Inorganic Materials Chemistry Jiri Pinkas Office C12/224

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 Lectures: PowerPoint presentations with recorded talk uploaded in the IS MU

Course grading:

- 3 graded homeworks during semester
- Short presentations on a selected topic concerning materials chemistry

• Written final **exam** (100 pts, minimum 50 pts to pass) Grading weights: final test 75%, homeworks 15%, presentation 10%.

### **Recommended Literature**

SCHUBERT, U., HÜSING, N., Synthesis of Inorganic Materials. Weinheim: Wiley-VCH

CALLISTER, W.D.J., Materials Science and Engineering, An Introduction. John Wiley and Sons

SMART, L., MOORE, E., Solid state chemistry : an introduction. 2nd ed. London: Chapman & Hall

INNOCENZI, Plinio, The Sol to Gel Transition, Springer International Publishing

WHITE, Mary Anne, Physical Properties of Materials, 2nd Edition, CRC Press

MUELLER, Ulrich, Inorganic Structural Chemistry, 2nd Edition, Wiley

### **Recommended Literature**

OZIN, G.A., ARSENAULT, A.C., CADEMARTIRI, L., Nanochemistry : a chemical approach to nanomaterials. 2nd ed. Cambridge: RSC Publishing, 2009.

CADEMARTIRI, L., OZIN, G.A., Concepts of nanochemistry. Edited by Jean-Marie Lehn. Weinheim: Wiley-VCH, 2009.

FAHLMAN, B. D. Materials Chemistry. Springer 2018.

LOWELL, S., SHIELDS J. E., Martin A. THOMAS, M. A., THOMMES, M. Characterization of Porous Solids and Powders: Surface Area, Pore Size and Density. Springer 2004.

MITTEMEIJER, E. J. Fundamentals of Materials Science. Springer 2011.

# **Materials in Human History**

**Historical perspective:** 

New materials bring advancement to societies

- Stone age
- Bronze age
- Iron age
- Silicon age



Crescent Axes. The top Syrian, the bottom Egyptian. about 1900 BC 4

# **Materials in Human History**

- 50 000 B.C. Iron oxide pigments Lascaux, Altamira
- 24 000 B.C. Ceramics fat, bone ash, clay
- 3 500 B.C. Cu metallurgy

**Glass, Egypt and Mesopotamia** 

- 3 200 B.C. Bronze
- 1 600 B.C. Iron metallurgy, Hittites
- 1 300 B.C. Steel
- **1 000 B.C.** Glass production, Greece, Syria
- 105 B.C. Paper, China
- 590 A.D. Gun powder, China
- 700 A.D. Porcelain, China



# Materials in Human History -Metals



FIG. 11.--EGYPTIAN GOLDSMITHS WASHING, MELTING AND WEIGHING GOLD, BENI HASAN, 1900 B.C.

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# Materials in Human History -**Ceramics**



# Development of Materials in Human History



# **Compounds vs. Materials**

- \* Chemical compounds single use (pharmaceuticals, fertilizers, fuels)
- \* Materials repeated or continual use - shaping





#### **Shaping of Macro and Micro Materials**



Ceramics Glasses Metals, Alloys Polymers Composites Semiconductors







#### **Shaping of Nanomaterials**



# **Classes of Materials**

Ceramics (oxides, carbides, nitrides, borides) Glasses (oxides, fluorides, chalcogenides, metallic) Metals, Alloys, Intermetallics **Polymers - inorganic, organic, hybrid** Semiconductors (Si, Ge, 13/15, 12/16 compounds) **Composites, Inorganic-Organic Hybrid Materials Zeolites, Layer and Inclusion Materials Biomimetic Materials, hydroxyapatite** 



**Carbon-based Materials: Fullerenes, Fullerene Tubes, Graphene** 12

#### **Three Classical Classes of Materials**

Metals	Ceramics	Polymers
Strong	Strong	Usually not strong
Ductile	Brittle	Very ductile
Electrical Conductor	Electrical Insulator	Electrical Insulator
Heat Conductor	Thermal Insulator	Thermal Insulator
Not transparent	May be transparent	Not transparent
Shiny	Heat Resistant	Low Densities

### **Properties of Materials**

A property = a material trait, the kind and magnitude of response to a specific stimulus

**Properties** 

Mechanical Electrical Thermal Magnetic Optical Deteriorative (corrosion) Catalytic Biocompatibility

#### **Materials Science:**

Studies relationships between the structure and properties of materials

#### **Materials Engineering:**

Designing and engineering the structure of a material to produce a predetermined set of properties



# **Materials Chemistry**

**Role of Materials Chemistry** 

- Synthesis of new materials new atom architecture
- Preparation of high purity materials
- Fabrication techniques for tailored morphologies (shapes and sizes)
- Fabrication of functional materials



# **Size of Particles**

#### Nanoparticles 1 – 100 nm Traditional materials > 1 mm









1µm



2µm

### Shapes of Natural and Synthetic Single Crystals

#### Calcite CaCO<sub>3</sub>

100 µm

#### **Cu-Ag nanoalloy**





\_\_\_\_10 μm

### **Onion-Like Particles**



#### **Functional Materials** Synthesis of porous SiO<sub>2</sub> nanospheres (8)HO. OH hydrolysis. condensation. 200nm (b) organic phase 00 0 water. n Patrick 2 µm

#### CB[6] = cucurbit[6]uril

ertactary.



#### **Dynamic wagging motion by Z-E interconversion**

00n



#### **Functional Materials**

#### **Dual-controlled nanoparticles exhibiting AND logic function**

(a) Excitation with 448 nm light induces the dynamic wagging motion of the nanoimpellers, but the nanovalves remain shut and the contents are contained

(b) Addition of NaOH opens the nanovalves, but the static nanoimpellers are able to keep the contents contained

(c = a + b) Simultaneous excitation with 448 nm light AND addition of NaOH causes the contents to be released

CB[6] = cucurbit[6]uril



# Functional Materials - Si<sub>3</sub>N<sub>4</sub> Si

#### Hexagonal

#### $\alpha$ modification



 $\beta$  modification



- Strong covalent bond (4.9 eV)
- Hardness (a-monocrystal, Vickers 21 GPa)
- Tensile Strength 1.5 GPa (β-whisker)
- Young modulus 350 GPa
- Decomposition temp. 1840 °C/1 atm N<sub>2</sub>
- Density 3.2 g cm<sup>-3</sup>

N

# Si<sub>3</sub>N<sub>4</sub> Ceramics



### **Microstructure of Materials**



### **Microstructure vs. Properties**



#### **Materials Chemistry**

Single crystals, defects, dopants, non-stoichiometry **Monoliths** Coatings Thin or thick films - singlecrystalline, polycrystalline, amorphous, epitaxial Fibers, Wires, Tubes **Powders – primary particles, aggregates, agglomerates** polycrystalline, amorphous, nanocrystalline (1-100 nm) **Porous materials** micropores (< 20 Å), mesopores (20-500 Å), macropores (> 500 Å) **Micropatterns** Nanostructures – spheres, hollow spheres, rods, wires, tubes, photonic crystals Self-assembly – supramolecular chemistry: rotaxenes, catenanes, cavitands, carcerands

### **Self-Assembling Monolayers**



#### STM on HOPG

#### **Materials Chemistry Tool Box**

**Direct reactions of solids – "heat-and-beat"** 

**Precursor methods** 

Chimie douce, soft-chemistry methods, synthesis of novel metastable materials, such as open framework phases

Ion-exchange methods, solution, melt

Intercalation: chemical, electrochemical, pressure, exfoliation-reassembly

Crystallization techniques, solutions, melts, glasses, gels, hydrothermal, molten salt, high P/T

Vapor phase transport, synthesis, purification, crystal growth, doping

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#### **Materials Chemistry Tool Box**

Electrochemical synthesis, redox preparations, anodic oxidation, oxidative polymerization

Preparation of thin films and superlattices, chemical, electrochemical, physical, self-assembling mono- and multilayers

Growth of single crystals, vapor, liquid, solid phase chemical, electrochemical

High pressure methods, hydrothermal, diamond anvils

Combinatorial materials chemistry, creation and rapid evaluation of gigantic libraries of related materials

### **STEM - Imaging at Nanoscale**



a) 2D EDX map of a Au@Ag nanocube. Based on a tilt series of 2D EDX maps the 3D reconstruction presented in (b) was obtained. The contrast in the 3D reconstruction is based on differences in chemical composition and it is clear that the core of the particle has an octahedral form.

### **TEM - Atomic Scale Imaging**



Atomic scale reconstruction of Au nanorods. a,b) Orthogonal slices through the atomic scale reconstruction of Au nanorods prepared using different surfactants. The side facets of these rods can be clearly recognized. c) Strain measurement along the major axis of the nanorod.