### C7800 Introduction to molecular modelling - seminar TSM Modeling of molecular structures

### Projects Project A - Water Dimer

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**C7800 Introduction to molecular modelling - seminar** 

# Water molecule Water molecule dimer

Quantum-chemical calculations



#### Water molecule

structure and energy

- basis set effect
- > properties

#### Water molecule dimer

- structure and energy
- interaction energy
  - basis set effect
- > properties



## To think about

- What effect does the basis set size have on the value of the absolute energy of a water molecule?
- What effect does the basis set size have on the value of the absolute energy of the water molecule dimer?
- What effect does the basis set size have on the value of the interaction energy of the water molecule dimer?

### How to process results

I recommend to summarize the results in the form of a brief protocol, which should have the following requirements:

- Name and surname, name of the project and date
- For each thematic area:
  - Brief summary of the topic, including a reaction scheme, if appropriate
    - Software used, including versions
    - Results (tables)
      - Tables
        - numbers aligned to the right
        - energies with 6 decimal points (au) or 2 decimal points (kcal/mol)
        - lengths with 4 decimal points (A)
        - angles with 1 decimal points (deg)
        - charges with 3 decimal points (au)
  - Discussion of results in context of the course
  - References (e.g., for experimental values)

# Water molecule

(Almost) individual project

### Reference manuals

- Gaussian (QM calculations)
- Nemesis (preparation, visualization)
- CBS Extrapolation to a complete basis set
- VMD (volumetric data visualization)
- Infinity (submitting jobs)

### Tasks

- 1) Create a model of water molecule and optimize its geometry using molecular mechanics.
- 2) Optimize the geometry of the water molecule using the HF/cc-pVDZ
- 3) Measure significant geometric parameters of the optimized geometry and compare them with the initial model.
- 4) Employing vibrational analysis, verify that the found geometry corresponds to the local minimum on the PES.
- 5) Perform energy and property calculation at the HF/cc-pVDZ level of theory and extract:
  - total energy
  - dipole moment
  - Mulliken and MK (Merz-Singh-Kollman) atomic charges
- 6) Repeat the calculation described in the point 5 on the same geometry for the following basis sets:
  - cc-pVTZ
  - cc-pVQZ
  - cc-pV5Z

# Solution

#### **Data organization**

- 00.input 01.opt 02.freq 03.props 01.cc-pVDZ 02.cc-pVTZ 03.cc-pVQZ 04.cc-pV5Z
- Create an initial geometry of water molecule in Nemesis (Project: Build Structure). Pre-optimize model geometry using molecular mechanics. Select such a force field (Geometry->Optimizer Setup), which in your opinion best describes its geometry.
- Save pre-optimized model geometry in the xyz format under the name water.xyz to a folder 00.input (File->Export Structure as...->OpenBabel). Next, save the input file for the program Gaussian (HF/cc-pVDZ, Geometry Optimization) with name opt.com to a folder 01.opt (File->Export Structure as...->Gaussian).
- 3) Run the calculation in the program Gaussian in the directory **01.opt** using the Infinity environment.

# Solution, cont.

- In the directory **01.opt**, open the file **opt.log** in the Nemesis program (Project: **Trajectory**, File->Import Trajectory from...-> Gaussian-> Geometry Optimization File).
  Analyze the course of optimization and the geometry of the optimized model.
- 5) Save the optimized geometry as an input file for Gaussian (HF/cc-pVDZ, Frequencies) with name **freq.com** to a folder **02.freq** (File->Export Structure as...->Gaussian). In the directory, run the calculation in the program Gaussian using the Infinity environment.
- 6) In the directory **02.freq**, open the file **opt.log** in the Nemesis program (Project: **Trajectory**, File->Import Trajectory from...-> Gaussian->Vibrations File). Does the geometry correspond to a local minima?
- 7) If so, successively save the optimized geometry in the directories 03.props/01.ccpVDZ,... When creating the input file **props.com** for Gaussian select "MK Charges" and the correct basis set. Run the calculations in Gaussian using the Infinity environment (All jobs can be submitted at once).
- 8) Analyze the calculated data and put them in the following tables.

### **Results** I

#### Geometry of water molecule



### **Results II**

#### Water molecule

Basis set	Cardinal number	E	Er
		[au]	[kcal/mol]
cc-pVDZ	2		0.0
cc-pVTZ	3		
cc-pVQZ	4		
cc-pV5Z	5	1	
CBS		/	
	he calculat nergy (E(RF		lative energy with

### **Results III**

#### Water molecule

Basis set	Cardinal number	Mulliken		ESP		μ
		q(H)	q(O)	q(H)	q(O)	[D]
cc-pVDZ	2					
cc-pVTZ	3					
cc-pVQZ	4					
cc-pV5Z	5					
CBS						

# Water molecule dimer

**Individual project** 

### Reference manuals

- Gaussian (QM calculations)
- Nemesis (preparation, visualization)
- CBS Extrapolation to a complete basis
- VMD (volumetric data visualization)
- Infinity (submitting jobs)

### Tasks

- 1) Create model of the water molecule dimer and optimize its geometry using molecular mechanics.
- 2) Optimize the geometry of the water molecule dimer using the HF/cc-pVDZ.
- 3) Measure significant geometric parameters of the optimized geometry and compare them with the initial model. Try to justify the observed differences.
- 4) Verify that the found geometry corresponds to the local minimum on the PES, using vibrational analysis.
- 5) With optimized geometry, perform the energy calculation for the bases:
  - cc-pVDZ
  - cc-pVTZ
  - cc-pVQZ
  - cc-pV5Z
- 6) For each base, determine the interaction energy between water molecules.
- 7) Determine the interaction energy extrapolated to CBS.
- 8) Display the electrostatic potential mapped to electron density calculated by the HF/ccpVDZ. Compare the potential with a water molecule.

# Solution

#### **Data organization**

00.input 01.opt 02.freq 03.props 01.cc-pVDZ 02.cc-pVTZ 03.cc-pVQZ 04.cc-pV5Z

The procedure is analogous to that of water monomer.

### **Results** I

#### Geometry of water molecule dimer



hydrogen bond

At your discretion, list other geometric parameters that best capture the difference between the two geometries.

#### Water molecule dimer

