A Quantum Chemical Study of Isomerization Step within Acetylene Bromoboration

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Haloboration reactions of alkynes represent key elements in many organic synthetic routes, as evidenced in a series of 21 papers by A. Suzuki through 1983-1992.¹ During an acetylene bromoboration reaction, (*Z*)–alkene is formed by the direct addition of boron tribromide to acetylene which undregoes a subsequent isomerization into (*E*)-dibromo(2-bromovinyl)borane cf. Figure 1.



Figure 1: Z/E isomerization of (Z)-dibromo(2-bromovinyl)borane.

Wang and Uchyiama have proposed a polar isomerization mechanism catalyzed by another BBr_3 molecule with a barrier of ca. 26 kcal mol⁻¹.² This is in contrast with recent experimental findings obtained in our department.³ The present study concentrates on an alternative mechanism initiated by a reaction of Z-alkene with the HBr molecule and with the radical Br., whose presence is expected due to the interaction with residual air humidity.

- [1] Suzuki, A. Heterocycles 80 (2010) 15.
- [2] Wang, C., Uchyiama, M. Eur. J. Org. Chem. (2012) 6548.
- [3] Polášek, J., Bachelor's thesis, MU Brno (2014).