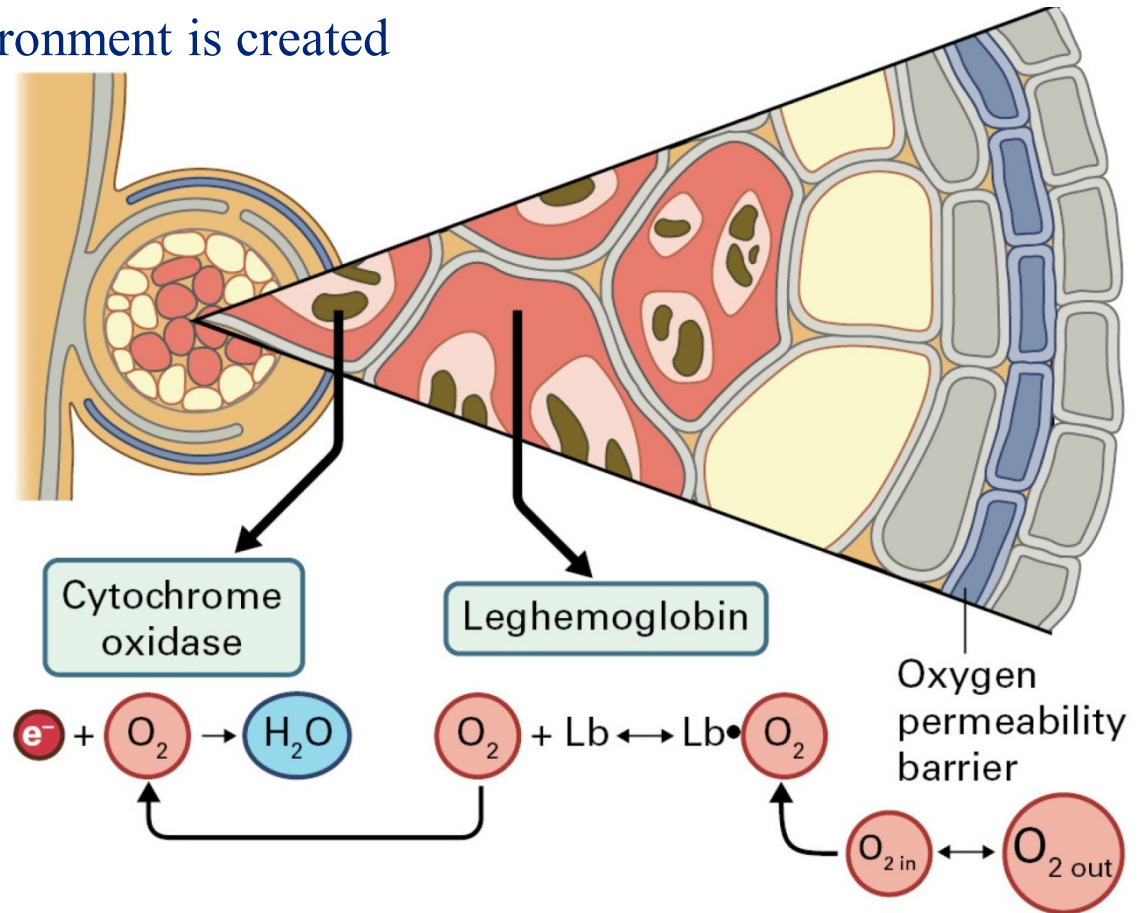
Nodule morphogenesis



In symbiosomes, bacteria differentiate into bacteroids



Microaerobic nodule environment is created

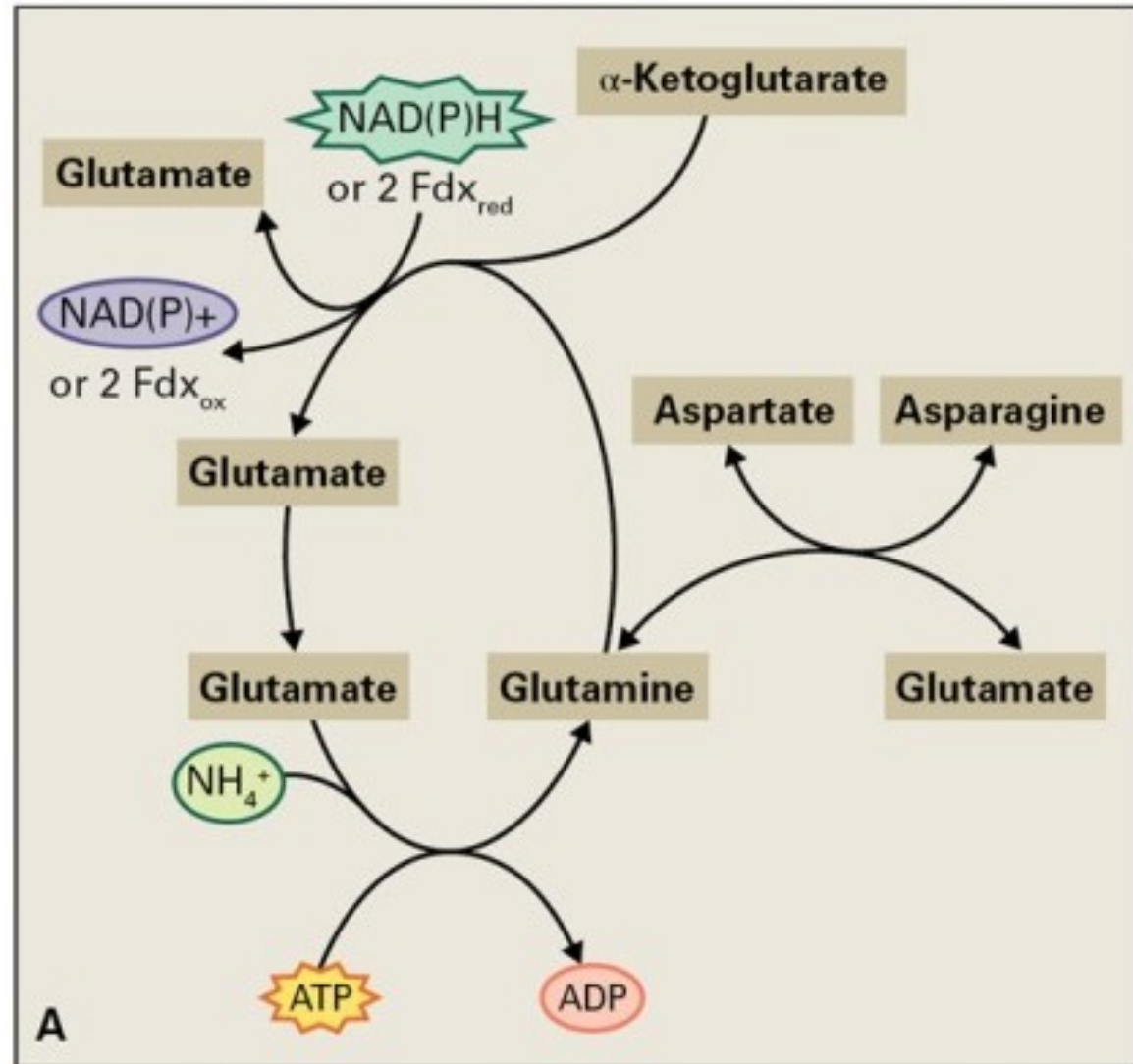


Carbon is provided to the bacteroids as dicarboxylic acids

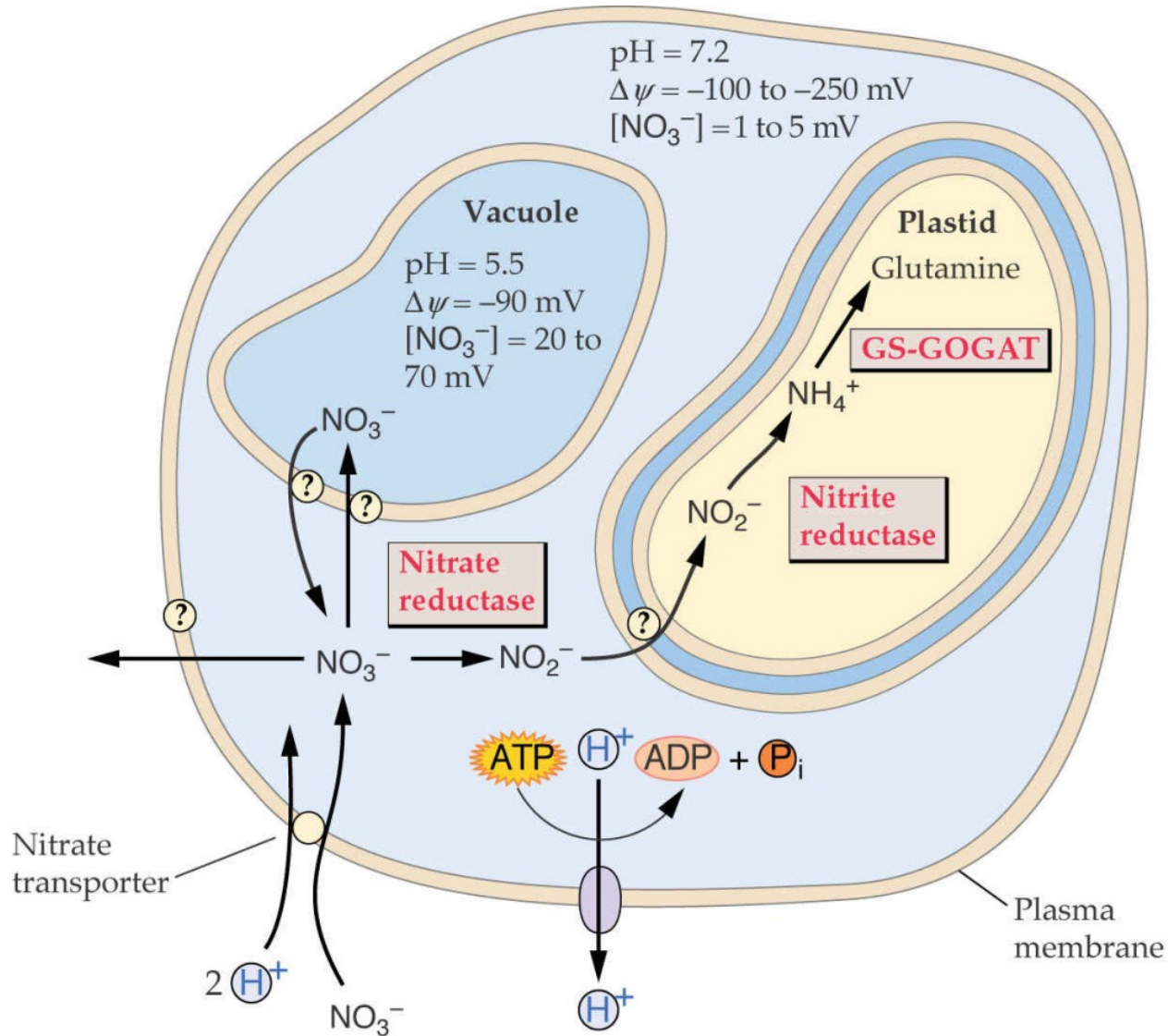
- oxidation of DCA provides ATP
- DCA carbon backbones are used for nitrogen transport

Ammonia assimilation

GS-GOGAT cycle



Nitrate assimilation



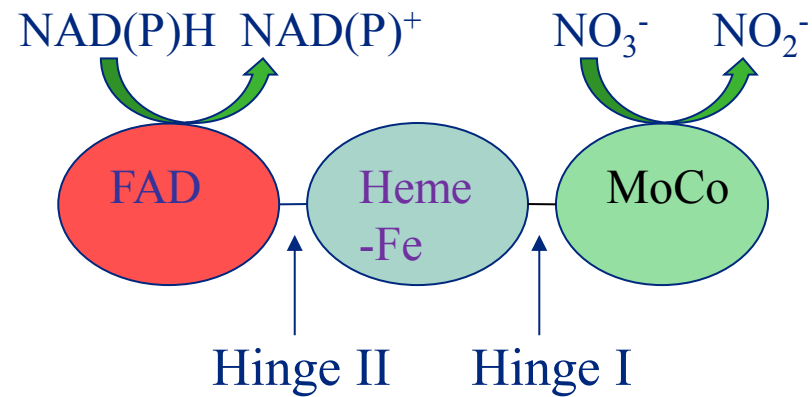
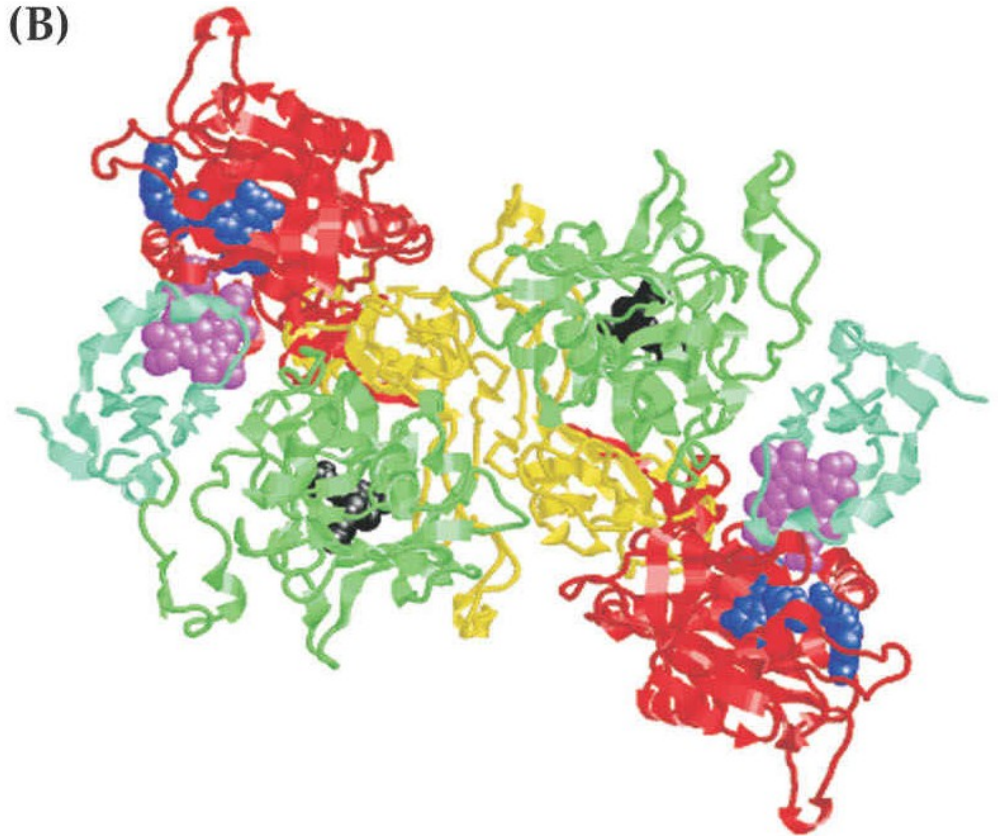
Nitrate reductase

NR reaction



NR homodimer

(B)

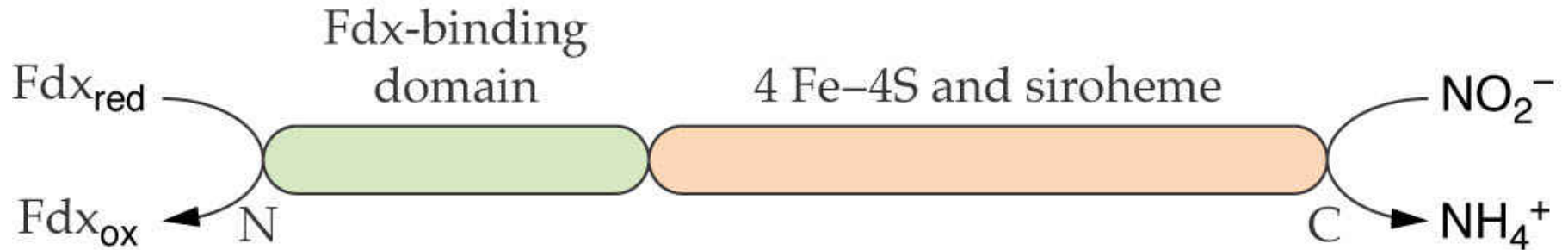


Nitrite reductase

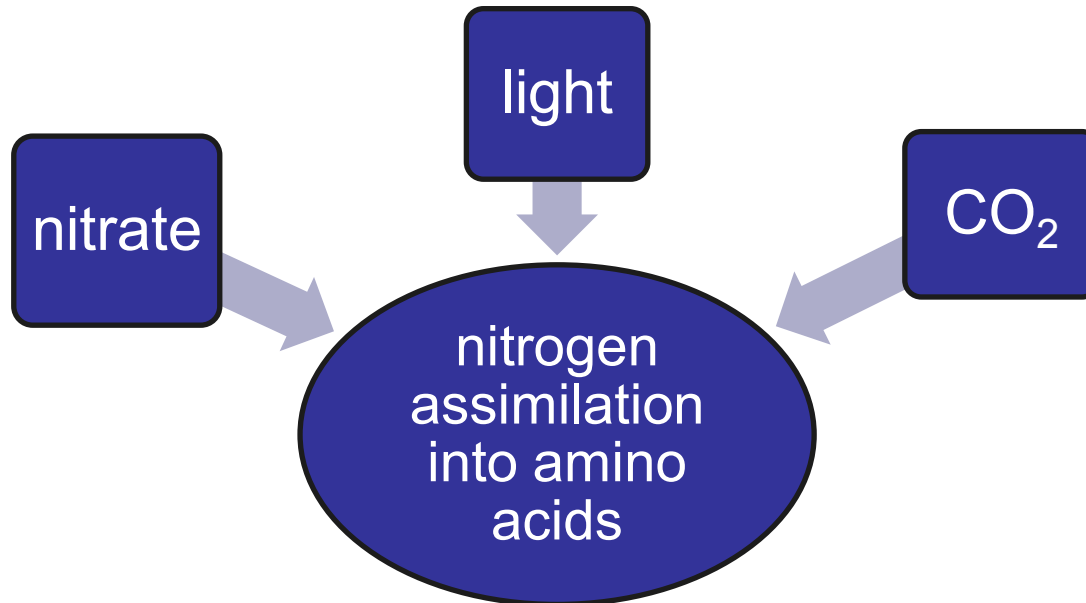
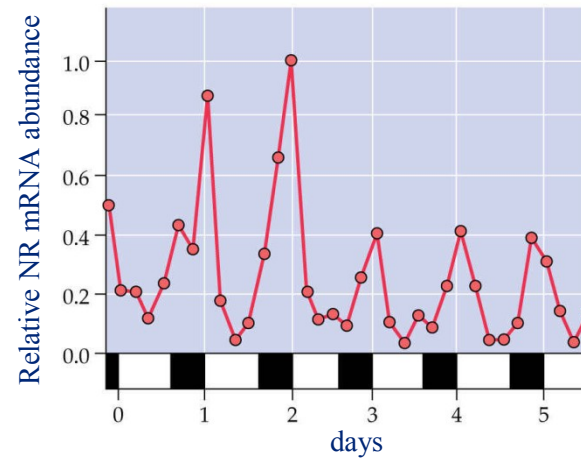
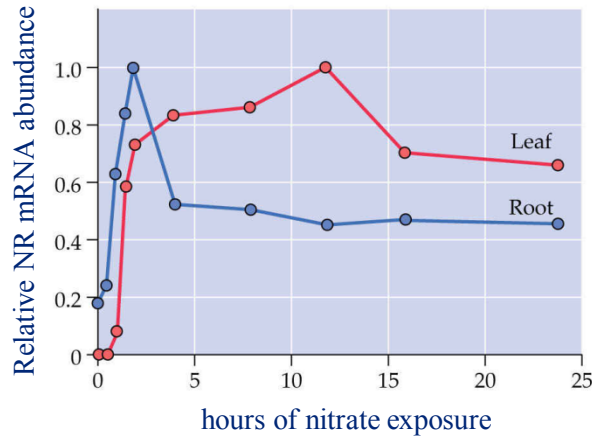
NiR reaction



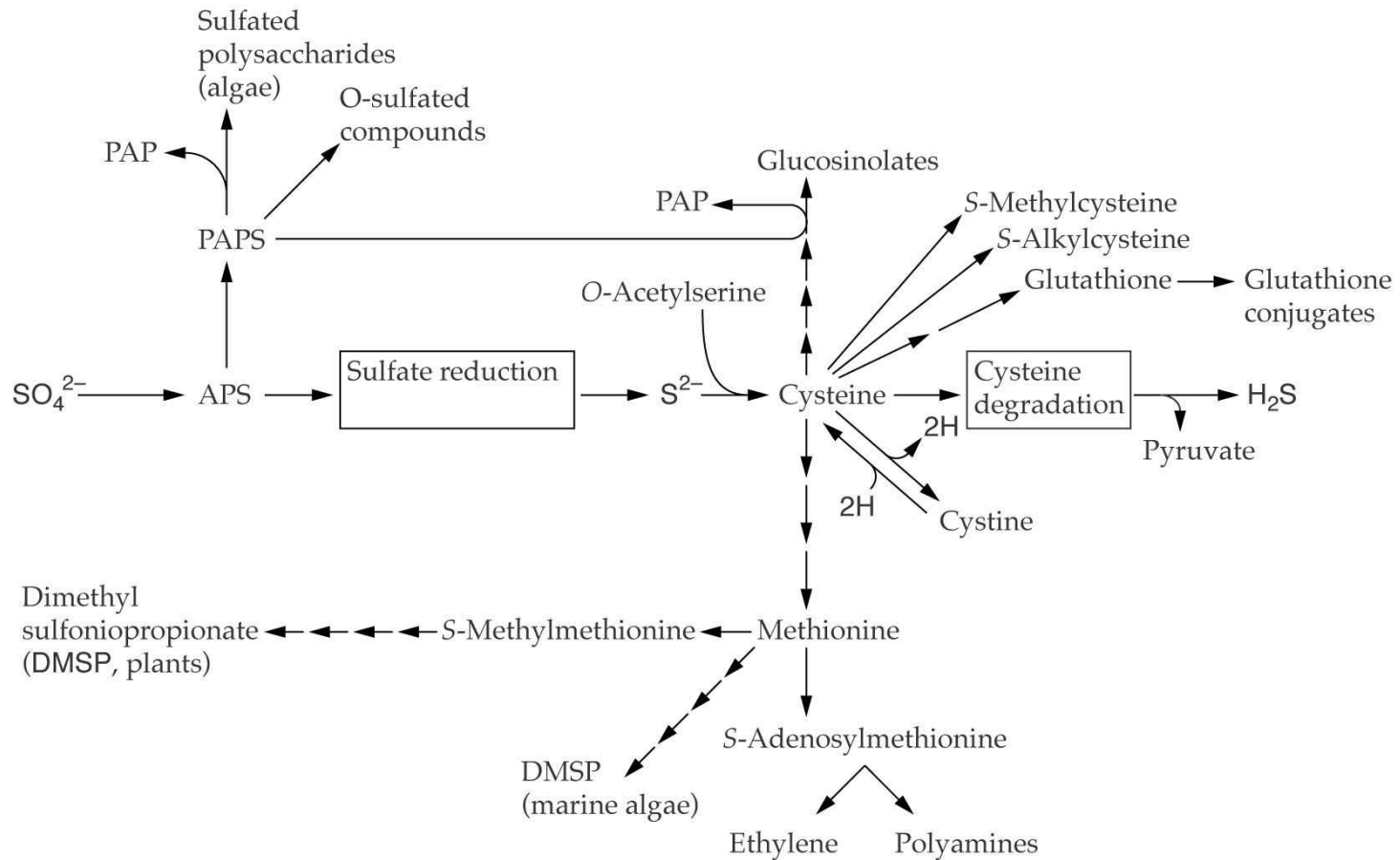
Plant nitrite reductase



Regulation of nitrate assimilation

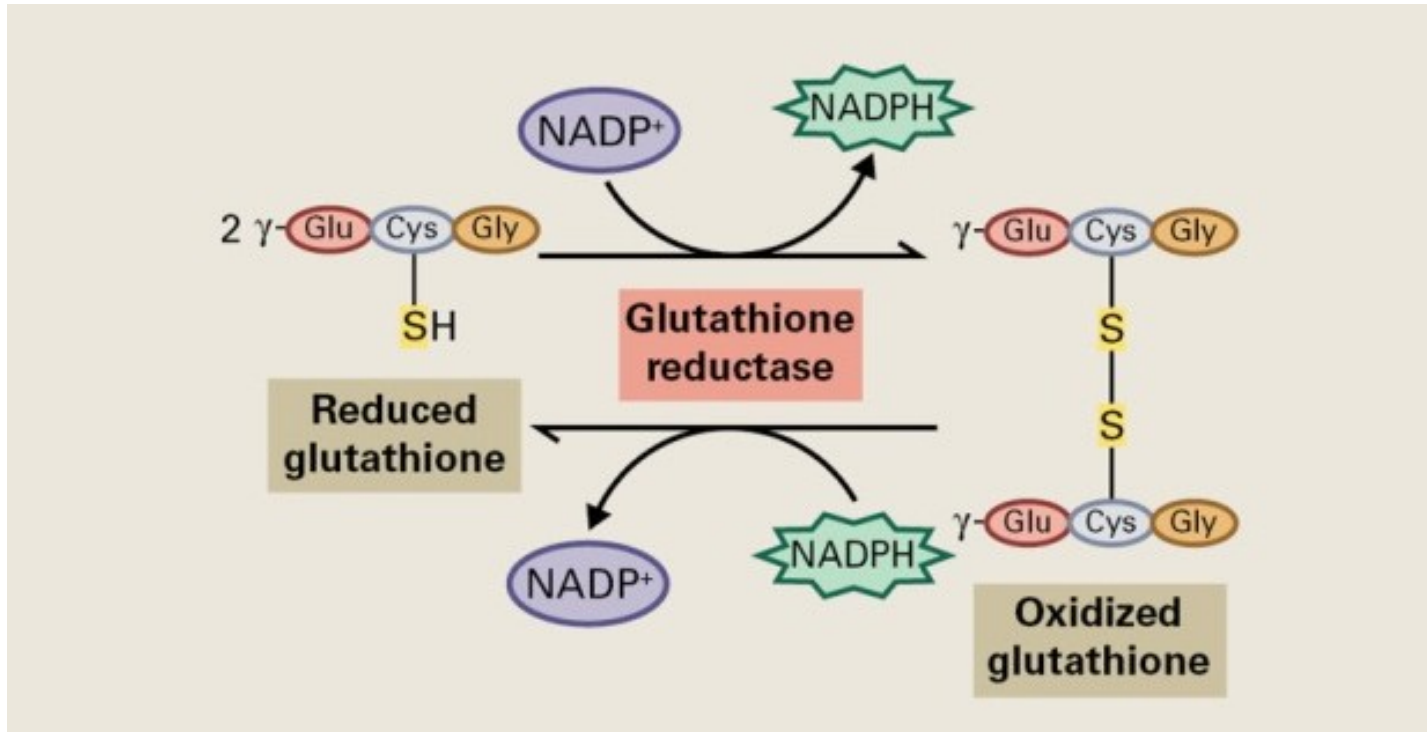


Sulfur

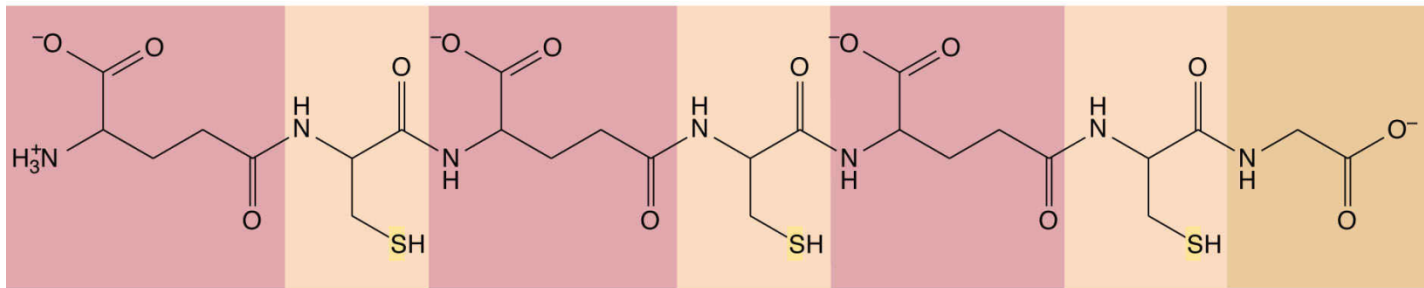


APS – 5-adenylylsulfate

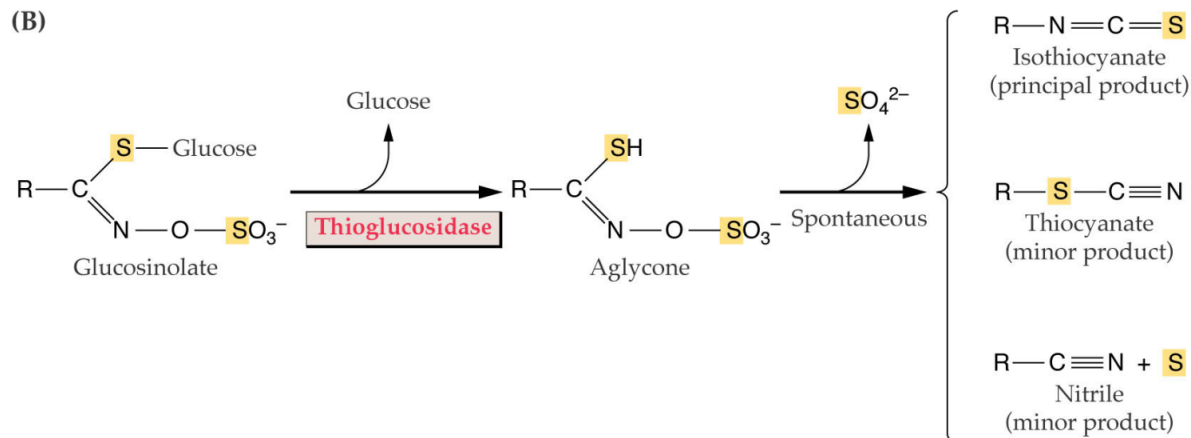
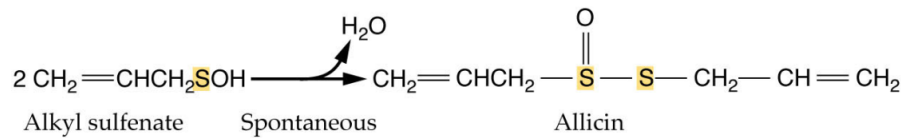
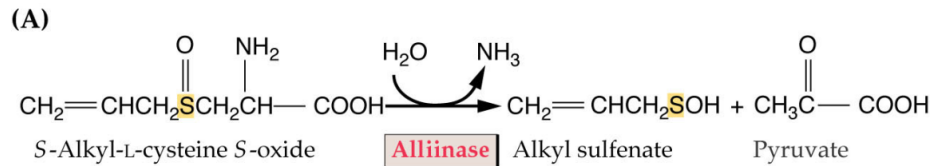
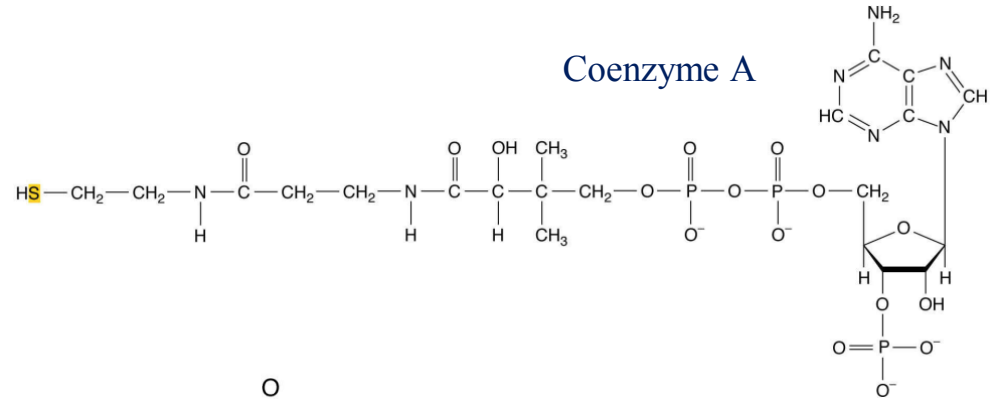
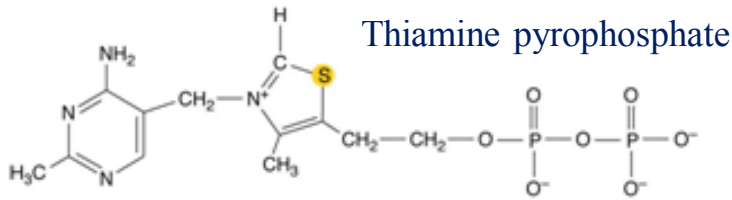
PAPS – 3-phosphoadenosine-5-phosphosulfate,

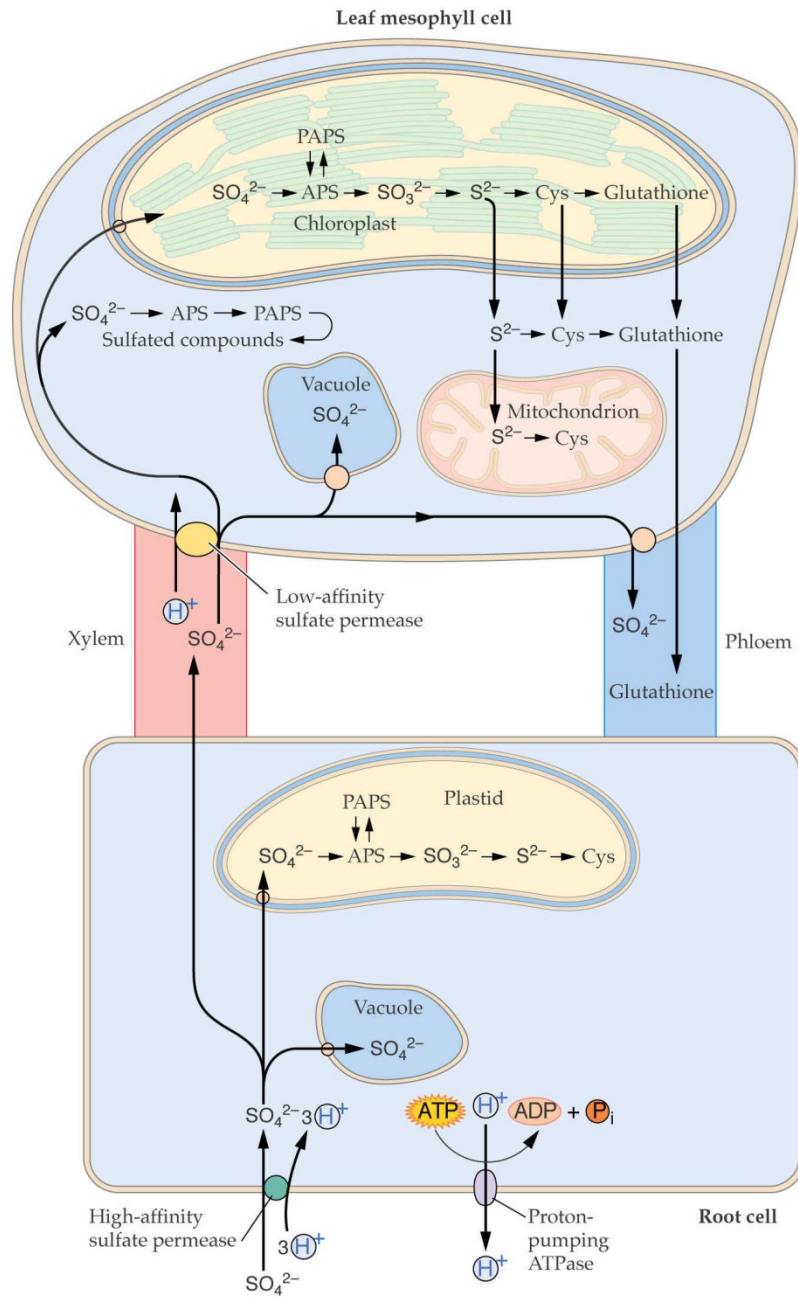


Phytochelatin molecule

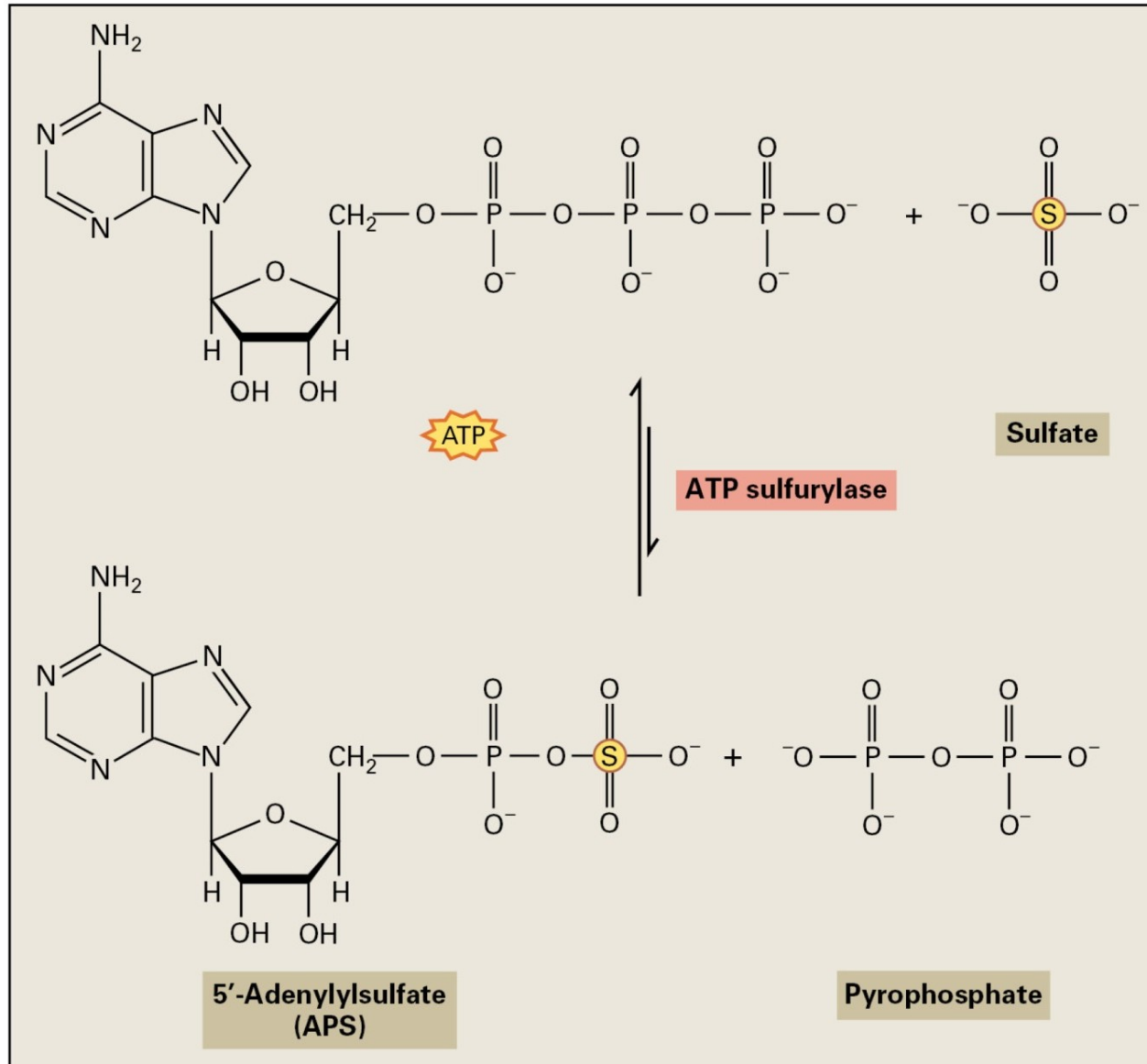


(γ-Glu-Cys)₃-Gly

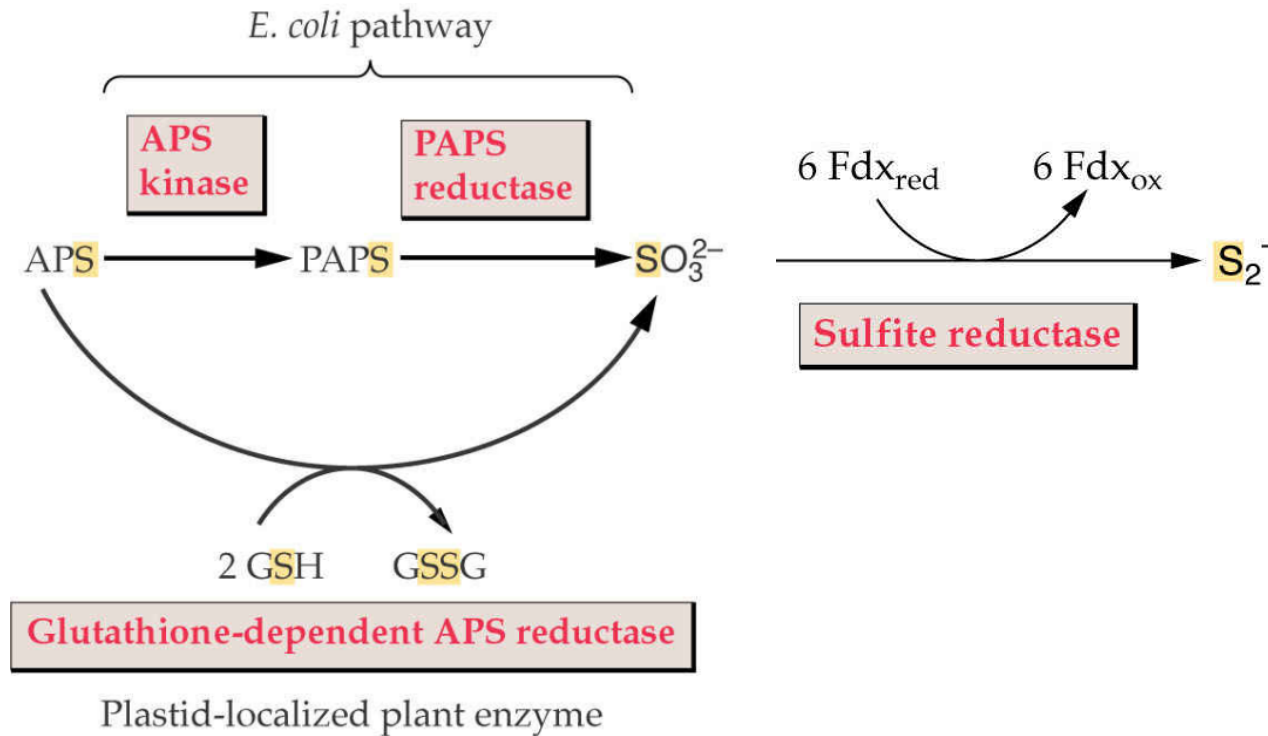




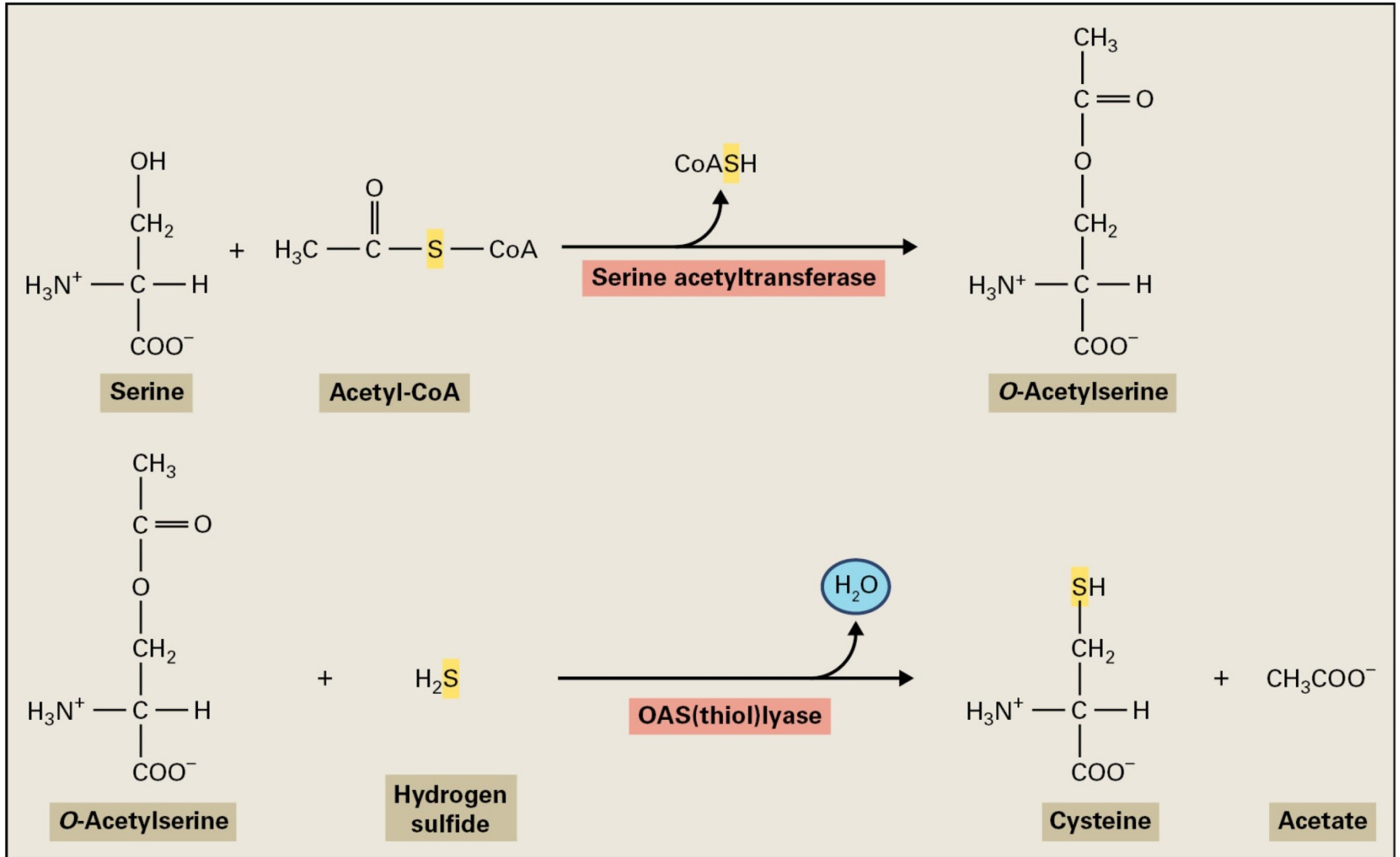
Sulfate activation



Sulfate reduction to sulfide



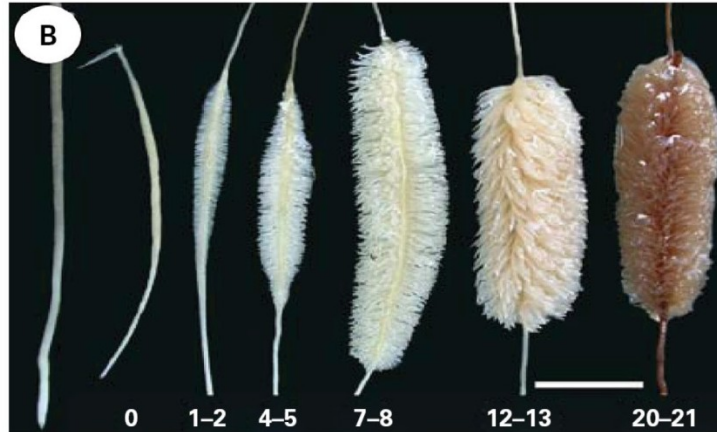
Cysteine synthesis



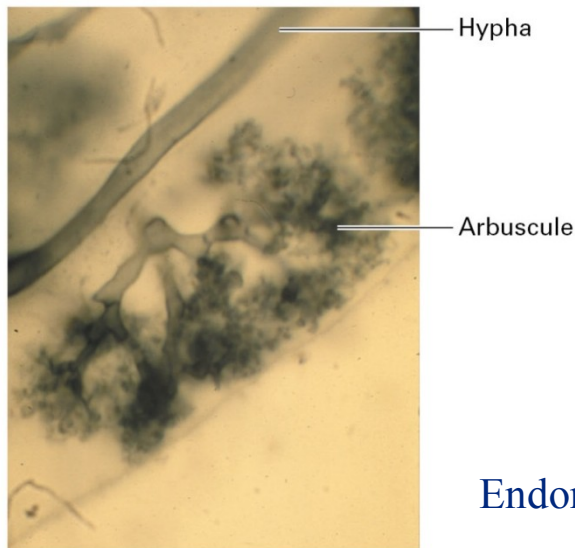
Regulation of sulfate assimilation

- Sulfur assimilation is not strongly regulated by light
the enzymes are also active in etiolated plants and do not demonstrate diurnal oscillations
- Sulfur assimilation is regulated by developmental stage
all the enzymes are highly active in young leaves and root tips
- Sulfur assimilation is regulated in response to the availability of sulfur
sulfur starvation results in the up-regulation of sulfate transport and APS reductase
- The content of reduced sulfur and nitrogen is strictly coordinated
- Sulfite and sulfide are not allowed to accumulate

Phosphorus



Root modifications in low Pi concentration



Endomycorrhizae

Phosphate functions:

- component of nucleic acids and phospholipids
- energy conversion (ATP)
- regulation